

Sizable excitonic effects undermining the photocatalytic efficiency of $\beta\text{-Cu}_2\text{V}_2\text{O}_7$

Julia Wiktor,^{*,†} Igor Reshetnyak,[†] Michal Strach,[‡] Mariateresa Scarongella,[‡]
Raffaella Buonsanti,[‡] and Alfredo Pasquarello[†]

Chaire de Simulation à l'Echelle Atomique (CSEA), Ecole Polytechnique Fédérale de Lausanne (EPFL), CH-1015 Lausanne, Switzerland, and Laboratory of Nanochemistry for Energy (LNCE), Ecole Polytechnique Fédérale de Lausanne (EPFL), CH-1951 Sion, Switzerland

E-mail: julia.wiktor@epfl.ch

*To whom correspondence should be addressed

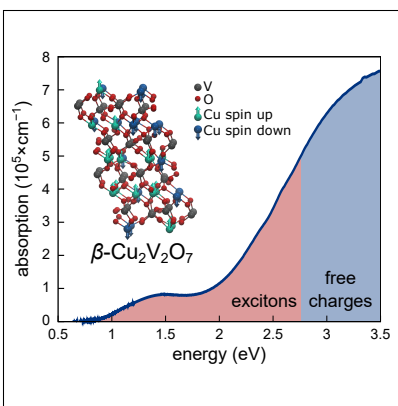
[†]Chaire de Simulation à l'Echelle Atomique (CSEA), Ecole Polytechnique Fédérale de Lausanne (EPFL), CH-1015 Lausanne, Switzerland

[‡]Laboratory of Nanochemistry for Energy (LNCE), Ecole Polytechnique Fédérale de Lausanne (EPFL), CH-1951 Sion, Switzerland

Abstract

Copper vanadates have been proposed as promising photoanodes for water-splitting photoelectrochemical cells, but their performance has recently been shown to be severely limited. To understand this behavior, we study the electronic structure and the optical properties of β - $\text{Cu}_2\text{V}_2\text{O}_7$ both experimentally and computationally. The measured absorption spectrum shows an absorption peak at 1.5 eV followed by the onset of an apparent continuum at 2.26 eV, as generally found for this class of materials. We perform calculations within the framework of the QSGW method and the Bethe-Salpeter equation, while including effects of magnetic ordering, nuclear quantum motion, and thermal vibrations. We demonstrate the occurrence of two kinds of excitons with high binding energies upon optical excitation in β - $\text{Cu}_2\text{V}_2\text{O}_7$, which account for the first absorption peak and the lower edge of the apparent continuum. These results provide one possible explanation for the low photocatalytic efficiencies of copper vanadates, despite the favorable size of their optical band gaps.

Graphical TOC Entry



Photocatalytic water splitting is regarded as a promising way of generating renewable energy.¹⁻³ To achieve high efficiencies of the reaction, it is essential to identify photoabsorbers with optimal properties. As a consequence, extensive efforts are being undertaken to identify suitable candidates. Recent high-throughput searches^{4,5} proposed copper vanadates as a promising class of materials for photocatalytic water splitting, mainly based on the favorable sizes of their band gaps. However, experimental studies have revealed that their performance is severely limited,⁶ even though their measured band gaps of about 2 eV⁵⁻⁷ suggest that cells based on these materials should potentially reach high efficiencies.⁸ For the widely studied vanadate BiVO₄, heterojunctions,⁹ cocatalyst deposition,¹⁰ gradient doping,¹¹ and crystallographic orientation¹² proved to be effective means to achieve photocurrent densities on par with the theoretical limit. Following the same rationale, a few experimental studies have been undertaken to optimize the morphology and doping of copper vanadates.^{13,14} However, such strategies have proven unsuccessful so far.

Though no definite conclusions have been drawn as to the origin of the low measured efficiencies in copper vanadates, it was suggested that high bulk recombination rates might be one of the underlying detrimental factors.¹⁴ Furthermore, a number of studies on copper vanadates have observed an unusual feature in the absorption spectrum, apparently unrelated to the band structure. This feature has remained poorly understood, possibly related to a localized ligand field excitation at the Cu²⁺ cation.¹⁵ While the corresponding states are generally not assumed to produce mobile charge carriers,¹⁵ this limited understanding nevertheless calls for a more thorough investigation of the optical properties of copper vanadates.

In the present study, we investigate, both computationally and experimentally, the electronic structure and the optical properties of β -Cu₂V₂O₇ (ziesite phase), as a representative of the copper vanadate class. We demonstrate that strong excitonic effects account for both the first peak in the absorption spectrum and the onset of the apparent continuum. These results explain the limited photocatalytic efficiency of photoelectrochemical cells based on

copper vanadates.

