

Biocides in Sewage Sludge: Quantitative Determination in Some Swiss Wastewater Treatment Plants

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Biocides are included in a wide range of products. In the present study, carbendazim (Car), Irgarol 1051[®] (Ir), octhilonone (Oc), permethrin (Per), tributyltin (TBT) and triphenyltin (TPT) with numerous industrial or domestic uses have been studied (Table 1). Data on consumption and uses of these substances are difficult to obtain. In addition, there is a lack of data on these biocides sources and release processes into the environment. Lassen et al. (2001) have determined a consumption of active substances in biocidal products between 3,600 to 5,530 tonnes per year in Denmark for 1998/99. In Switzerland, such information is not available. Biocides released by private households and the industry may enter wastewater streams directly or they may end up in wastewater after aerial deposition on impervious surfaces and runoff into sewer systems. In wastewater treatment plants (WWTPs), lipophilic substances are mainly adsorbed onto sewage sludge. Sludge is therefore an appropriate matrix for the characterization of the release of numerous xenobiotic compounds (Oeberg et al. 2002; Chassot 1995). To describe the sources of pollutants and their distribution in the environment through the study of sewage sludge, additional informations such as the type of technology used for wastewater treatment, the collector system, socio-economic features and the type of industry present is required. Therefore, all samples of this study have been taken from WWTPs of a Swiss monitoring network with characterized observation sites (WWTP and the corresponding catchment). To date, only very little data on occurrence of biocides in sludge is available. Thus, the present study aims to determine (i) the concentrations of biocides in sewage sludge, (ii) the specific loads of biocides in sewage sludge sampled in different types of monitoring sites and (iii) their main sources.

MATERIALS AND METHODS

The monitoring network comprises three types of sites described by Kupper et al. (2004). Sites of type A include a separate sewer system and a rural catchment without industrial activities apart from a few craft industries. Sites B exhibit the same characteristics but the catchment has a combined sewer system and more numerous craft industries. Sites of type C have a combined sewer system and an urban catchment with craft industries and some types of industrial activities. In January and May 2001, samples of sewage sludge were collected in the 12 different WWTPs of this monitoring network as described by Kupper et al. (2004).

Table 1. Name, N°CAS and applications of the measured biocides

Carbendazim (10605-21-7, benzimidazole, fungicide) preservative for masonry, fibrous or polymerised material, coating, paint, wood and non-alimentary products, disinfectant, agriculture	Irgarol 1051[®] (28159-98-0, triazine, algicide) emulsion paint and antifouling paint
Octhilinone (26530-20-1, isothazolo- <i>n</i> -one, fungicide) preservative of non-alimentary products, paint, leather and leather products, preservation of cooling water, heating and processing systems, pest control	Permethrin (52645-53-1, pyrethroid, insecticide) repellent/insecticide for clothing, household insect foggers and sprays, flea dips and sprays for cats and dogs, ornamental garden and turf products, termite treatment, lice shampoos and body lotion for scabies control, wood preservative, agriculture
TBT (688-73-3, organotin, fungicide, algicide) preservative for wood, timber and material (stone, leather, paper, textile), antifouling paints, disinfection, cooling systems, dispersion paints, PVC	TPT (668-34-8, organotin, fungicide) moth proofing of textiles, in antifouling paints and in agriculture

