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Stable Dopant-Free Asymmetric Heterocontact Silicon Solar Cells with Efficiencies Above 20%

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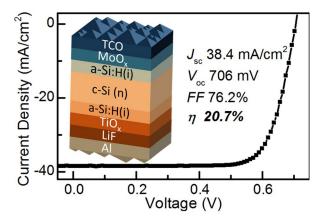
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Abstract

Development of new device architectures and process technologies are of tremendous interest in crystalline silicon (c-Si) photovoltaics to drive enhanced performance and/or reduced processing cost. In this regard, an emerging concept with a high efficiency potential is to employ low/high work function metal compounds or organic materials to form asymmetric electron and hole heterocontacts. This paper demonstrates two important milestones in advancing this burgeoning concept. Firstly, a high-performance, low-temperature, electron-selective heterocontact is developed, comprised of a surface passivating *a*-Si:H layer, a protective TiO_x interlayer and a low work function LiF_x/Al outer electrode. This is combined with a MoO_x hole-selective heterocontact to demonstrate a cell efficiency of 20.7% – the highest value for this cell class to date. Secondly, we show that this cell passes a standard stability test by maintaining >95% of its original performance after 1000 hours of unencapsulated damp-heat exposure, indicating its potential for longevity.



TOC Figure

In recent times, there has been a significant increase in the use of metal oxides,¹⁻⁶ fluorides,⁷⁻⁹ sulphides,¹⁰ and organic materials^{11,12} as carrier selective heterocontacts for crystalline silicon (c-Si) photovoltaic (PV) devices. This research stream has been motivated by potential advantages associated with fabrication simplicity and cost reduction. Such materials can be deposited at low temperature (< 200°C) using simple techniques, to form full-area heterocontacts with optical characteristics tailored for either the sunward- or rear-side of a solar cell. These heterocontacts can also overcome or reduce losses common to other c-Si cell architectures—for example, parasitic absorption or heavy impurity doping losses^{7,13-15} increasing the practical efficiency limit of this structure. Most efforts so far have focused on substituting one such heterocontact into an otherwise conventional c-Si cell,^{16–20} demonstrating, in many cases, clear performance or fabrication advantages. The ultimate extension of this concept is to use a set of asymmetric heterocontacts in a single cell structure, sometimes referred to as the dopant-free asymmetric heterocontact or DASH cell. In our previous study, we presented a record 19.4% efficient DASH solar cell,7 utilizing MoOx and LiFx based heterocontacts with thin amorphous silicon (a-Si:H) interfacial passivation layers. Although promising for a first proof-ofconcept, it is important to demonstrate that higher conversion efficiencies can be achieved, in line with the suggested higher efficiency potential of this architecture. Further, for a new technology to be considered in a field such as c-Si PV, it must satisfy additional requirements related to thermal steps during cell and module fabrication and to device longevity in operation. Therefore, in this study the DASH cell structure is revisited with a particular emphasis on simultaneously improving the device efficiency and stability. Modifications to the structure and fabrication allow us to show for the first time that this technology is compatible with efficiencies greater than 20%. We also

show that un-encapsulated DASH devices can pass an accelerated environmental test designed to simulate the expected damp-heat stressors presented to a solar cell over its lifetime.

To increase the DASH cell performance, improvements must be made simultaneously to both the electron and hole heterocontacts. A recent study conducted by co-authors has shown the thermal stability of the hole heterocontact can be improved via an additional annealing step prior to the MoO_x deposition.²¹ In this study, we focus on the electron side, aiming to develop a thermally robust rear heterocontact. The electron-selective heterocontact of our first-generation DASH cell utilized a low work function (~1 nm) LiF_x / Al outer stack to efficiently extract electrons. When applied to c-Si the low work function induces downward band-bending, encouraging electrons to the surface. To improve the stability of this contact, here we integrate thin oxide protective layers to prevent interaction between the thin a-Si:H passivation layer and the LiF_x / Al layers, without causing a significant impediment to electron flow. Four candidate oxides are trialled in this application: Titanium oxide (TiO_x) , Tantalum oxide (Ta_2O_x) , Hafnium oxide (HfO_x) and Aluminium oxide (Al_2O_x). All are deposited via atomic layer deposition (ALD) at temperatures $\leq 150^{\circ}$ C (further details can be found in Table 1). These materials are chosen to study the influence of conduction band offset on the electron contact performance. To firstly quantify the conduction band offset, Figure 1 presents the measured optoelectronic properties of TiO_x , Ta_2O_x , HfO_x and Al_2O_x thin films (~15 nm) deposited on polished c-Si wafers. The work function and valence band spectrum, measured by X-ray Photoelectron Spectroscopy (XPS), are presented in Figure 1a and b, respectively. The oxygen and metal core levels are also measured by XPS, revealing the stoichiometry of TiO_x (x = 2.02), Ta₂O_x (x = 5.0), HfO_x (x = 1.93) and Al₂O_x (x = 2.98). This is accompanied by the refractive indices (n, k) presented in Figure 1c, extracted from spectroscopic ellipsometry. Implicit within this modelling is a fitting of the optical bandgap