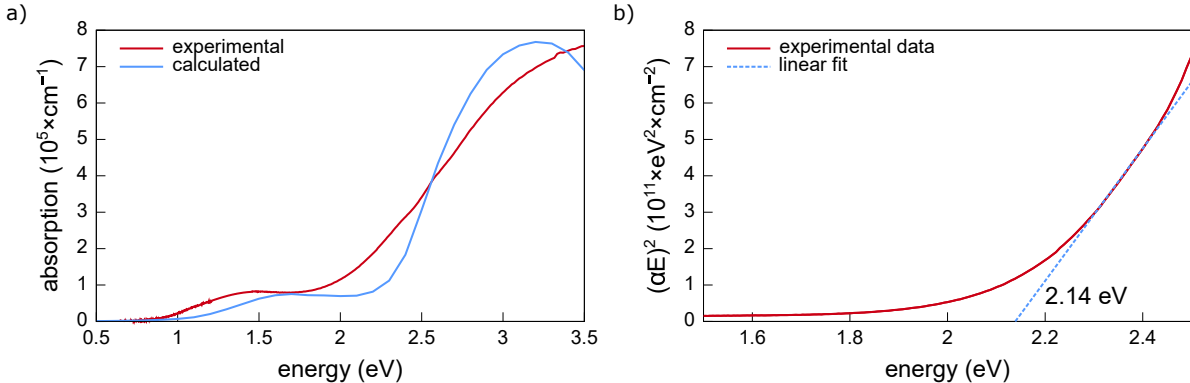


Figure 1: a) Experimental absorption coefficient compared to that achieved through our computational scheme. b) Experimental Tauc plot.

We first determine experimentally the optical absorption spectrum of $\beta\text{-Cu}_2\text{V}_2\text{O}_7$ using UV-vis spectroscopy. Thin films of $\text{Cu}_2\text{V}_2\text{O}_7$ were prepared using the seeded growth approach. The sample composition was verified by X-ray diffraction measurements, as detailed in the Supporting Information (SI). The corresponding absorption spectrum is given in Fig. 1 a). The raw data contained severe noise around 1.4 eV due to a change of lamps in our dual light source spectrometer during the measurement (cf. SI), and the curve was treated with a locally weighted scatterplot smoothing (LOESS) method. Additionally, the background was removed to ease comparison. The measured absorption spectrum shows a first peak at about 1.5 eV and the onset of an apparent continuum close to 2 eV. These features are generally observed in the class of copper oxides with Cu^{2+} ions,^{5-7,15,16} corroborating their intrinsic origin. From the corresponding Tauc plot [Fig. 1 b)], we estimate a value of 2.14 eV for this onset.

While steady-state spectra are routinely used to evaluate the absorption properties of semiconductors, they do not provide any information about the nature of the generated charge carriers. To evaluate the electronic properties underlying the observed features, we resort to a first-principles modeling scheme. From the computational point of view, modeling a complex material such as a copper vanadate requires special care. First, the occurrence

of the 2+ oxidation state of Cu leads to the presence of unpaired electrons and to magnetic ordering. In particular, β -Cu₂V₂O₇ is known to be paramagnetic at room temperature,^{17,18} which presents a challenge for first-principles modeling.^{19,20} Moreover, it has recently been observed that the electronic structure of an analogous vanadate, BiVO₄, is strongly affected by thermal vibrations and nuclear quantum motions.²¹ These effects might similarly influence the electronic structure of β -Cu₂V₂O₇, and thus need to be properly accounted for in the calculations. Finally, the full treatment of the optical properties of the material requires the consideration of excitonic effects, which generally involve advanced schemes such as the Bethe-Salpeter equation.

We begin the theoretical analysis by studying the electronic structure of β -Cu₂V₂O₇ in the antiferromagnetic ground state at 0 K.²² We perform calculations within many-body perturbation theory through the use of the self-consistent QSG \tilde{W} technique. An efficient exchange-correlation kernel is included to account for vertex corrections, as described in Ref. 23. This approach has been shown to yield band gaps in very good agreement with experiment.^{23,24} We refer the reader to the SI for computational details. In Fig. 2, we show the density of states (DOS) as calculated with various electronic structure schemes. The top of the valence band results from a mixture of O 2*p*, V 3*d*, and Cu 3*d* orbitals, while the lowest conduction band states essentially consist of Cu 3*d* states and give rise to an isolated peak in the DOS. This is consistent with the *d*⁹ occupation of the Cu 3*d* states in the Cu²⁺ ion. The QSG \tilde{W} calculation gives an indirect band gap of 3.56 eV, whereas the direct band gap is found to lie slightly higher at 3.63 eV. The semilocal Perdew-Burke-Ernzerhof (PBE) functional yields a much lower band gap of 0.72 eV, but also a much larger separation between the isolated peak of empty Cu 3*d* states and the higher lying conduction band states. We note that the QSG \tilde{W} band gap at 0 K (3.56 eV) is much higher than the onset at 2.14 eV of the apparent continuum in the experimental spectrum (Fig. 1). However, following recent work on BiVO₄,²¹ one expects additional effects, such as nuclear quantum motions, thermal vibrations, and excitonic effects, to lower the theoretical estimate. To address larger

supercells preserving the accuracy of the $\text{QSG}\tilde{W}$ scheme, it is convenient to make use of a hybrid functional. For this purpose, we use the Perdew-Burke-Ernzerhof (PBE0) hybrid functional²⁵ with a mixing parameter α set to 0.225 in order to closely reproduce the DOS calculated in $\text{QSG}\tilde{W}$ (see Fig. 2).