Permethrin (95.5 %), octhilinone (96.5%), TBT chloride (96.5%) were obtained from Dr. Ehrenstorfer, Augsburg, Germany. TPT chloride (98.0 %), NaCl, Na₂SO₄, H₂SO₄, NaOH, K₂CO₃ and silica gel 40 (activated 12h at 350 °C) were supplied from Merck, Dietikon, Switzerland. Carbendazim (99.8 %) and Irgarol 1051[®] (97.6%) were supplied by Promochem, Wesel, Germany and by Ciba Speciality Chemicals, Basel, Switzerland, respectively. All solvents were super-purity solvents from Romil, Cambridge, England or analytical-grade from Merck, Dietikon, Switzerland. Florisil (100-200 mesh, activated 2h at 650°C) was purchased from Fluka, Buchs, Switzerland. Pentafluorobenzyl bromide (PFB-Br) was used as derivatization agents from Aldrich, Buchs, Switzerland. Three-ml SiOH cartridges containing 500 mg of unmodified silica were from Macherey-Nagel Chromabond, Düren, Germany, while aminopropyl-bonded silica was from Supelco, Buchs, Switzerland. Basic alumina (50-200 µm, activated 16h at 350 °C) by Alltech, Deerfield, USA. For the purification on Gel Permeation Chromatography, a 600/20-mm glass-column fitted with a 5 ml sample loop and packed with 100 g of Bio-Beads SX3 (BioRad, Hercules, USA) has been used. One cm I.D. glass columns conditioned with hexane were used for the clean-up. Due to the different chemical characteristics of the studied biocides, three different methods were required. Reagent blanks were prepared and analysed prior to applying any of the methods. The analytical method for organotin compounds has been described by Becker van Slooten et al. (1994).

For Irgarol 1051[®] and octhilinone, 60 g of bulk sludge mixed with 5 g of NaCl were shaken with 30 ml of pentane:dichloromethane (4:1, v:v) for 30 min. The supernatant was dried over Na₂SO₄ and extraction was repeated twice. This extract, reduced to 1 ml in isoctane, was cleaned-up on a column prepared with 4 g of florisil and Na₂SO₄. Four fractions were eluted: 20 ml of hexane discarded, 30 ml of hexane:acetone (9:1, v:v) with Irgarol 1051[®], 10 ml of dichloromethane:acetone (7:3,v:v) discarded and 20 ml of dichloromethane:acetone (7:3, v:v) containing octhilinone. The second fraction was concentrated and cleaned-up on a 8-g basic

alumina column. Irgarol 1051[®] was eluted with 30 ml of hexane/acetone (9:1, v:v) after washing with 20 ml hexane and 30 ml hexane:acetone (9:1, v:v). Octhilinone's fraction was transferred on a glass cartridge with 1 g of aminopropyl-bonded silica sorbent. The cartridge was rinsed with 4 ml of hexane/acetone (90:10, v:v) and octhilinone was eluted with 6 ml of hexane/acetone (4:1). The extracts were re-dissolved in isooctane for analysis.

For carbendazim and permethrin, 70 g of sample (with 5 ml of phosphate buffer and 4 g of NaCl) was shaken with 25 ml of hexane:ethyl acetate (1:1, v:v) for 30 min. Extraction was repeated twice with 15 ml of solvent. Hexane was evaporated to adjust the volume at 30 ml with ethyl acetate. The extract was washed with 10 ml of water at pH adjusted to 12-13 with NaOH 5N, 10 ml of MilliQ water and then twice with 10 ml of H₂SO₄ (0.025M). The acidic phase was washed with 15 ml of ethyl acetate. The two organic phases (containing permethrin) were combined for purification on GPC. The two acidic phases (containing carbendazim) were combined, adjusted to basic pH with NaOH (5N) and extracted three times with 10 ml of ethyl acetate. This phase containing carbendazim was re-dissolved in 1 ml of acetonitrile for derivatization with 100 µl of K₂CO₃ 25% (w/w) in MilliQ water and 100 µl of PFB-Br 15% (w/w) in acetonitrile for 3 h at 60°C. Toluene was added to evaporate acetonitrile and then percolated on a SiOH cartridge conditioned with hexane. To eliminate PFB-Br the cartridge was washed with 8 ml of toluene:hexane (3:17, v:v). Carbendazim was recovered with 4 ml of acetone:hexane (1:4, v:v) after discarding the two first ml. The volume of the organic phase containing permethrin was reduced to 1 ml and 5 ml of hexane:dichloromethane (1:1, v:v) were added before injecting into the GPC. The analytes were collected in the fraction between 175 ml and 225 ml of hexane:dichloromethane (1:1, v:v) at a 5 ml/min flow. The extract was purified on a column prepared with 5 g of silica gel, 4 g of florisil (5% of water), 4 g of Na₂SO₄. Permethrin was recovered with 30 ml of acetone:hexane (1:4, v:v) after discarding the first 15 ml. Sulfur was eliminated on 1g of copper activated with 3 ml of nitric acid (7N), rinsed with MilliQ water until neutral pH, conditioned with 3 ml of acetone and 3 ml of hexane.