 E_{g} , which is included for each oxide in the bottom plot. Combining the extracted E_{g} with the XPS results in Figure 1a and b allows an estimation of the band positions of these oxides relative to *c*-Si. As shown in Figure 1d an increasing conduction band offset can be expected for the four materials, $TiO_{x} < Ta_{2}O_{x} < HfO_{x} < Al_{2}O_{x}$. The $Ta_{2}O_{x}$ and TiO_{x} films show the lowest conduction band offsets and hence they should present the smallest impediment to electron flow.

To test this electrically, TiO_x , Ta_2O_x , HfO_x and Al_2O_x films of ~3 nm are inserted on the rear side of 2×2 cm² solar cells with the structure shown in Figure 2a. These cells, referred to here as hybrid cells, utilize a standard silicon heterojunction (SHJ) sunward-side hole contact for simplicity,²² further details of which can be found in the Methods section. Contrary to our previous studies, a sub-nanometer layer of LiF_x (~0.6 nm) is used between the metal oxide and Al layers, as it was found to provide the lowest resistivities for this heterocontact stack. The photovoltaic current density-voltage JV behaviour, measured under standard 1-sun conditions (25°C, 100 mW/cm² AM 1.5G spectrum), is shown in Figure 2b. A clear trend of decreasing fill factor FF, due to increasing series resistance R_s , is observed, in line with the increasing conduction band offset discussed above. Series resistance values extracted from these JV curves are included in Figure 2b. The results indicate that, among these materials, TiO_x produces the lowest R_s , likely due to its small conduction band offset with c-Si and hence this material is studied in greater detail. The potential of TiO_x for electron extraction / injection has been demonstrated on a number of optoelectronic devices, including solar cells made from a range of absorber materials.^{23–25} The use of TiO_x electron heterocontacts in c-Si solar cells has expanded rapidly in recent years, being trialled directly,^{12,26} with passivating layers,^{18,27} and with low work function metals.^{16,28} The heterocontact explored here is unique in the combination of passivation, protection and low work function from the a-Si:H, TiO_x and LiF_x / Al layers. This provides a stable, full-area contact

deposited via simple means with high rear reflection. Figure 2c shows *JV* curves measured on hybrid cells with different TiO_x thicknesses ranging from 6 nm to 1.5 nm. The three solid lines correspond to the *JV* curves of cells cut from the same wafer and processed in parallel, showing that the thinnest film of 1.5 nm exhibits the smallest series resistance. Thinner films were not trialled due to the likelihood of incomplete surface coverage, and cells made without any TiO_x (not shown) had lower open circuit voltage V_{oc} and *FF* values than those with the 1.5 nm TiO_x film. The dotted line represents the best of five devices measured on a full (uncut) wafer processed using the same *a*-Si:H / TiO_x (1.5 nm) / LiF_x (~0.6 nm) / Al rear electron contact. This hybrid cell reaches an efficiency of 20.9%, supported by a V_{oc} of 713 mV and a *FF* of 78.1%.