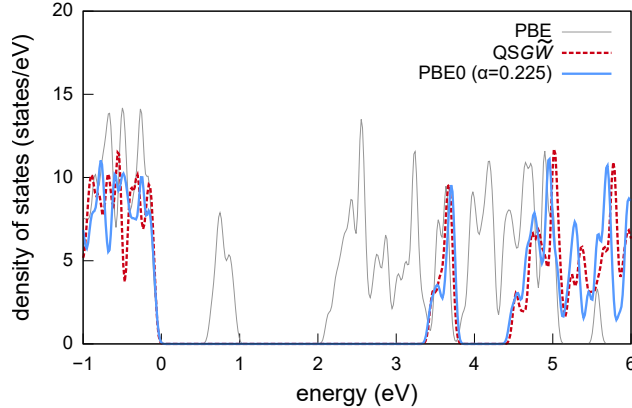


Figure 2: Density of states for the antiferromagnetic ground state of $\beta\text{-Cu}_2\text{V}_2\text{O}_7$ at 0 K, as calculated using the PBE, $\text{QSG}\tilde{W}$, and $\text{PBE0}(\alpha = 0.225)$ methods. The energies are referred to the top of the valence band for each calculation. We use a broadening of 0.05 eV to generate the plots.

$\beta\text{-Cu}_2\text{V}_2\text{O}_7$ has been shown to be paramagnetic at room temperature.^{17,18} To describe the paramagnetic state within density functional theory, we use the Disordered Local Moment model,^{26,27} in which a random distribution of atomic magnetic moments is assumed. Here, we consider only collinear magnetic moments, which is appropriate in this case as spin-orbit coupling is not dominant.²⁸ To achieve a random distribution of spin-up and spin-down Cu atoms in $\beta\text{-Cu}_2\text{V}_2\text{O}_7$, we generate a special quasirandom structure (SQS)²⁹ containing 176 atoms in the supercell, as represented in Fig. 3 a). This cell contains 32 Cu atoms, 16 of them with spin up and 16 with spin down. When we compare the density of states of the antiferromagnetic and paramagnetic states calculated with the PBE functional at a temperature of 0 K, we find that the magnetic structure of the paramagnetic state leads to a broadening of the width of the empty Cu $3d$ states, consequently reducing the band gap by 0.18 eV.

Next, we study the influence of thermal disorder and nuclear quantum effects (NQEs) on

the band gap of $\beta\text{-Cu}_2\text{V}_2\text{O}_7$. We carry out path-integral molecular dynamics (PIMD) at the PBE level for 5 ps at 300 K, within the NVT ensemble. The simulation is performed for the paramagnetic supercell containing 176 atoms. To achieve an accurate description of the electronic structure, we perform subsequent calculations with the hybrid functional PBE0 ($\alpha = 0.225$)²⁵ on 300 configurations regularly spaced in time from the PIMD trajectory. The density of states calculated for the selected paramagnetic configurations is shown in Fig. 3 b), where it is compared to the corresponding DOS of the antiferromagnetic ground state at 0 K, achieved with the same hybrid functional but neglecting NQEs. From the DOS, we extract a room-temperature fundamental band gap of 2.76 eV, smaller than the value at 0 K (3.56 eV) by almost 0.8 eV. We note that in the DOS of the paramagnetic structure at 300 K the empty Cu 3*d* states merge with the higher lying conduction states. This qualitative aspect is missed in the semilocal description at 0 K, undermining the reliability of such a scheme in high-throughput searches of new materials.

