

# Supporting Information

## Z-Selective synthesis of vinyl boronates through Fe-catalyzed alkyl radical addition

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## General Information

NMR spectra were recorded on a 400 MHz instrument at ambient temperature in  $\text{CDCl}_3$  or  $\text{CD}_2\text{Cl}_2$  as solvent.  $^1\text{H}$  NMR chemical shifts ( $\delta$ , ppm) were measured relative to tetramethylsilane (TMS) signal in  $\text{CDCl}_3$  (0.00 ppm) unless otherwise stated. Splitting patterns are designated as s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet; br, broad.  $^{13}\text{C}$  NMR chemical shifts ( $\delta$ , ppm) are reported relative to  $\text{CDCl}_3$  signal (77.16 ppm) unless otherwise stated.

Unless otherwise noted, all chemicals were commercially available and were used as received without further purifications. Solvents were purified using a two-column solid-state purification system and transferred to glove box without exposure to air by the aid of a Straus flask. Zn powder (<10 $\mu$ , 98%+) was purchased from Aldrich. Anhydrous dimethylacetamide (DMA) (99.8% purity) was commercially purchased and stored under nitrogen. Iron(II) triflate ( $\text{FeOTf}_2$ , 98% purity) was purchased from Aldrich or Acros. All the Z/E ratios are determined by  $^1\text{H}$  NMR analysis after purification of the compounds.

## Experimental protocols

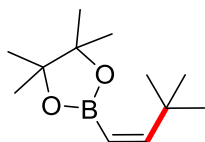
### General procedure for the reaction with tertiary and secondary alkyl halides (General procedure A)

Zinc powder (25.5 mg, 0.39 mmol) was put in a 10 mL dry vial equipped with a stirring bar inside the glovebox. DMA (0.25 mL, 0.5 M) and TMSI (0.52 mg 0.0026 mmol) were added to this vial. The mixture was vigorously shaken until the disappearance of the white fume. FeOTf<sub>2</sub> (9.2 mg, 20 mol%), ethynyl boronic pinacolester (20 mg, 0.13 mmol) and the alkyl halide (0.39 mmol) were added. The vial was then sealed and its content was allowed to stir overnight (16-20 h) at 50°C. The mixture was quenched with water. *n*-Dodecane (29.5 μL, 0.13 mmol) was added as an internal standard, and the mixture was extracted into ca. 4 mL of ethyl acetate. The organic layer was analyzed by GC-MS. It was then dried over sodium sulfate and the solvent was evaporated under reduced pressure. The crude mixture was purified by column chromatography (ethyl acetate/hexanes).

### General procedure for the reaction with primary alkyl halides (General procedure B)

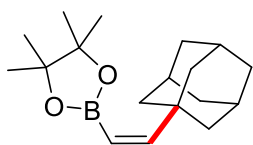
Zinc powder (25.5 mg, 0.39 mmol) was put in a 10 mL dry vial equipped with a stirring bar inside the glovebox. DMA (0.25 mL, 0.5 M) and TMSI (0.52 mg 0.0026 mmol) were added to this vial. The mixture was vigorously shaken until the disappearance of the white fume. The iron complex **4** (5 mol%), ethynyl boronic pinacolester (20 mg, 0.13 mmol) and the alkyl halide (0.39 mmol) were added. The vial was then sealed and its content was allowed to stir overnight (16-20 h) at 50°C. The mixture was quenched with water. *n*-Dodecane (29.5 μL, 0.13 mmol) was added as an internal standard, and the mixture was extracted into ca. 4 mL of ethyl acetate. The organic layer was analyzed by GC-MS. It was then dried over sodium sulfate and the solvent was evaporated under reduced pressure. The crude mixture was purified by column chromatography (ethyl acetate/hexanes).

### Synthesis of (Z)-2-(3,3-dimethylbut-1-en-1-yl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (3a)



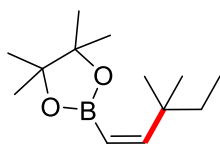
The synthesis was done following the general procedure A. The crude residue was purified via column chromatography over silica gel (hexanes:ethyl acetate 9:1). Off white oil (25.9mg, 95%, Z:E: 44:1).  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR data were consistent with literature.<sup>1</sup>

### Synthesis of 2-((Z)-2-((-adamantan-1-yl)vinyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (3b)



The synthesis was done following the general procedure A. The crude residue was purified via column chromatography over silica gel (hexanes:ethyl acetate 9:1). Off white oil (34.1mg, 91%, Z:E: 28:1).  $^1\text{H}$  NMR (400 MHz, Chloroform-d)  $\delta$  6.07 (d,  $J = 15.1$  Hz, 1H), 5.19 (d,  $J = 15.1$  Hz, 1H), 1.99 (d,  $J = 4.3$  Hz, 3H), 1.72 (dd,  $J = 11.9, 4.3$  Hz, 12H), 1.31 (s, 12H).  $^{13}\text{C}$  NMR (101 MHz, Chloroform-d)  $\delta$  159.44, 83.35, 41.96, 36.79, 28.55, 24.83. HRMS (APPI/LTQ-Orbitrap)  $m/z$ :  $[\text{M}]^+$  Calcd for  $\text{C}_{18}\text{H}_{29}\text{BO}_2^+$  288.2255; Found 288.2253.

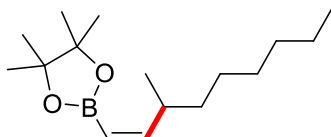
### Synthesis of (Z)-2-(3,3-dimethylpent-1-en-1-yl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (3c)



The synthesis was done following the general procedure A. The crude residue was purified via column chromatography over silica gel (hexanes:ethyl acetate 9:1). Off white oil (11.7mg, 40%, Z:E: 18:1).  $^1\text{H}$  NMR (400 MHz, Chloroform-d)  $\delta$  6.19 (d,  $J = 15.0$  Hz, 1H), 5.27 (d,  $J = 15.1$  Hz, 1H), 1.40 (q,  $J = 7.5$  Hz, 2H), 1.30 (s, 12H), 1.07 (s, 3H), 0.84 (t,  $J = 7.5$  Hz, 3H).  $^{13}\text{C}$  NMR (101

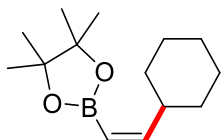
MHz, Chloroform-d)  $\delta$  158.57, 83.27, 38.23, 35.60, 26.96, 24.81, 8.97.  $^{11}\text{B}$  NMR (128 MHz, Chloroform-d)  $\delta$  30.76. HRMS (APPI/LTQ-Orbitrap)  $m/z$ :  $[\text{M} + \text{H}]^+$  Calcd for  $\text{C}_{13}\text{H}_{26}\text{BO}_2^+$  225.2020; Found 225.2025.

### Synthesis of (Z)-4,4,5,5-tetramethyl-2-(3-methylnon-1-en-1-yl)-1,3,2-dioxaborolane (3d)



The synthesis was done following the general procedure A. The crude residue was purified via column chromatography over silica gel (hexanes:ethyl acetate 9:1). Off white oil (26.6mg, 77%, Z:E: 13:1).  $^1\text{H}$  NMR (400 MHz, Chloroform-d)  $\delta$  6.15 (dd,  $J = 13.5, 9.8$  Hz, 1H), 5.26 (d,  $J = 13.5$  Hz, 1H), 2.90 (td,  $J = 6.7, 3.4$  Hz, 1H), 1.27 (p,  $J = 4.3$  Hz, 25H), 0.96 (d,  $J = 6.6$  Hz, 3H), 0.89 (t,  $J = 6.9$  Hz, 3H).  $^{13}\text{C}$  NMR (101 MHz, Chloroform-d)  $\delta$  161.15, 82.67, 37.21, 35.96, 31.90, 29.34, 27.27, 24.87, 24.72, 22.69, 21.22, 14.10. HRMS (APPI/LTQ-Orbitrap)  $m/z$ :  $[\text{M}]^+$  Calcd for  $\text{C}_{16}\text{H}_{31}\text{BO}_2^+$  266.2412; Found 266.2412.

### Synthesis of (Z)-2-(2-cyclohexylvinyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (3e)

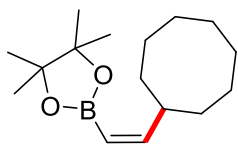


The synthesis was done following the general procedure A. The crude residue was purified via column chromatography over silica gel (hexanes:ethyl acetate 9:1). Off white oil (22.7mg, 74%, Z:E: 8:1).  $^1\text{H}$  NMR (400 MHz, Chloroform-d)  $\delta$  6.27 (dd,  $J = 13.5, 9.3$  Hz, 1H), 5.24 (dd,  $J = 13.6, 0.9$  Hz, 1H), 2.73 (dtd,  $J = 14.1, 11.3, 10.7, 3.6$  Hz, 1H), 1.84 – 1.53 (m, 8H), 1.28 (d,  $J = 1.3$  Hz, 12H).  $^{13}\text{C}$  NMR (101 MHz, Chloroform-d)  $\delta$  160.59, 82.74, 40.59, 33.33, 26.02, 25.76, 24.81. HRMS (APPI/LTQ-Orbitrap)  $m/z$ :  $[\text{M}]^+$  Calcd for  $\text{C}_{14}\text{H}_{25}\text{BO}_2^+$  236.1942; Found 236.1941.

Synthesis of (Z)-2-(2-cyclohexylvinyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (3e) in a gram scale

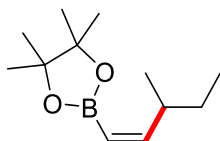
Zinc powder (941.96 mg, 14.4 mmol) was put in a 10 mL dry vial equipped with a stirring bar inside the glovebox. DMA (9.2 mL, 0.5 M) and TMSI (19.19 mg 0.095 mmol) were added to this vial. The mixture was vigorously shaken until the disappearance of the white fume. FeOTf<sub>2</sub> (339.8 mg, 20 mol%), ethynyl boronic pinacolester (1 g, 4.8 mmol) and the alkyl halide (3.028 g, 14.4 mmol) were added. The vial was then sealed and its content was allowed to stir overnight (16-20 h) at 50°C. The mixture was quenched with water. The mixture was extracted with ethyl acetate, the combined organic layers were dried over sodium sulfate and the solvent was evaporated under reduced pressure. The crude mixture was purified by column chromatography (ethyl acetate/hexanes 1:9) Off white oil (793.4mg, 70%, Z:E: 8:1) .

#### Synthesis of (Z)-2-(2-cyclooctylvinyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (3f)



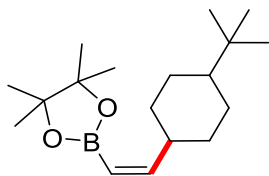
The synthesis was done following the general procedure A. The crude residue was purified via column chromatography over silica gel (hexanes:ethyl acetate 9:1). Off white oil (19.2mg, 56%, Z:E: 20:1). <sup>1</sup>H NMR (400 MHz, Chloroform-d) δ 6.34 (d, J = 3.6 Hz, 1H), 5.17 (d, J = 13.3 Hz, 1H), 3.07 (qt, J = 9.6, 3.4 Hz, 1H), 1.72 – 1.47 (m, 14H), 1.28 (s, 12H). <sup>13</sup>C NMR (101 MHz, Chloroform-d) δ 161.92, 82.72, 39.32, 33.18, 26.76, 25.38, 24.88. <sup>11</sup>B NMR (128 MHz, Chloroform-d) δ 29.91. HRMS (APPI/LTQ-Orbitrap) m/z: [M]<sup>+</sup> Calcd for C<sub>16</sub>H<sub>29</sub>BO<sub>2</sub>+ 264.2255; Found 264.2255.

#### Synthesis of (Z)-4,4,5,5-tetramethyl-2-(3-methylpent-1-en-1-yl)-1,3,2-dioxaborolane (3g)



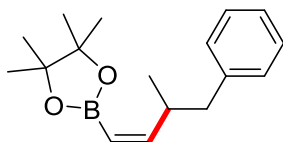
The synthesis was done following the general procedure A. The crude residue was purified via column chromatography over silica gel (hexanes:ethyl acetate 9:1). Off white oil (19.7mg, 72%, Z:E: 13:1).  $^1\text{H}$  NMR (400 MHz, Chloroform- $d$ )  $\delta$  6.13 (dd,  $J = 13.5, 9.8$  Hz, 1H), 5.27 (d,  $J = 13.5$  Hz, 1H), 2.80 (tt,  $J = 9.0, 6.1$  Hz, 1H), 1.33 – 1.20 (m, 12H), 0.95 (d,  $J = 6.6$  Hz, 3H), 0.84 (t,  $J = 7.5$  Hz, 3H).  $^{13}\text{C}$  NMR (101 MHz, Chloroform- $d$ )  $\delta$  160.71, 82.71, 37.68, 29.96, 24.89, 24.70, 20.92, 11.78.  $^{11}\text{B}$  NMR (128 MHz, Chloroform- $d$ )  $\delta$  29.96. HRMS (nanochip-ESI/LTQ-Orbitrap)  $m/z$ :  $[\text{M} + \text{H}]^+$  Calcd for  $\text{C}_{12}\text{H}_{24}\text{BO}_2^+$  211.1864; Found 211.1864.

**Synthesis of (Z)-2-(2-(4-(tert-butyl)cyclohexyl)vinyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (3h)**



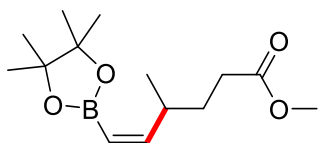
The synthesis was done following the general procedure A. The crude residue was purified via column chromatography over silica gel (hexanes:ethyl acetate 9:1). Off white oil (37.2mg, 98%, Z:E: 9:1).  $^1\text{H}$  NMR (400 MHz, Chloroform- $d$ )  $\delta$  6.83 (dd,  $J = 13.7, 9.4$  Hz, 1H), 5.36 (d,  $J = 13.6$  Hz, 1H), 1.88 – 1.62 (m, 6H), 1.61 – 1.53 (m, 4H), 1.28 (d,  $J = 2.7$  Hz, 12H), 0.86 (s, 9H).  $^{13}\text{C}$  NMR (101 MHz, Chloroform- $d$ )  $\delta$  158.28, 82.71, 48.36, 34.59, 32.49, 27.51, 24.80, 22.14, 14.11.  $^{11}\text{B}$  NMR (128 MHz, Chloroform- $d$ )  $\delta$  29.61. HRMS (APPI/LTQ-Orbitrap)  $m/z$ :  $[\text{M} + \text{H}]^+$  Calcd for  $\text{C}_{18}\text{H}_{34}\text{BO}_2^+$  293.2646; Found 293.2648.

**Synthesis of (Z)-4,4,5,5-tetramethyl-2-(3-methyl-4-phenylbut-1-en-1-yl)-1,3,2-dioxaborolane (3i)**



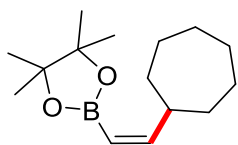
The synthesis was done following the general procedure A. The crude residue was purified via column chromatography over silica gel (hexanes:ethyl acetate 9:1). Off white oil (14.1mg, 40%, Z:E: 7:1).  $^1\text{H}$  NMR (400 MHz, Chloroform-d)  $\delta$  7.32 – 7.17 (m, 5H), 6.28 (dd,  $J$  = 13.5, 9.6 Hz, 1H), 5.29 (d,  $J$  = 13.4 Hz, 1H), 3.31 (dq,  $J$  = 9.6, 6.9 Hz, 1H), 2.61 (ddd,  $J$  = 61.7, 13.3, 7.2 Hz, 2H), 1.27 (d,  $J$  = 2.2 Hz, 11H), 0.98 (d,  $J$  = 6.5 Hz, 3H).  $^{13}\text{C}$  NMR (101 MHz, Chloroform-d)  $\delta$  160.16, 140.77, 129.32, 127.96, 125.64, 82.79, 43.82, 37.80, 24.86, 20.19.  $^{11}\text{B}$  NMR (128 MHz, Chloroform-d)  $\delta$  29.89. HRMS (APPI/LTQ-Orbitrap)  $m/z$ :  $[\text{M}]^+$  Calcd for  $\text{C}_{17}\text{H}_{25}\text{BO}_2^+$  272.1942; Found 272.1951.

### Synthesis of methyl (S,Z)-4-methyl-6-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)hex-5-enoate (3j)



The synthesis was done following the general procedure A. The crude residue was purified via column chromatography over silica gel (hexanes:ethyl acetate 8:2). Off white oil (13.24mg, 38%, Z:E: 11:1).  $^1\text{H}$  NMR (400 MHz, Chloroform-d)  $\delta$  6.11 (dd,  $J$  = 13.5, 10.0 Hz, 1H), 5.33 (d,  $J$  = 13.5 Hz, 1H), 4.32 (dd,  $J$  = 8.2, 4.6 Hz, 2H), 3.66 (s, 3H), 3.02 – 2.85 (m, 1H), 2.30 (t,  $J$  = 7.9 Hz, 2H), 1.82 – 1.64 (m, 2H), 1.27 (s, 12H), 1.01 (d,  $J$  = 6.5 Hz, 3H).  $^{13}\text{C}$  NMR (101 MHz, Chloroform-d)  $\delta$  174.41, 159.30, 130.90, 128.82, 82.87, 51.38, 35.87, 32.23, 31.93, 24.87, 24.75, 21.16.  $^{11}\text{B}$  NMR (128 MHz, Chloroform-d)  $\delta$  29.93. HRMS (APPI/LTQ-Orbitrap)  $m/z$ :  $[\text{M} + \text{H}]^+$  Calcd for  $\text{C}_{14}\text{H}_{26}\text{BO}_4^+$  269.1919; Found 269.1917.

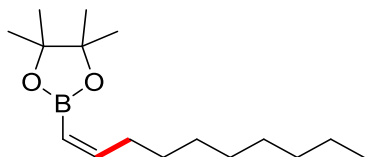
### Synthesis of (Z)-2-(2-cycloheptylvinyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (3k)





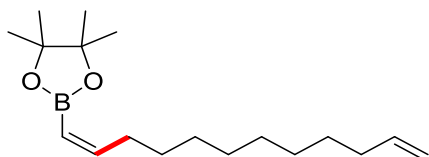
The synthesis was done following the general procedure A. The crude residue was purified via column chromatography over silica gel (hexanes:ethyl acetate 9:1). Off white oil (23.0mg, 70.8%, Z:E: 13:1).  $^1\text{H}$  NMR (400 MHz, Chloroform-d)  $\delta$  6.36 (dd,  $J = 13.4, 9.7$  Hz, 1H), 5.18 (d,  $J = 13.4$  Hz, 1H), 2.92 (tq,  $J = 9.8, 5.0, 3.8$  Hz, 1H), 1.77 – 1.40 (m, 7H), 1.28 (s, 17H).  $^{13}\text{C}$  NMR (101 MHz, Chloroform-d)  $\delta$  161.28, 82.72, 41.93, 35.12, 28.60, 26.58, 24.83.  $^{11}\text{B}$  NMR (128 MHz, Chloroform-d)  $\delta$  29.67. HRMS (APPI/LTQ-Orbitrap)  $m/z$ :  $[\text{M} + \text{H}]^+$  Calcd for  $\text{C}_{15}\text{H}_{28}\text{BO}_2^+$  251.2177; Found 251.2178.

### Synthesis of (Z)-2-(dec-1-en-1-yl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (3l)



The synthesis was done following the general procedure B. The crude residue was purified via column chromatography over silica gel (hexanes:ethyl acetate 9:1). Off white oil (26.6mg, 77%, Z:E: 10:1).  $^1\text{H}$  NMR (400 MHz, Chloroform-d)  $\delta$  6.46 (dt,  $J = 15.0, 7.8$  Hz, 1H), 5.35 (d,  $J = 13.5$  Hz, 1H), 2.41 (q,  $J = 7.3$  Hz, 2H), 1.28 (d,  $J = 8.7$  Hz, 24H), 0.90 (d,  $J = 7.4$  Hz, 3H).  $^{13}\text{C}$  NMR (101 MHz, Chloroform-d)  $\delta$  155.28, 82.76, 32.20, 31.93, 29.71, 29.47, 29.41, 29.29, 29.08, 24.84, 24.02, 22.70, 14.12.  $^{11}\text{B}$  NMR (128 MHz, Chloroform-d)  $\delta$  34.43, 29.93. HRMS  $m/z$ :  $[\text{M} + \text{H}]^+$  Calcd for  $\text{C}_{16}\text{H}_{32}\text{BO}_2^+$  267.2490; Found 267.2485.

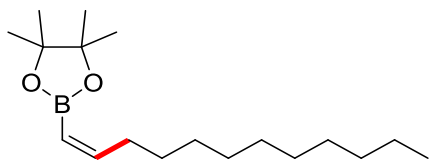
### Synthesis of (Z)-2-(dodeca-1,11-dien-1-yl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (3m)



The synthesis was done following the general procedure B. The crude residue was purified via column chromatography over silica gel (hexanes:ethyl acetate 9:1). Off white oil (23.2mg, 61%, Z:E: 10:1).  $^1\text{H}$  NMR (400 MHz, Chloroform-d)  $\delta$  6.44 (dt,  $J = 14.2, 7.3$  Hz, 1H), 5.82 (ddt,  $J = 16.9, 10.1, 6.7$  Hz, 1H), 5.33 (dt,  $J = 13.4, 1.4$  Hz, 1H), 5.04 – 4.89 (m, 2H), 2.04 (qd,  $J = 7.6, 3.8$

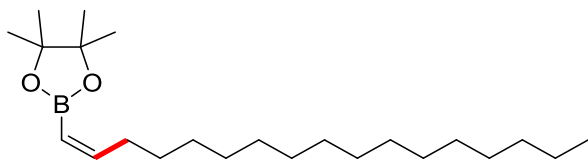
Hz, 2H), 1.45 – 1.20 (m, 27H).  $^{13}\text{C}$  NMR (101 MHz, Chloroform-d)  $\delta$  155.21, 139.22, 114.05, 82.79, 33.83, 32.39, 29.40, 29.34, 29.13, 28.95, 24.81, 23.98.  $^{11}\text{B}$  NMR (128 MHz, Chloroform-d)  $\delta$  34.52, 29.97. HRMS (APPI/LTQ-Orbitrap)  $m/z$ :  $[\text{M}]^+$  Calcd for  $\text{C}_{18}\text{H}_{33}\text{BO}_2^+$  292.2568; Found 292.2573.

### Synthesis of (Z)-2-(dodec-1-en-1-yl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (3n)



The synthesis was done following the general procedure B. The crude residue was purified via column chromatography over silica gel (hexanes:ethyl acetate 9:1). Off white oil (13.4mg, 35%, Z:E: 13:1).  $^1\text{H}$  NMR (400 MHz, Chloroform-d)  $\delta$  6.45 (dt,  $J = 14.3, 7.5$  Hz, 1H), 5.34 (dt,  $J = 13.5, 1.4$  Hz, 1H), 2.41 (qd,  $J = 7.3, 1.4$  Hz, 2H), 1.28 (d,  $J = 3.8$  Hz, 28H), 0.90 (t,  $J = 6.7$  Hz, 3H).  $^{13}\text{C}$  NMR (101 MHz, Chloroform-d)  $\delta$  155.28, 82.76, 32.20, 31.93, 29.66, 29.63, 29.48, 29.45, 29.37, 29.07, 24.84, 24.81, 22.69, 14.12.  $^{11}\text{B}$  NMR (128 MHz, Chloroform-d)  $\delta$  30.02. HRMS (APPI/LTQ-Orbitrap)  $m/z$ :  $[\text{M} + \text{H}]^+$  Calcd for  $\text{C}_{18}\text{H}_{36}\text{BO}_2^+$  295.2803; Found 295.2798.

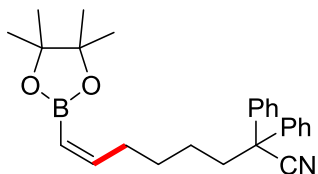
### Synthesis of (Z)-2-(heptadec-1-en-1-yl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (3o)



The synthesis was done following the general procedure B. The crude residue was purified via column chromatography over silica gel (hexanes:ethyl acetate 9:1). Off white oil (14.7mg, 31%, Z:E: 9:1).  $^1\text{H}$  NMR (400 MHz, Chloroform-d)  $\delta$  6.45 (dt,  $J = 14.2, 7.5$  Hz, 1H), 5.35 (dd,  $J = 13.5, 1.5$  Hz, 1H), 2.49 – 2.35 (m, 2H), 1.29 (q,  $J = 6.2, 4.9$  Hz, 38H), 0.90 (t,  $J = 6.6$  Hz, 3H).  $^{13}\text{C}$  NMR (101 MHz, Chloroform-d)  $\delta$  155.29, 82.76, 43.93, 32.45, 32.20, 31.94, 29.71, 29.67, 29.47, 29.43, 29.37, 29.18, 29.08, 24.84, 24.82, 22.70, 14.12.  $^{11}\text{B}$  NMR (128 MHz, Chloroform-d)  $\delta$  34.55,

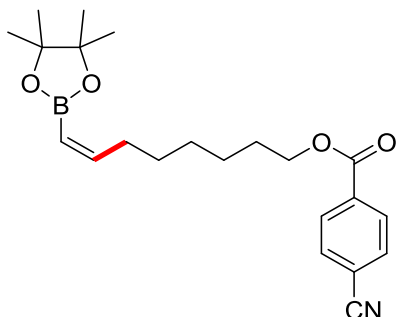
30.04. HRMS (APPI/LTQ-Orbitrap) m/z: [M + H]<sup>+</sup> Calcd for C<sub>24</sub>H<sub>48</sub>BO<sub>2</sub><sup>+</sup> 379.3742; Found 379.3735.

**Synthesis of (Z)-2,2-diphenyl-8-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)oct-7-enenitrile (3p)**



The synthesis was done following the general procedure B. The crude residue was purified via column chromatography over silica gel (hexanes:ethyl acetate 8:2). Off white oil (31.8mg, 61%, Z:E: 10:1). <sup>1</sup>H NMR (400 MHz, Chloroform-d) δ 7.47 – 7.24 (m, 10H), 6.39 (dt, J = 14.2, 7.5 Hz, 1H), 5.35 (d, J = 13.5 Hz, 1H), 2.48 – 2.33 (m, 7H), 1.68 – 1.43 (m, 9H), 1.25 (s, 12H). <sup>13</sup>C NMR (101 MHz, Chloroform-d) δ 154.26, 140.38, 128.94, 128.80, 128.56, 128.35, 127.96, 127.76, 126.88, 126.83, 82.84, 51.80, 39.46, 31.63, 29.27, 24.83. <sup>11</sup>B NMR (128 MHz, Chloroform-d) δ 29.95. HRMS (APPI/LTQ-Orbitrap) m/z: [M + H]<sup>+</sup> Calcd for C<sub>26</sub>H<sub>33</sub>BNO<sub>2</sub><sup>+</sup> 402.2599; Found 402.2599.

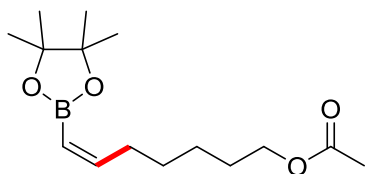
**Synthesis of (Z)-8-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)oct-7-en-1-yl 4-cyanobenzoate (3q)**



The synthesis was done following the general procedure B. The crude residue was purified via column chromatography over silica gel (hexanes:ethyl acetate 8:2). Off white oil (10.7mg, 21.5%, Z:E: 8:1). <sup>1</sup>H NMR (400 MHz, Chloroform-d) δ 8.21 – 8.10 (m, 2H), 7.77 (dd, J = 8.4, 1.8 Hz, 2H), 6.44 (dt, J = 14.2, 7.4 Hz, 1H), 5.36 (dt, J = 13.5, 1.3 Hz, 1H), 4.42 – 4.33 (m, 2H), 2.49 – 2.38 (m, 2H), 1.80 (p, J = 7.2, 6.8 Hz, 4H), 1.54 – 1.37 (m, 10H), 1.28 (s, 14H). <sup>13</sup>C NMR (101

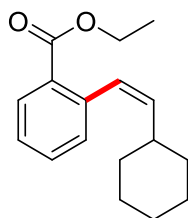
MHz, Chloroform-d)  $\delta$  154.79, 134.32, 132.20, 130.06, 118.02, 116.29, 82.81, 65.98, 32.03, 29.26, 28.63, 28.55, 25.79, 24.85.  $^{11}\text{B}$  NMR (128 MHz, Chloroform-d)  $\delta$  29.99. HRMS (APPI/LTQ-Orbitrap)  $m/z$ :  $[\text{M} + \text{K}]^+$  Calcd for  $\text{C}_{22}\text{H}_{30}\text{BNO}_4\text{K}^+$  422.1899; Found 422.1833.

### Synthesis of (Z)-7-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)hept-6-en-1-yl acetate (3r)



The synthesis was done following the general procedure B. The crude residue was purified via column chromatography over silica gel (hexanes:ethyl acetate 9:1). Off white oil (9.28mg, 25.3%, Z:E: 11:1).  $^1\text{H}$  NMR (400 MHz, Chloroform-d)  $\delta$  6.43 (dt,  $J = 14.2, 7.4$  Hz, 1H), 5.36 (dt,  $J = 13.5, 1.3$  Hz, 1H), 4.07 (t,  $J = 6.8$  Hz, 2H), 2.42 (qd,  $J = 7.2, 1.3$  Hz, 2H), 2.06 (d,  $J = 1.9$  Hz, 3H), 1.71 – 1.57 (m, 4H), 1.27 (d,  $J = 9.1$  Hz, 15H).  $^{13}\text{C}$  NMR (101 MHz, Chloroform-d)  $\delta$  171.20, 154.65, 82.81, 64.59, 31.90, 29.40, 29.21, 28.94, 28.60, 28.34, 25.89, 25.24, 24.83, 21.01.  $^{11}\text{B}$  NMR (128 MHz, Chloroform-d)  $\delta$  33.68, 29.93. HRMS (APPI/LTQ-Orbitrap)  $m/z$ :  $[\text{M}]^+$  Calcd for  $\text{C}_{15}\text{H}_{27}\text{BO}_4$  282.1997; Found 282.2792.

### Synthesis of ethyl (Z)-2-(2-cyclohexylvinyl)benzoate (5)



To a solution of ethyl 2-bromobenzoate (64.6 mg, 0.282 mmol) and potassium carbonate (116.92 mg, 0.846 mmol) in degassed dioxane/water (1.6 mL, 1/1 ratio) was added molecule **3a** (100 mg, 0.423 mmol) and tetrakis(triphenylphosphine)palladium(0) (16.34 mg, 0.0142 mmol) at room temperature. Then the mixture was heated to 95°C for 1 hour. After quenching with water, the mixture was extracted with ethyl acetate and concentrated, obtaining the crude product, which was

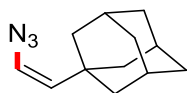
purified by chromatography (hexanes/ethyl acetate 4:1) to afford a pure compound **5** as a off yellow liquid (56.3 mg, yield: 60%).  $^1\text{H}$  NMR (400 MHz, Chloroform-d)  $\delta$  7.94 (dd,  $J = 7.8, 1.4$  Hz, 1H), 7.47 (dd,  $J = 7.6, 1.4$  Hz, 1H), 7.36 – 7.26 (m, 2H), 6.75 (d,  $J = 11.6$  Hz, 1H), 5.54 (t,  $J = 10.9$  Hz, 1H), 4.36 (q,  $J = 7.1$  Hz, 2H), 2.36 – 2.17 (m, 1H), 1.77 – 1.60 (m, 7H), 1.39 (t,  $J = 7.1$  Hz, 3H), 1.33 – 1.27 (m, 3H), 1.18 (d,  $J = 7.4$  Hz, 5H).  $^{13}\text{C}$  NMR (101 MHz, Chloroform-d)  $\delta$  167.39, 139.37, 137.77, 131.41, 130.54, 130.15, 126.81, 126.50, 60.80, 36.87, 33.24, 25.99, 25.62, 14.29. HRMS (ESI/QTOF)  $m/z$ :  $[\text{M} + \text{Na}]^+$  Calcd for  $\text{C}_{17}\text{H}_{22}\text{NaO}_2^+$  281.1512; Found 281.1515.