As discussed above, the primary role of the TiO_x interlayer is to improve the stability of the device. To investigate this effect, a set of four transfer-length-method (TLM) samples was prepared on n-type *c*-Si (1 Ω cm) for the extraction of contact resistivity ρ_c . These four test samples, were fabricated with the following contact stacks: *i. a*-Si:H / TiO_x (1.5 nm) / LiF_x (~0.6 nm) / Al; *ii. a*-Si:H / LiF_x (~0.6 nm) / Al; *iii.* TiO_x (1.5 nm) / LiF_x (~0.6 nm) / Al; and *iv*. LiF_x (~0.6 nm) / Al. Figure 3a shows the measured ρ_c of the four samples as a function of annealing temperature in air. The measurements were taken after successive 10 minute anneals at incrementally increasing temperatures between room temperature and 300°C; hence the last point corresponds to a cumulative annealing time of 50 min. It can be seen that the addition of the 1.5 nm TiO_x layer does not increase the as-deposited ρ_c . This is true for both contacts with and without the a-Si:H layers with ρ_c values of ~70 m Ω cm² and 4 m Ω cm², respectively. More importantly, the TiO_x layer improves the thermal stability of ρ_c drastically for annealing temperatures up to 300°C. It should be noted that TLM samples fabricated without the LiF_x interlayer (not shown) exhibited significantly higher ρ_c values than those shown in Figure 3a. A second stability test, conducted on two hybrid solar cells with and without a 1.5 nm TiO_x layer, is presented in Figure 3b. Whereas the V_{oc} of the cell without TiO_x degrades drastically as the annealing temperature increases, a significant improvement in the passivation stability is achieved as a result of the TiO_x interlayer addition. The enhanced stability results presented in Figure 3a for the ρ_c and in Figure 3b for the V_{oc} are likely due to the prevention of intermixing between the *a*-Si:H and the LiF_x / Al layers. TiO_x protective layers have been utilized in a number of silicon based structures, including photoelectrochemical cells as a protective film in harsh chemical environments,²⁹ and in microelectronics and solar cell devices as a high temperature diffusion barrier to metals and dopants.^{30,31} Without this layer, at elevated temperatures the angstrom-scale LiF_x (~0.6 nm) may not provide sufficient protection of the thin a-Si:H passivation layer from the overlying Al layer. A number of studies have shown metal-induced crystallization of *a*-Si:H at low temperatures upon contact with Al,^{32–34} a process which will increase interface recombination.

Finally, the optimized *a*-Si:H / TiO_x / LiF_x / Al electron heterocontact is combined with a front, *a*-Si:H / MoO_x / indium tin oxide (ITO) hole heterocontact into a full $2\times 2 \text{ cm}^2$ DASH cell with the structure shown in Figure 4a. The front MoO_x heterocontact fabrication is conducted in line with the suggestions of Ref 21 to improve its thermal stability. The 1-sun *JV* results of the stability enhanced DASH is provided in Figure 4b, showing that a conversion efficiency of 20.7% has been achieved, supported by a V_{oc} of 706 mV and a *FF* of 76.1%. This is the first demonstration of a DASH cell above 20%. A quantum efficiency analysis of this cell is also included in Figure 4c, showing the measured external quantum efficiency (EQE), reflection and internal quantum efficiency (IQE), taken in-between Ag metal grid fingers. After accounting for a contact fraction of ~3%, the integration of the EQE and the solar spectrum product gives a J_{sc} of 38.8 mA/cm².

This value agrees well with that obtained directly from the light JV of 38.4 mA/cm² supporting the accuracy of the value. To further test the stability of the electron and hole contact structures, a representative un-encapsulated DASH cell is subjected to a standard 1000-hour damp-heat test at 85°C and 85% relative humidity(RH). Table 1 shows the measured characteristic parameters of this device before and after the 1000-hour period, revealing that 96% of the original performance has been maintained.

In this article we have simultaneously improved the benchmark efficiency and stability of c-Si DASH solar cells. We have found that the thermal stability of the electron-selective heterocontacts can be improved by introducing a protective TiO_x interlayer between the *a*-Si:H passivation layer and the low work function LiF_x / Al electrode. This second-generation DASH cell has been proven to maintain > 95% of its performance after 1000 hours of unencapsulated damp-heat exposure. Importantly, the improved stability does not come at the expense of performance. On the contrary, a conversion efficiency of 20.7% has been demonstrated, the highest value of any DASH solar cell presented to date. These developments in efficiency and stability pave the way for the DASH cell design to become a viable contender for high performance, low-cost c-Si PV.