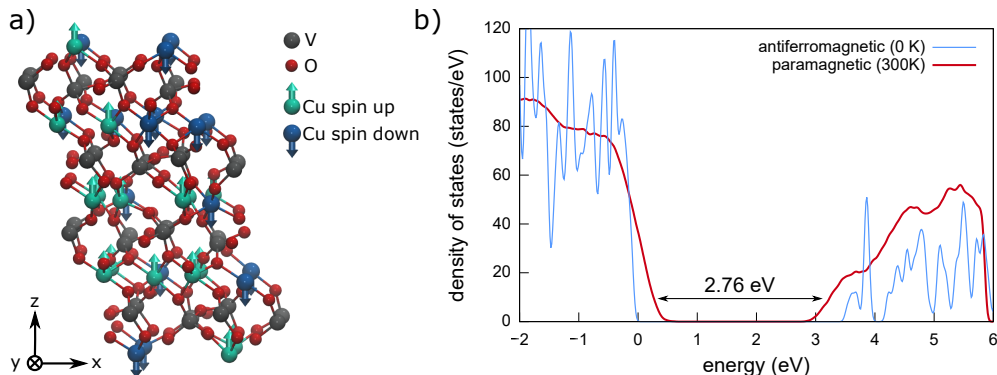


Figure 3: a) Paramagnetic supercell of $\beta\text{-Cu}_2\text{V}_2\text{O}_7$ containing a quasirandom distribution of spin-up (green) and spin-down (blue) Cu atoms, based on a special quasirandom structure (SQS). b) Density of states of $\beta\text{-Cu}_2\text{V}_2\text{O}_7$ at 300 K, in the paramagnetic state. The calculation uses the PBE0($\alpha = 0.225$) hybrid functional and includes NQEs. For comparison, we also show the DOS of the antiferromagnetic ground state, calculated with the same functional at 0 K but neglecting NQEs. The DOS are aligned using the Cu 3*s* states. The energies are referred to the top of the valence band of the antiferromagnetic state. We use a broadening of 0.05 eV to generate the plots.

Optical absorption spectra can be used to extract the optical band gap of $\beta\text{-Cu}_2\text{V}_2\text{O}_7$. This property differs from the fundamental band gap by the electron-hole interaction.

To account for excitonic effects, we solve the Bethe-Salpeter equation (BSE)³⁰ using the ABINIT³¹⁻³³ and DP-CODE³⁴ packages. We calculate the absorption spectra on 20 instantaneous configurations from the PIMD simulation. In this way, our description accounts for the paramagnetic spin structure, the thermal vibrations, and the nuclear quantum motions. Since the treatment of the 176-atom cell is demanding, we use wave functions achieved at the PBE level and a scissor operator of 2.8 eV, which is chosen to reproduce the QSGW band gap of the antiferromagnetic ground state at 0 K. In the SI, the validity of this approach is analyzed and the computational details are provided.

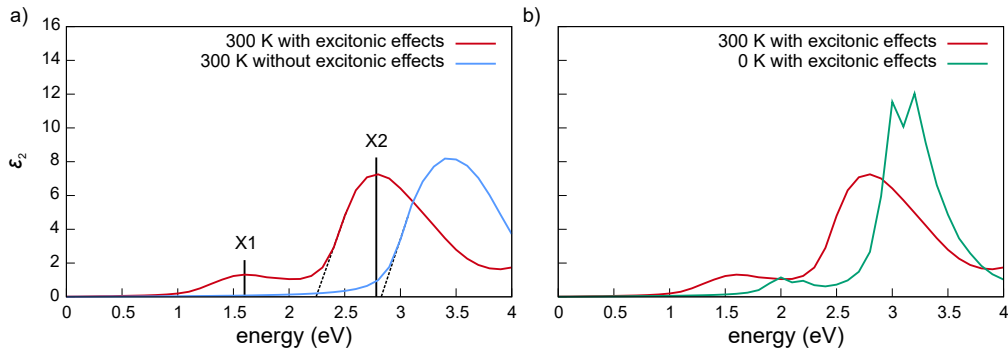


Figure 4: a) Comparison between the imaginary part of the dielectric function (ε_2) calculated with the BSE for the paramagnetic ground state at 300 K (red) and the corresponding result obtained in the RPA (blue). The results include NQEs. b) Comparison between the imaginary part of the dielectric function calculated with the BSE for the paramagnetic ground state at 300 K (including NQEs) and the result achieved at 0 K for the antiferromagnetic state (green).

We compare in Fig. 4 a) the imaginary part of the dielectric function ε_2 calculated by solving the BSE equation with that obtained within the random phase approximation (RPA), which does not include excitonic effects. The RPA curve shows a peak composed of excitations to Cu 3d states at 3.5 eV with an onset at 2.76 eV, consistent with the fundamental band gap in Fig. 3. The comparison with the BSE result shows that the excitonic effects are sizable in β -Cu₂V₂O₇. We distinguish two peaks below the fundamental band gap at 1.6 and 2.7 eV, denoted as X1 and X2, respectively. Since these excitonic features correspond to excitations to Cu 3d states, we refer their peak positions with respect to the RPA peak and

find respective binding energies of 1.9 and 0.8 eV. We determine the onset of the feature X2 to be at 2.26 eV. This shows a sizable shift of 0.5 eV with respect to the RPA one, in accord with shifts of similar size observed in other complex copper oxides, such as CuAlO_2 .^{35,36} This can be considered as the minimal energy that needs to be supplied to overcome the electron-hole binding energy and to achieve mobile carriers. We remark that there is a nontrivial interplay between atomic displacements and excitonic effects. This can be inferred from Fig. 4 b), where ϵ_2 calculated for the antiferromagnetic ground state at 0 K and for the paramagnetic state at 300 K are compared. A simplistic approach consisting in applying the broadening due to atomic displacements observed in Fig. 3 to the result at 0 K could not give the correct one at 300 K.