### Synthesis of 1-((Z)-2-bromovinyl)adamantine (**6**)



Molecule **3b** (100 mg, 0.35 mmol) was dissolved in 4 mL of diethyl ether and cooled to  $-20^\circ\text{C}$ . A solution of  $\text{Br}_2$  (55.9 mg, 0.35 mmol) in DCM (0.35 mL) was added dropwise over 15 minutes, then the mixture was stirred for additional 15 minutes. A solution of NaOMe (44 mg, 0.77 mmol) in MeOH (0.26 mL) was added and after 30 minutes the reaction was quenched with benzoic acid (20 equiv.) in DCM (3.4 mL). Purification by flash chromatography (hexanes/ethyl acetate from 0% to 20% of EA). Colorless liquid (71.7mg, yield 85%).  $^1\text{H}$  NMR (400 MHz, Chloroform-d)  $\delta$  6.10 (d,  $J = 13.7$  Hz, 1H), 5.93 (d,  $J = 13.7$  Hz, 1H), 2.02 (t,  $J = 3.1$  Hz, 3H), 1.79 – 1.60 (m, 10H).  $^{13}\text{C}$  NMR (101 MHz, Chloroform-d)  $\delta$  148.73, 102.44, 41.61, 37.71, 36.62, 28.20. HRMS (APPI/LTQ-Orbitrap)  $m/z$ :  $[\text{M}]^+$  Calcd for  $\text{C}_{12}\text{H}_{17}^+$  161.1330; Found 161.1322.

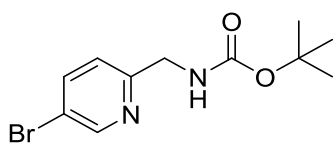
### Synthesis of 1-((Z)-2-azidovinyl)adamantine (**7**)



Molecule **3b** (100 mg, 0.35 mmol),  $\text{NaN}_3$  (34 mg, 0.53 mmol) and  $\text{CuSO}_4$  (33.5 mg, 0.21 mmol) were dissolved in 3 mL of MeOH and stirred at room temperature for one night. The mixture is then concentrate under reduced pressure, dissolved in DCM and washed with water and brine. The residue is dried with anhydrous sodium sulfate, concentrated and purified by flash chromatography

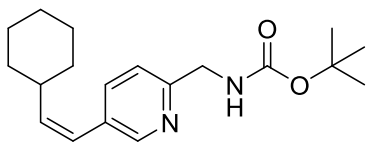
to obtain a white solid (61 mg, 85.8%).  $^1\text{H}$  NMR (400 MHz, Chloroform- $d$ )  $\delta$  5.97 (d,  $J$  = 8.5 Hz, 1H), 4.56 (d,  $J$  = 8.5 Hz, 1H), 2.05 – 1.90 (m, 3H), 1.79 – 1.57 (m, 10H), 1.45 – 1.17 (m, 3H).  $^{13}\text{C}$  NMR (101 MHz, Chloroform- $d$ )  $\delta$  130.22, 123.05, 42.14, 36.77, 34.88, 28.60. HRMS (APPI/LTQ-Orbitrap)  $m/z$ :  $[\text{M}]^+$  Calcd for  $\text{C}_{12}\text{H}_{17}\text{N}^+$  175.1356; Found 175.1355.

### Synthesis of tert-butyl ((5-bromopyridin-2-yl)methyl)carbamate (**13**)



5-bromopicolinonitrile (5 g, 27 mmol),  $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$  (10 mol%),  $\text{BocO}_2$  (2 equiv.) and methanol (100 mL) were put in a flask which was then cooled to  $0^\circ\text{C}$ . Sodium boron hydride (7 equiv.) was then added portion-wise over 2h. After one additional hour of stirring at  $15^\circ\text{C}$ , the mixture was poured into ice-water and extracted with ethyl acetate. The combined organic layers were washed with brine, dried with anhydrous sodium sulfate and concentrated under reduced pressure. The residue was purified by column chromatography giving the desired product as a white solid (4.46g, yield:57.2%).  $^1\text{H}$  NMR (400 MHz, Methanol- $d_4$ )  $\delta$  8.57 (d,  $J$  = 2.4 Hz, 1H), 7.97 (dd,  $J$  = 8.4, 2.4 Hz, 1H), 7.32 (d,  $J$  = 8.4 Hz, 1H), 4.32 (s, 2H), 3.33 (dt,  $J$  = 3.3, 1.7 Hz, 1H), 1.48 (s, 9H).  $^{13}\text{C}$  NMR (101 MHz, Methanol- $d_4$ )  $\delta$  157.82, 157.07, 149.40, 139.66, 122.47, 118.55, 79.15, 44.72, 27.34. Melting point:  $92\text{-}94^\circ\text{C}$ .

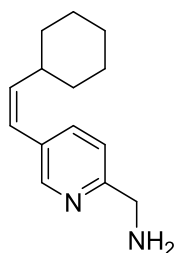
### Synthesis of tert-butyl (Z)-((5-(2-cyclohexylvinyl)pyridin-2-yl)methyl)carbamate (**14**)



To a solution of **13** (30.44 mg, 0.106 mmol) and potassium carbonate (49.2 mg, 0.356 mmol) in degassed 1,2-dimethoxyethane/ethanol/water (0.8 mL, 1/0.5/1 ratio) was added molecule **3a** (50 mg, 0.212 mmol) and tetrakis(triphenylphosphine)palladium(0) (34.78 mg, 0.142 mmol) at room

temperature. Then the mixture was heated to 95°C for 1 hour. After quenching with water, the mixture was extracted with ethyl acetate and concentrated, obtaining the crude product, which was purified by chromatography (hexanes/ethyl acetate 4:1) to afford a pure compound **14** as a colorless liquid (48.1 mg, yield: 72%). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz): δ 8.45 (1H, s), 7.53 (1H, d), 7.24 (1H, d), 6.24 (1H, d), 5.62 (1H, t), 4.44 (2H, d), 2.48 (1H, m), 1.69 (5H, m), 1.46 (9H, s), 1.21 (6H, m), 5.53-5.61 (1H,bs). <sup>13</sup>C NMR (101 MHz, Chloroform-d) δ 156.00, 155.21, 148.88, 141.15, 136.42, 132.16, 122.94, 121.13, 79.48, 45.58, 37.09, 33.10, 29.70, 28.42, 25.91, 25.56. HRMS (APPI/LTQ-Orbitrap) m/z: [M + H]<sup>+</sup> Calcd for C<sub>19</sub>H<sub>29</sub>N<sub>2</sub>O<sub>2</sub><sup>+</sup> 317.2224; Found 317.2222.

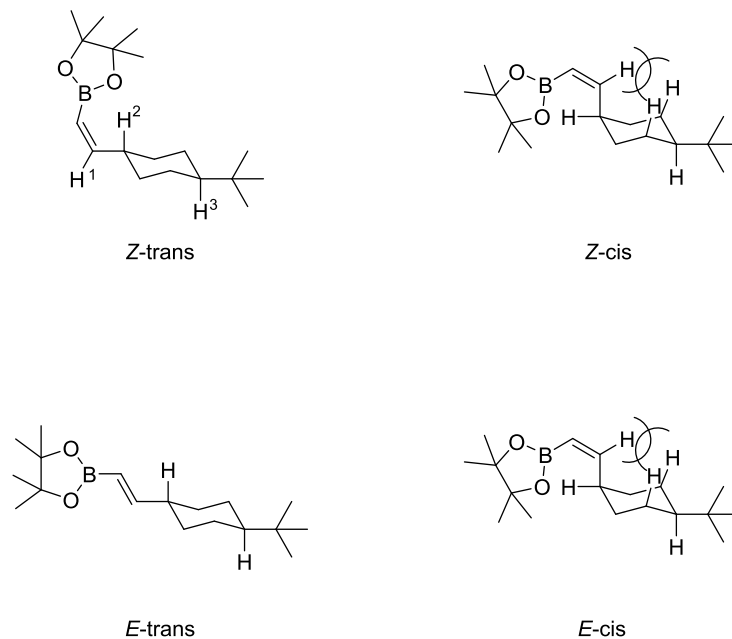
### Synthesis of (Z)-(5-(2-cyclohexylvinyl)pyridin-2-yl)methanamine (**9**)



A solution of **14** (150 mg, 0.5 mmol) in DCM (5 mL) and trifluoroacetic acid (4 mL) was stirred for 45 minutes at room temperature, after which the solvent was evaporated. Ethyl ether (10 mL) was added, and the solvent was evaporated again. The residue was dissolved in DCM (20 mL), washed with saturated aqueous of sodium bicarbonate, dried with anhydrous sodium sulfate, and the solvent was evaporated to give product **9** in a quantitative yield as a green liquid (100 mg, quantitative yield). <sup>1</sup>H NMR (400 MHz, Chloroform-d) δ 8.47 (s, 1H), 7.53 (dd, J = 8.0, 2.1 Hz, 1H), 7.31 – 7.22 (m, 1H), 6.25 (d, J = 11.7 Hz, 1H), 5.61 (dd, J = 11.7, 10.2 Hz, 1H), 4.00 (s, 2H), 2.49 (tdd, J = 10.7, 7.3, 3.2 Hz, 1H), 2.02 (d, J = 21.9 Hz, 2H), 1.78 – 1.63 (m, 5H), 1.36 – 1.11 (m, 5H). <sup>13</sup>C NMR (101 MHz, Chloroform-d) δ 159.39, 149.00, 140.97, 136.39, 131.86, 123.05, 120.80, 37.09, 33.11, 29.70, 25.91, 25.57. HRMS (ESI/QTOF) m/z: [M + H]<sup>+</sup> Calcd for C<sub>14</sub>H<sub>21</sub>N<sub>2</sub><sup>+</sup> 217.1699; Found 217.1701.

## Analysis of the cis/trans ratio of compound **3h**

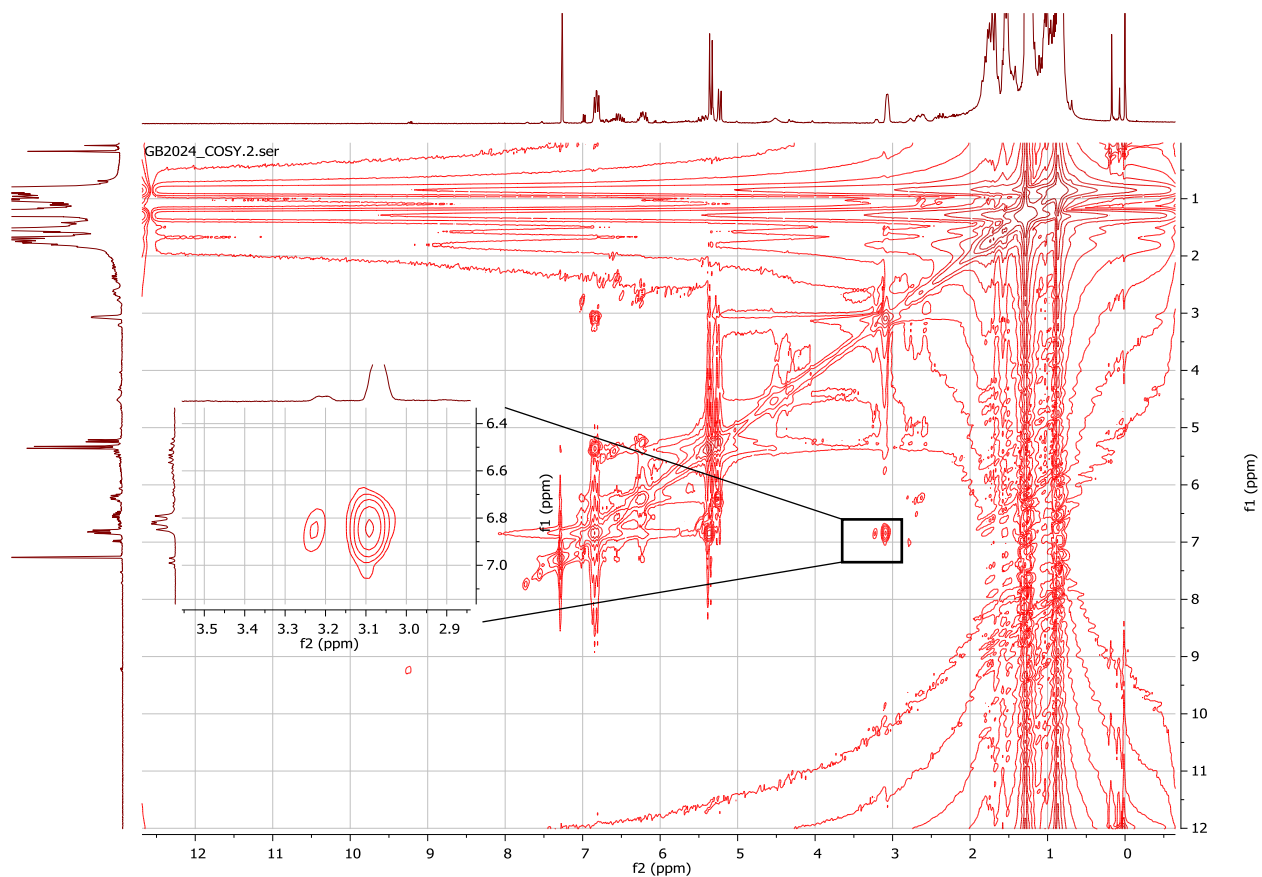
Compound **3h** was obtained as a mixture of different isomers. (Figure S1).



**Figure S1.** Four different possible isomers of compound **3h**

From <sup>1</sup>H NMR analysis it was possible to know the *E/Z* ratio of the double bond. A COSY-NMR analysis was used to determine the ratio of cis- and trans-isomer of **Z-3h** (Figure S2)





**Figure S2.** COSY spectrum of molecule **3h**

At 3.07 ppm (Figure S2) is the signal of the allylic hydrogen (H2, Figure S1) that has a correlation with the signal of the  $sp^2$  hydrogen of the double bond (H1, Figure S1). Another peak at 3.20 ppm also correlates with the same  $sp^2$  hydrogen. The ratio of the two allylic hydrogens is 14:1. Thus, the reaction has a diastereoselectivity of 14:1.

According to a previous report, the *cis* or *trans* isomer might be assigned by comparing the difference of the  $^1\text{H}$  NMR shift of H2 and H3 (Figure S1).<sup>2</sup> Unfortunately, in the present case, the H3 signal overlaps with the other signals of the cyclohexyl moiety and cannot be identified. Nevertheless, previous reports of substituted cyclohexanes show that a hydrogen on a cyclohexyl ring is more shielded if it is in axial position than in an equatorial position.<sup>3,4</sup> Moreover, according to a previous report,<sup>2</sup> reaction of 4-substituted cyclohexyl radical should give rise to the more stable *trans* isomer. Therefore we assign the major product as the *trans*-isomer.

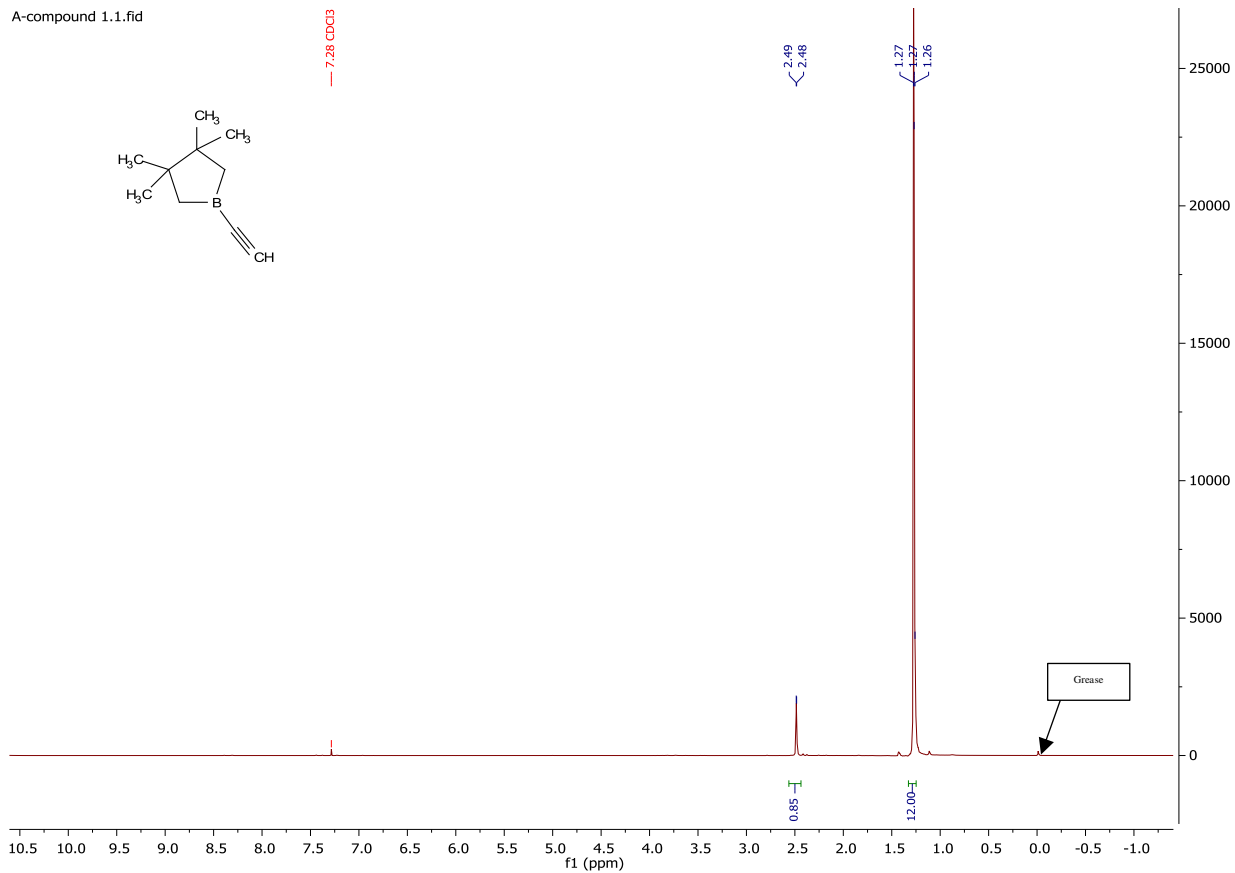
## References

- (1) Ohmura, T.; Yamamoto, Y.; Miyaura, N. Rhodium- or Iridium-Catalyzed Trans-Hydroboration of Terminal Alkynes, Giving (Z)-1-Alkenylboron Compounds. *J. Am. Chem. Soc.* **2000**, *122* (20), 4990–4991. <https://doi.org/10.1021/ja0002823>.
- (2) Perez Garcia, P. M.; Di Franco, T.; Orsino, A.; Ren, P.; Hu, X. Nickel-Catalyzed Diastereoselective Alkyl–Alkyl Kumada Coupling Reactions. *Org. Lett.* **2012**, *14* (16), 4286–4289. <https://doi.org/10.1021/ol302067b>.
- (3) Abraham, R. J.; Byrne, J. J.; Griffiths, L.; Perez, M. <sup>1</sup>H Chemical Shifts in NMR: Part 23, the Effect of Dimethyl Sulphoxide versus Chloroform Solvent on <sup>1</sup>H Chemical Shifts. *Magn. Reson. Chem.* **2006**, *44* (5), 491–509. <https://doi.org/10.1002/mrc.1747>.
- (4) Hanessian, S.; Dhanoa, D. S.; Beaulieu, P. L. Synthesis of Carbocycles from ω-Substituted α,β-Unsaturated Esters via Radical-Induced Cyclizations. *Can. J. Chem.* **1987**, *65* (8), 1859–1866. <https://doi.org/10.1139/v87-312>.

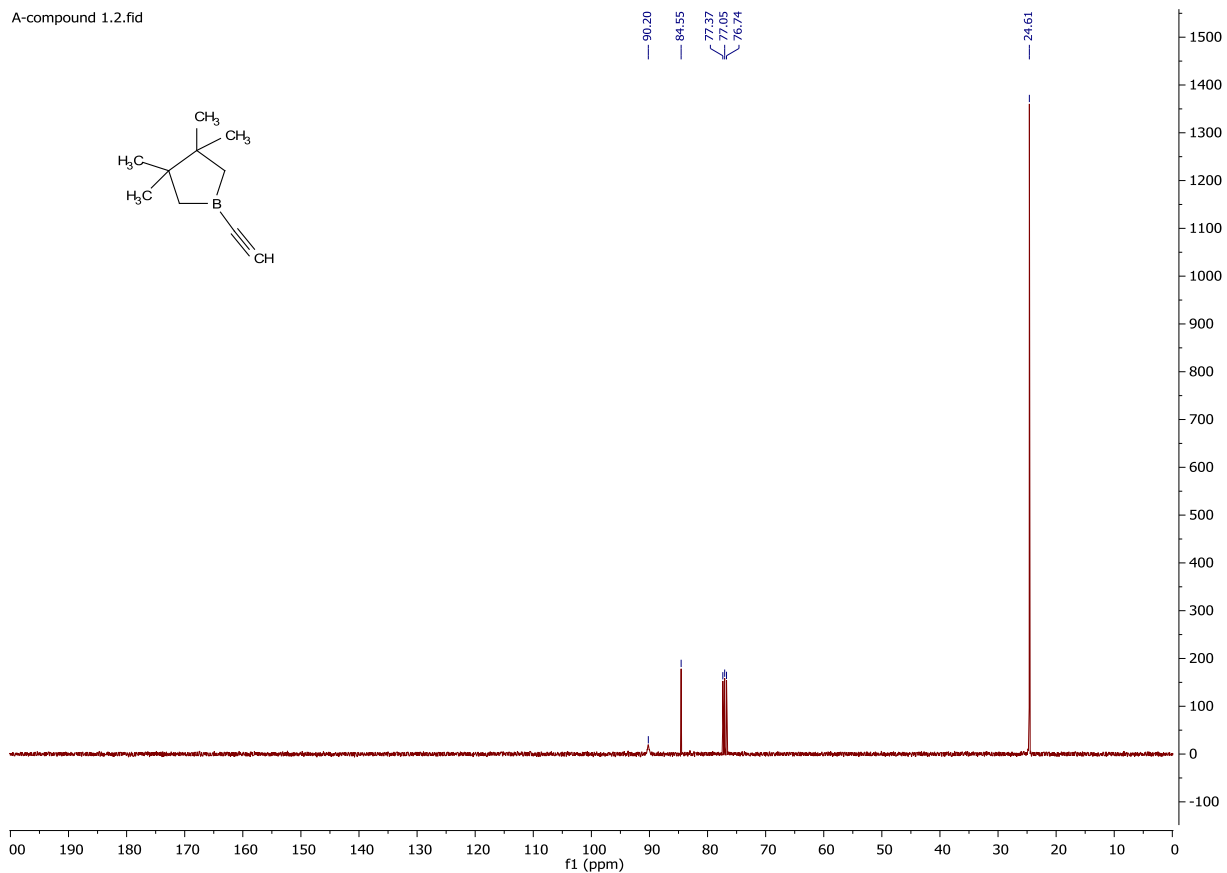
# NMR spectra

## Molecule 1

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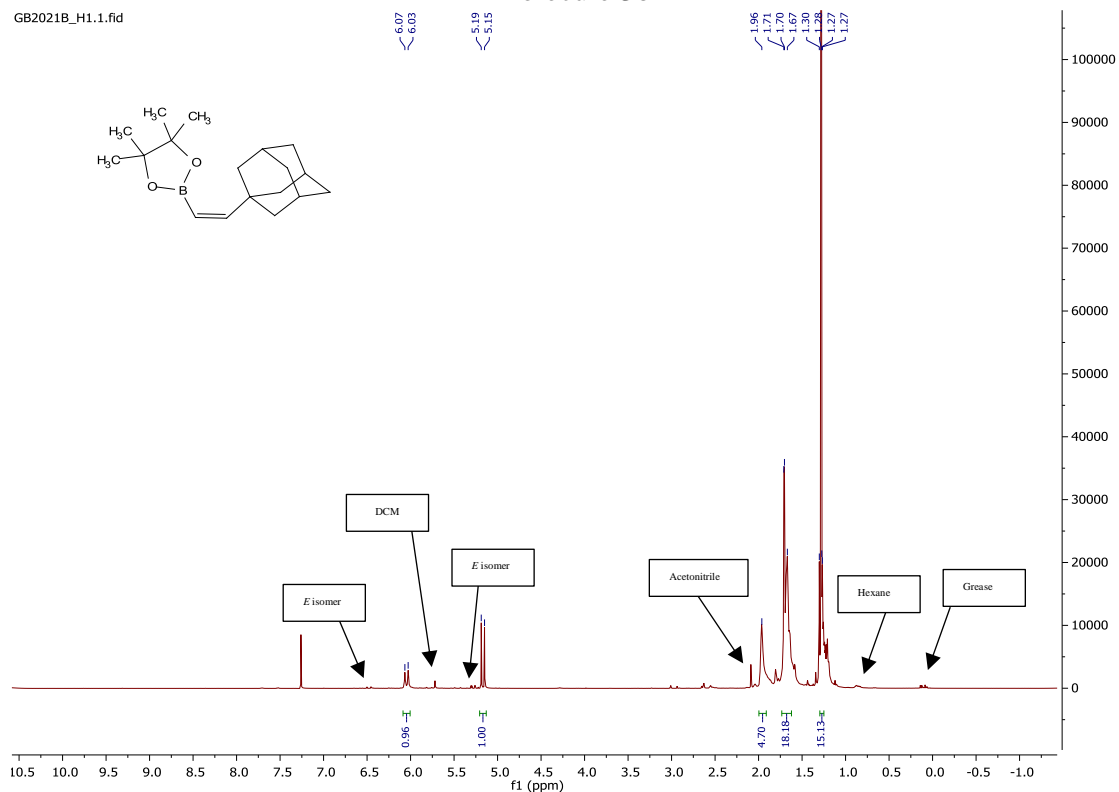


A-compound 1.2.fid

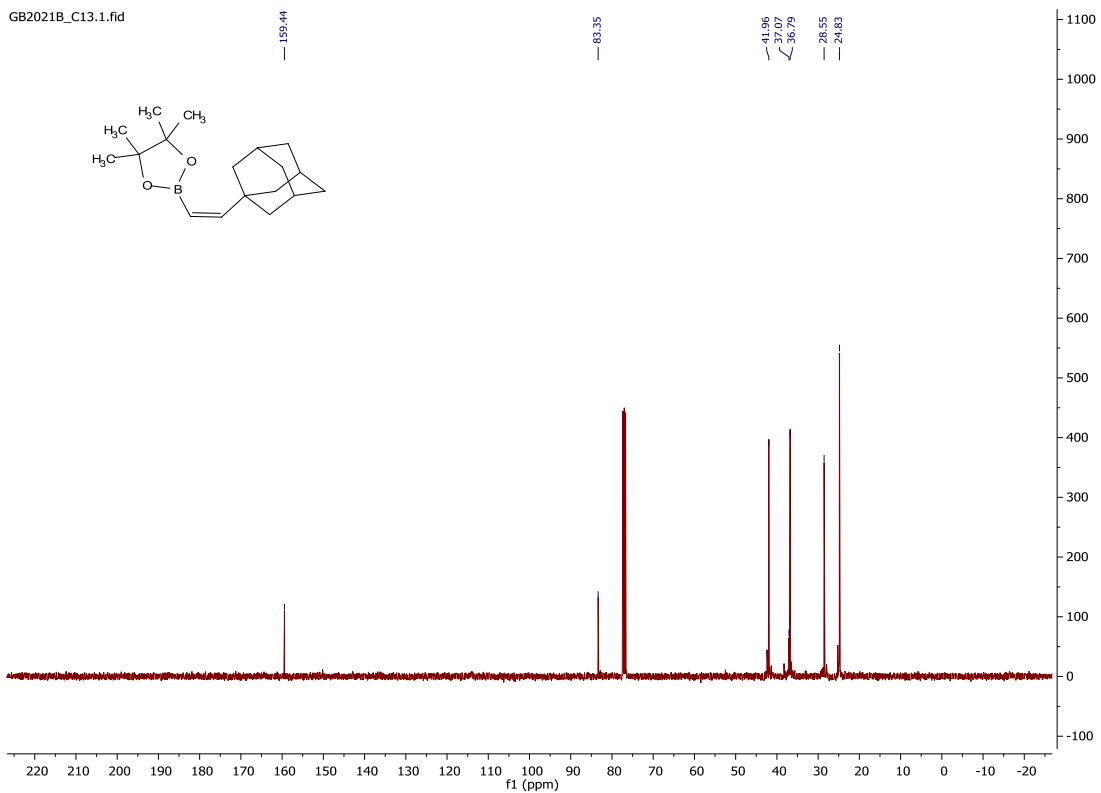


### Molecule 3b

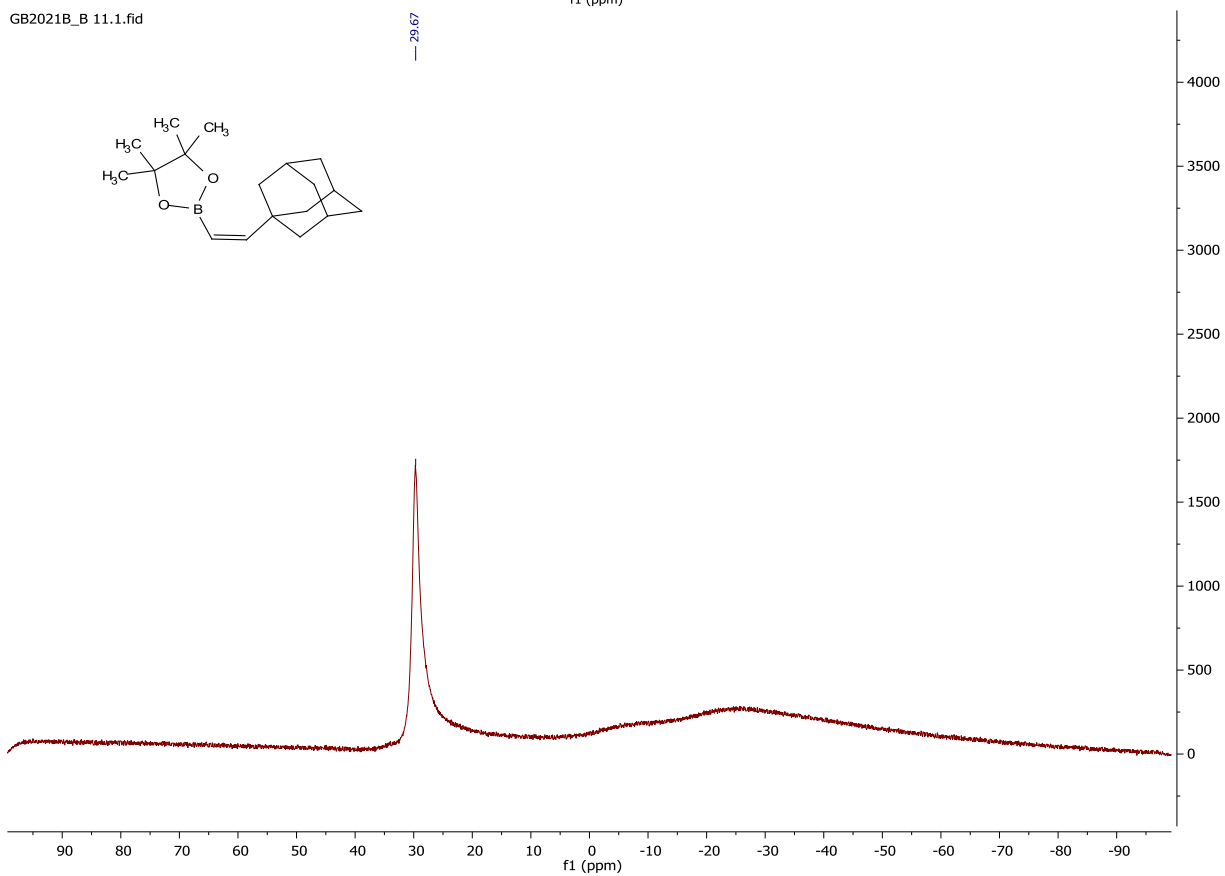
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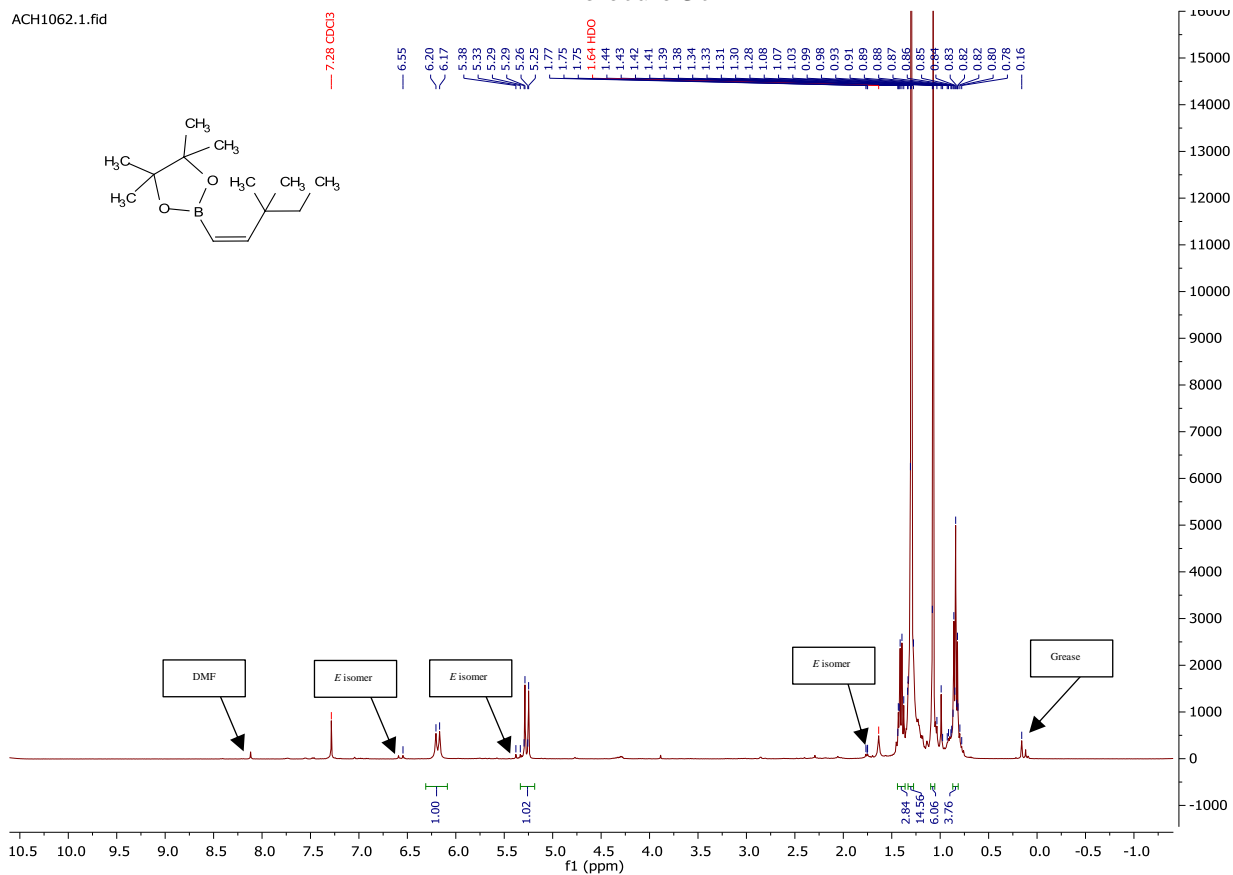