Materials characterization:

Samples for X-ray Photoelectron Spectroscopy (XPS) and Spectroscopic Ellipsometry (SE) were fabricated by depositing thin films of TiO_x, Ta₂O_x, HfO_x and Al₂O_x on a polished n⁺ silicon wafer which was given a short 5% HF etch prior to deposition. The thin films were deposited by atomic-layer-deposition (ALD), with a chamber temperature of \leq 150°C. Further

details can be found in Table 1. XPS measurements were performed in a Kratos spectrometer with an Al monochromatic X-ray source. All measurements were performed on thin films (~15 nm) without electron gun neutralization. Charging is known to be a general issue in XPS, particularly for insulating films. Efforts were made to minimize charging by reducing incident X-ray exposure during the secondary electron cutoff measurement, and core level and valence band spectra were referenced to a C 1s peak at 284.8eV. For the secondary electron cutoff measurement, a sputter cleaned Au sample was included in each run as a reference. The ranges provided for the Fermi energy, conduction band and valence band of each material in the schematic of Figure 1d are based off the measured variation in XPS results. SE measurements were also taken on ~15nm films with a JA Wollam M 2000 spectroscopic ellipsometer and the films were fitted with Tauc-Lorentz models, the optical bandgaps of which are listed in Table 1.

Device fabrication and characterization.

Contact structures were fabricated according to a transfer-length-method (TLM) design. Samples with and without the thin (~5nm) *a*-Si:H passivation layer were deposited on 4 Ω cm and 1 Ω cm n-type float zone (FZ) wafers, respectively. The *a*-Si:H was deposited via plasma enhanced chemical vapor deposition (PECVD). Half of these received a thin 1.5 nm TiO_x protective layer following which they were all deposited with a LiF_x (0.6 nm) / Al electrode, through a shadow mask to achieve the TLM pattern. The dark current-voltage *IV* behaviour was measured as a function of pad spacing using a Keithley 2400 sourcemeter and the contact resistivity was extracted from this relationship according to Ref 35. Annealing of contact samples was performed on a hotplate in air at temperatures between 100°C and 300°C.

The hybrid cell structures utilized in Figures 2 and 3 were fabricated on double side textured, float zone, 4 Ω cm n-type wafers with a thickness of ~250 μ m. The front hole

heterocontact consists of a stack of intrinsic (~5 nm) and boron doped (~10 nm) *a*-Si:H layers. This is capped with an indium tin oxide (ITO) layer which was sputtered through a shadow mask to define a 2×2 cm² area. The *a*-Si:H layers were deposited via plasma enhanced chemical vapor deposition (PECVD) and the TCOs via DC sputtering. The front Ag grid, with a surface fraction of ~3%, is achieved by screen printing a low temperature Ag paste and curing. For the rear contact, an intrinsic thin *a*-Si:H layer (~5nm) is coated with either TiO_x, Ta₂O_x, HfO_x or Al₂O_x thin films deposited via ALD using the precursors listed in Table 1. These are capped with a low work function LiF_x (~0.6 nm) / Al (~200 nm) electrode deposited via thermal evaporation. Hybrid cell annealing was performed in a quartz tube furnace between 100°C and 300°C.

The 2×2 cm² DASH cell was also fabricated on a double side textured 4 Ω cm n-type wafer with a thin (~5 nm) passivating *a*-Si:H layer on either side. This was subjected to a short anneal at 250°C prior to contact deposition in line with the suggestions of Ref 21. Next, the front-side was deposited with ~5 nm of thermally evaporated MoO_x followed by ~70 nm of sputtered indium tin oxide (ITO) and finally low temperature Ag paste was screen printed and cured at 190°C. The rear side was deposited with 1.5 nm of ALD TiO_x followed by the thermally evaporated LiF_x (~0.6 nm) / Al electrode.

One sun *JV* analysis of these cells was performed under standard conditions (25°C, 100 mW/cm², AM 1.5G spectrum) using a calibrated solar simulator and an aperture mask to avoid peripheral absorption. The external quantum efficiency and reflection were measured in an inhouse built setup. An Espec LHU humidity chamber, held at 85°C and 85% RH, was used for the humidity exposure test.

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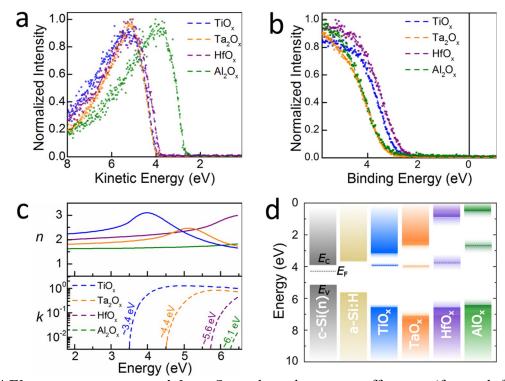


Figure 1 | Electron contact materials, a, Secondary electron cutoff energy (for work function), b, valence band spectrum, and c, refractive indices (top: real part n, bottom: imaginary part k) of different metal oxide protective films. d, potential band position diagram for the electron heterocontact materials using information from a-c.