GC/ECD was used for the quantitative analyses of permethrin (HP 6890 Series) and GC/MS (Varian 1200 L) for carbendazim (m/z:181/292/492), Irgarol 1051[®] (m/z: 82/196/238/253) and octhilinone (m/z: 101/114/213). TBT and TPT were analysed on a GC/PFPD- mode P (Varian star 3400 CX). Table 2 shows the operating conditions. All concentrations were determined using external standards and corrected for the recovery. Organotin results refer to the TBT and TPT as the ion (Bu₃Sn⁺ and Phe₃Sn⁺).

In order to test the validity of the methods, a routine sample recovery measurement have be made as described by Hess et al. (1995) and Wegener et al. (1997). Samples of sludge were spiked with the biocides at four concentrations: 50%, 100%, 150% and 200% of their initial concentration (carbendazim, permethrin, TBT, TPT) or at about 5, 10, 15 and 20 times the detection limit for Irgarol 1051[®] and octhilinone (not found in the sample). These samples were taken through the procedures. The slope of the regression of the experimental quantities expressed as a function of the theoretical quantities gives the recovery. The standard deviation of the slope corresponds to the standard deviation of the recovery. The limit of detection (signal-to-noise ratio of 3) of the whole analytical procedure was calculated with a spiked sam-

Table 2. Conditions of the chromatographic analysis for the detection of the biocides in sewage sludge

	Injection system	Column/Program
Car	Varian 1079 PTV 85°C (0.2 min)	DB-5 - ϕ 0.25 x 60 m, 0.25 μ m 80°C (0.5 min)-50°C/min to 150°C (1min) 2.5°C/min to 240°C (20 min)
Ir	100°C/min to 250°C	DB-5 - ϕ 0.25 x 60 m, 0.25 μ m 80°C (1 min) -8°C/min to 190°C (30 min)
Oc	Varian 1079 PTV 250°C	DB-5 - ϕ 0.25 x 60 m, 0.25 μ m 80°C (1 min)-40 °C/min to 150°C-3 °C/min to 220 °C
Per	On column	DB-5 - ϕ 0.25 x 60 m, 0.25 μ m 80°C (0.5 min)-50 °C/min to 150°C 150°C (1 min)-2.5°C/min to 285°C
TBT TPT	Splitless 250°C	HP-1 - ϕ 0.20 x 25 m, 0.33 μ m 40°C (1 min)-10°C/min to 280°C (15 min)

ple and was adjusted to the recovery rate. Suitable recoveries were obtained: carbendazim, 85.4% \pm 9.6; Irgarol 1051[®], 95.0% \pm 10.1; othililnone 76.9% \pm 1.4, cis-permethrin, 78.5% \pm 9.7; trans-permethrin, 82.3% \pm 9.7; TBT, 106.5% \pm 18.6 and TPT, 91.4% \pm 8.4. The limits of detection were (ng/kg of bulk sludge): carbendazim, 12.5; Irgarol 1051[®], 65.8; othililnone, 160; cis-permethrin, 9.1; trans-permethrin, 8.7; TBT, 323 and TPT, 201. A sample has been analysed three times: the RSD was 4%, 18% and 19 % for carbendazim, cis-permethrin and trans-permethrin. As suitable recoveries and repetability have been found with these methods, single analysis has been carried out for each sample.