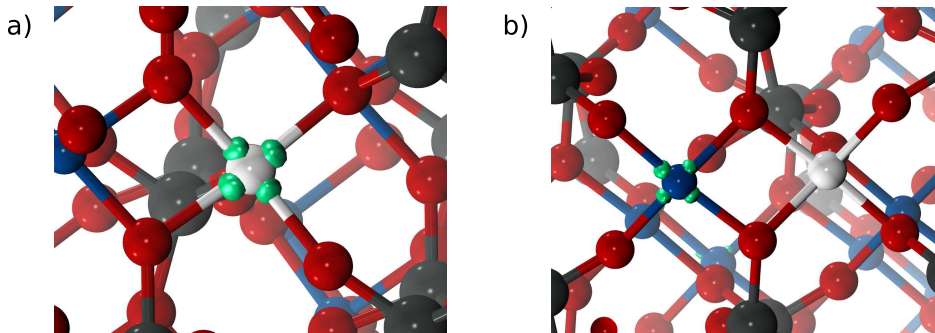


Figure 5: Representation of the wave functions of the lowest lying excitonic features (a) X1 and (b) X2 in $\beta\text{-Cu}_2\text{V}_2\text{O}_7$. Cu, O, and V atoms are in blue, red, and gray, respectively. To illustrate the two-particle wave functions of the excitons, the spatial coordinate of the hole has been fixed at the Cu atom marked in white, while the orbital of the electron wave function is displayed by a green contour.

To analyze the nature of these two excitonic states, we plot the corresponding wave functions in Fig. 5. We observe that the first peak X1 is dominated by on-site Cu-Cu transitions between occupied and empty $3d$ states. These transitions are forbidden in a spherically symmetric environment, but acquire oscillation strength due to the crystal field, which is already present at 0 K in $\beta\text{-Cu}_2\text{V}_2\text{O}_7$ and is further enhanced at 300 K. The excitonic feature X2, at higher energies, corresponds to excitations in which the electron and the hole reside in Cu $3d$ orbitals of neighboring atoms. Hence, the two excitonic features bear from

transitions to empty Cu 3*d*, as one could infer from the low-lying conduction band states in the DOS (Fig. 3). The broadening due to the atomic motion is then responsible for merging the X2 feature with transitions to higher lying conduction band states.

To make a comparison with experiment, we construct the frequency-dependent absorption coefficient $\alpha(\omega)$

$$\alpha(\omega) = \frac{4\pi\omega}{c} \sqrt{\frac{\sqrt{\varepsilon_1^2(\omega) + \varepsilon_2^2(\omega)} - \varepsilon_1(\omega)}{2}}, \quad (1)$$

where c is the speed of light, ε_2 the imaginary part of the calculated dielectric function of the paramagnetic state at 300 K (given in Fig. 4) and ε_1 the corresponding real part (given in the SI). The calculated $\alpha(\omega)$ is compared to our measured absorption spectrum in Fig. 1 a). The agreement between theory and experiment is remarkable, the peak position of X1 and the onset of X2 being both well reproduced. This confirms the validity of the underlying theoretical scheme and highlights the importance of all steps taken to achieve the present description.

The good agreement between theory and experiment also allows us to draw important conclusions as to the interpretation of the features in the absorption spectrum of β -Cu₂V₂O₇ and to the general suitability of this class of materials in photocatalytic devices. Our results indicate that the peculiar feature associated with the lowest-energy transitions originates from on-site transitions at the Cu²⁺ sites, corresponding to excitons of high binding energy. However, since the weak absorption in correspondence to this feature has not been considered in efficiency assessments, its origin is inconsequential to the estimation of the performance of this semiconductor. As for the higher lying excitations, the present results indicate that the apparent onset of the continuum actually corresponds to transitions to bound excitons in which the photogenerated charges are localized on neighboring Cu sites. Therefore, these charges are not available for catalytic reactions at the surface, unless an additional energy of at least 0.5 eV is supplied to split the charges. This value is much higher than found for other metal oxides, such as BiVO₄, for which an excitonic binding energy of only 0.11 eV

has been calculated,²¹ and undermines the viability of using β -Cu₂V₂O₇ in a PEC cell. The significant excitonic binding energy might contribute to the high onset potentials measured for pristine Cu-V-O samples.^{7,13}