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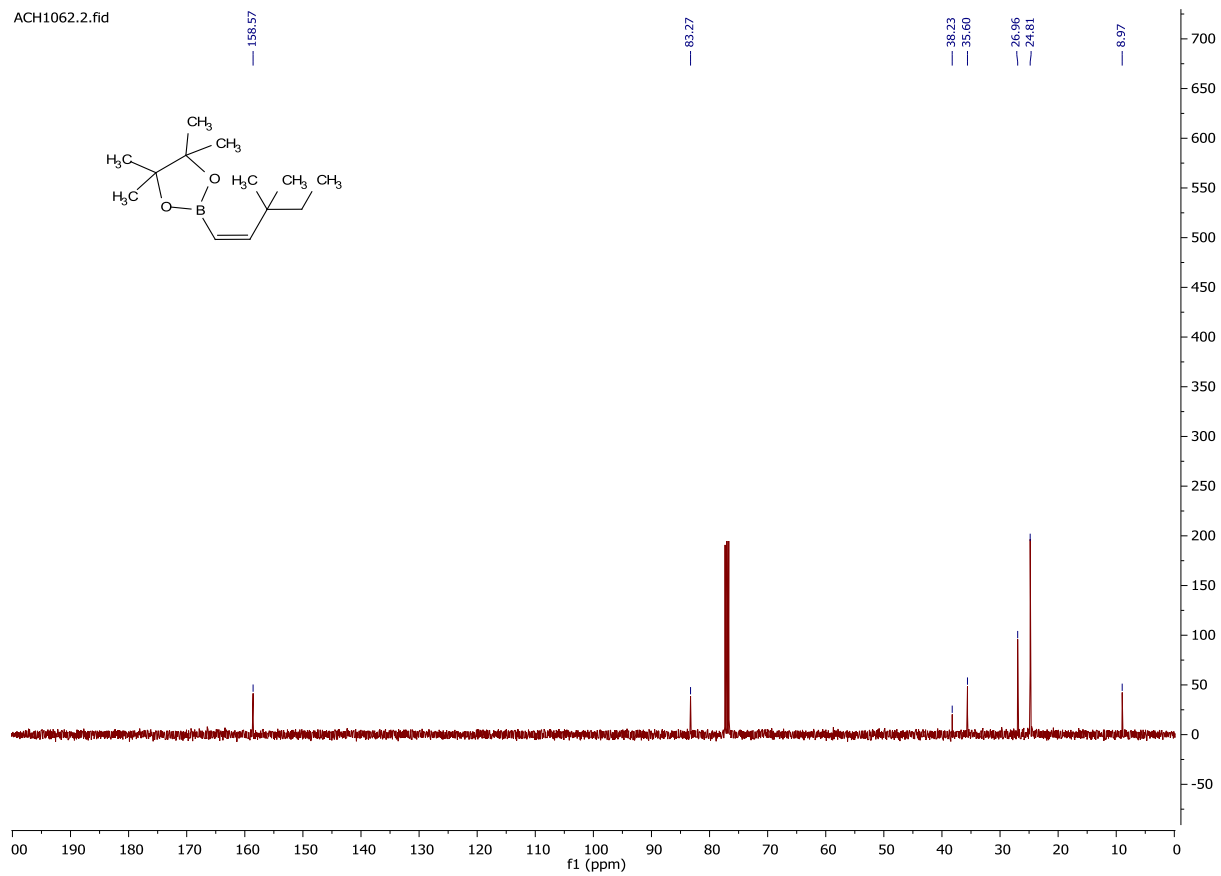


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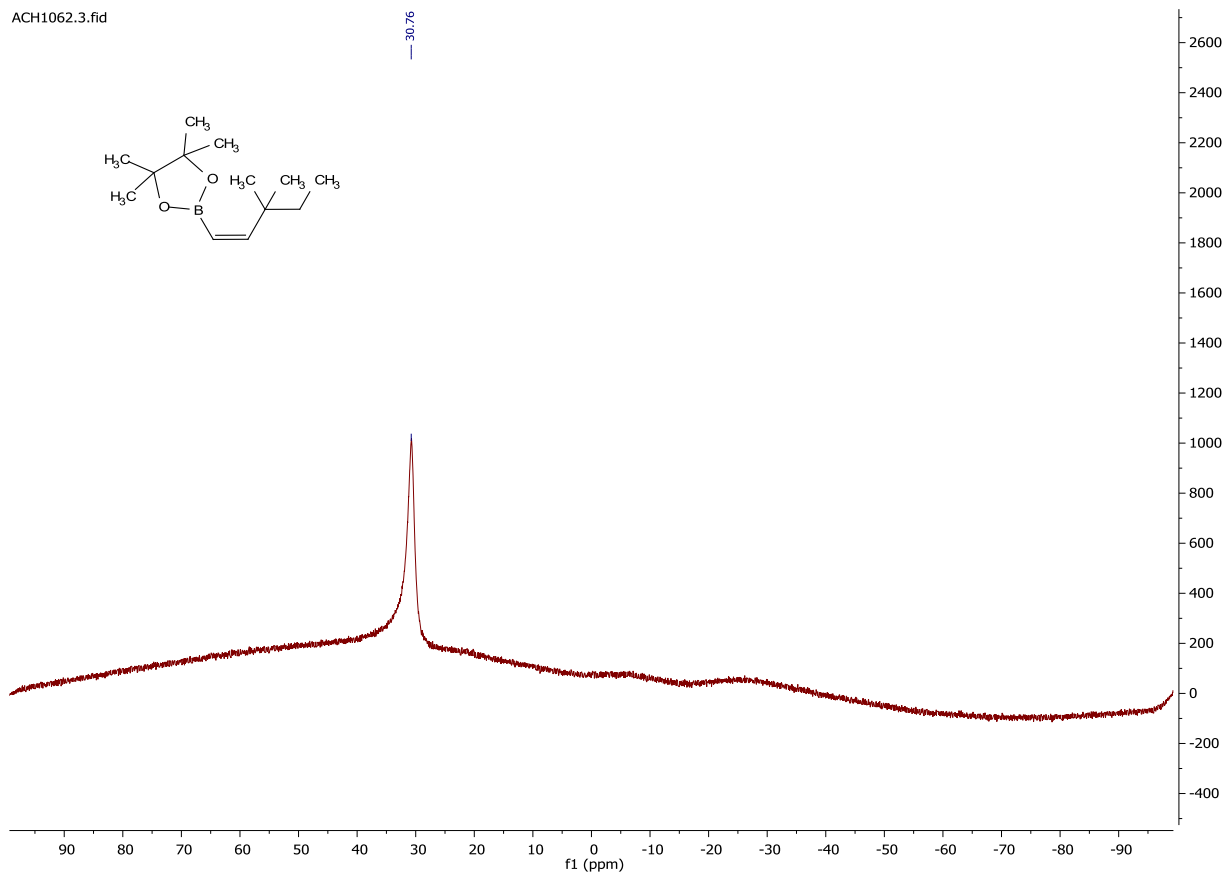
### Molecule 3c



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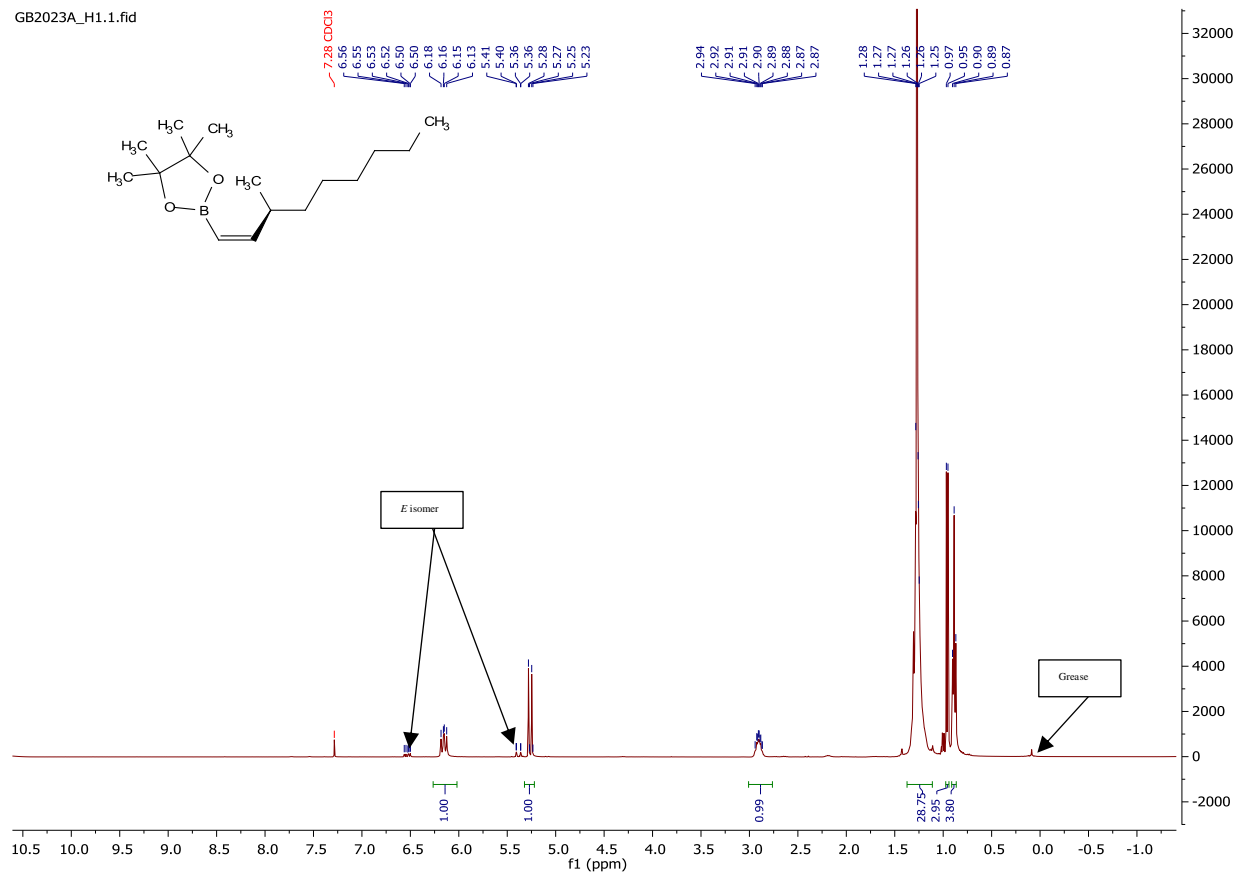
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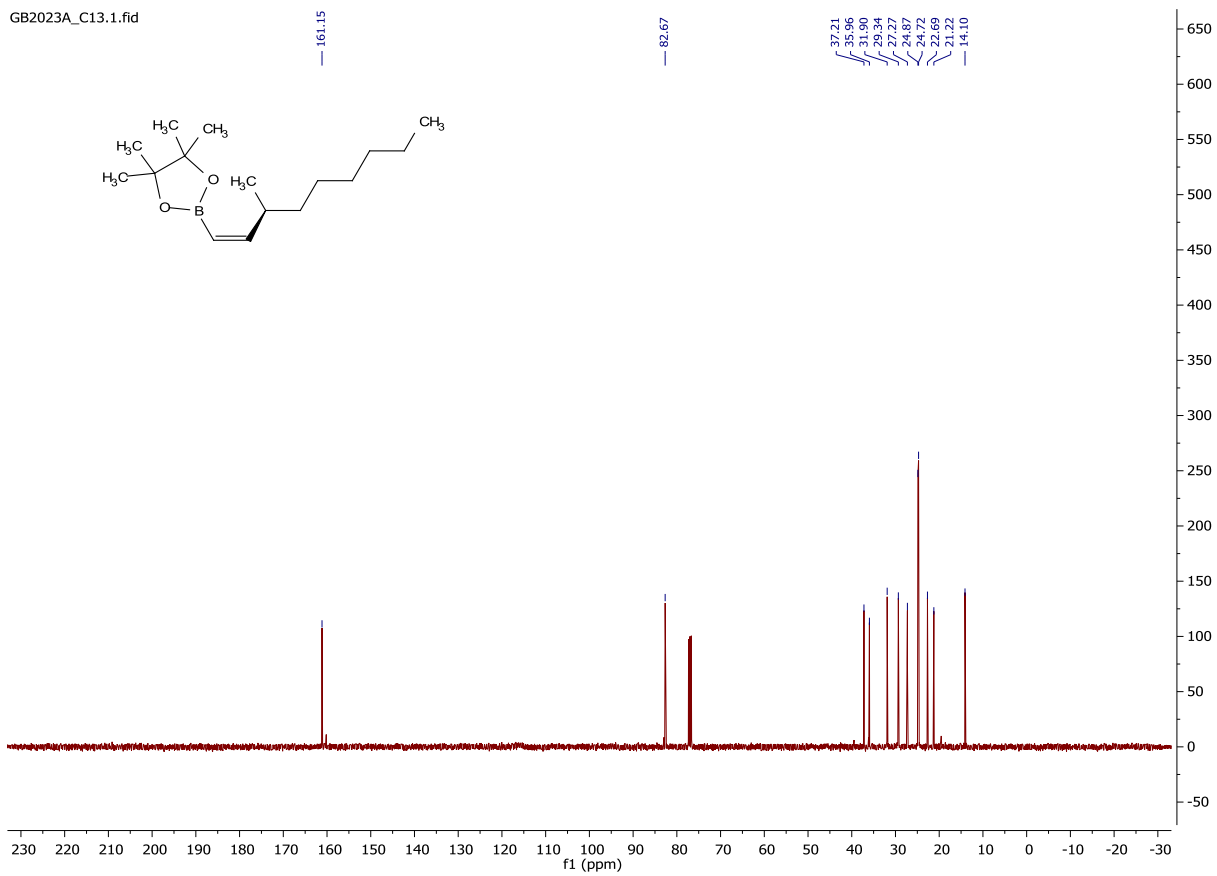


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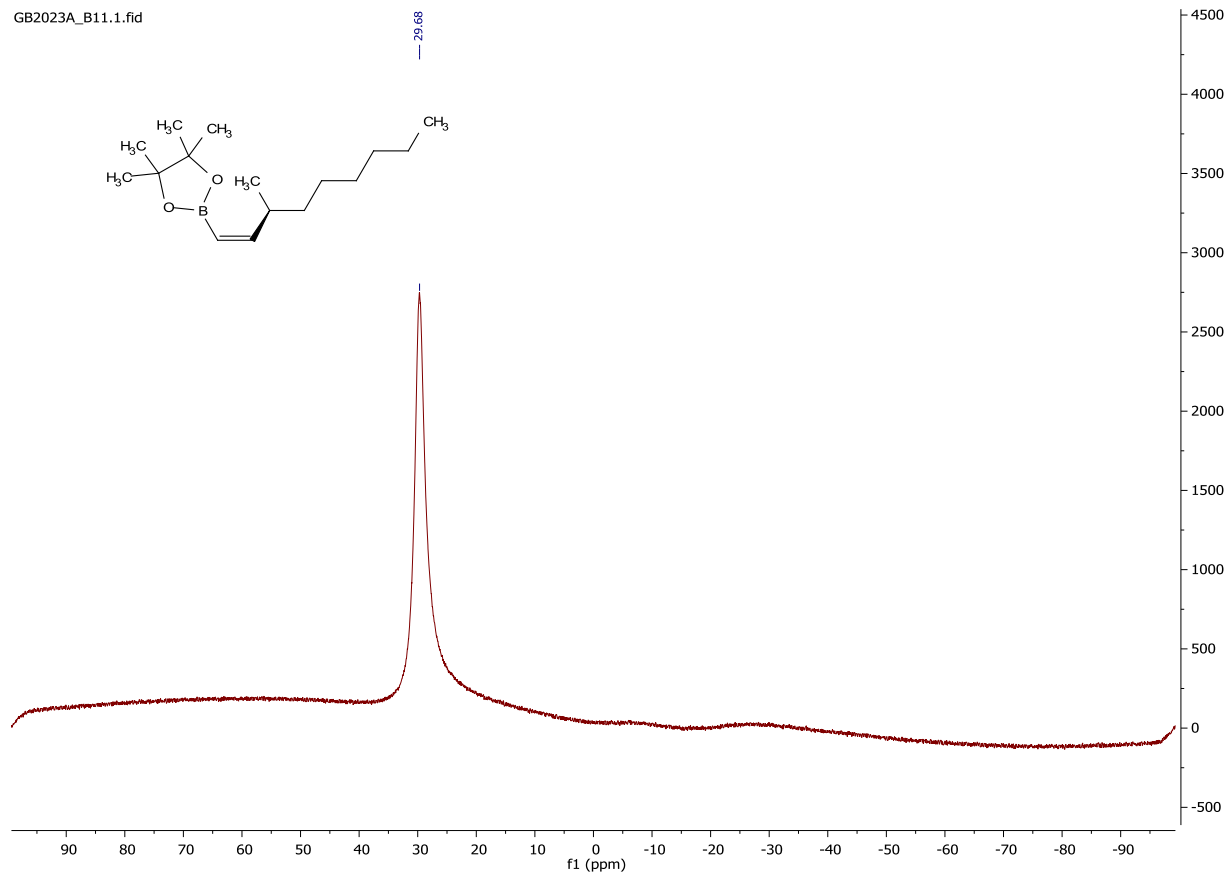
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GB2023A\_C13.1.fid



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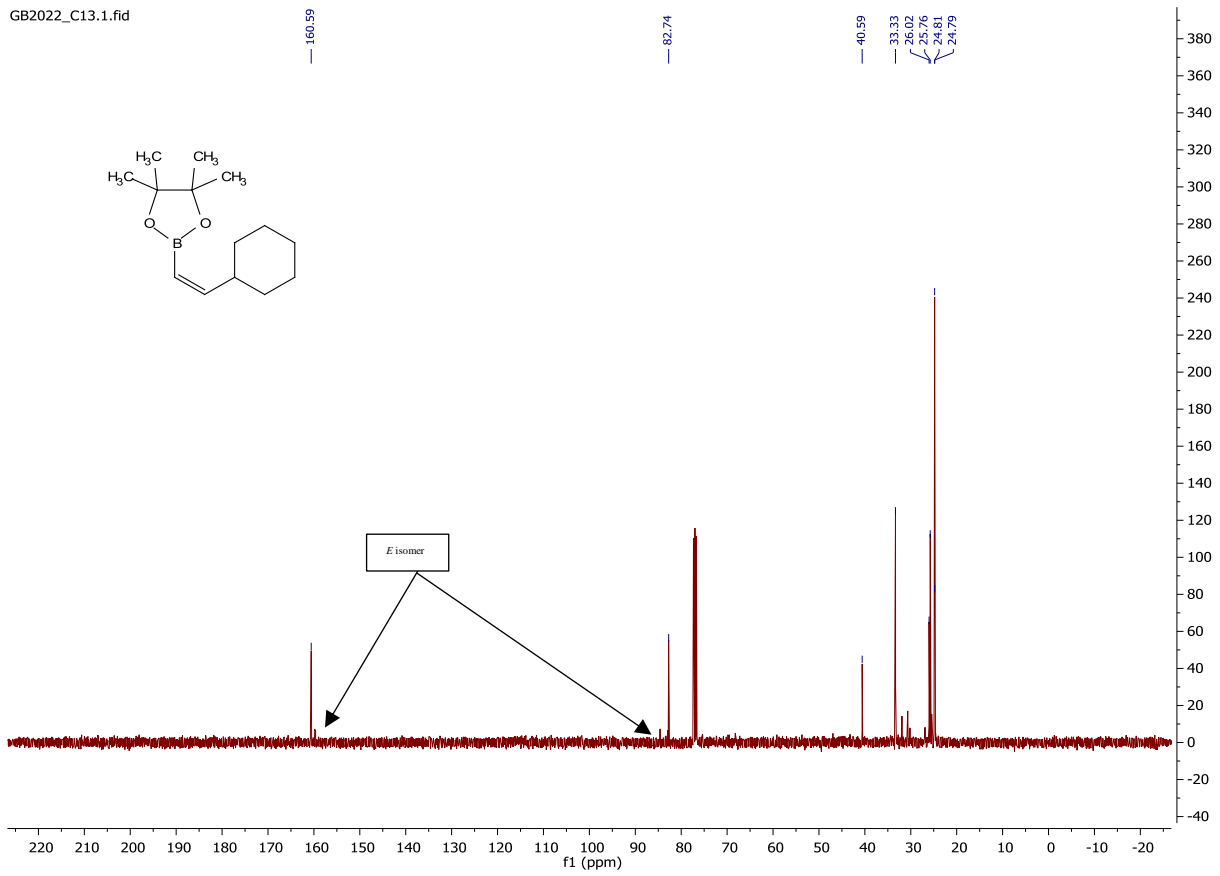


GB2022\_H1.1.fid

### Molecule 3e

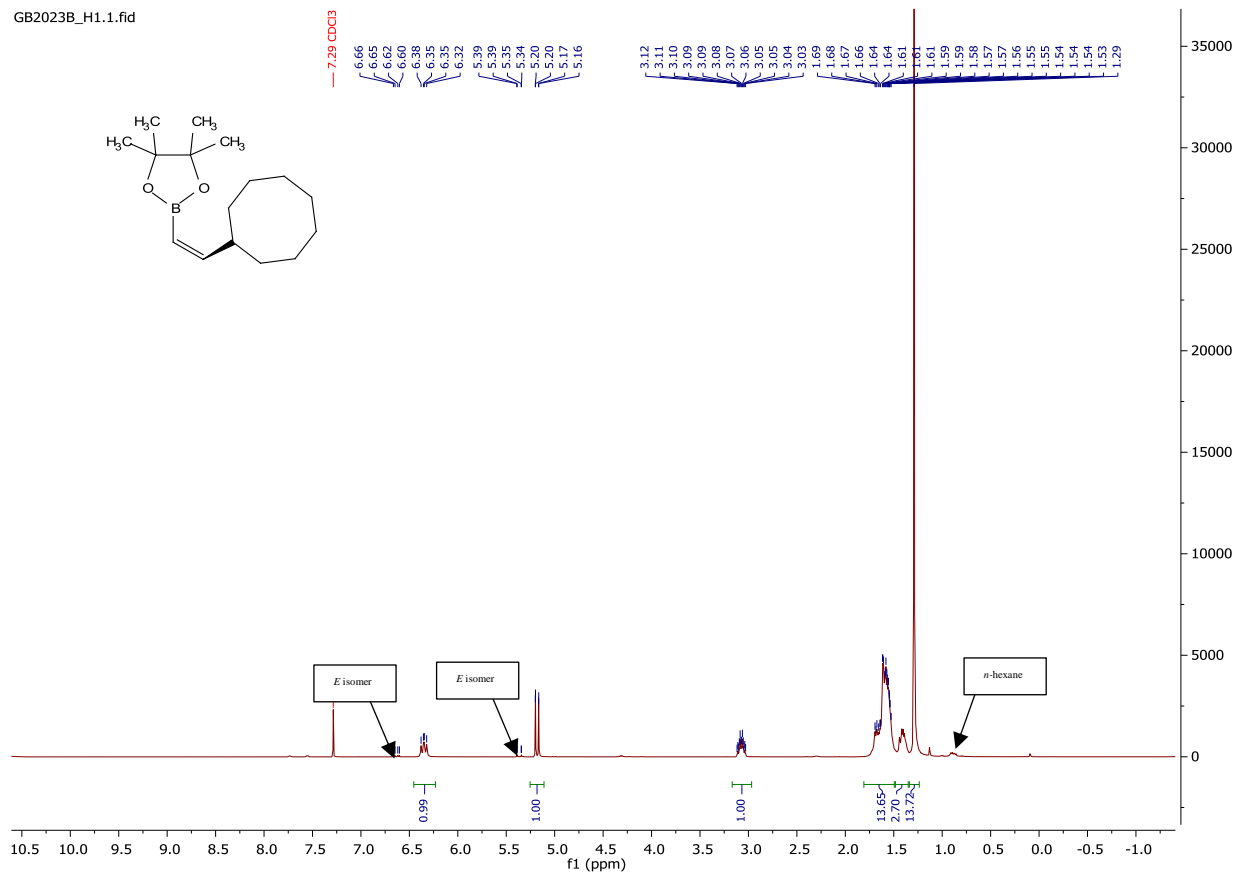


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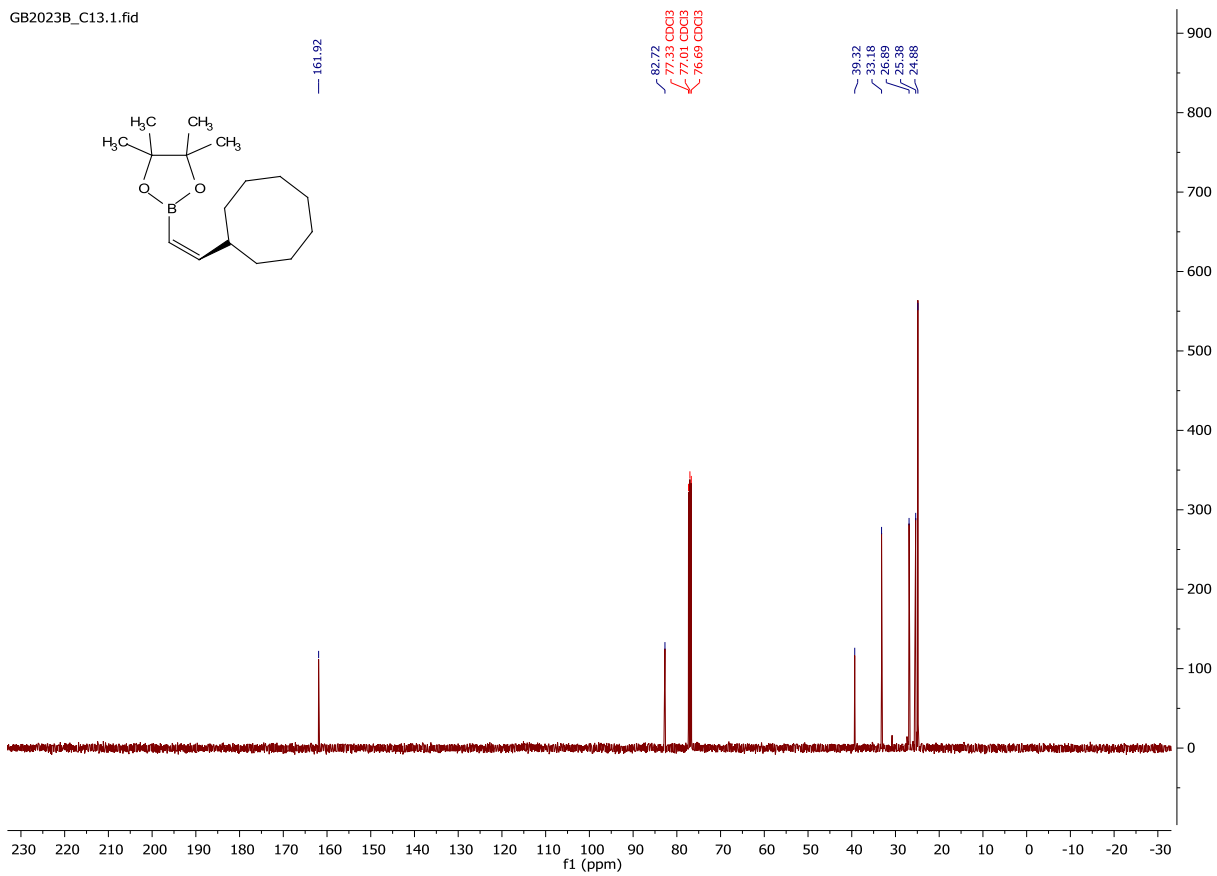