RESULTS AND DISCUSSION

The concentrations of both sampling periods are shown in Table 3. All substances, except for othililnone that was not detected, were found at a level of μ g/kg dry weight (d.w.) Although present in all samples, measured concentrations of carbendazim were low (mean of 6.8 μ g/kg d.w.). Irgarol 1051[®] was found in 7 out of 24 samples at low concentrations only (mean of 2.9 μ g/kg d.w.). It is used as an algicide in emulsion paints for walls and its presence in sewage sludge may be the result of leaching during rain events or misuse during handling of equipment containing Irgarol 1051[®].

Both permethrin isomers were detected in all sludge samples (mean of 55.3 μ g/kg d.w. for cis-isomer and 42.8 μ g/kg d.w. for trans-isomer). In most cases the concentration of the cis-isomer was higher than the trans-isomer's. This cis/trans ratio is comparable to ratios observed in other environmental matrices: 1.6 in water and 1.5 in moss (Hancock et al. 1997). This might be explained by a lower degradability of the cis isomer. The study of Sharom and Salomon (1981) on degradation of permethrin in water and flooded sediment has shown that the biological degradability and chemical hydrolysis of the cis-isomer is lower. Commercial formulas and applications have distinct cis/trans ratios; as the cis/trans ratio depends on the rates of degradation, it cannot be used as a marker to identify the sources of permethrin in sewage sludge. The concentrations of permethrin of this study are lower by a factor 100 than those found in surveys carried out in the UK (Woodhead 1983; Rogers et

Table 3. Concentration of biocides in sewage sludge of WWTPs of different monitoring sites ($\mu\text{g}/\text{kg}$ d.w.)

Sites	Date	DSS	DM	Car	Ir	Cis-Per	Trans-Per	Cis/trans	TBT	TPT
A11	23 Jan	110	4.3	1.7	nd	16.8	10.0	1.7	125.0	282.6
	09 May	230	2.5	1.6	nd	39.1	22.6	1.7	22.9	17.6
A12	11 Jan	70	1.7	2.9	nd	59.5	36.7	1.6	180.4	nd
	07 May	190	1.4	9.2	nd	71.3	37.2	1.9	172.2	nd
A15	16 Jan	120	2.8	2.2	nd	50.3	12.6	4.0	115.9	12.8
	07 May	230	2.3	1.9	nd	34.2	27.7	1.5	23.6	nd
A16	16 Jan	120	4.0	2.1	nd	7.2	2.9	2.5	193.0	nd
	07 May	230	3.4	4.9	nd	13.5	23.9	0.6	43.8	nd
B11	11 Jan	40	6.9	8.1	nd	80.1	102.5	0.8	505.1	33.0
	07 May	60	7.2	5.2	nd	74.6	81.4	0.9	82.8	4.3
B14	16 Jan	nd	6.6	9.2	nd	127.0	88.6	1.4	137.9	39.6
	09 May	60	7.6	2.1	nd	133.6	78.9*	1.7	18.6	12.2
B16	21/02	60	3.0	21.1	20.1	64.8	88.1	0.7	38.5	nd
	08 May	30	4.5	5.0	nd	36.0	26.8	1.3	24.2	nd
B25	17 Jan	90	6.5	21.3	30.4	55.6	30.3	1.8	648.5	nd
	08 May	190	5.6	11.3	3.3	82.7	65.6	1.3	137.0	9.6
C2	16 Jan	60	6.6	4.3	nd	38.2	45.0	0.8	323.8	35.1
	09 May	120	5.4	2.9	nd	20.5	33.2	0.6	46.2	nd
C4	17 Jan	90	5.1	5.7	1.5	22.3	14.3	1.6	165.8	33.4
	08 May	190	5.5	5.8	nd	53.7	37.3	1.4	102.6	5.4
C5	17 Jan	100	8.0	5.4	1.6	27.4	23.4	1.2	125.8	nd
	08 May	nd	6.3	5.8	nd	44.9	43.4	1.0	30.9	nd
C9	17 Jan	30	3.7	3.7	7.2	62.7	34.5	1.8	166.2	nd
	08 May	30	3.1	20.1	5.9	110.3	64.8	1.7	117.3	nd
Mean			6.8	2.9	55.3	42.8	1.5	147.8	20.2	
Median			5.1	0.0	52.0	35.6	1.5	121.2	0.0	
Minimum			1.6	nd	7.2	2.9	0.6	18.6	nd	
Maximum			21.3	30.4	133.6	102.5	4.0	648.5	282.6	

nd : not detected (equal to zero for the calculation of the mean and the median)