To conclude, we obtained an exceptionally good agreement between the measured and modeled optical absorption spectra of β -Cu₂V₂O₇, a representative of the class of copper vanadate compounds, presently widely investigated in view of photocatalytical applications. Our theoretical modeling relied on state-of-the-art computational methods, including path-integral molecular dynamics, self-consistent many-body perturbation theory, and the Bethe-Salpeter equation. Our results highlight the occurrence of excitons of high binding energy in the absorption spectrum as one of the reasons undermining the performance of β -Cu₂V₂O₇ as a photoanode in a water splitting cell. As this limitation directly stems from the properties of the Cu²⁺ ions, these conclusions generalize to the entire class of copper vanadates. Hence, our work suggests that electron-hole interactions between photogenerated carriers should be systematically accounted for in future searches for new materials for water splitting.

Acknowledgement

The authors acknowledge financial support from the Swiss National Science Foundation (SNSF) (Grant No. 200020-172524). This work has been realized in relation to the National Center of Competence in Research (NCCR) “Materials’ Revolution: Computational Design and Discovery of Novel Materials (MARVEL)” of the SNSF. We used computational resources of CSCS and SCITAS-EPFL.

Supporting Information Available

Experimental details on sample preparation, X-ray diffraction and UV-vis spectroscopy measurements, computational details on QSGW and BSE calculations, and additional Ref. [37–46](#)

References

- (1) Fujishima, A.; Honda, K. Electrochemical Photolysis of Water at a Semiconductor Electrode. *Nature* **1972**, *238*, 37–38.
- (2) Maeda, K.; Teramura, K.; Lu, D.; Takata, T.; Saito, N.; Inoue, Y.; Domen, K. Photocatalyst Releasing Hydrogen from Water. *Nature* **2006**, *440*, 295–295.
- (3) Walter, M. G.; Warren, E. L.; McKone, J. R.; Boettcher, S. W.; Mi, Q.; Santori, E. A.; Lewis, N. S. Solar Water Splitting Cells. *Chem. Rev.* **2010**, *110*, 6446.
- (4) Zhou, L.; Yan, Q.; Shinde, A.; Guevarra, D.; Newhouse, P. F.; Becerra-Stasiewicz, N.; Chatman, S. M.; Haber, J. A.; Neaton, J. B.; Gregoire, J. M. High Throughput Discovery of Solar Fuels Photoanodes in the CuO–V₂O₅ System. *Adv. Energy Mater.* **2015**, *5*.
- (5) Yan, Q.; Yu, J.; Suram, S. K.; Zhou, L.; Shinde, A.; Newhouse, P. F.; Chen, W.; Li, G.; Persson, K. A.; Gregoire, J. M. et al. Solar Fuels Photoanode Materials Discovery by Integrating High-Throughput Theory and Experiment. *Proc. Natl. Acad. Sci.* **2017**, *114*, 3040–3043.
- (6) Jiang, C.-M.; Segev, G.; Hess, L. H.; Liu, G.; Zaborski, G.; Toma, F. M.; Cooper, J. K.; Sharp, I. D. Composition-Dependent Functionality of Copper Vanadate Photoanodes. *ACS Appl. Mater. Interfaces* **2018**, 10627–10633.
- (7) Gadiyar, C.; Strach, M.; Schouwink, P.; Loiudice, A.; Buonsanti, R. Chemical Transformations at the Nanoscale: Nanocrystal-Seeded Synthesis of β -Cu₂V₂O₇ with Enhanced Photoconversion Efficiencies. *Chem. Sci.* **2018**, 5658–5665.
- (8) Werner, J. H.; Kolodinski, S.; Queisser, H. J. Novel Optimization Principles and Efficiency Limits for Semiconductor Solar Cells. *Phys. Rev. Lett.* **1994**, *72*, 3851.