GB2023B\_H1.1.fid

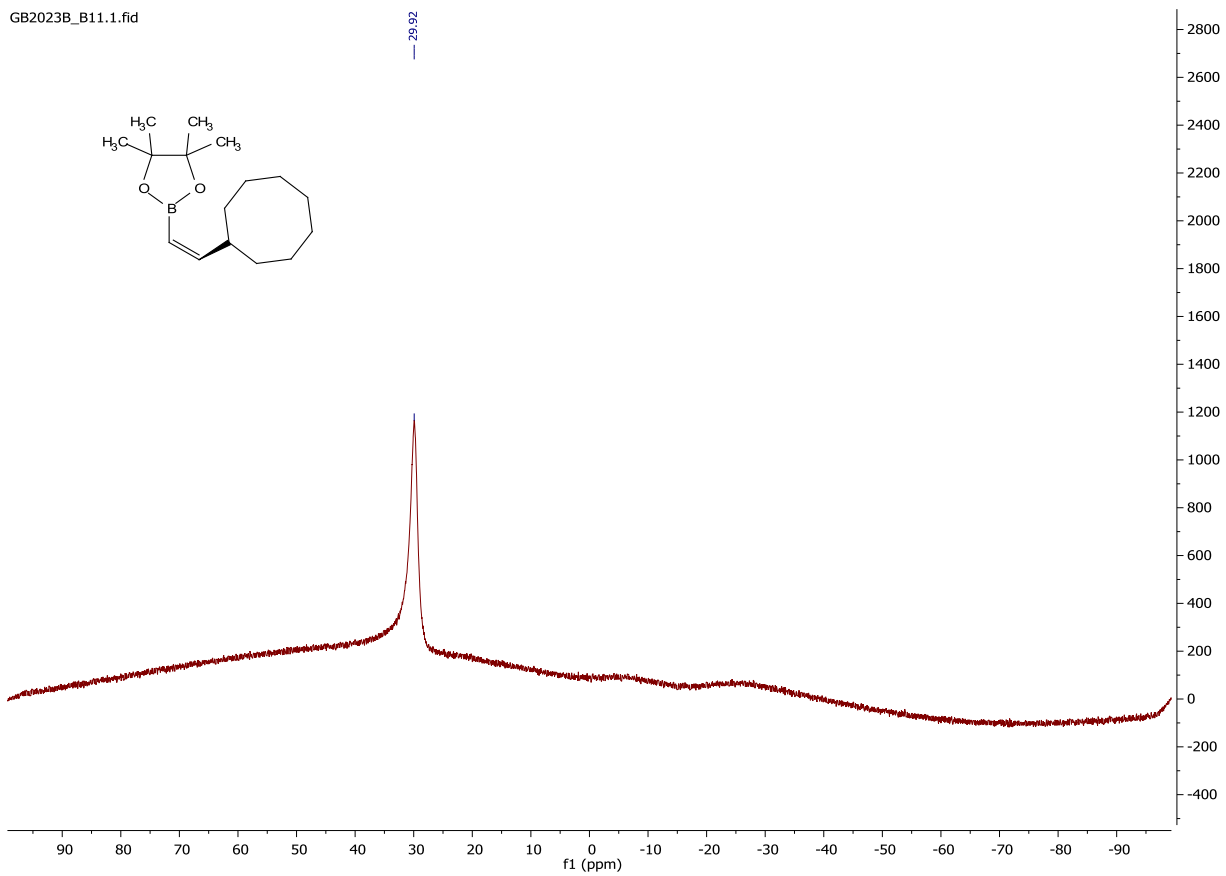
### Molecule 3f



GB2023B\_C13.1.fid



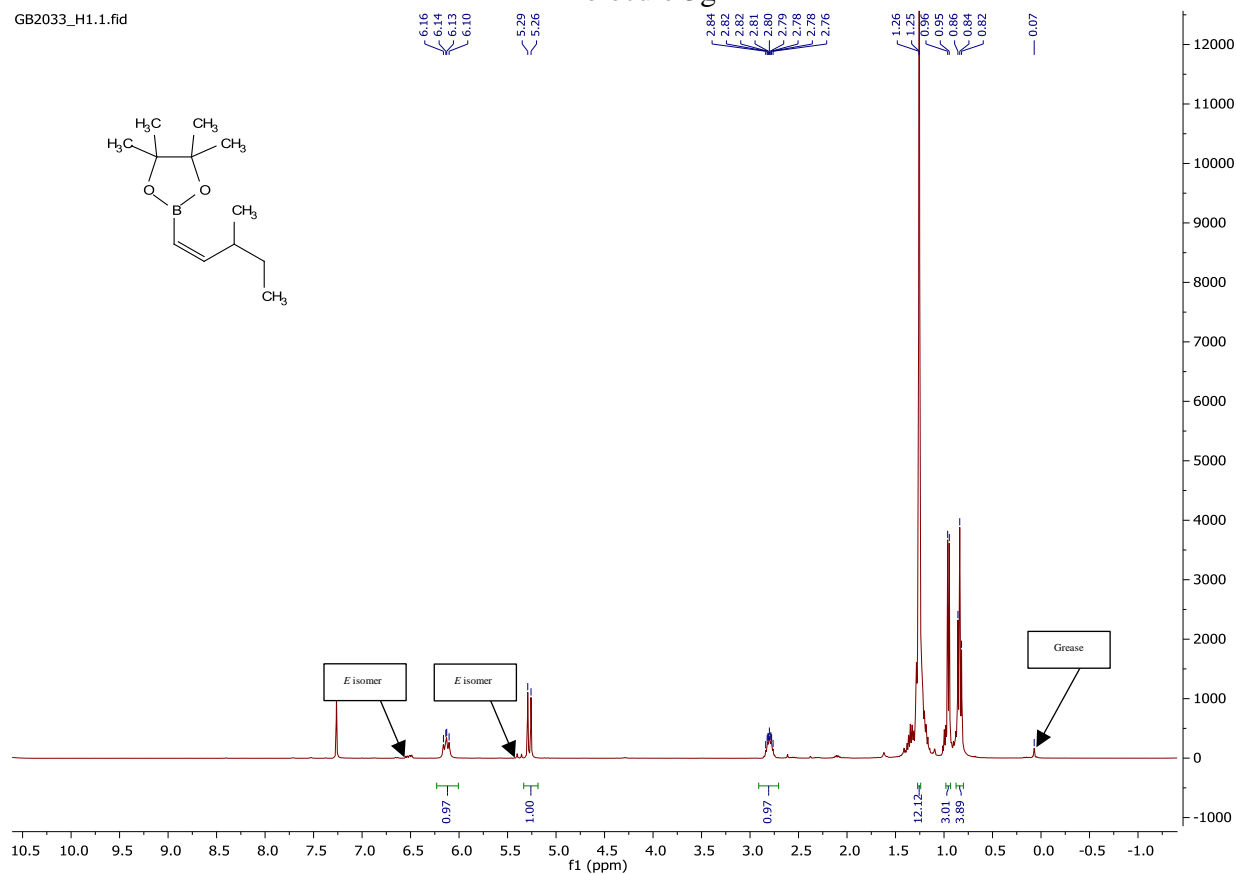
GB2023B\_B11.1.fid



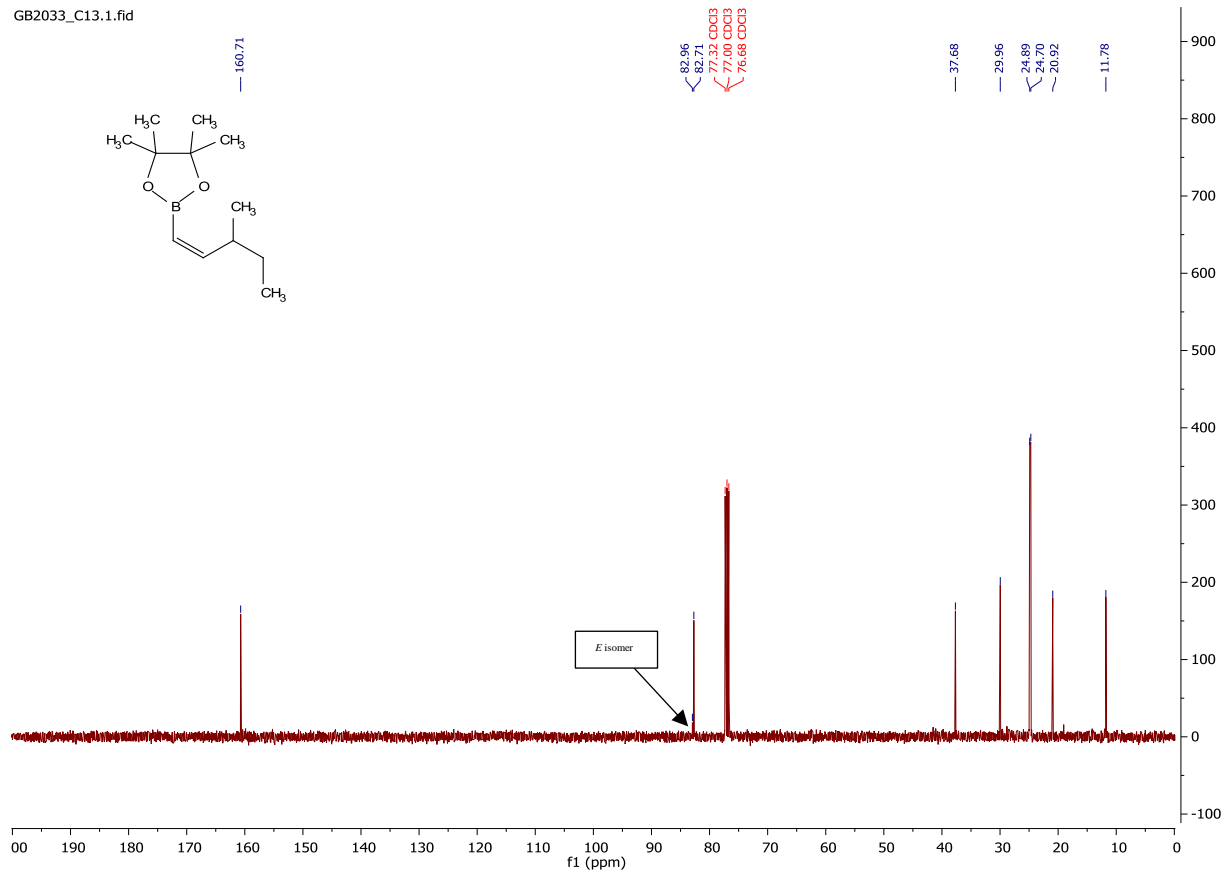


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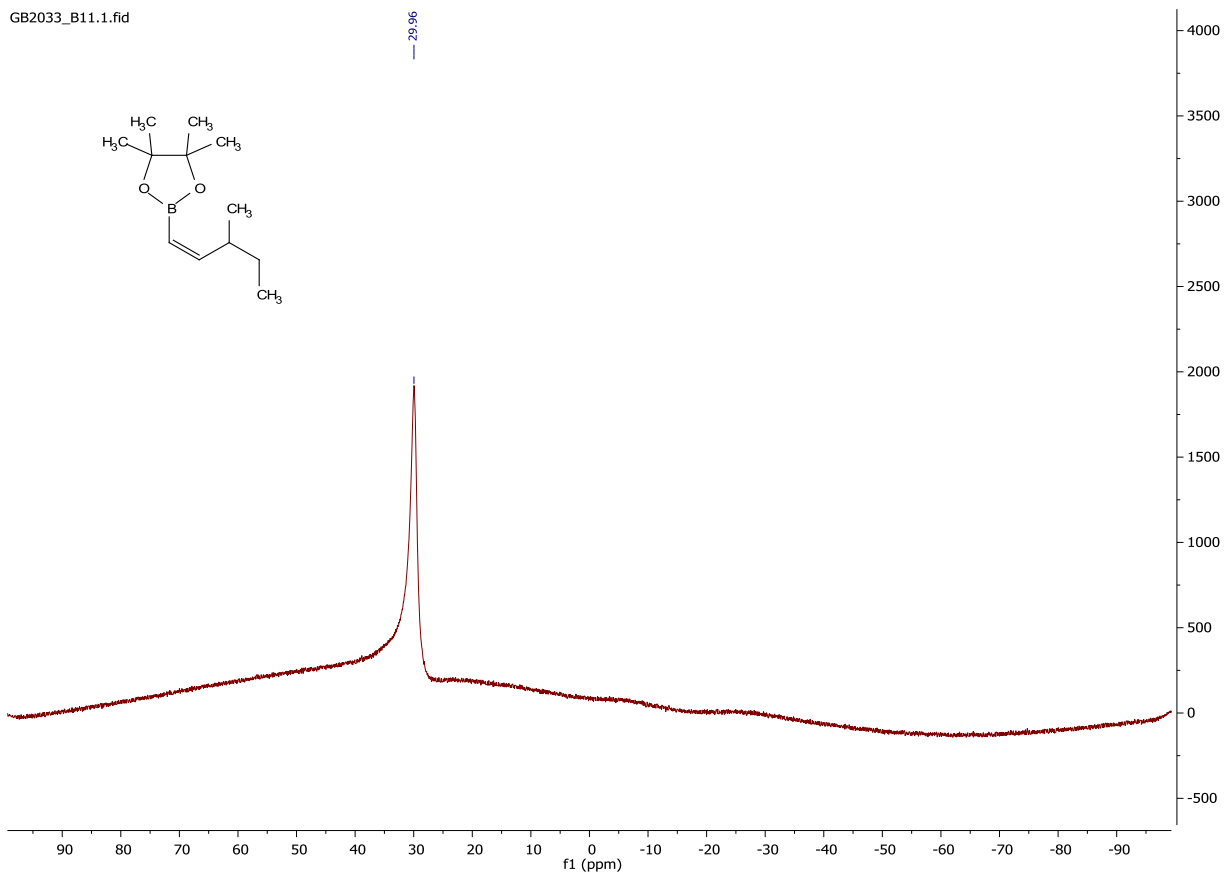
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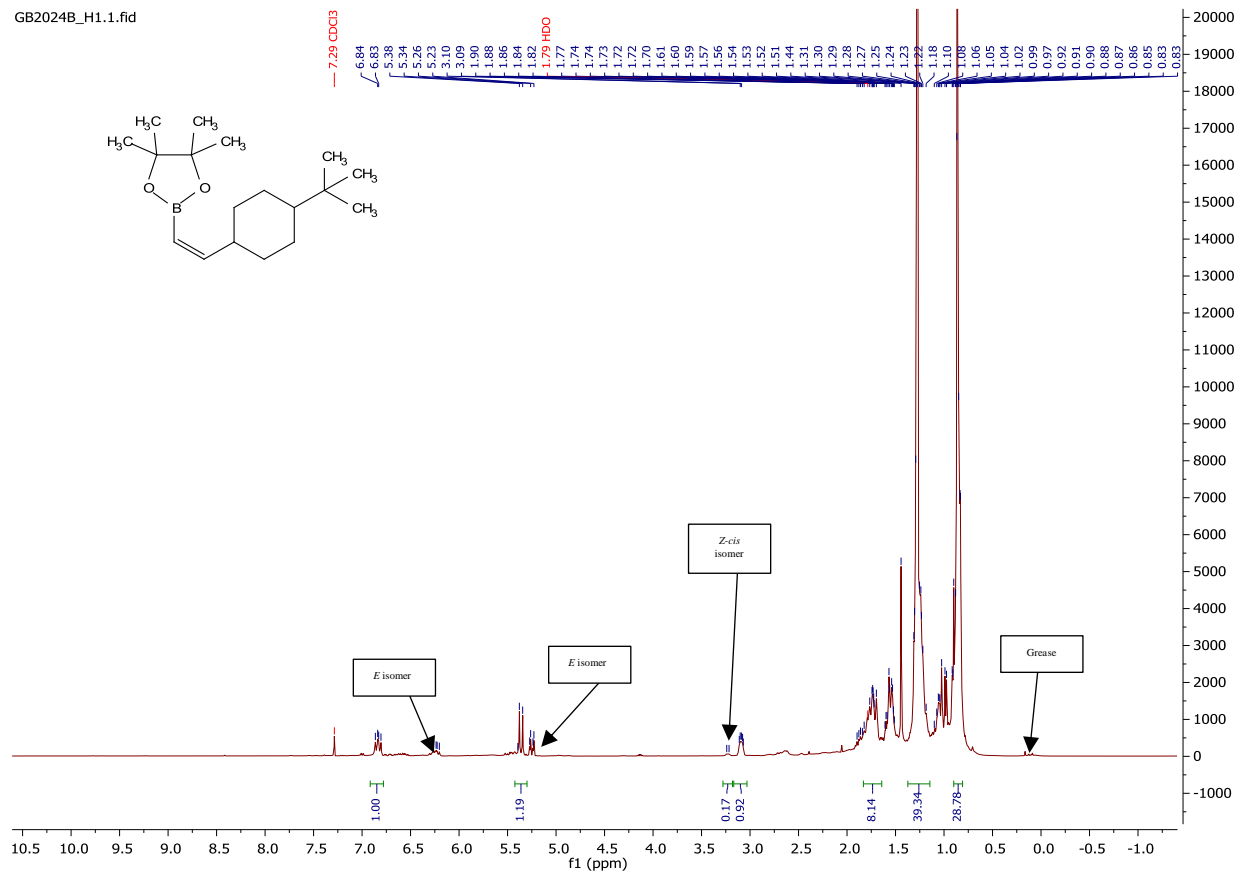


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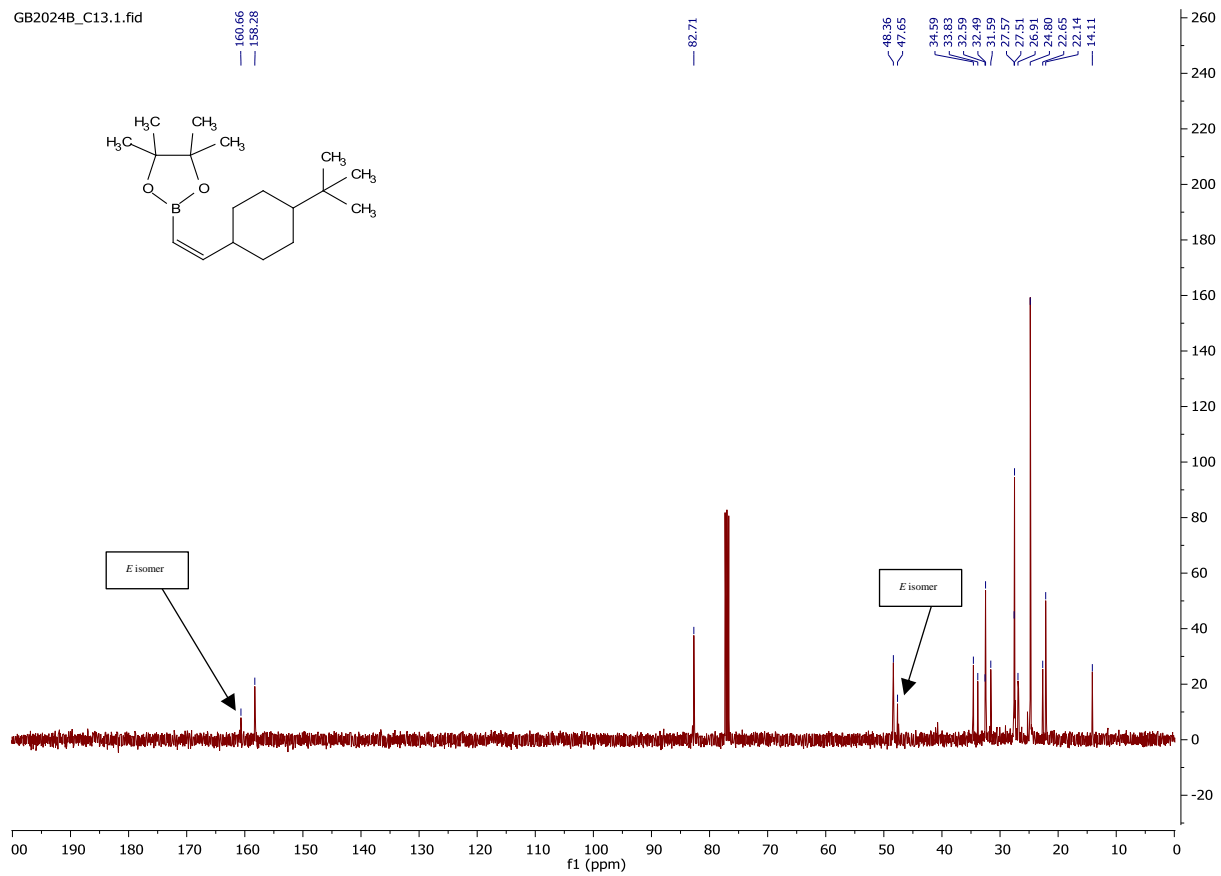