DSS: duration of sludge stocking (days), DM: dry matter (%)

*confirmation of the concentrations by GC/MS

al. 1989). This difference can be explained by the release from point sources like textile industries, which are not present in the sites of the present study, by a different permethrin usage of the two countries or by a decreasing consumption in recent years. Additionally, the technology used for wastewater treatment might play a significant role for degradation of permethrin.

Among all compounds studied here, the highest concentrations were found for TBT (mean of $147.8 \mu\text{g TBT}^+/\text{kg}$ d.w.). TPT was detected in 11 samples at contents below $40 \mu\text{g TPT}^+/\text{kg}$ d.w. except for one sample from site A11 ($282.6 \mu\text{g TPT}^+/\text{kg}$ d.w.). As the use of TPT acetate is authorized in Switzerland, its presence in sewage sludge could be explained by agricultural applications: surface runoff from treated soils and spray drift during applications of fungicides with subsequent dry or wet deposition on impervious surfaces, followed by washoff into sewer systems during rain events. The high concentration of TPT in sewage sludge of site A11, which has a separate system and is located in a rural area, could be explained by cleaning of agricultural equipment (i.e. measuring utilities) or improper disposal of a product

containing TPT. It seems to be time limited, as the sample collected in May has shown lower concentrations of TPT than the January sample. A survey in 1995 in 25 WWTP showed higher concentrations with an average of 1100 $\mu\text{g TBT}^+/\text{kg d.w.}$ and 500 $\mu\text{g TPT}^+/\text{kg d.w.}$ (Fent 1996b). Nevertheless, compared to the concentrations measured by Becker van Slooten et al. (1994), the TBT and TPT amounts are in the same range with contents of approximately 200 $\mu\text{g TBT}^+/\text{kg d.w.}$ and a maximum of 2150 $\mu\text{g/kg d.w.}$ and TPT detected in 9 out of 54 samples with a maximum of 3400 $\mu\text{g/kg d.w.}$ The highest concentrations of TBT and TPT observed in the present study are lower by a factor 3 and 12 respectively compared to the highest concentrations of TBT and TPT found by Becker van Slooten et al. (1994). This might indicate a decrease in consumption. In a more recent survey (Bancon-Montigny et al. 1999) TBT was found in concentrations at the same level (71-225 $\mu\text{g TBT}^+/\text{kg d.w.}$).

It cannot be excluded that wastewater and sludge treatment influence degradation of biocides. In order to determine if hygienisation of sludge enhances degradation of TBT, permethrin and carbendazim, the mean concentrations of disinfected (WWTPs B25, C2, C4, C5) and not disinfected sludge (WWTPs B11, B14, C9) were compared with a directional Wilcoxon-Mann-Whitney test. Both groups of WWTPs include an activated sludge process and are thus comparable with regard to wastewater treatment. Concentrations for permethrin in disinfected sludge (mean of 159.5 $\mu\text{g/kg d.w.}$) were significantly lower than concentrations in not-disinfected sludge (346.3 $\mu\text{g/kg d.w.}$) at the 95 % level of significance ($U=0, \alpha = 0,014$). This indicates that hygienisation might enhance the degradation process of permethrin. However, the difference was not significant for the other compounds.

Concentrations of compounds in sewage sludge do not provide enough information to fully understand their release into wastewater. For instance, high loads of organic matter in wastewater induce a high sludge production which results in dilution of the concentrations of the studied compounds. Specific loads in sewage sludge per inhabitant connected to the WWTP per year account for this effect. The method has been described by Kupper et al. (2004). Figure 1 gives specific loads calculated for carbendazim, permethrin and TBT. The statistical significance of differences between the specific loads of sites A, B and C was determined by a directional Wilcoxon-Mann-Whitney test (Table 4).