- (9) Zhou, Y.; Zhang, L.; Lin, L.; Wygant, B. R.; Liu, Y.; Zhu, Y.; Zheng, Y.; Mullins, C. B.; Zhao, Y.; Zhang, X. et al. Highly Efficient Photoelectrochemical Water Splitting from Hierarchical $\text{WO}_3/\text{BiVO}_4$ Nanoporous Sphere Arrays. *Nano Letters* **2017**, *17*, 8012–8017, PMID: 29185764.
- (10) Abdi, F. F.; Firet, N.; van de Krol, R. Efficient BiVO_4 Thin Film Photoanodes Modified with Cobalt Phosphate Catalyst and W-doping. *Chem. Cat. Chem.* **5**, 490–496.
- (11) Abdi, F.; Han, L.; Smets, A.; Zeman, M.; Dam, B.; van de Krol, R. Efficient Solar Water Splitting by Enhanced Charge Separation in a Bismuth Vanadate-Silicon Tandem Photoelectrode. *Nat. Commun.* **2013**, *4*, 1–7.
- (12) Han, H. S.; Shin, S.; Kim, D. H.; Park, I. J.; Kim, J. S.; Huang, P.-S.; Lee, J.-K.; Cho, I. S.; Zheng, X. Boosting the Solar Water Oxidation Performance of a BiVO_4 Photoanode by Crystallographic Orientation Control. *Energy Environ. Sci.* **2018**, *11*, 1299–1306.
- (13) Guo, W.; Chemelewski, W. D.; Mabayoje, O.; Xiao, P.; Zhang, Y.; Mullins, C. B. Synthesis and Characterization of CuV_2O_6 and $\text{Cu}_2\text{V}_2\text{O}_7$: Two Photoanode Candidates for Photoelectrochemical Water Oxidation. *J. Phys. Chem. C* **2015**, *119*, 27220–27227.
- (14) Lumley, M. A.; Choi, K.-S. Investigation of Pristine and (Mo, W)-Doped $\text{Cu}_1\text{V}_6\text{O}_6$ for Use as Photoanodes for Solar Water Splitting. *Chem. Mater.* **2017**, *29*, 9472–9479.
- (15) Jiang, C.-M.; Farmand, M.; Wu, C. H.; Liu, Y.-S.; Guo, J.; Drisdell, W. S.; Cooper, J. K.; Sharp, I. D. Electronic Structure, Optoelectronic Properties, and Photoelectrochemical Characteristics of $\gamma\text{-Cu}_3\text{V}_2\text{O}_8$ Thin Films. *Chem. Mater.* **2017**, *29*, 3334–3345.
- (16) Lhermitte, C. R.; Bartlett, B. M. Advancing the Chemistry of CuWO_4 for Photoelectrochemical Water Oxidation. *Acc. Chem. Res.* **2016**, *49*, 1121–1129.

- (17) Ponomarenko, L.; Vasil'ev, A.; Antipov, E.; Velikodny, Y. A. Magnetic Properties of $\text{Cu}_2\text{V}_2\text{O}_7$. *Physica B* **2000**, *284*, 1459–1460.
- (18) He, Z.; Ueda, Y. Paramagnetic Anisotropy and Spin-Flop Transition in Single Crystals of the Quasi-One-Dimensional System $\beta\text{-Cu}_2\text{V}_2\text{O}_7$. *Phys. Rev. B* **2008**, *77*, 052402.
- (19) Dorado, B.; Garcia, P. First-principles DFT+ U modeling of actinide-based alloys: Application to paramagnetic phases of UO_2 and (U, Pu) mixed oxides. *Phys. Rev. B* **2013**, *87*, 195139.
- (20) Abrikosov, I. A.; Ponomareva, A.; Steneteg, P.; Barannikova, S.; Alling, B. Recent Progress in Simulations of the Paramagnetic State of Magnetic Materials. *Curr. Opin. Solid State Mater. Sci.* **2016**, *20*, 85–106.
- (21) Wiktor, J.; Reshetnyak, I.; Ambrosio, F.; Pasquarello, A. Comprehensive Modeling of the Band Gap and Absorption Spectrum of BiVO_4 . *Phys. Rev. Mater.* **2017**, *1*, 022401.
- (22) Yashima, M.; Suzuki, R. O. Electronic Structure and Magnetic Properties of Monoclinic $\beta\text{-Cu}_2\text{V}_2\text{O}_7$: A GGA+U Study. *Phys. Rev. B* **2009**, *79*, 125201.
- (23) Chen, W.; Pasquarello, A. Accurate Band Gaps of Extended Systems Via Efficient Vertex Corrections in *GW*. *Phys. Rev. B* **2015**, *92*, 041115.
- (24) Wiktor, J.; Rothlisberger, U.; Pasquarello, A. Predictive Determination of Band Gaps of Inorganic Halide Perovskites. *J. Phys. Chem. Lett.* **2017**, *8*, 5507–5512.
- (25) Perdew, J. P.; Ernzerhof, M.; Burke, K. Rationale for Mixing Exact Exchange with Density Functional Approximations. *J. Chem. Phys.* **1996**, *105*, 9982–9985.
- (26) Hubbard, J. The Magnetism of Iron. *Phys. Rev. B* **1979**, *19*, 2626.
- (27) Hasegawa, H. Single-site Spin Fluctuation Theory of Itinerant-Electron Systems with Narrow Bands. II. Iron and nickel. *J. Phys. Soc. Jpn* **1980**, *49*, 963–971.