# Molecule 3h

GB2024B\_H1.1.fid

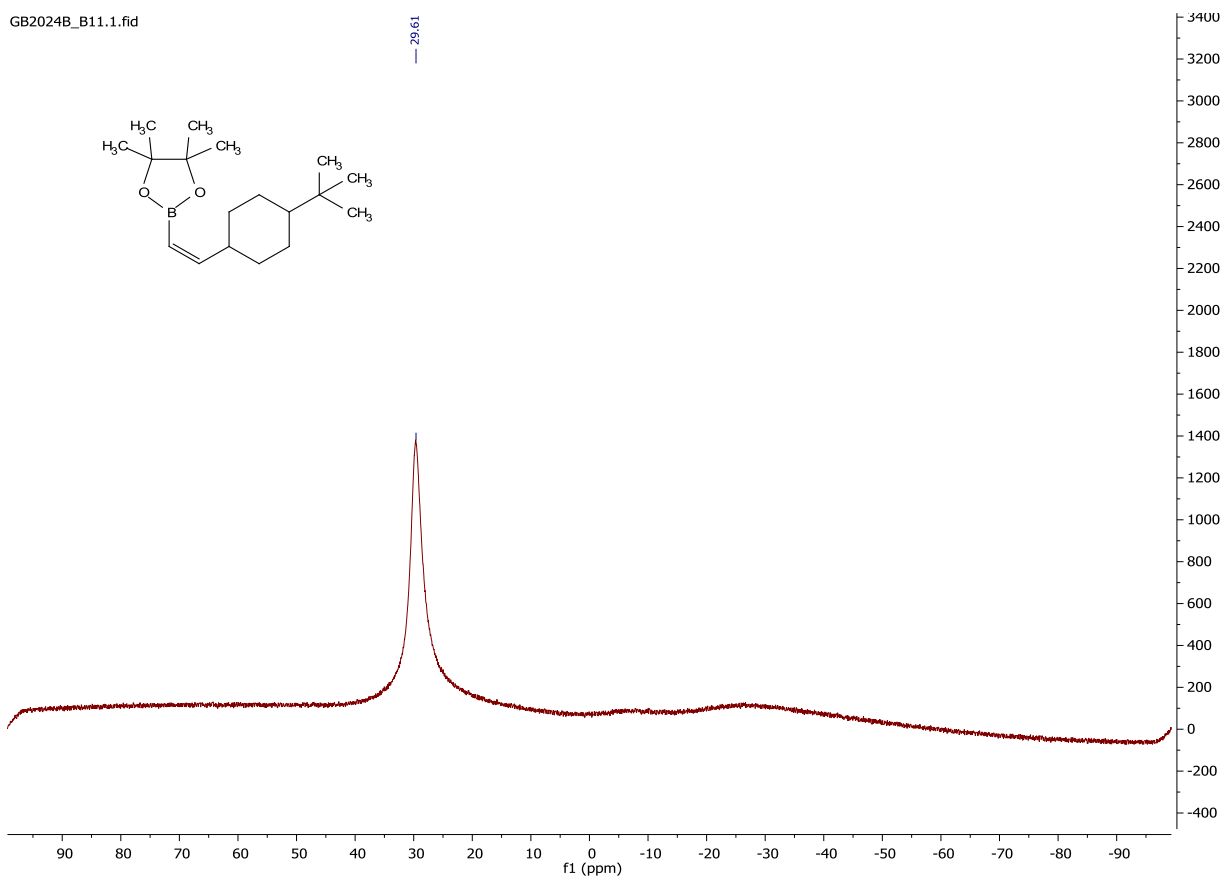


GB2024B\_C13.1.fid



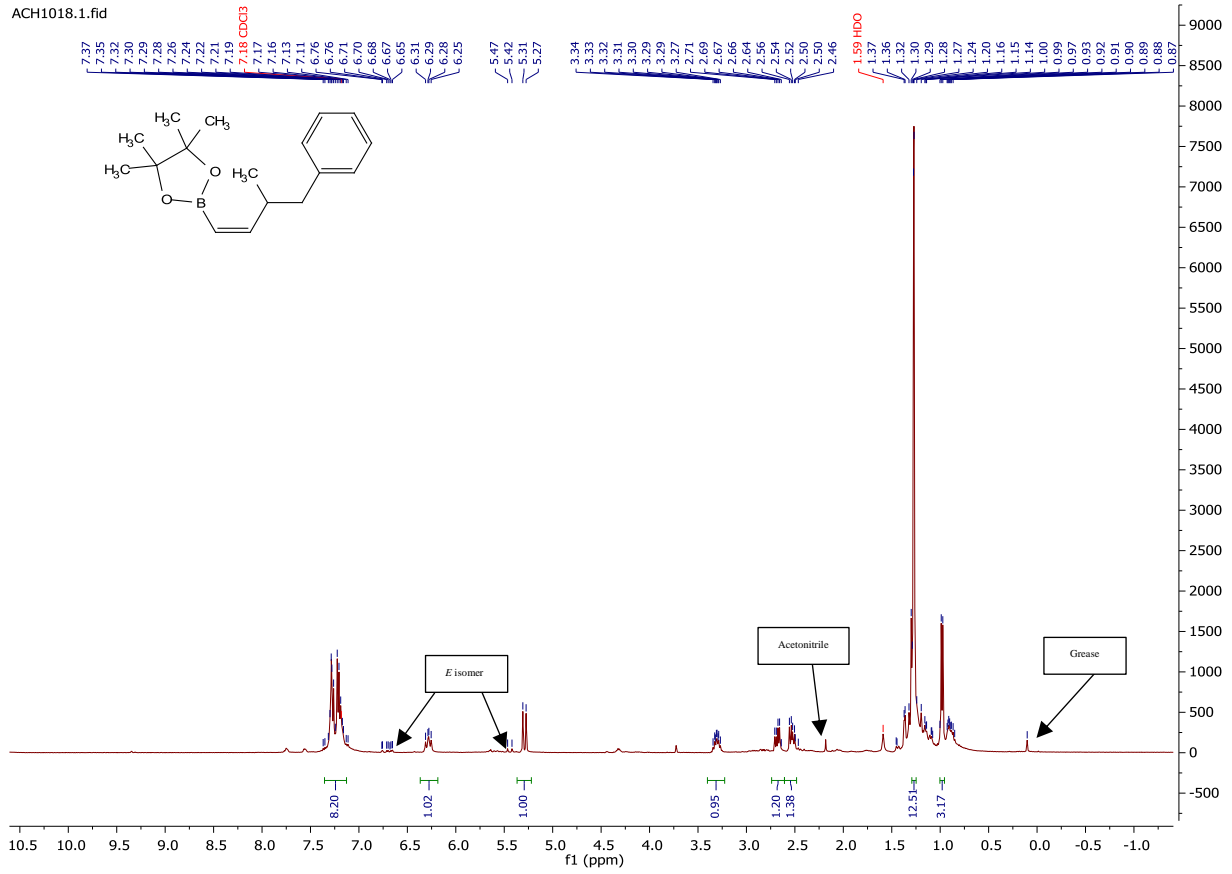
GB2024B\_B11.1.fid

29.61

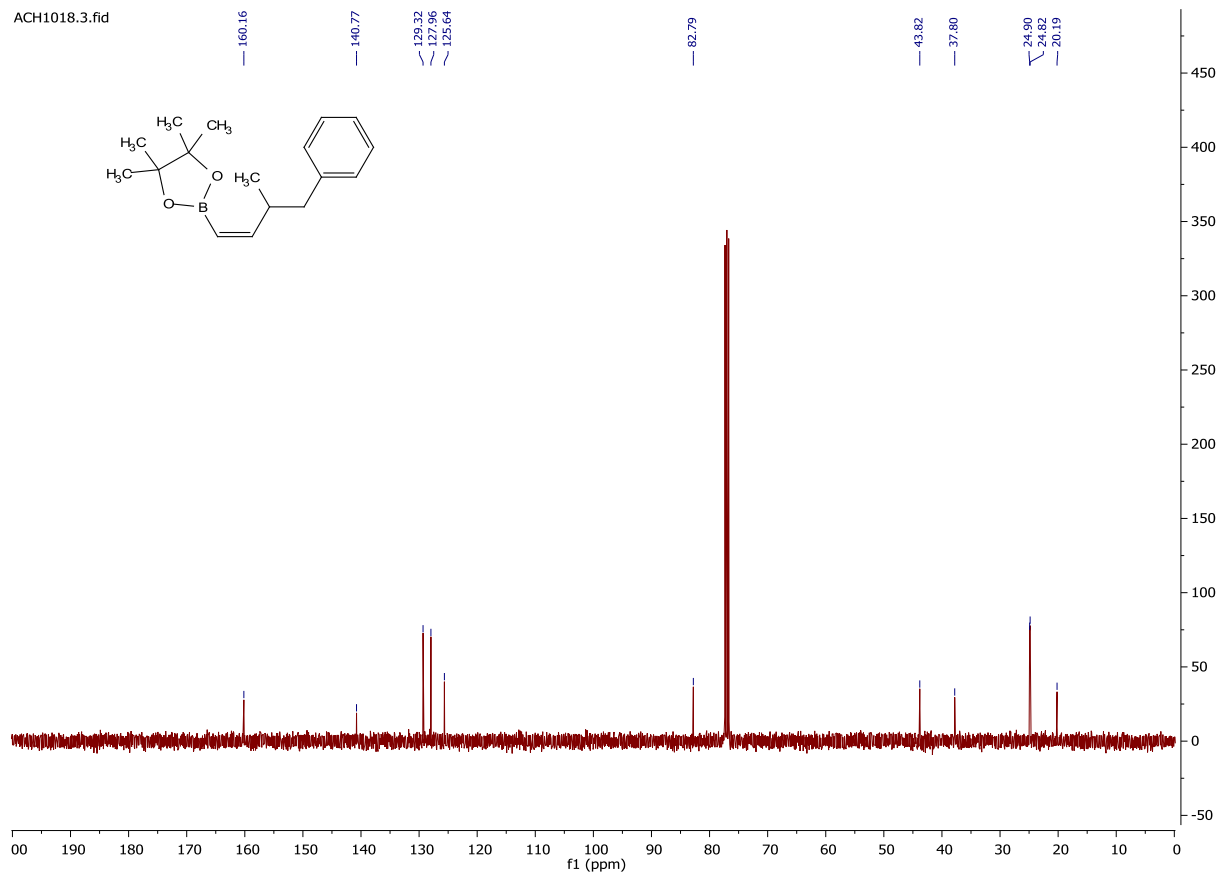


# Molecule 3i

ACH1018.1.fid



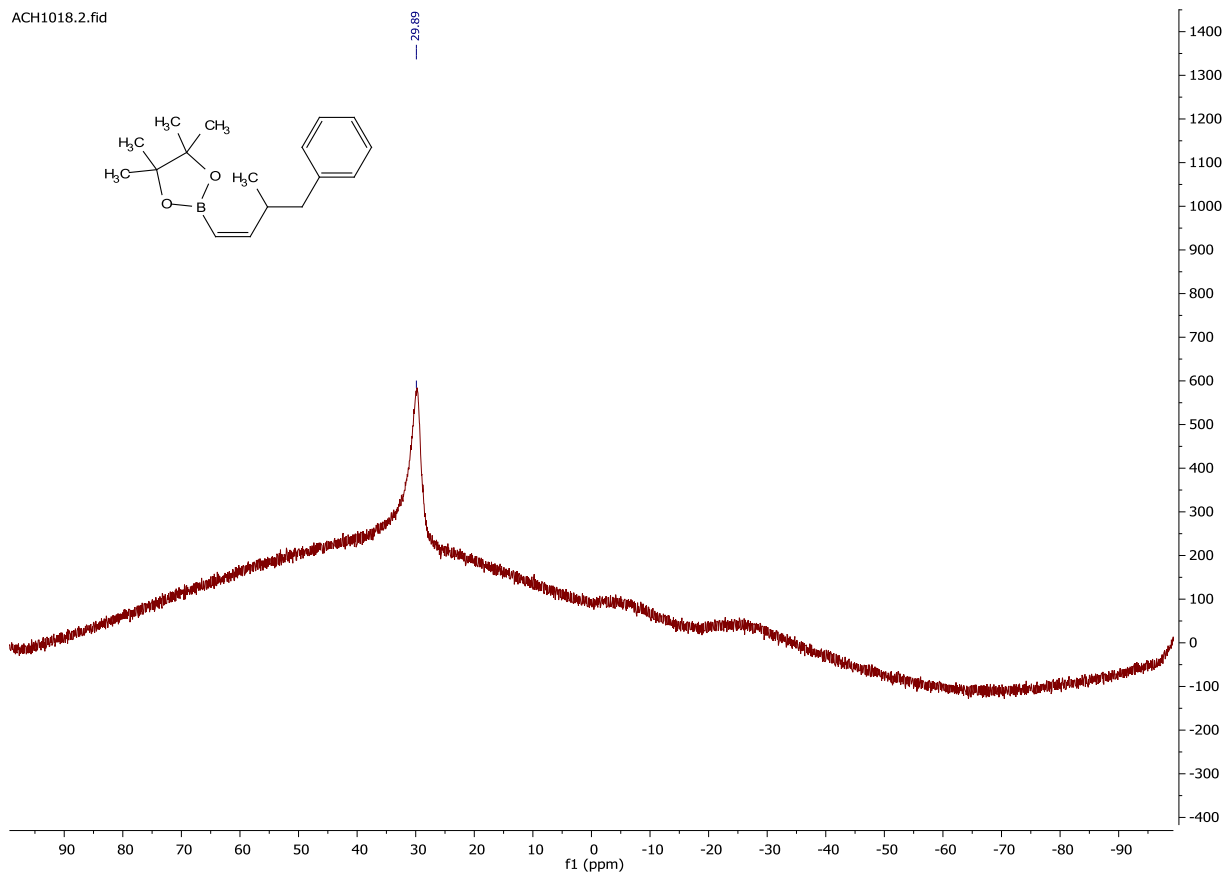
ACH1018.3.fid



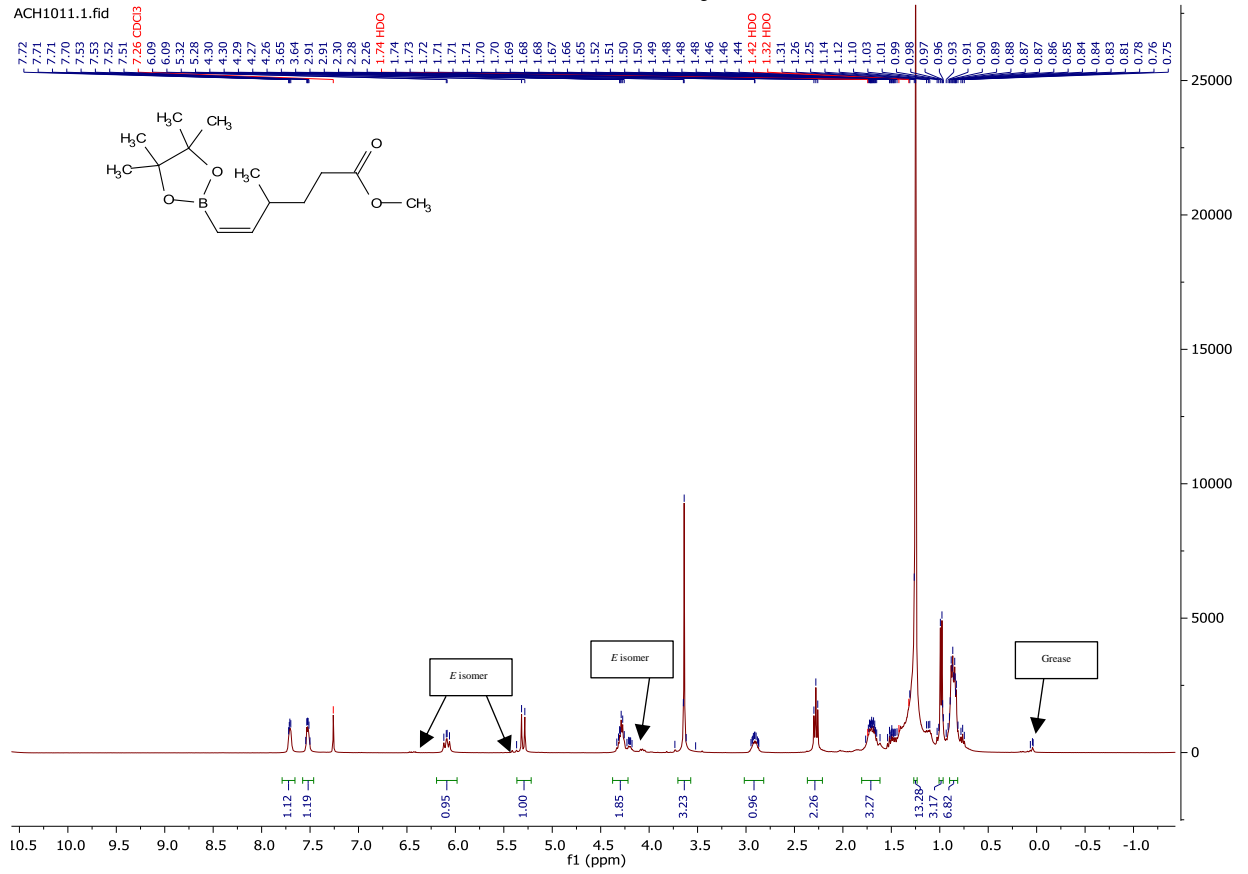


ACH1018.2.fid

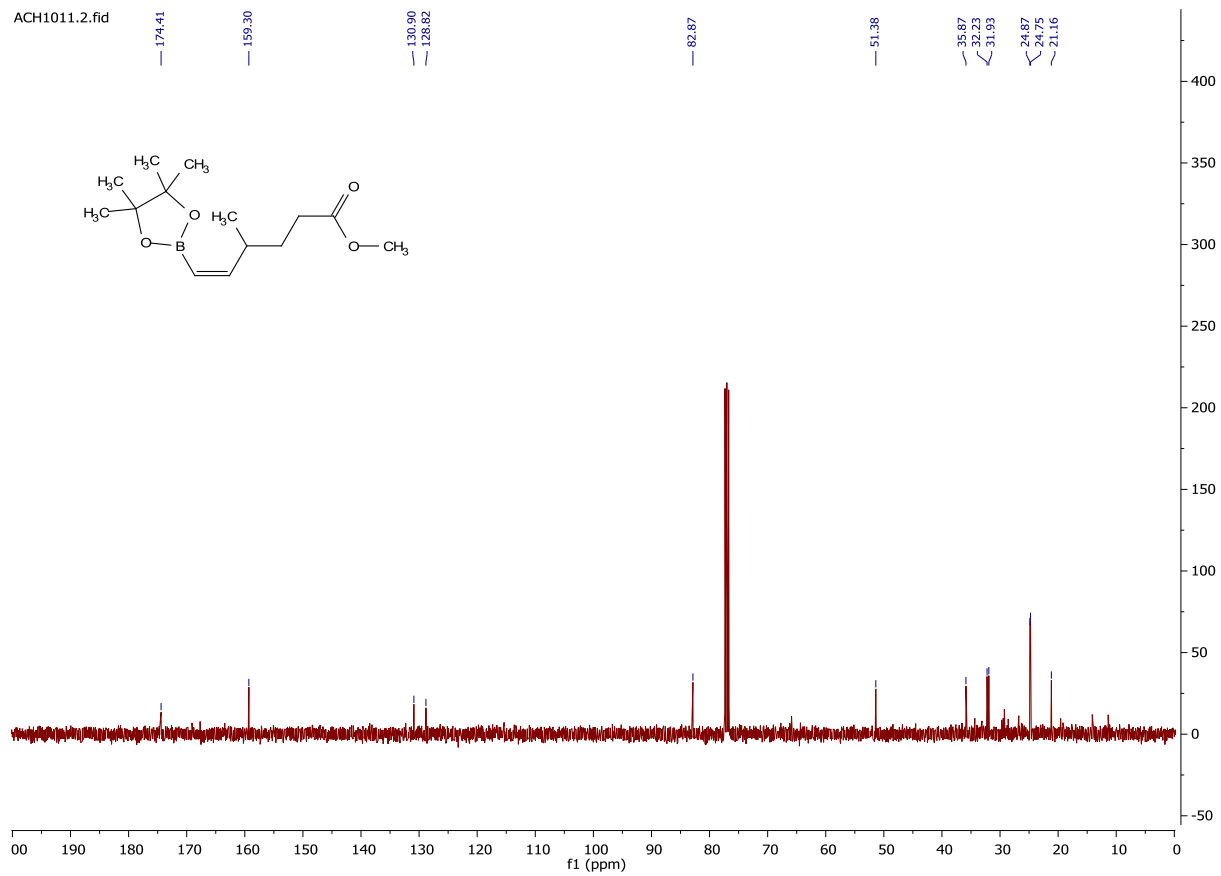
29.89



# Molecule 3j

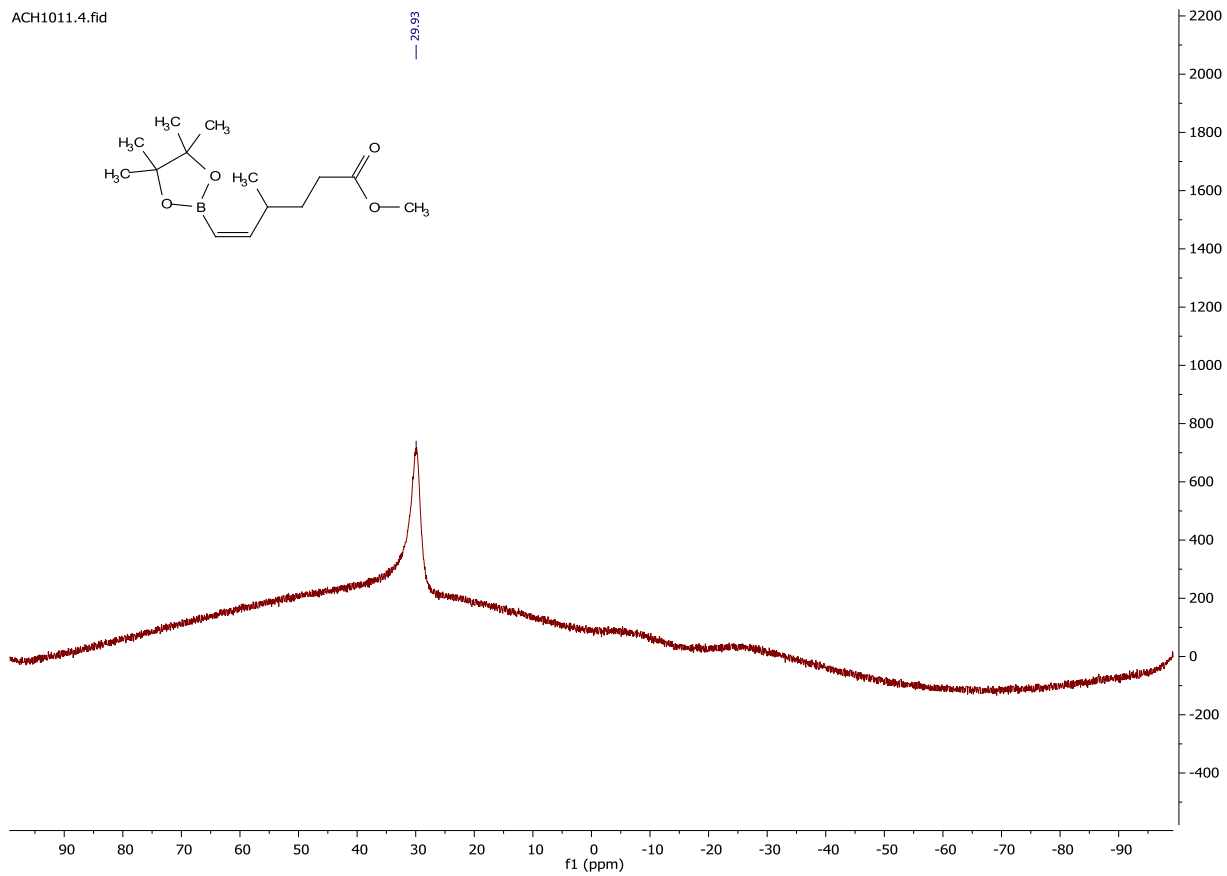


ACH1011.2.fid



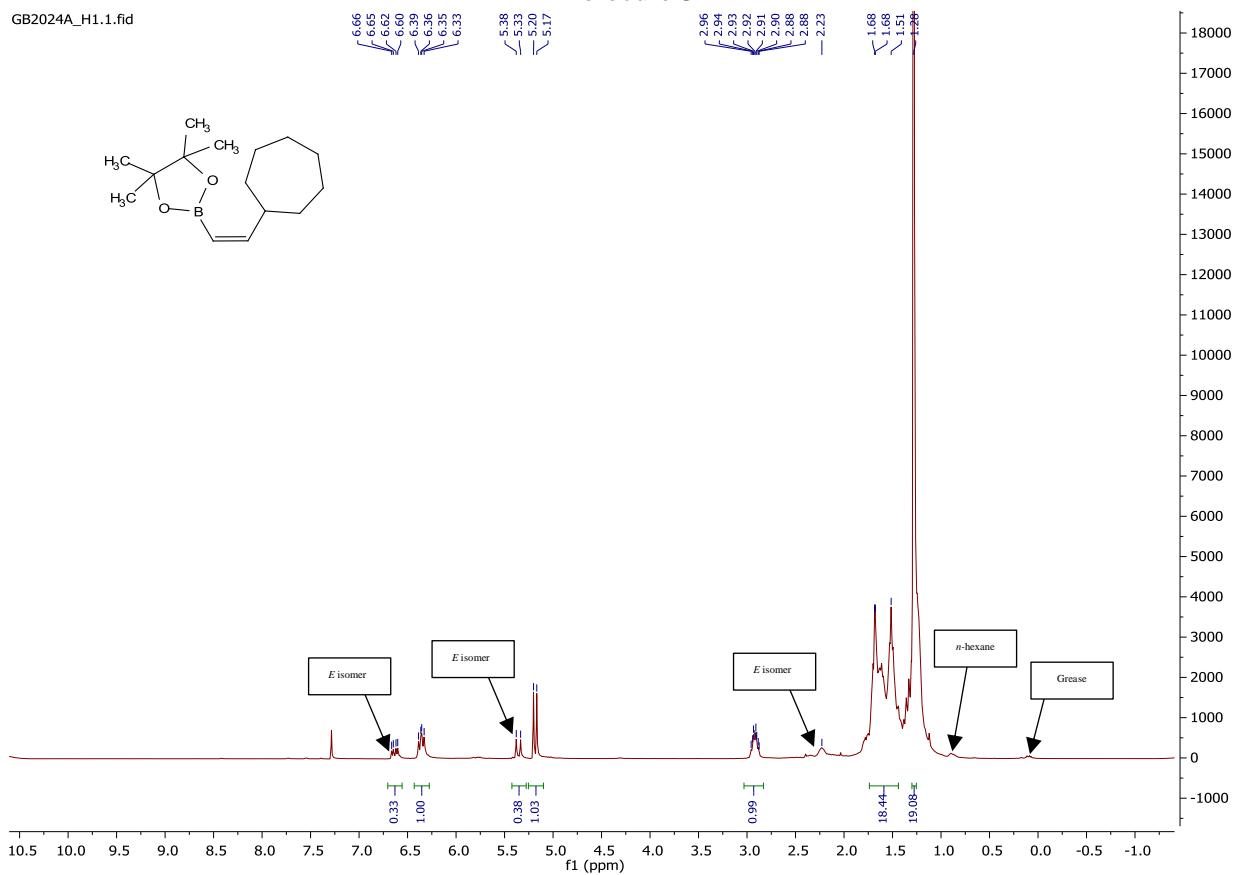
ACH1011.4.fid

29.93

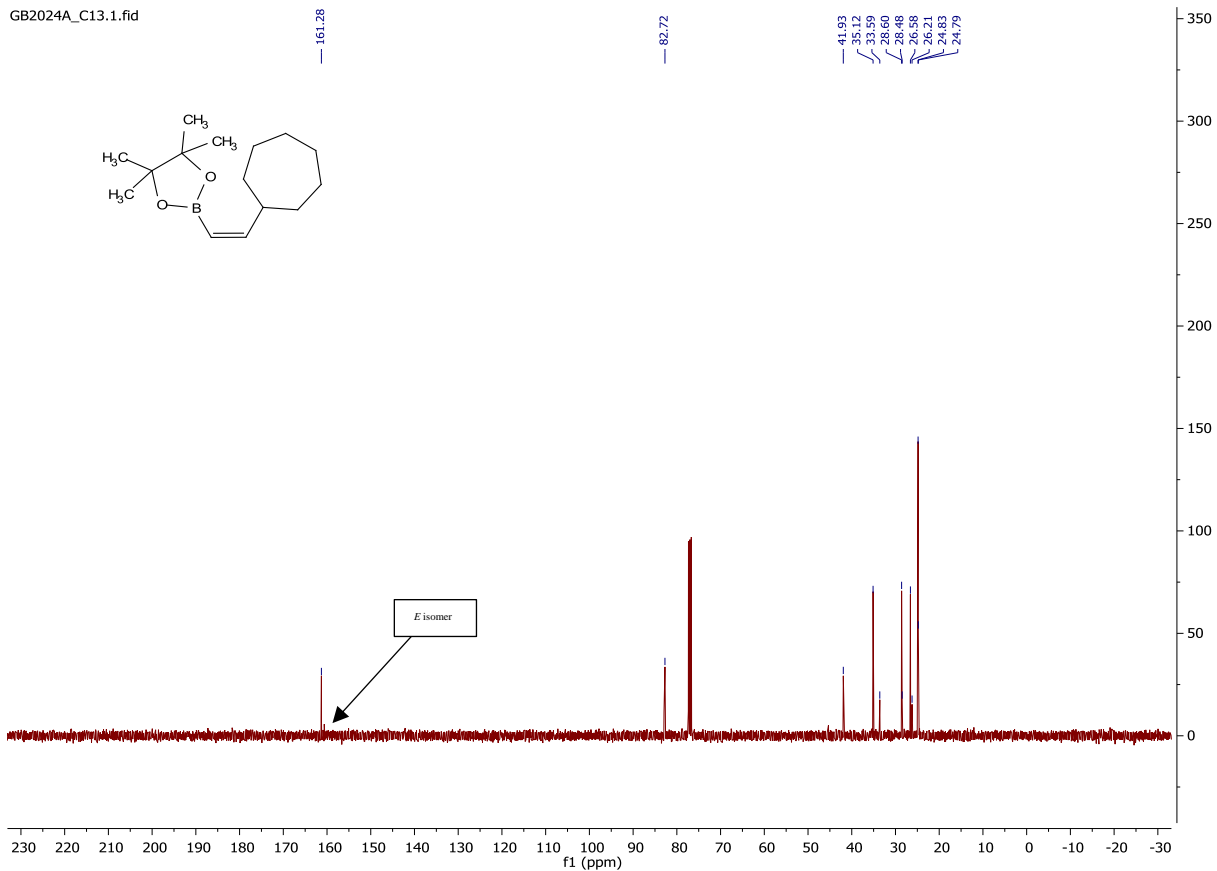


GB2024A\_H1.1.fid

### Molecule 3k

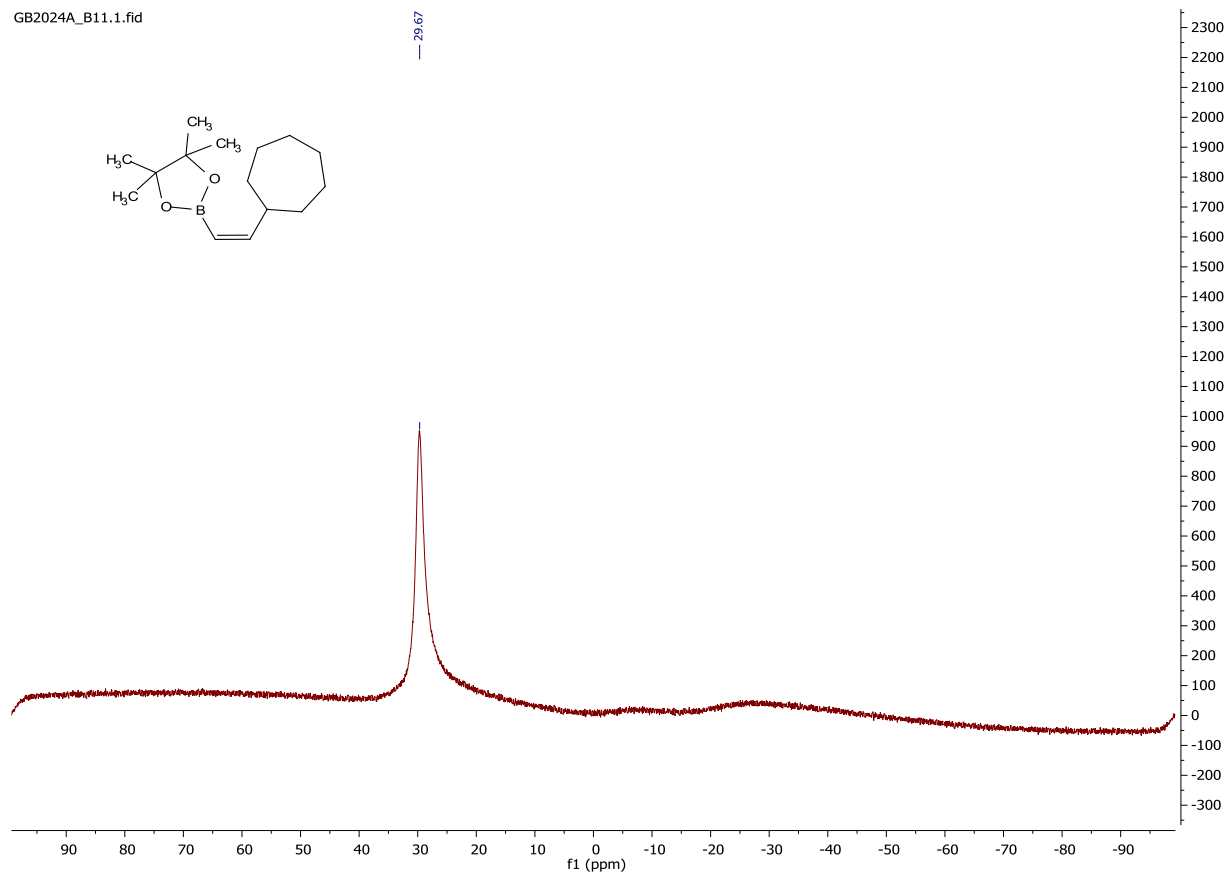
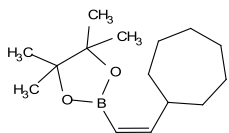