Table 4. U and significance level (α) of the non-parametric Wilcoxon-Mann-Whitney test

	B>A		C>A		C>B	
	U	α	U	α	U	α
Carbendazim	0	0.014*	0	0.014*	7	0.443
Permethrin	1	0.029*	1	0.029*	8	0.557
TBT	7	0.443	0	0.014*	6	0.343

* significantly superior at $\alpha = 0.05$

The presence of biocides in sludge of domestic origin (sites A) are probably the result of preservative's leaching during cleaning and washing of treated wood, textile, paper or surfaces. Biocides might also disperse into wastewater due to improper

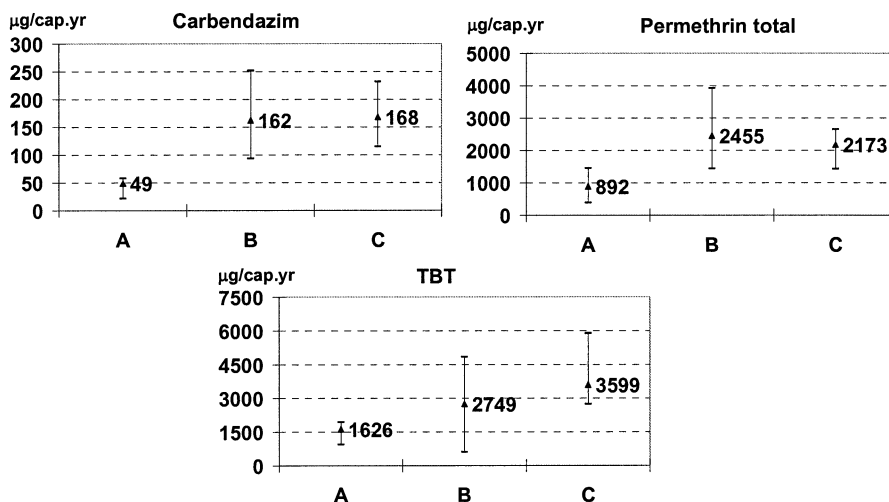


Figure 1. Minimum, mean and maximum of specific loads ($\mu\text{g}/\text{connected inhabitant (cap.)}\cdot\text{year}$) on different types of monitoring sites.

disposal of biocidal products. A possible source of TBT may be leaching from PVC products and dispersion paints (Fent 1996a). The considerable loads of permethrin might be due to its use in veterinary products against fleas or in insecticides for private households.

Except for TBT, specific loads for sites of type B are significantly higher than loads from sites of type A ($\alpha = 0.05$). It seems likely that surface runoff contributes to the load in sewage sludge. This could be explained by leaching during rain events of substances used for purposes such as wood protector and preservative in outdoor paints, or spraying drifts on impervious surfaces and the use in private gardens of products containing carbendazim, permethrin or benomyl, a substance quickly converted to carbendazim. Evaporation of biocides after use may cause ubiquitous distribution in the environment. According to Tomlin (1997), carbendazim and permethrin exhibit low vapour pressure ($<0.1 \text{ mPa}$ at 20°C). They are thus expected to occur in the particulate phase of the atmosphere. The contamination of sewage sludge might thus be the consequence of dry or wet deposition and runoff from impervious surfaces into the sewer systems.

Differences between loads of sites B and C were not significant for the three biocides. It can be concluded that industrial activities don't induce additional loads. As mentioned earlier, hygienisation, which is present on most WWTPs of sites of type C, might enhance degradation of permethrin, potentially masking a release of this compound by the industry. Further investigations are required to assess the role of industrial activities for the release of permethrin. For TBT, there is a significant difference between specific loads of sites of type A and sites of type C, but not between sites of type A and B. Thus, it seems that industrial wastewater is a potential source. As no data on the consumption of biocides in Switzerland are available, we are unable to compare the total loads found in this study with the amounts used. The comparison with other countries is also difficult because of a lack of data and the

differences in terms of legislation and uses of the compounds on the market.

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