- (28) Gyorffy, B.; Pindor, A.; Staunton, J.; Stocks, G.; Winter, H. A First-Principles Theory of Ferromagnetic Phase Transitions in Metals. *J. Phys. F: Met. Phys.* **1985**, *15*, 1337.
- (29) Zunger, A.; Wei, S.-H.; Ferreira, L.; Bernard, J. E. Special Quasirandom Structures. *Phys. Rev. Lett.* **1990**, *65*, 353.
- (30) Cutkosky, R. E. Solutions of a Bethe-Salpeter Equation. *Phys. Rev.* **1954**, *96*, 1135–1141.
- (31) Gonze, X.; Beuken, J.-M.; Caracas, R.; Detraux, F.; Fuchs, M.; Rignanese, G.-M.; Sindic, L.; Verstraete, M.; Zerah, G.; Jollet, F. et al. First-Principles Computation of Material Properties: The ABINIT Software Project. *Comput. Mater. Sci.* **2002**, *25*, 478 – 492.
- (32) Gonze, X.; Amadon, B.; Anglade, P.-M.; Beuken, J.-M.; Bottin, F.; Boulanger, P.; Bruneval, F.; Caliste, D.; Caracas, R.; Côté, M. et al. ABINIT: First-Principles Approach to Material and Nanosystem Properties. *Comput. Phys. Commun.* **2009**, *180*, 2582 – 2615.
- (33) Gonze, X.; Jollet, F.; Araujo, F. A.; Adams, D.; Amadon, B.; Applencourt, T.; Audouze, C.; Beuken, J.-M.; Bieder, J.; Bokhanchuk, A. et al. Recent Developments in the ABINIT Software Package. *Comput. Phys. Commun.* **2016**, *205*, 106 – 131.
- (34) Olevano, V.; Reining, L.; Sottile, F. DP-Code, [Http://www.dp-code.org/](http://www.dp-code.org/). <http://www.dp-code.org/>.
- (35) Laskowski, R.; Christensen, N. E.; Blaha, P.; Palanivel, B. Strong Excitonic Effects in CuAlO₂ Delafossite Transparent Conductive Oxides. *Phys. Rev. B* **2009**, *79*, 165209.
- (36) Vidal, J.; Trani, F.; Bruneval, F.; Marques, M. A.; Botti, S. Effects of Electronic and Lattice Polarization on the Band Structure of Delafossite Transparent Conductive Oxides. *Phys. Rev. Lett.* **2010**, *104*, 136401.

- (37) Hamann, D. Optimized Norm-Conserving Vanderbilt Pseudopotentials. *Phys. Rev. B* **2013**, *88*, 085117.
- (38) Hugues, J. M.; Brown, M. A. The Crystal Structure of Ziesite, β - $\text{Cu}_2\text{V}_2\text{O}_7$, a Thortveitite-Type Structure with a Non-Linear X-O-X Inter-Tetrahedral Bond. *Neu. Jb. Mineral. Mh.* **1989**, 41–47.
- (39) Perdew, J. P.; Burke, K.; Ernzerhof, M. Generalized Gradient Approximation Made Simple. *Phys. Rev. Lett.* **1996**, *77*, 3865.
- (40) Lebègue, S.; Arnaud, B.; Alouani, M.; Bloechl, P. Implementation of an All-Electron GW Approximation Based on the Projector Augmented Wave Method Without Plasmon Pole Approximation: Application to Si, SiC, AlAs, InAs, NaH, and KH. *Phys. Rev. B* **2003**, *67*, 155208.
- (41) VandeVondele, J.; Krack, M.; Mohamed, F.; Parrinello, M.; Chassaing, T.; Hutter, J. Quickstep: Fast and Accurate Density Functional Calculations Using a Mixed Gaussian and Plane Waves Approach. *Comput. Phys. Commun.* **2005**, *167*, 103 – 128.
- (42) Ceriotti, M.; Bussi, G.; Parrinello, M. Nuclear Quantum Effects in Solids Using a Colored-Noise Thermostat. *Phys. Rev. Lett.* **2009**, *103*, 030603.
- (43) VandeVondele, J.; Hutter, J. Gaussian Basis Sets for Accurate Calculations on Molecular Systems in Gas and Condensed Phases. *J. Chem. Phys.* **2007**, *127*, 114105.
- (44) Goedecker, S.; Teter, M.; Hutter, J. Separable Dual-Space Gaussian Pseudopotentials. *Phys. Rev. B* **1996**, *54*, 1703–1710.
- (45) Fuchs, F.; Rödl, C.; Schleife, A.; Bechstedt, F. Efficient $\mathcal{O}(N^2)$ Approach to Solve the Bethe-Salpeter Equation for Excitonic Bound States. *Phys. Rev. B* **2008**, *78*, 085103.
- (46) Hung, L.; Bruneval, F.; Baishya, K.; Ögüt, S. Benchmarking the GW Approximation

and Bethe–Salpeter Equation for Groups IB and IIB Atoms and Monoxides. *J. Chem. Theory Comput.* **2017**, *13*, 2135–2146.