GB2024A\_C13.1.fid



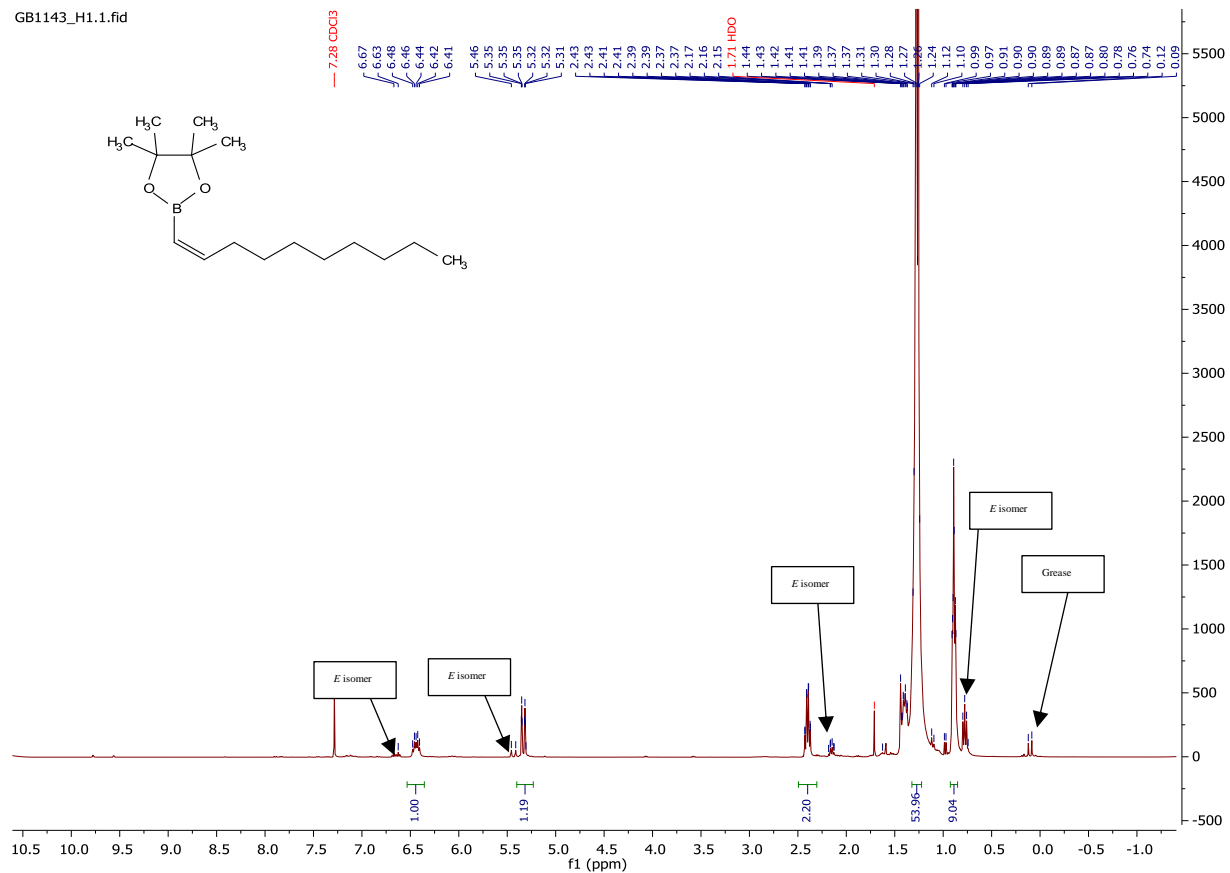
GB2024A\_B11.1.fid

29.67



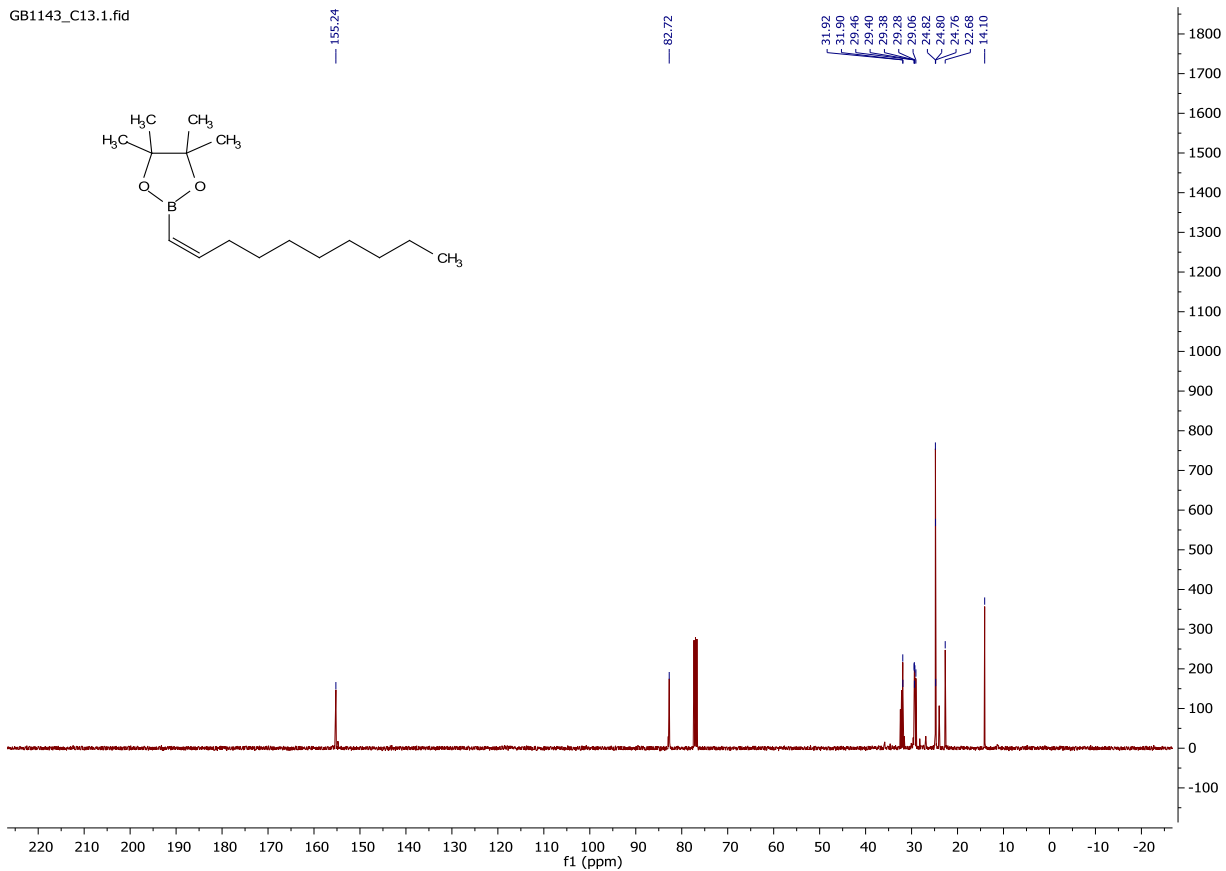
# Molecule 31

GB1143\_H1.1.fid

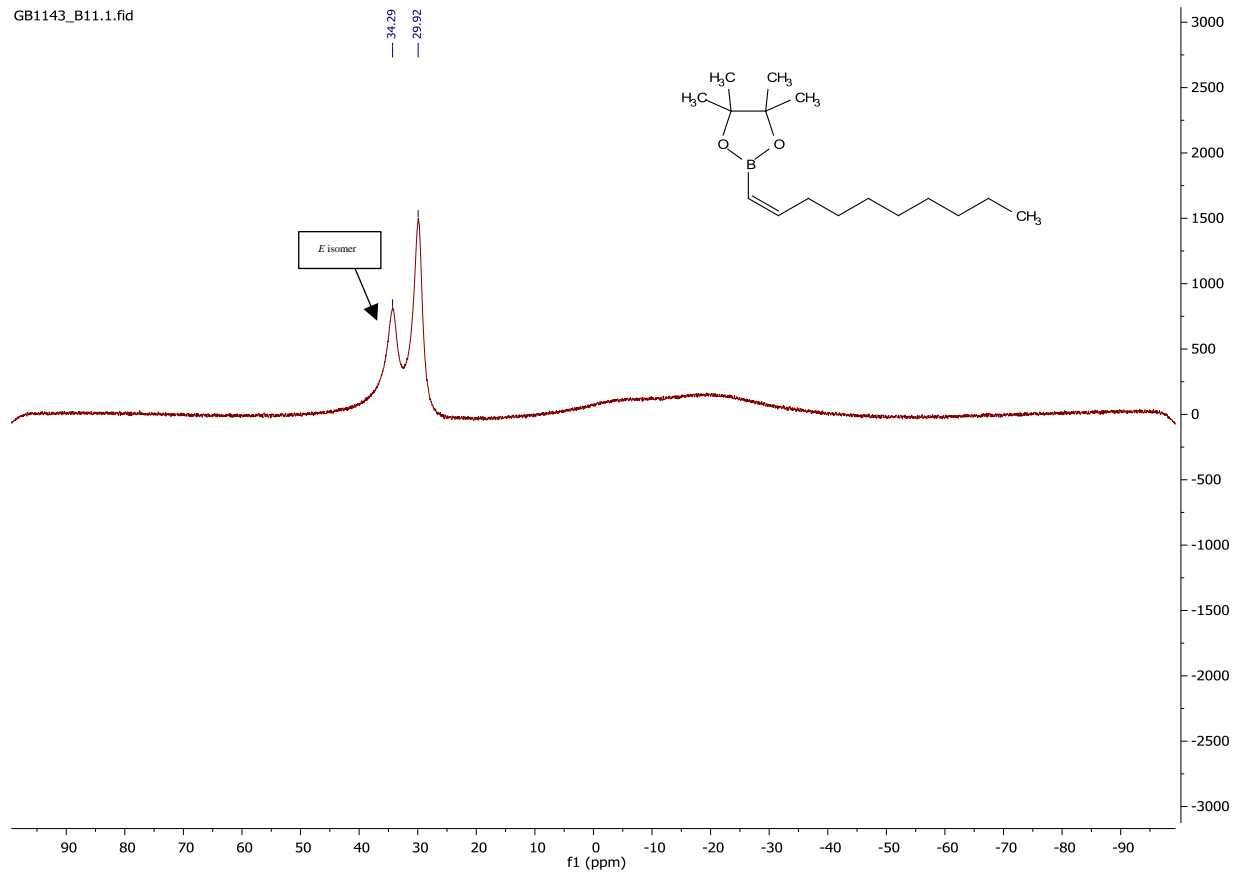




GB1143\_C13.1.fid

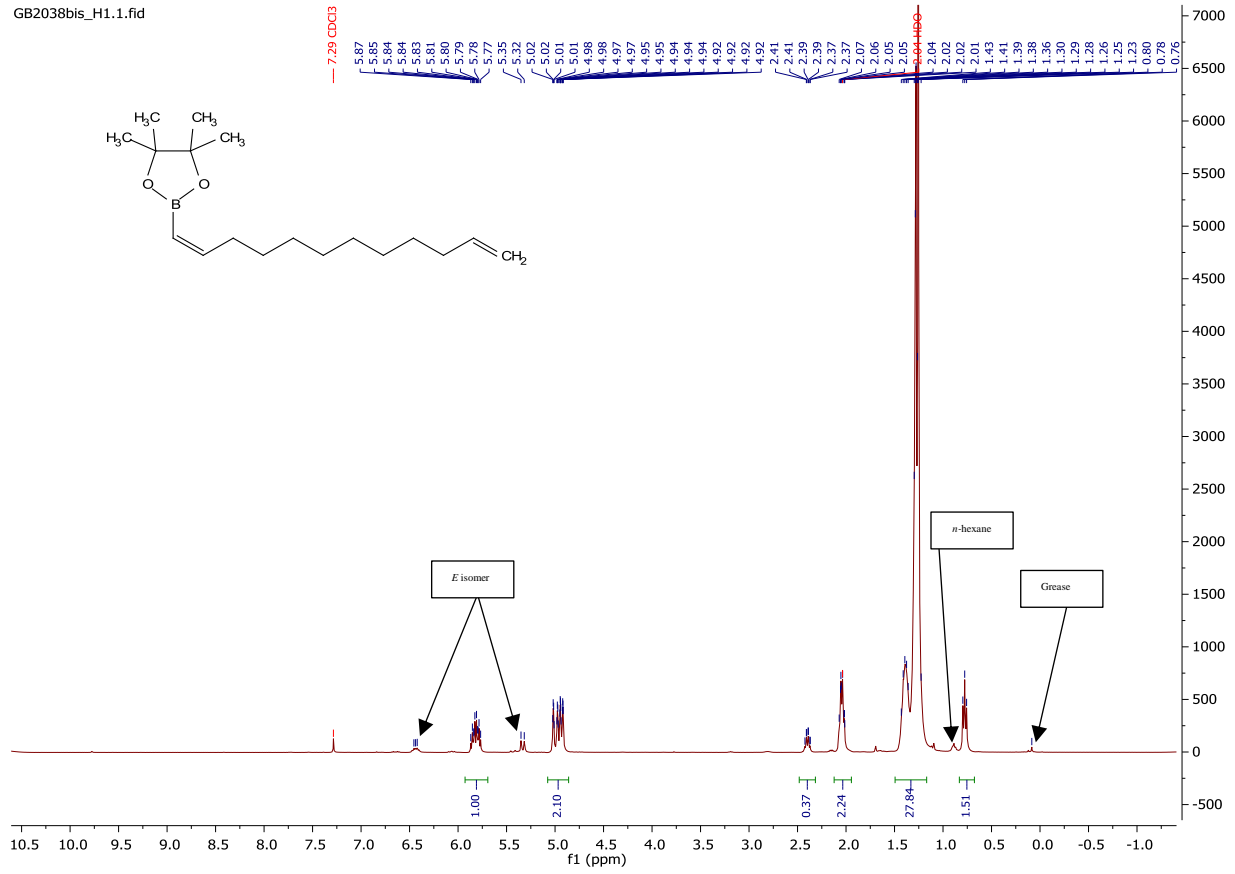


GB1143\_B11.1.fid

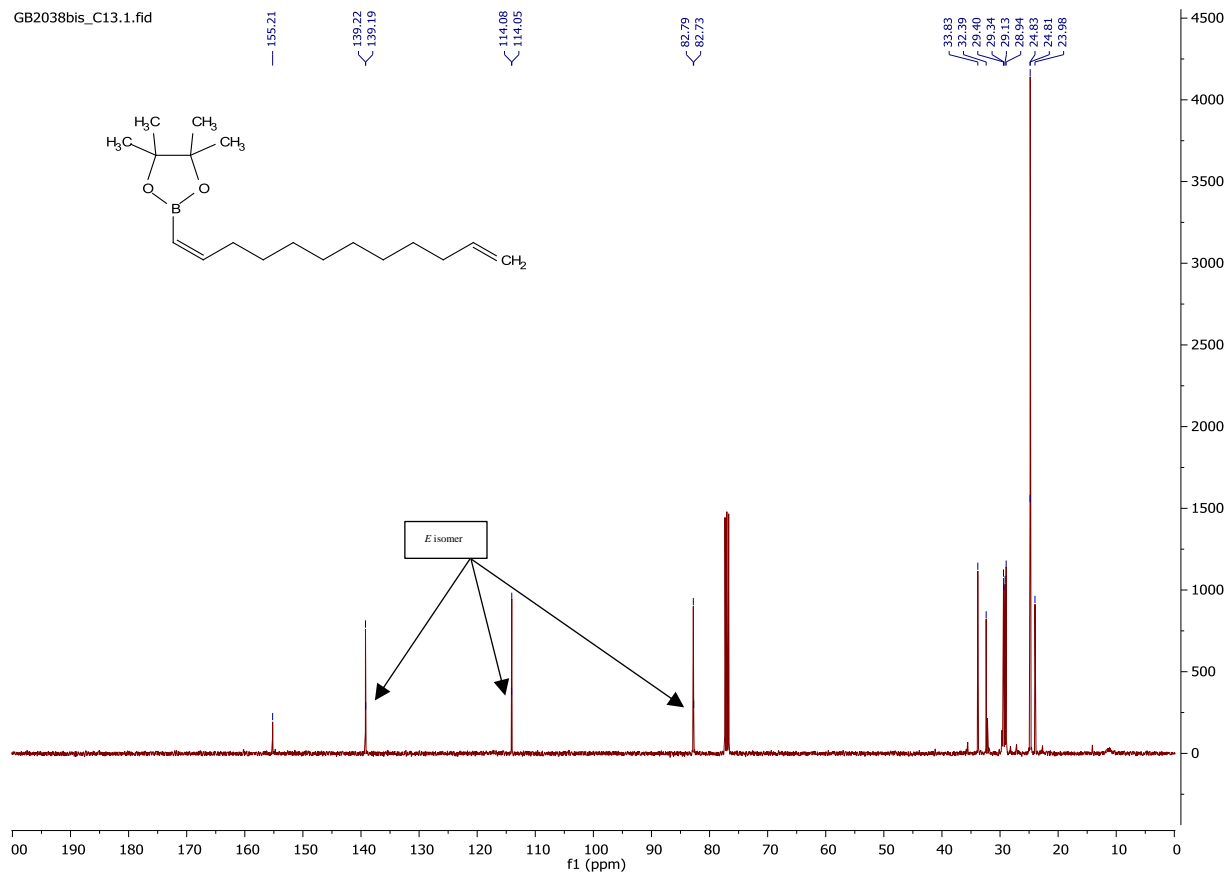


# Molecule 3m

GB2038bis\_H1.1.fid

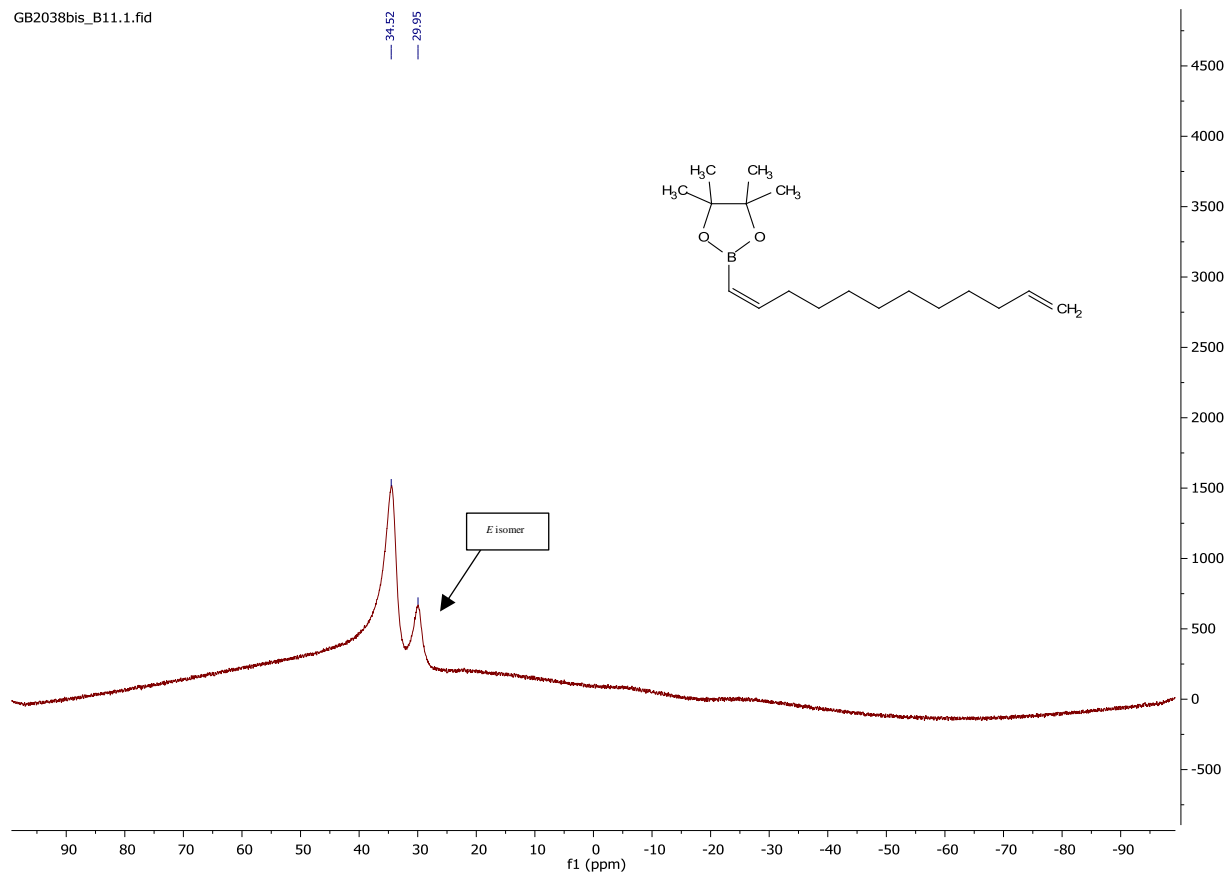


GB2038bis\_C13.1.fid



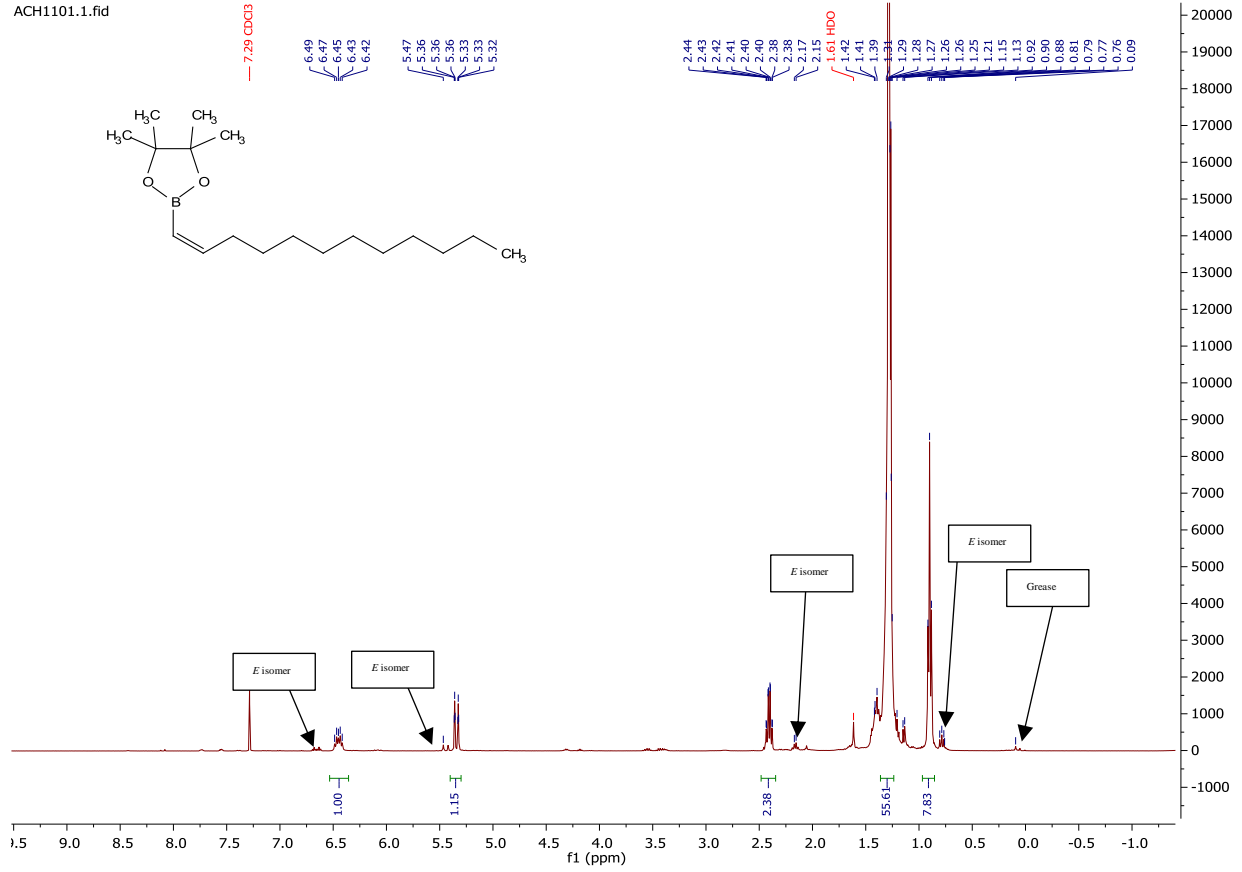
GB2038bis\_B11.1.fid

34.52  
29.95

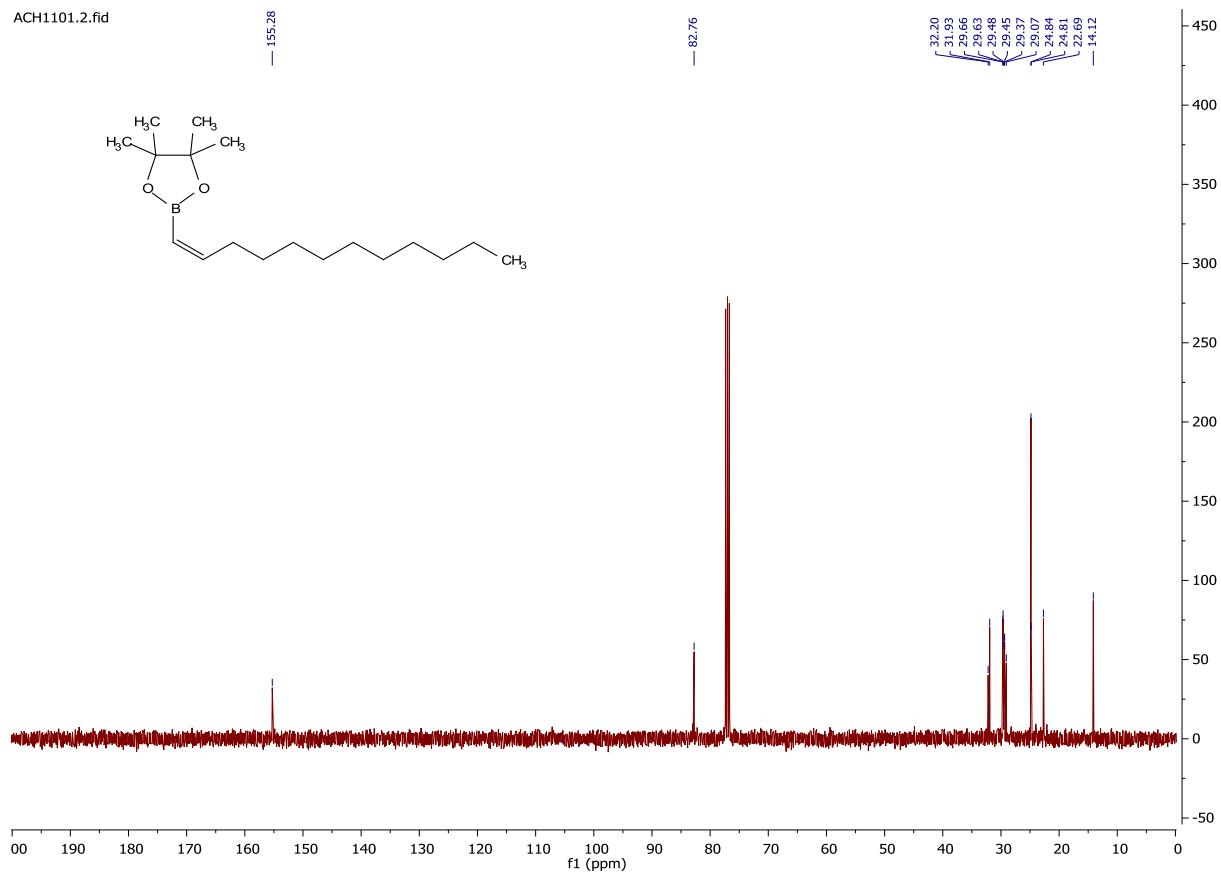


# Molecule 3n

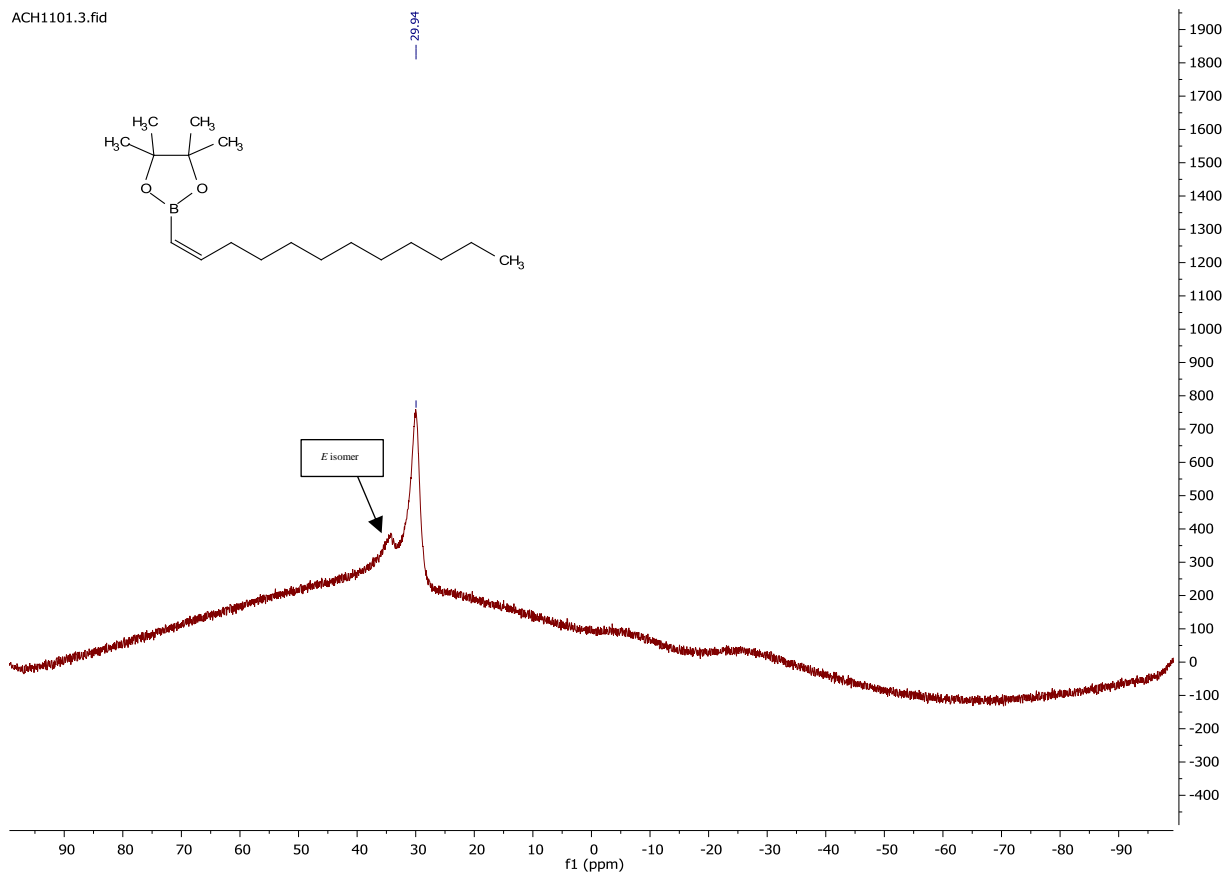
ACH1101.1.fid



ACH1101.2.fid



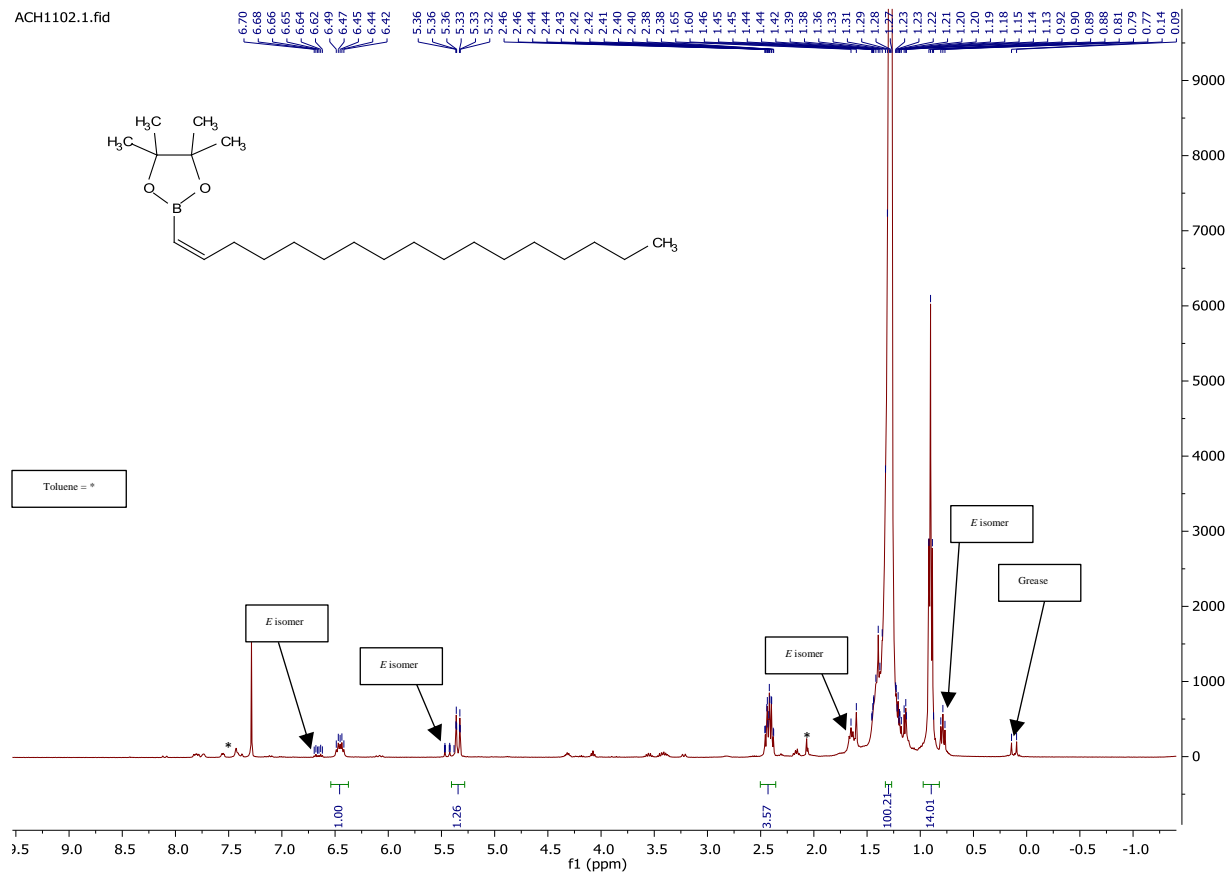
ACH1101.3.fid



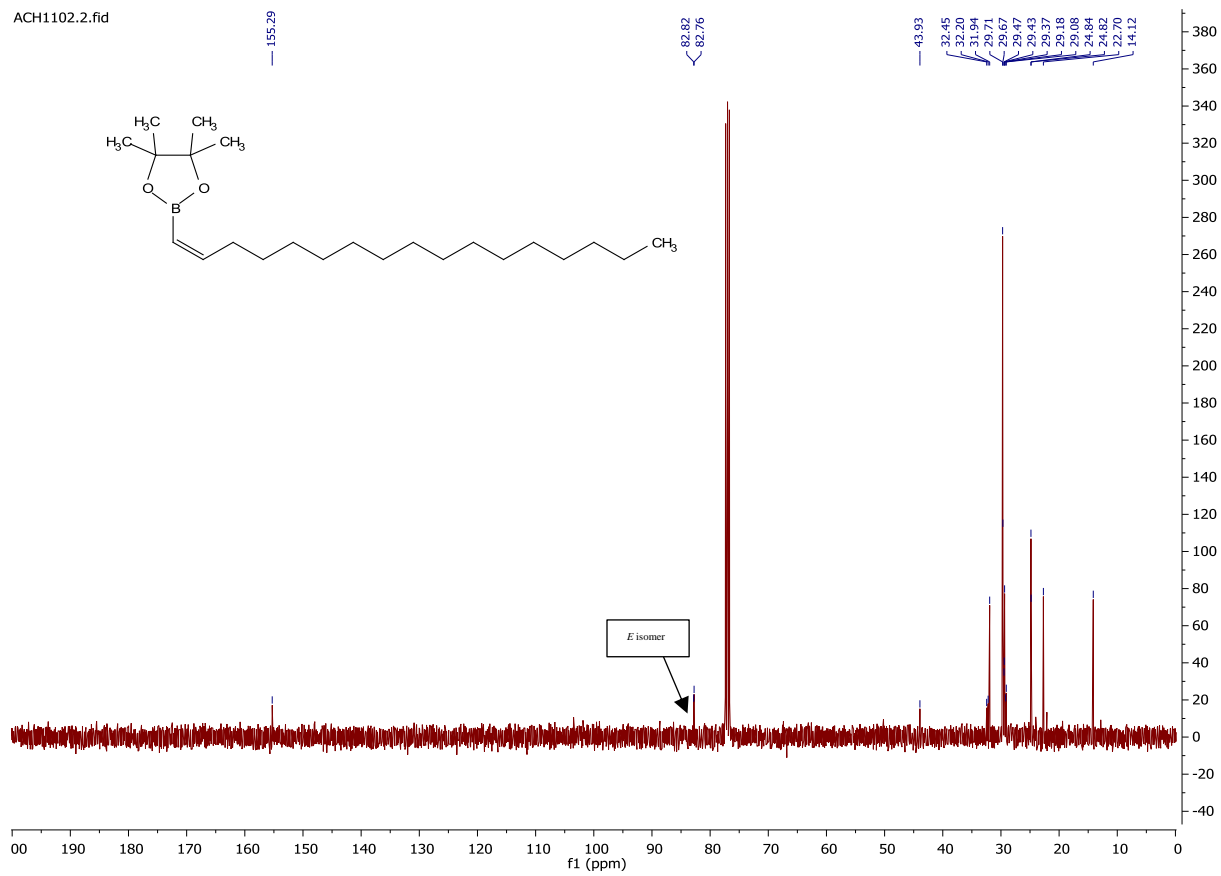


# Molecule 3o

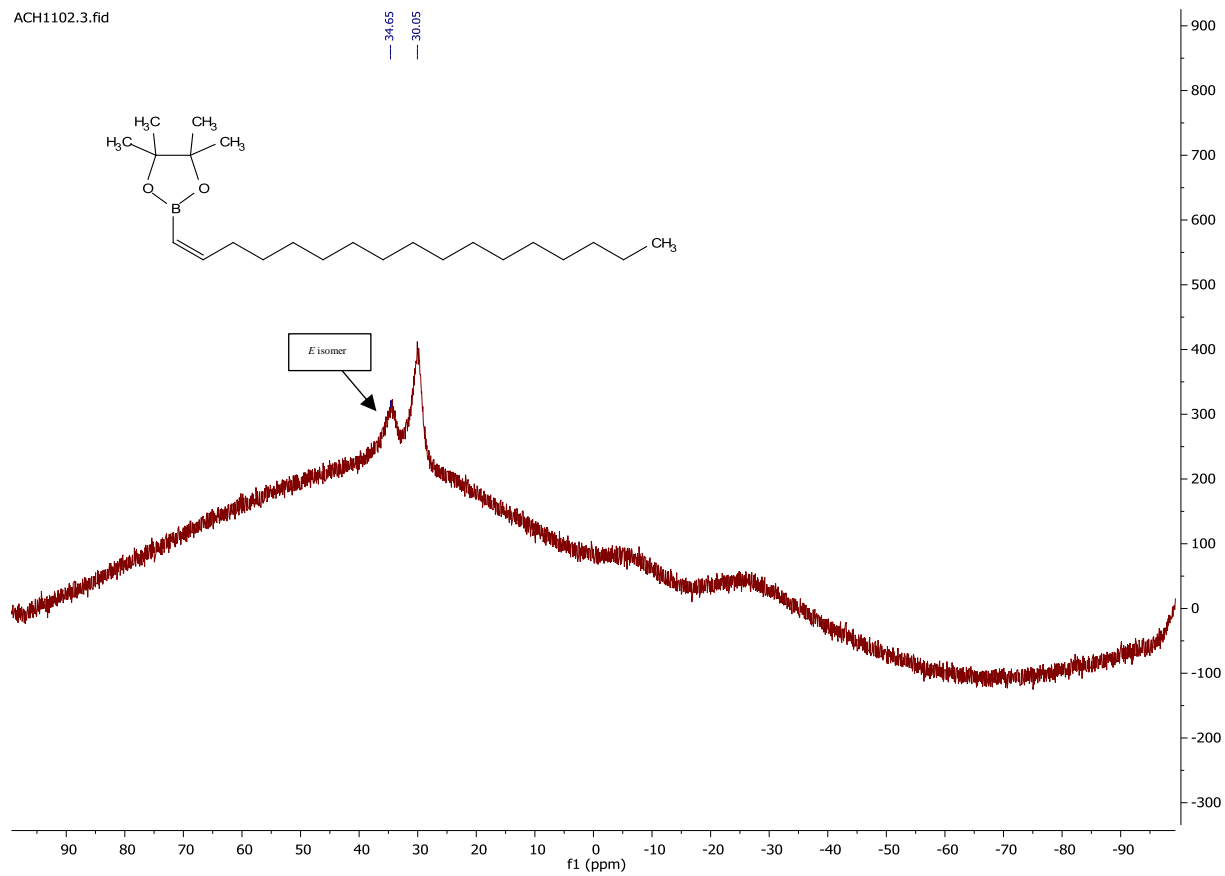
ACH1102.1.fid



ACH1102.2.fid

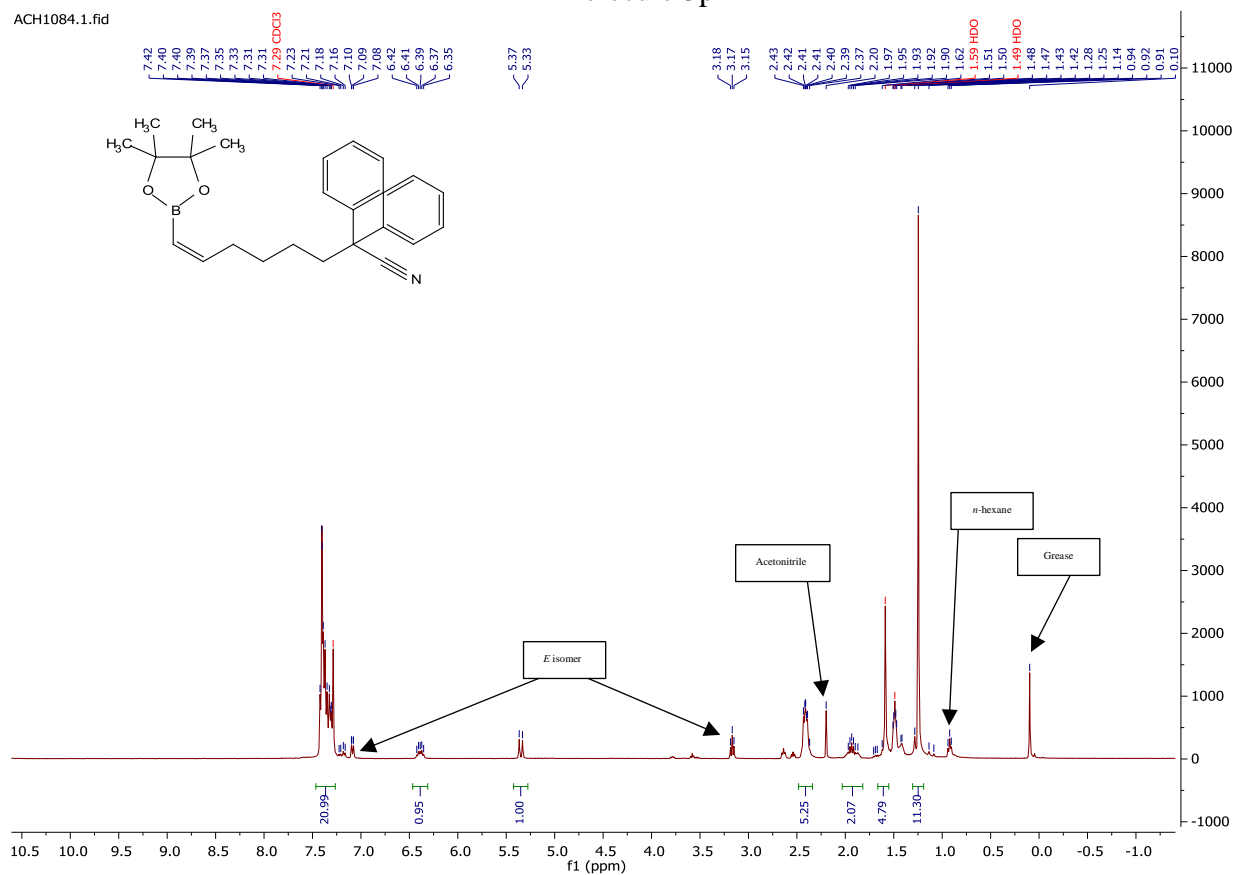


ACH1102.3.fid

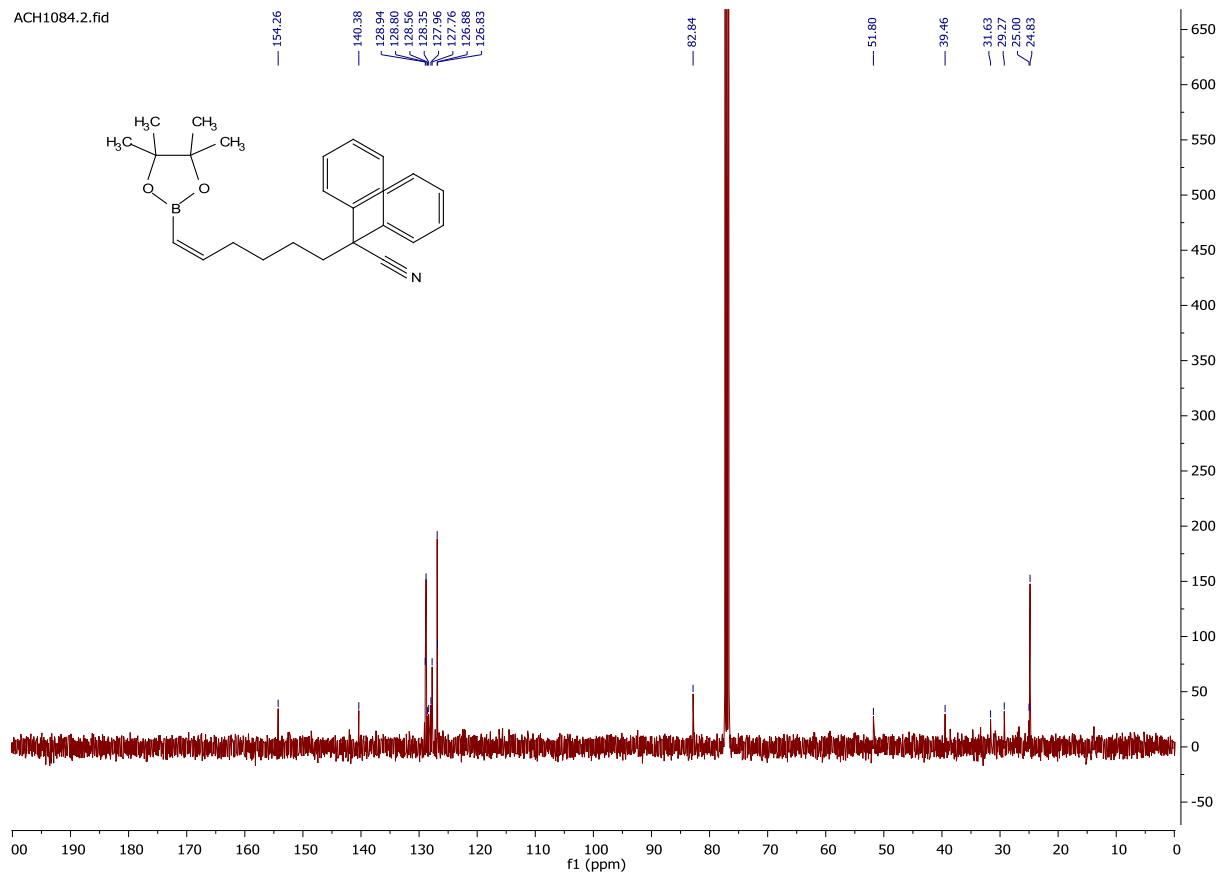


# Molecule 3p

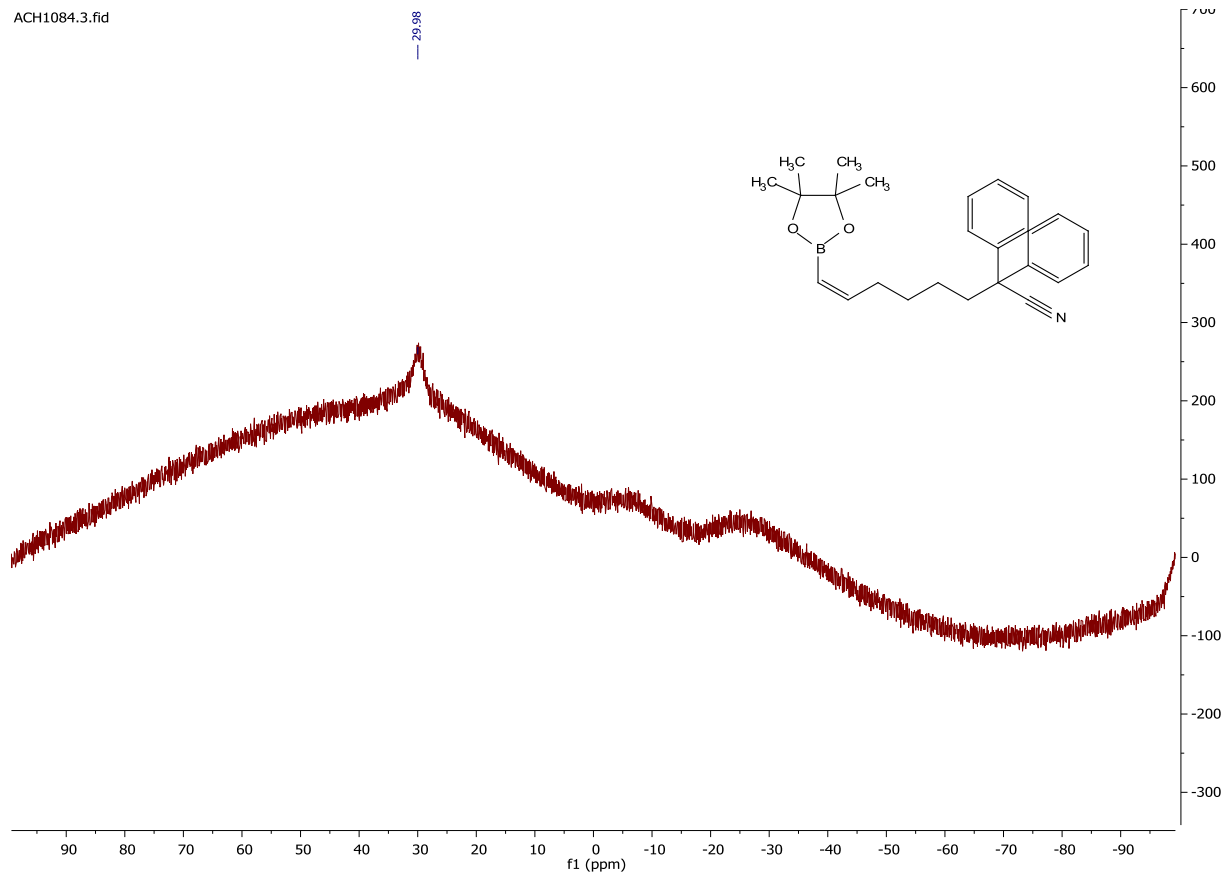
ACH1084.1.fid



ACH1084.2.fid

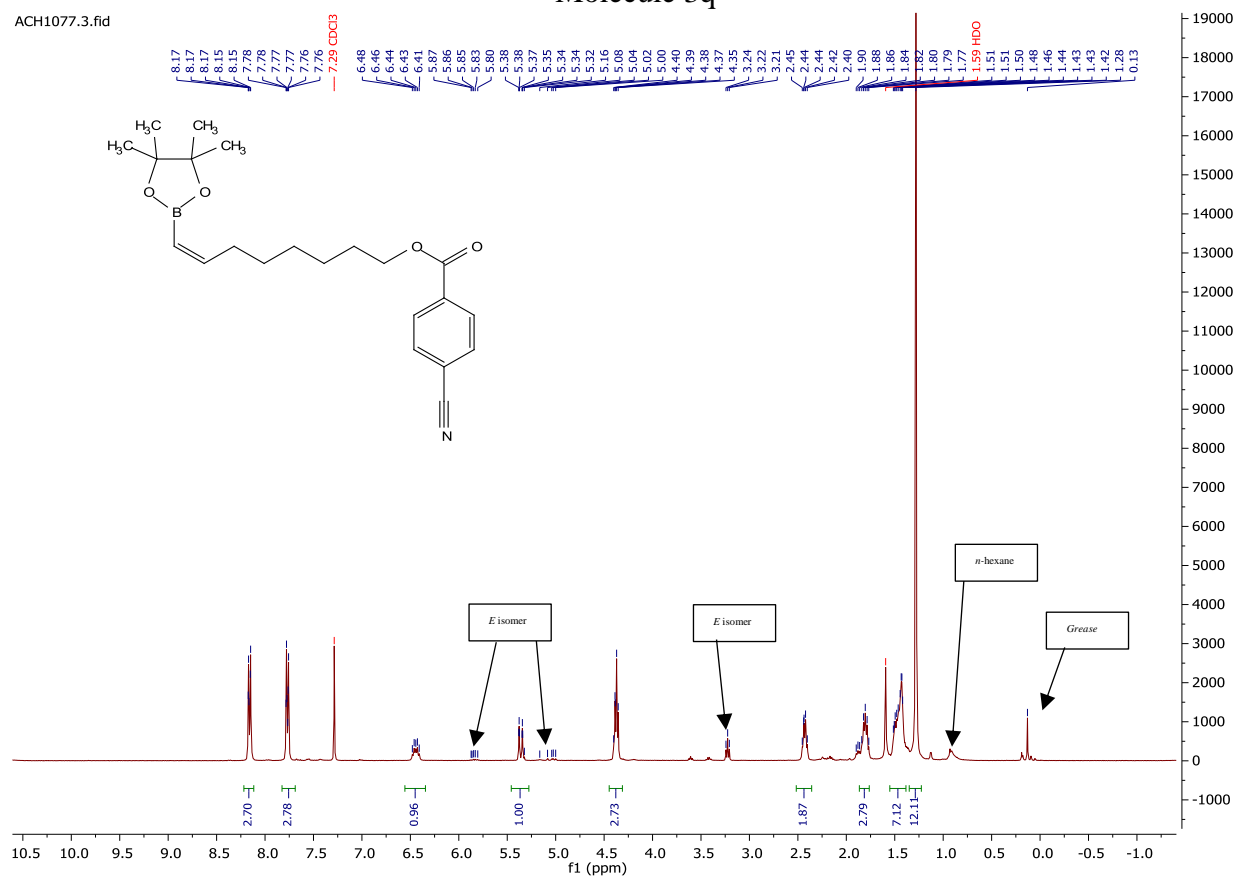


ACH1084.3.fid

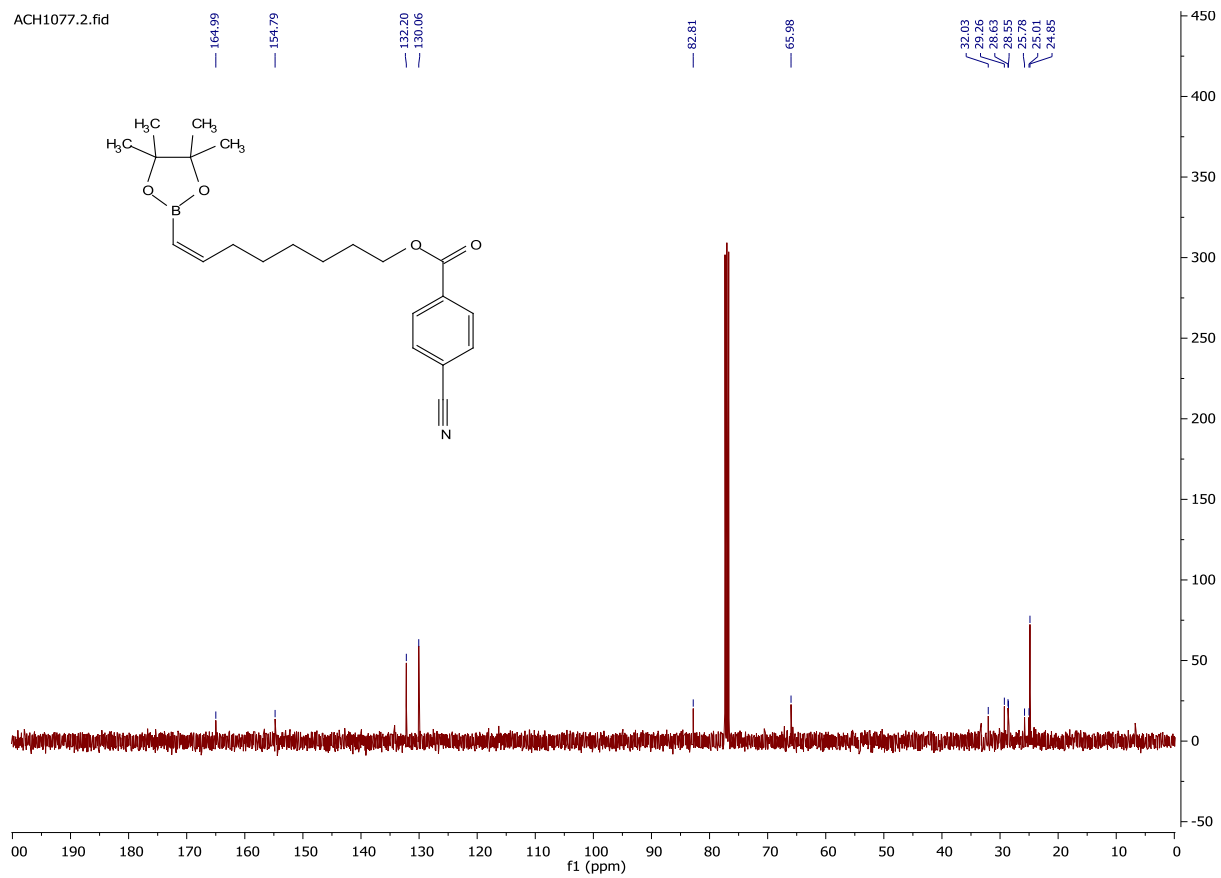


# Molecule 3q

ACH1077.3.fid

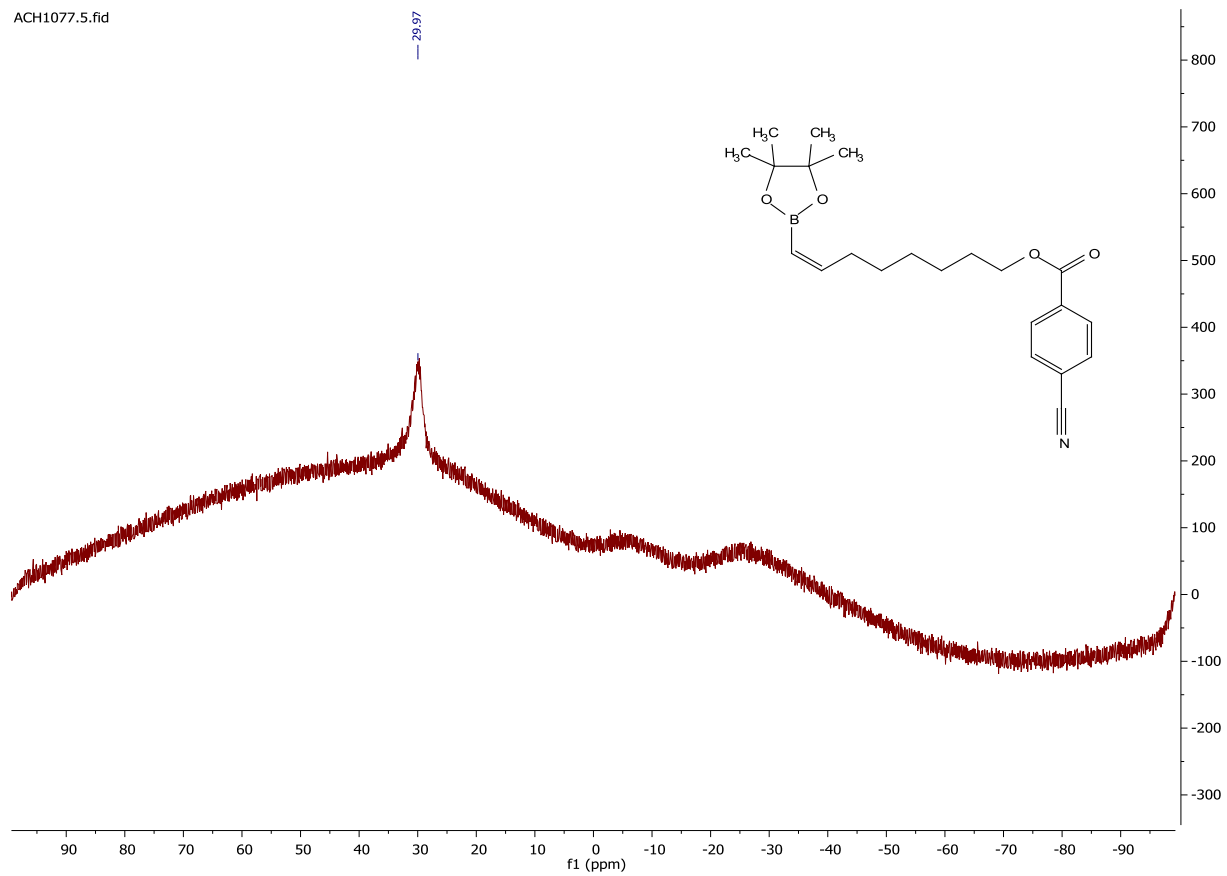


ACH1077.2.fid



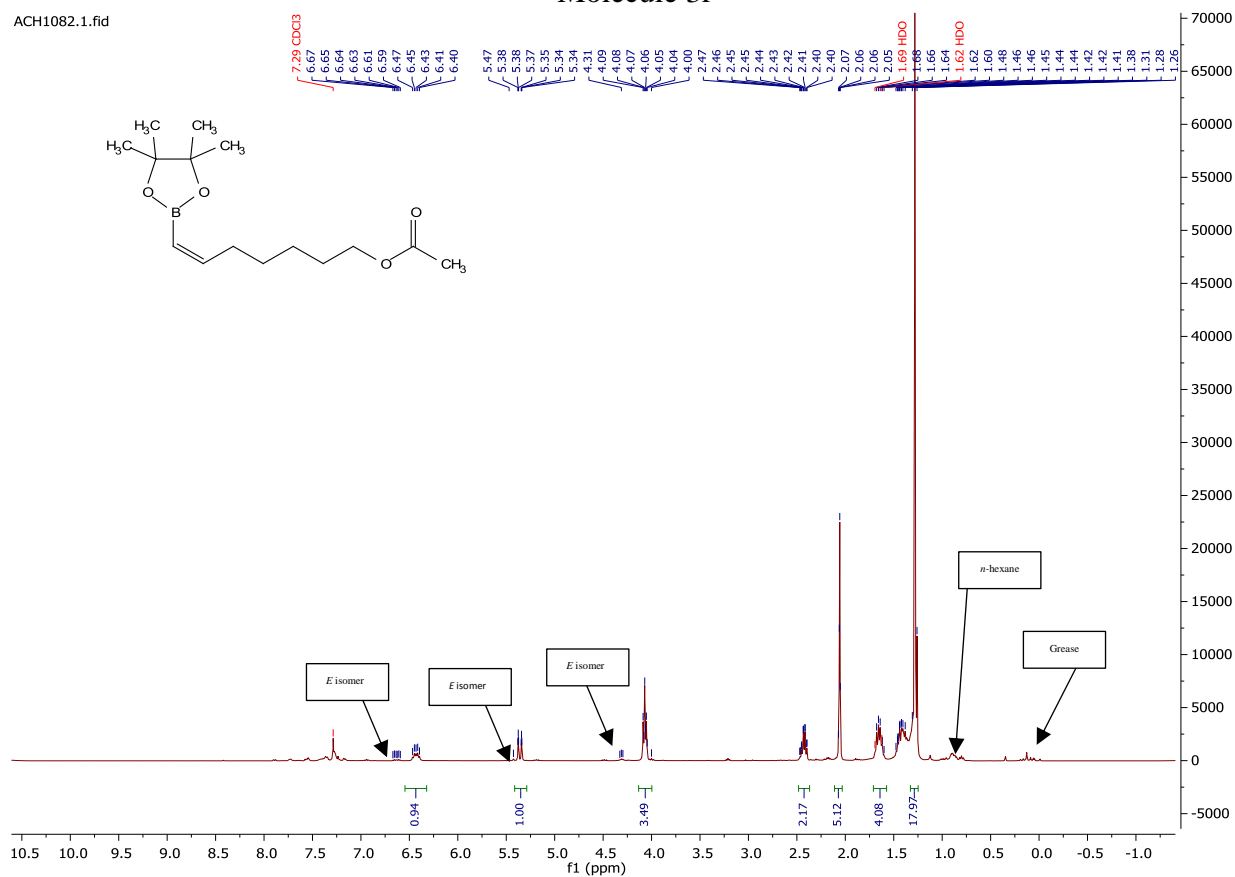


ACH1077.5.fid

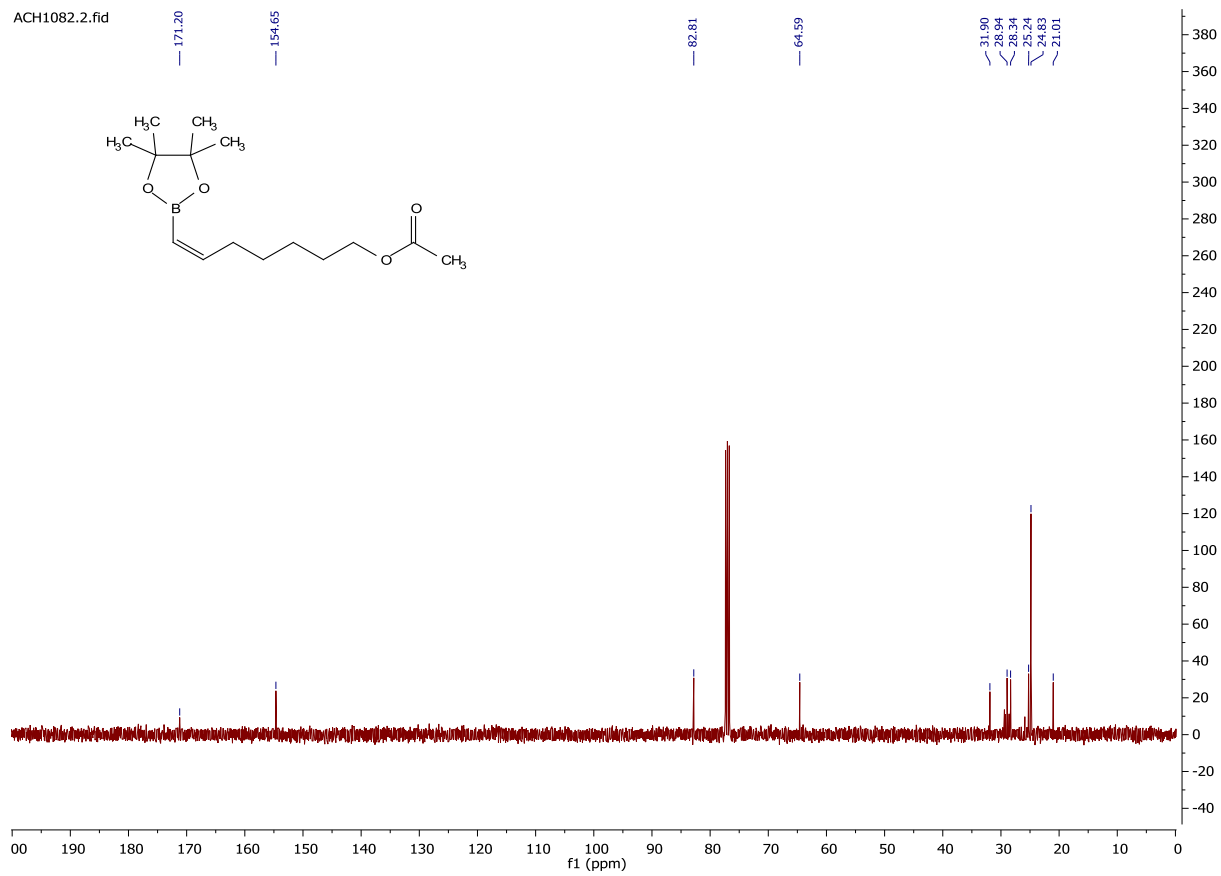


# Molecule 3r

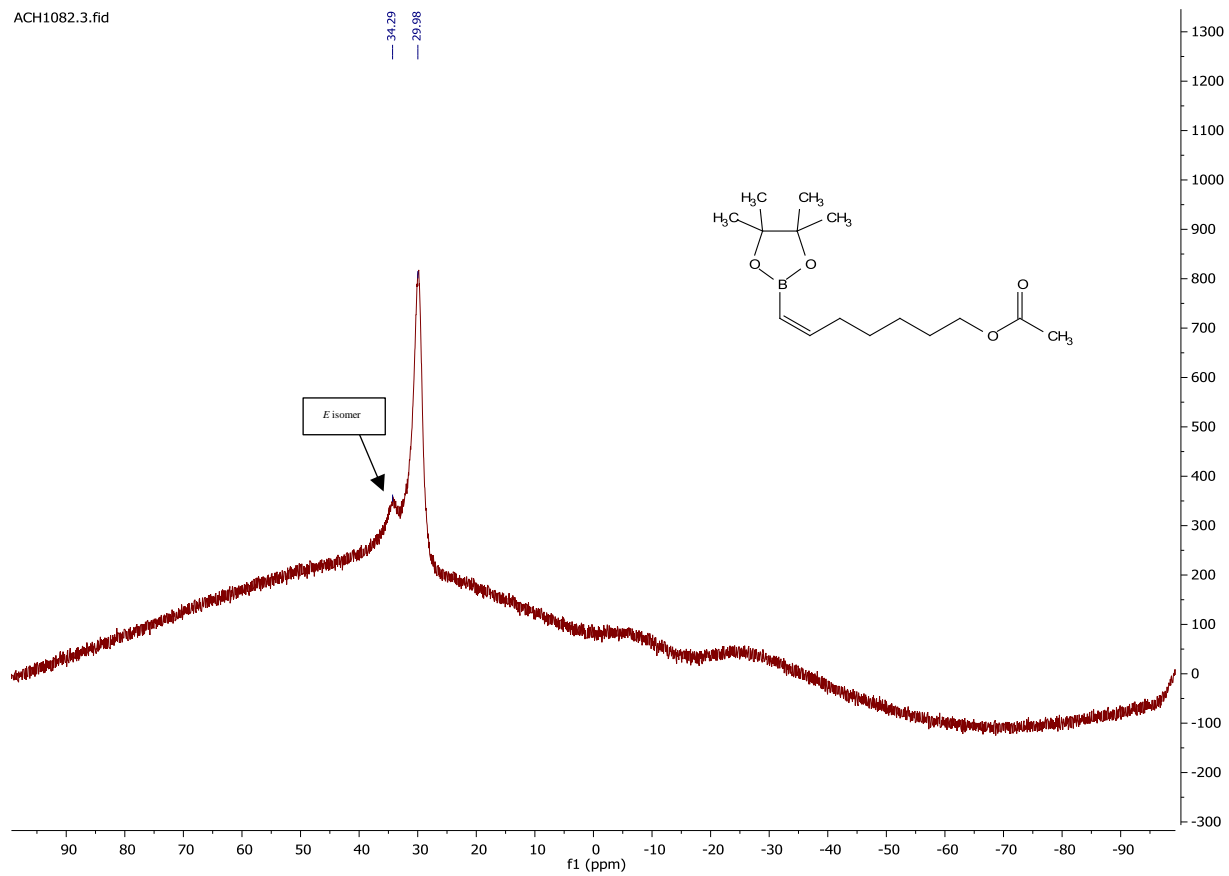
ACH1082.1.fid



ACH1082.2.fid

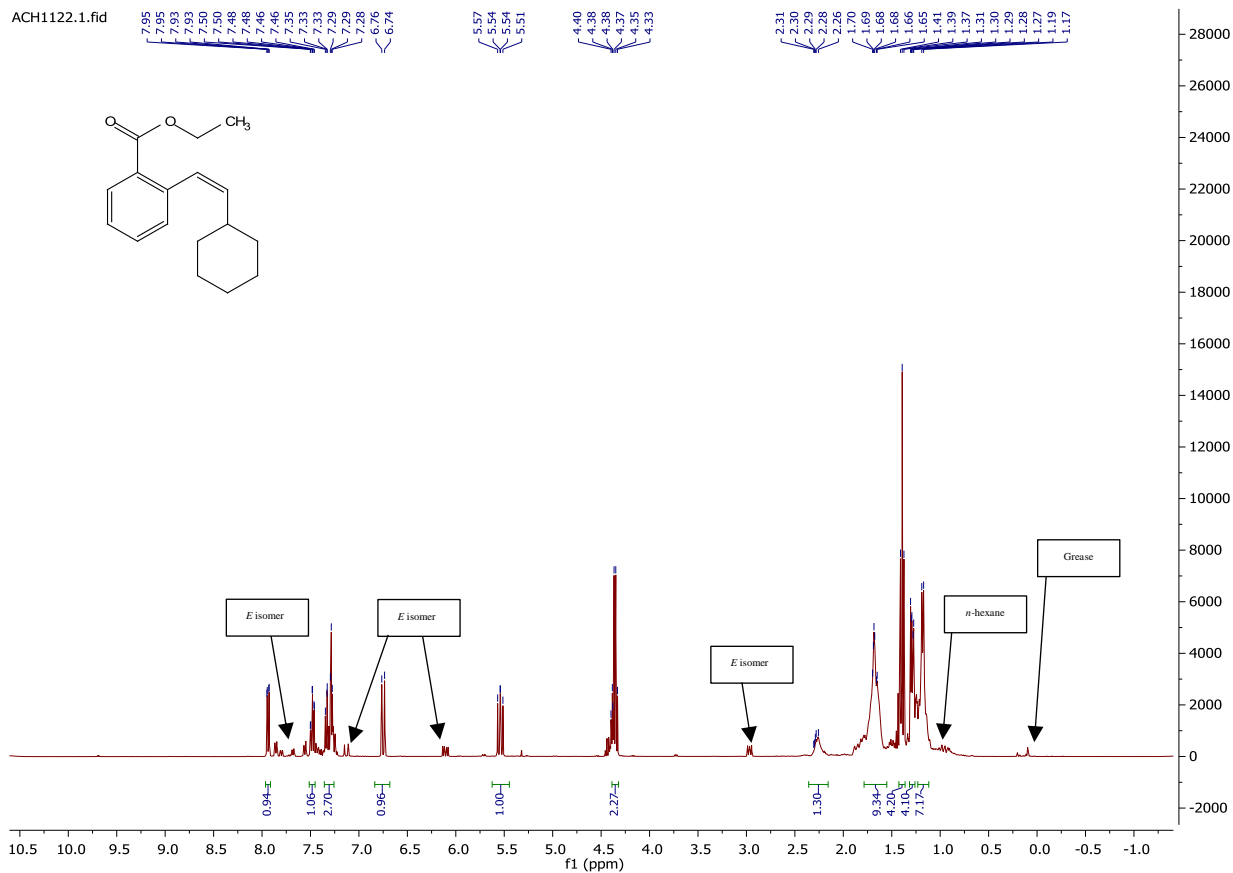


ACH1082.3.fid

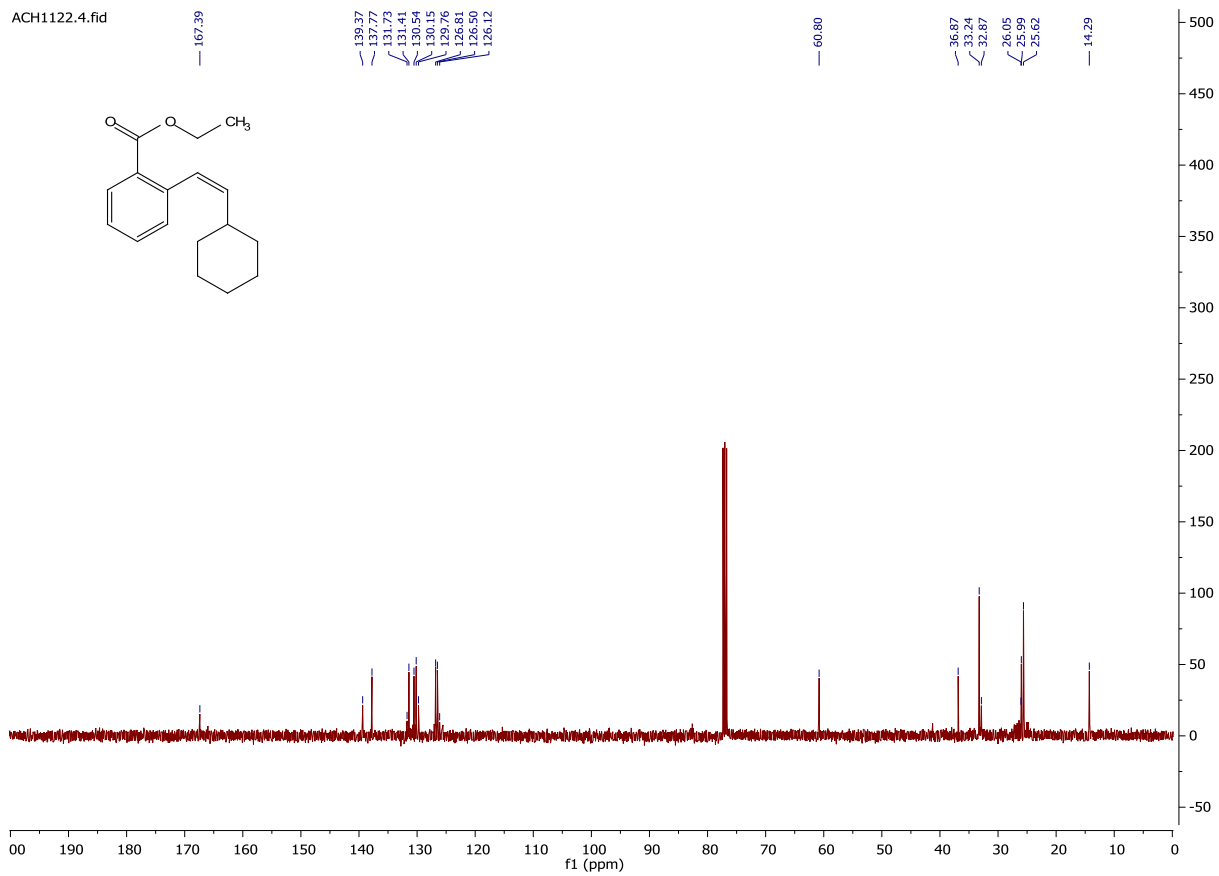


ACH1122.1.fid

### Molecule 5

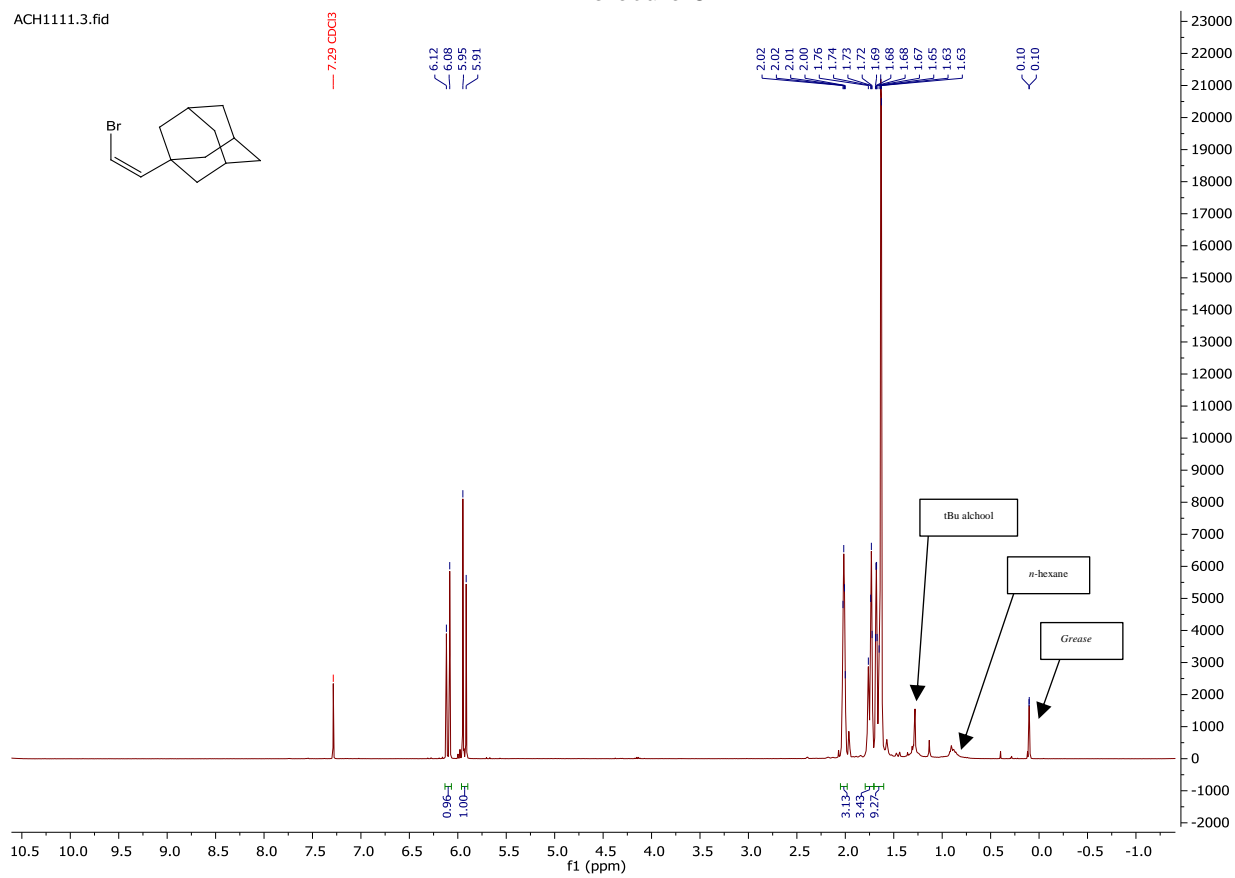


ACH1122.4.fid

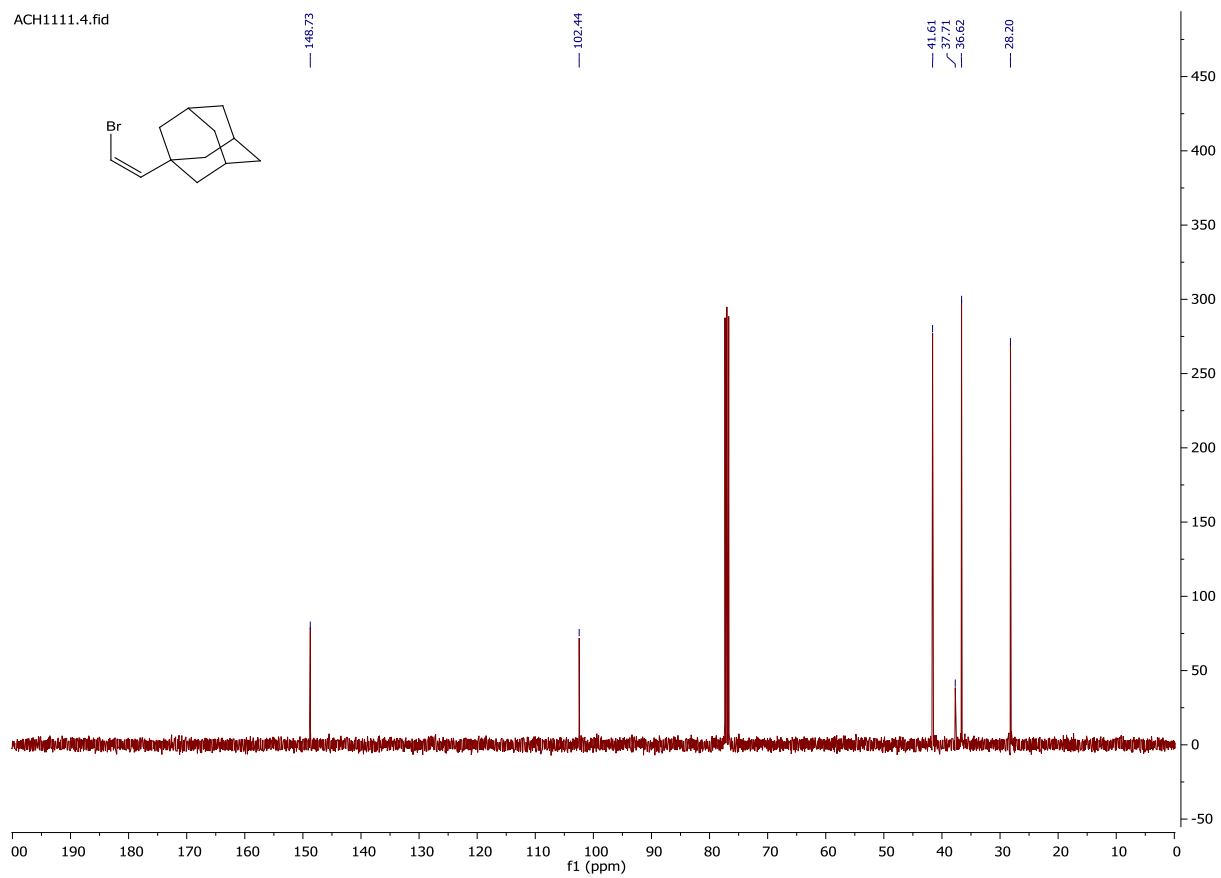


ACH11111.3.fid

# Molecule 6



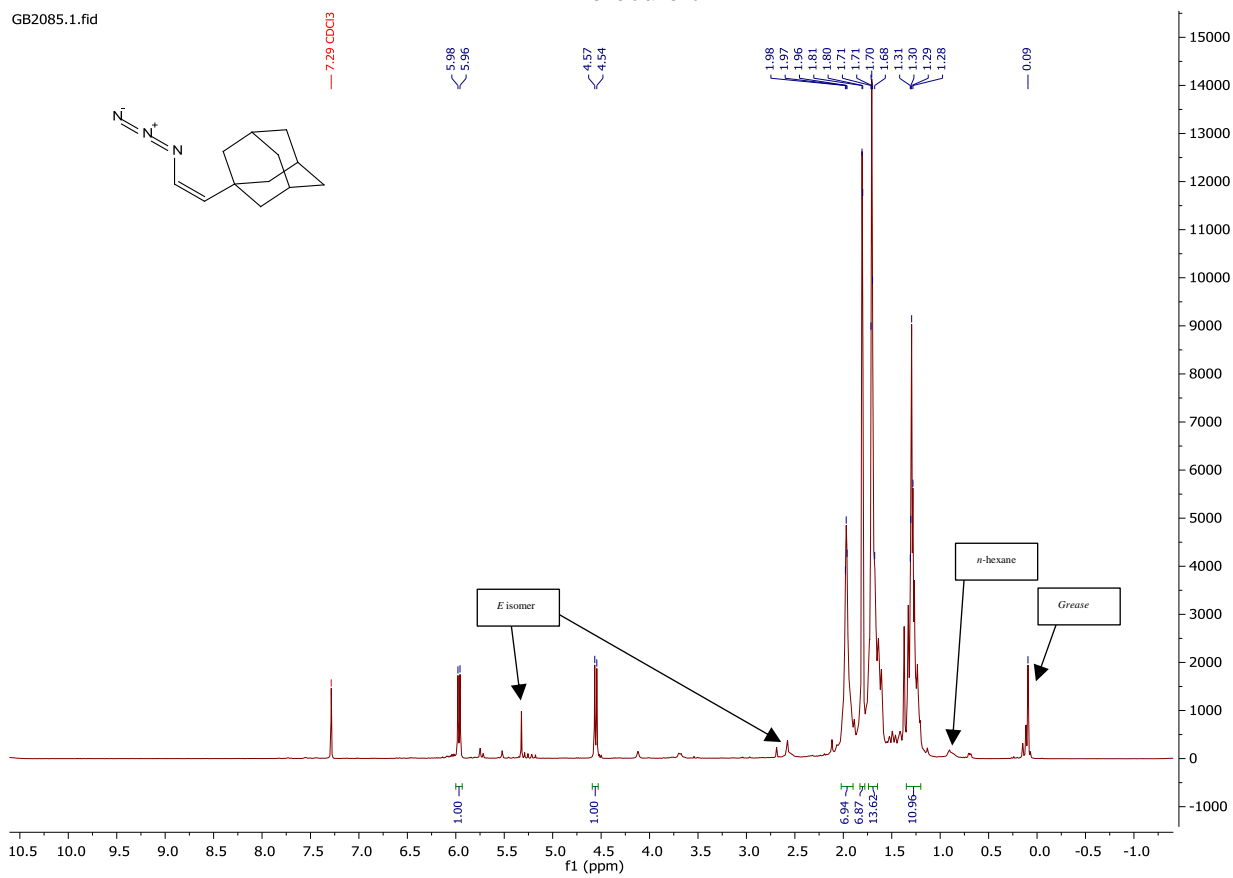
ACH1111.4.fid



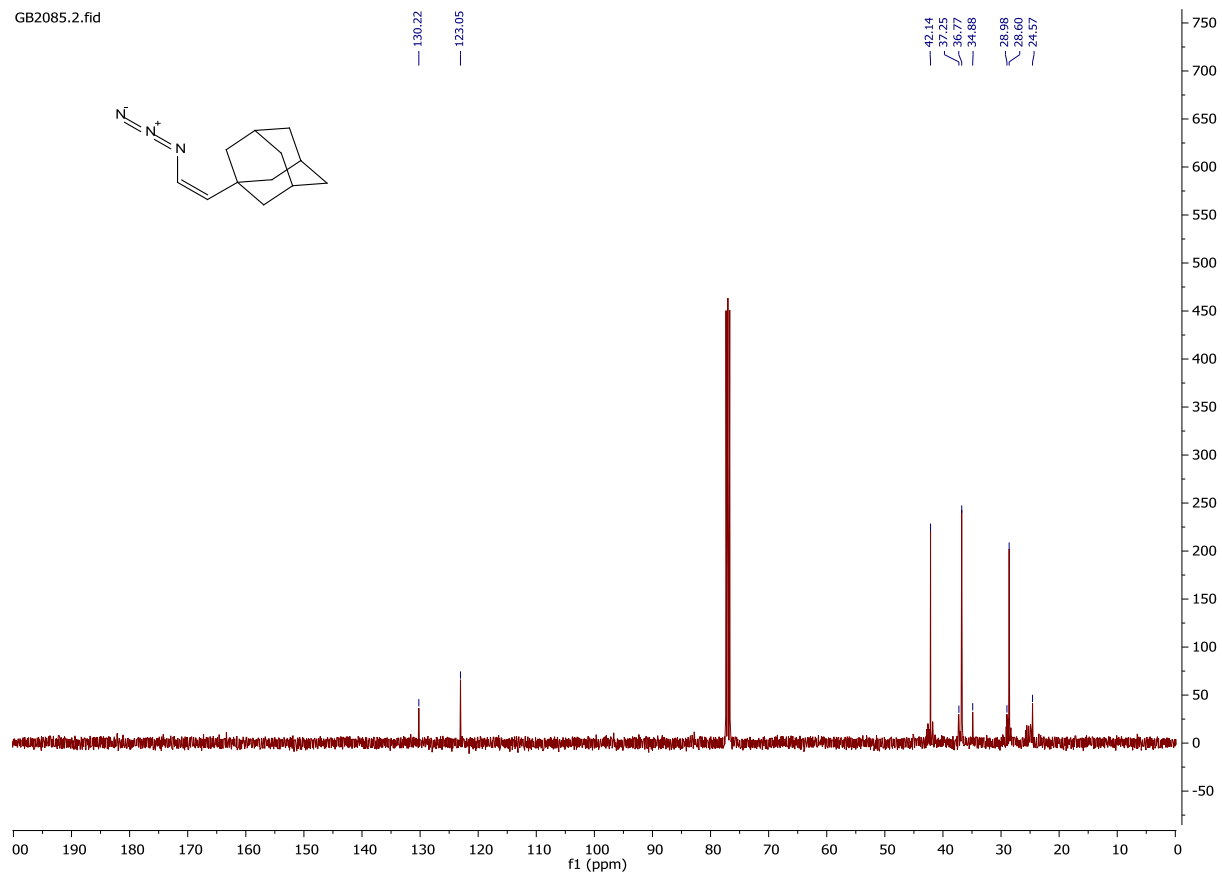
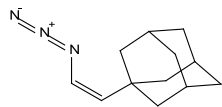


# Molecule 7

GB2085.1.fid

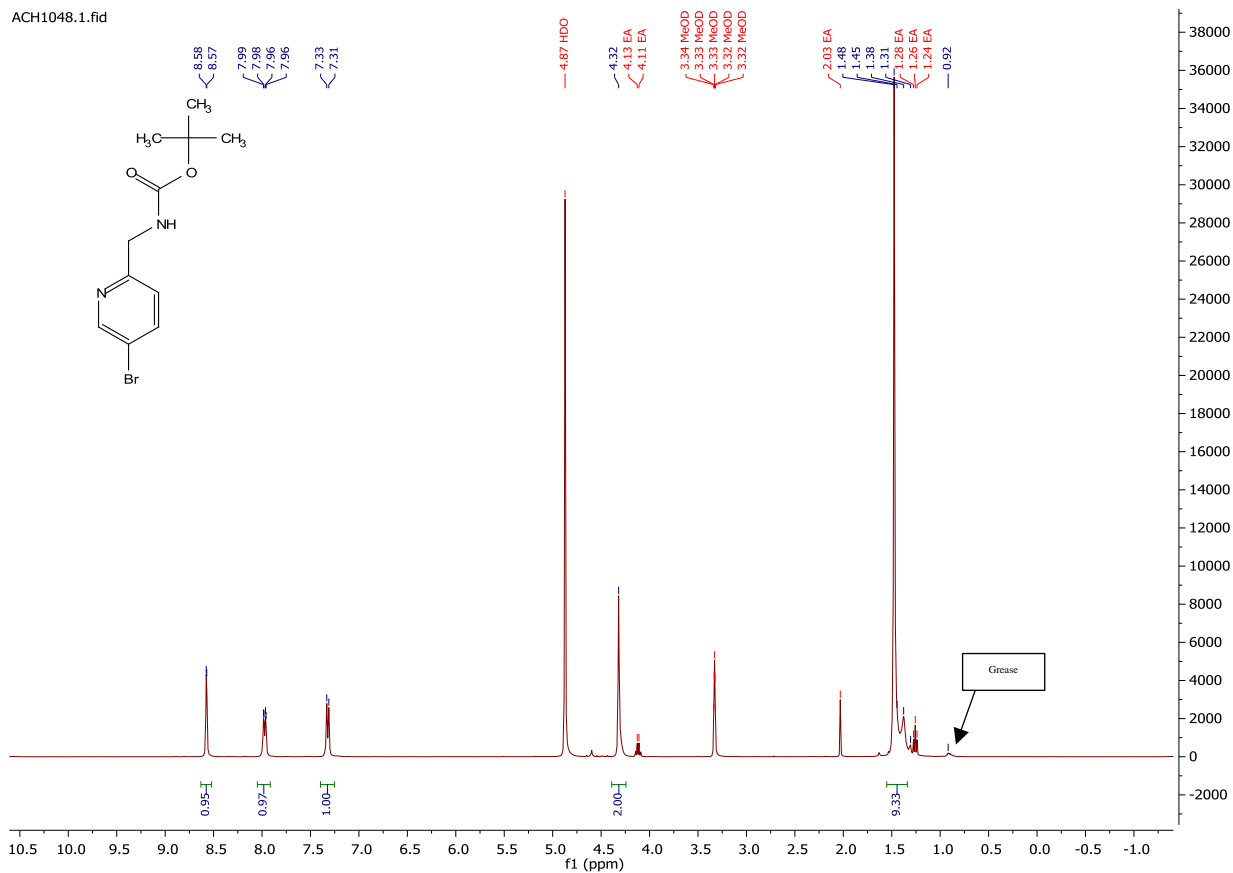


GB2085.2.fid

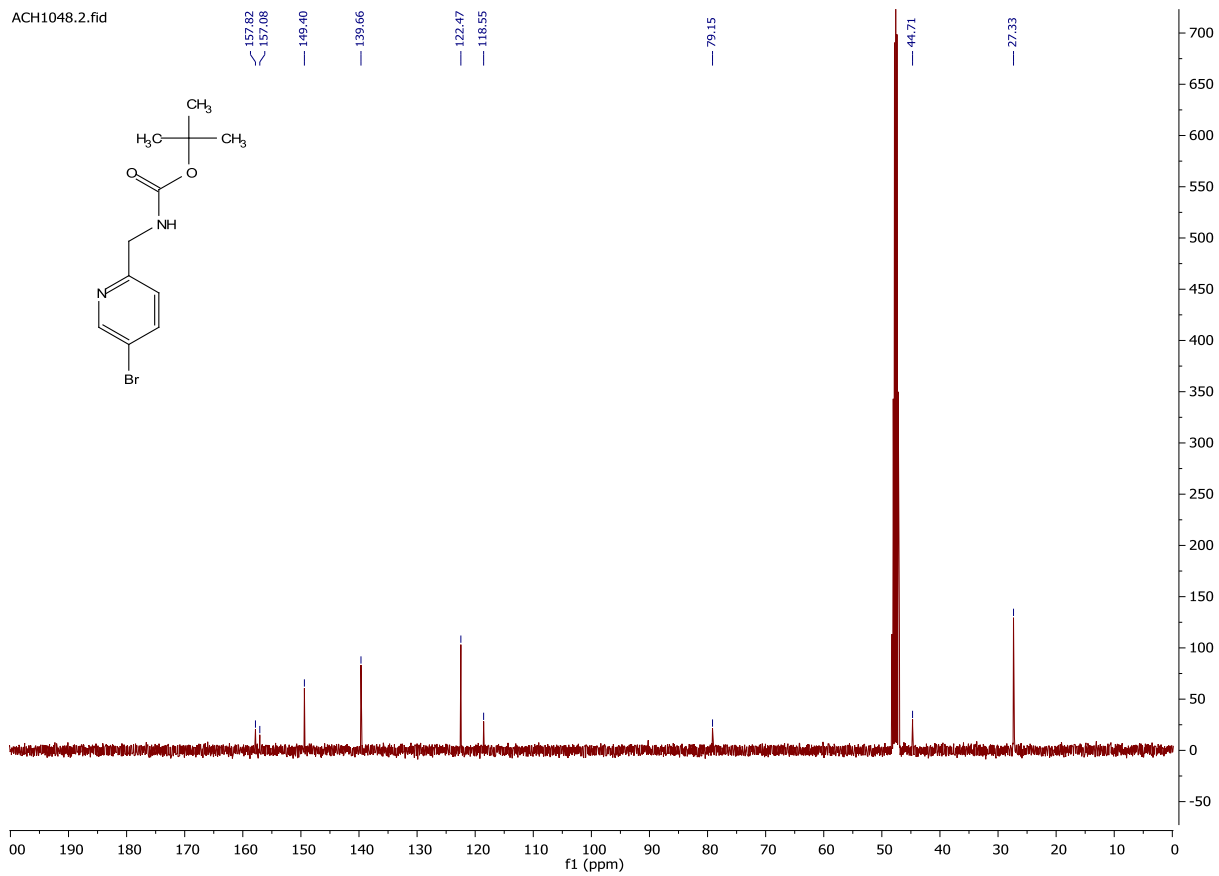


ACH1048.1.fid

### Molecule 13

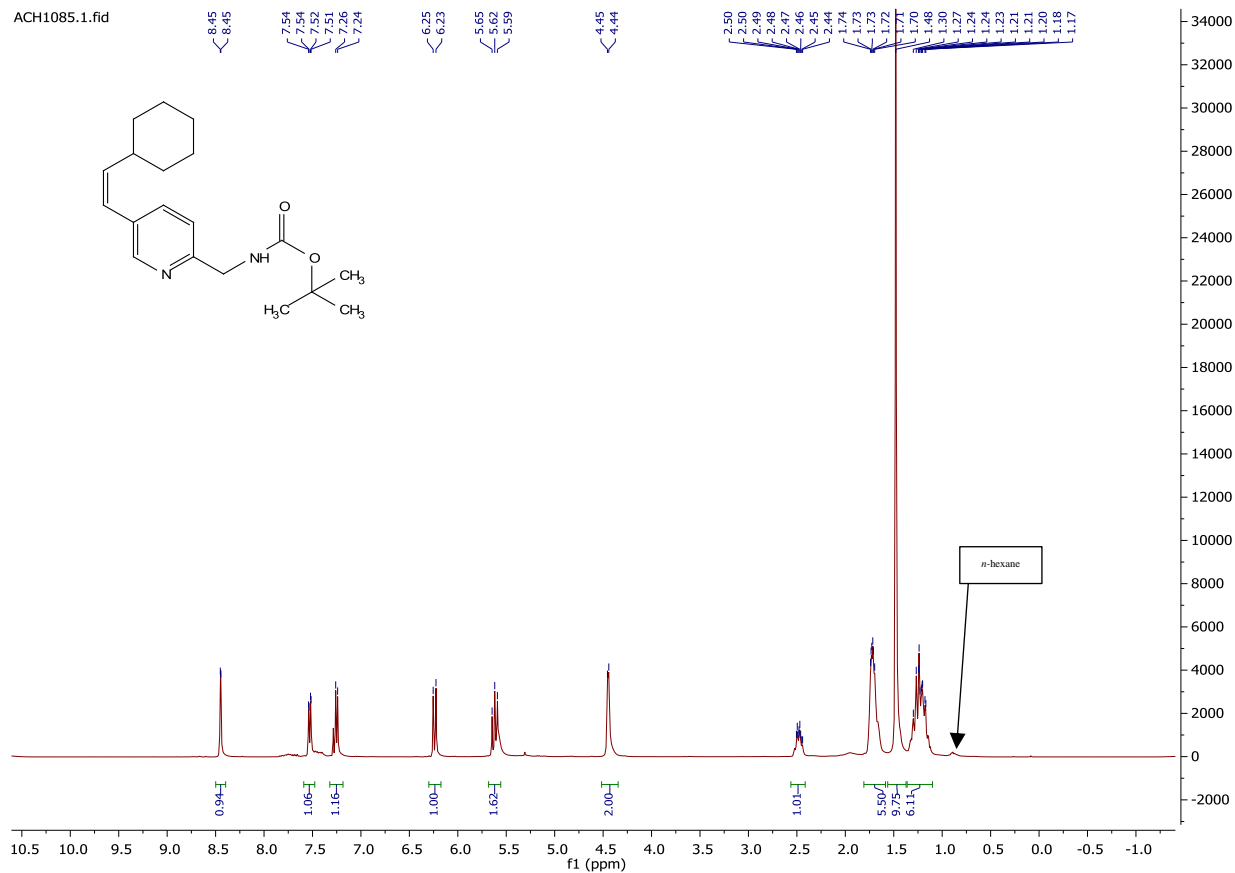


ACH1048.2.fid

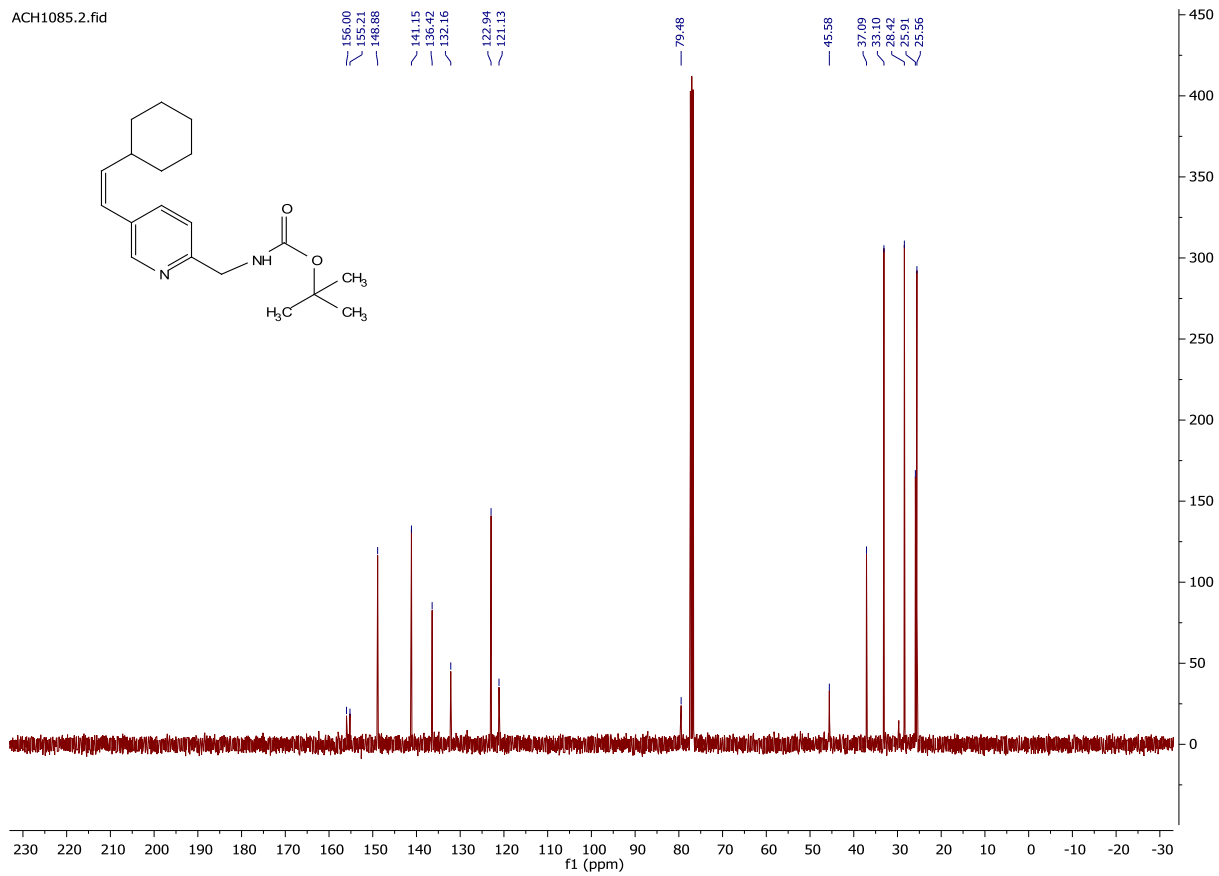


ACH1085.1.fid

### Molecule 14

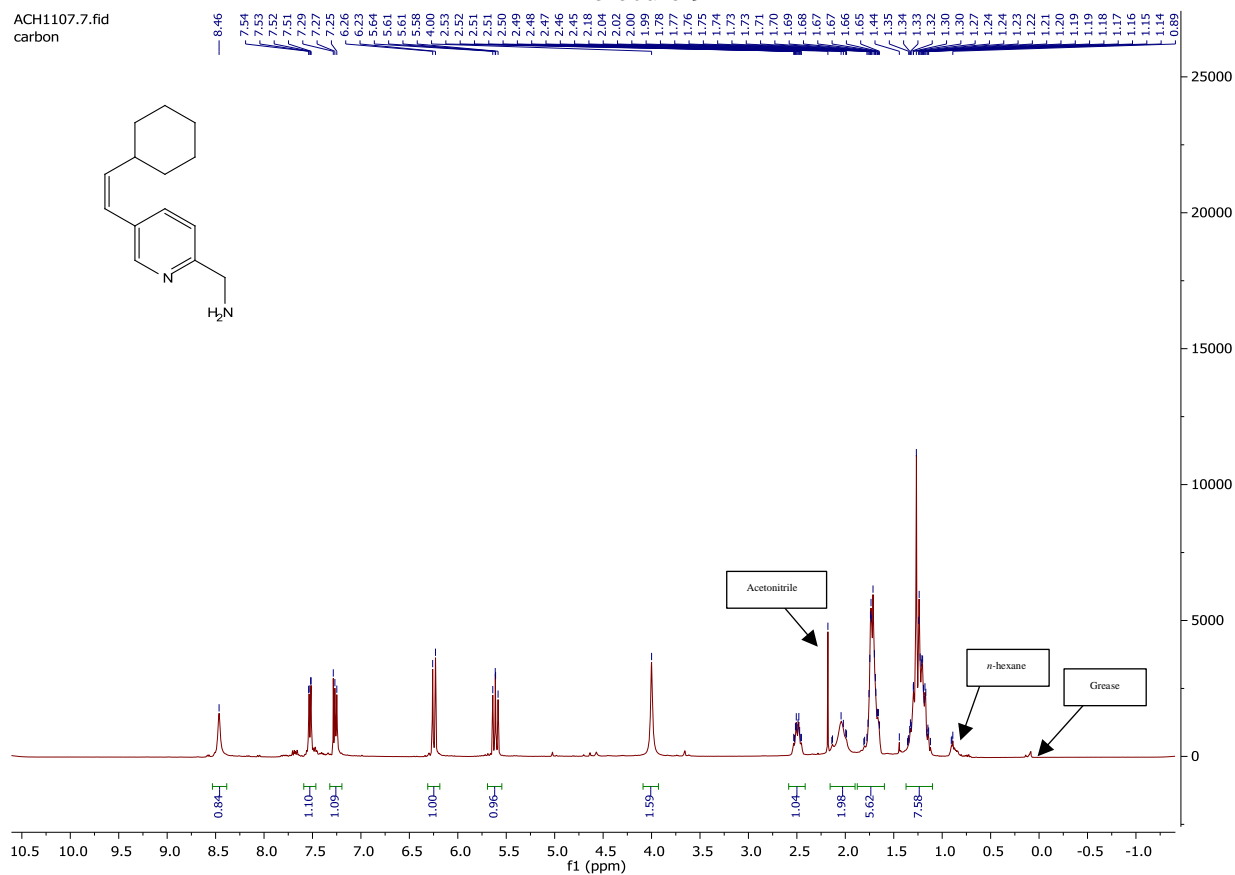


ACH1085.2.fid



# Molecule 9

ACH1107.7.fid  
carbon



ACH1107.8.fid  
carbon