Material	Metal precursor	Oxidizing precursor	Extracted <i>E</i> g (eV)	Oxygen : metal ratio
TiO _x	Titanium isopropoxide	Water	3.45	2.02
Ta ₂ O _x	Tantalum ethoxide	Water	4.4	2.5
HfO_x	Tetrakis(diethylamido)hafnium	Water	5.65	1.93
AI_2O_x	Trimethylaluminum	Water	6.08	1.49

 Table 1 | Deposition conditions and properties of atomic-layer-deposited metal oxide protective layers.

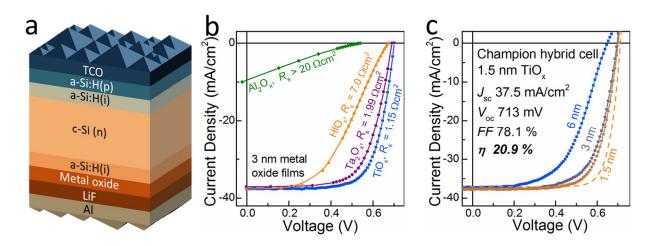


Figure 2 | Electron contact optimization, a, structure of hybrid cell used for contact optimization, b, 1-sun light JV behaviour of hybrid cells with 3 nm protective metal oxide layers, c, 1-sun light JV behaviour of hybrid cells with different TiO_x thicknesses showing an optimum thickness of 1.5 nm.

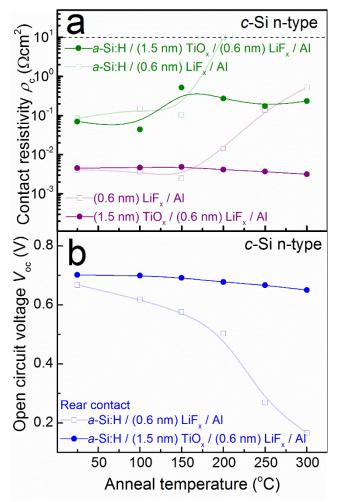


Figure 3 | Electron contact stability. a, ρ_c as a function of sequential 10 minute anneals at increasing temperatures for samples with/without *a*-Si:H and TiO_x layers. b, V_{oc} progression of hybrid cells after sequential 10 minute anneals at increasing temperatures for cells with/without the TiO_x layer.

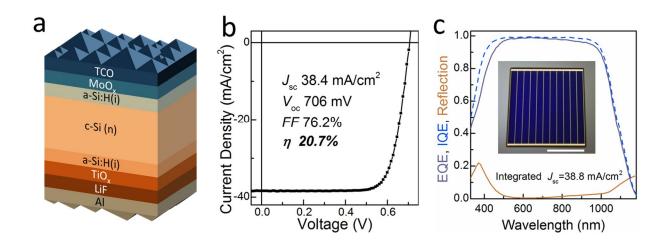


Figure 4 | **DASH cell record. a**, structure of the DASH cell, **b**, 1-sun light *JV* behaviour of champion DASH cell, **c**, external quantum efficiency (EQE), internal quantum efficiency (IQE) and reflection of champion DASH cell, the inset shows a photograph of the cell design and the integrated J_{sc} value. The white scale bar is ~1 cm.

	V _{oc} (V)	FF (%)	J _{sc} (mA/cm²)	η (%)
As fabricated	698	73.5	38.2	19.6
1000 hrs at 85°C, 85% RH	696	71.3	38.2	19.0

 Table 2 | Damp heat stability. 1-sun JV characteristics of a representative DASH cell measured

 before and after 1000 hours of unencapsulated exposure to 85°C, 85% RH.

References

- Battaglia, C.; Yin, X.; Zheng, M.; Sharp, I. D.; Chen, T.; McDonnell, S.; Azcatl, A.; Carraro, C.; Ma, B.; Maboudian, R.; *et al.* Hole Selective MoO_x Contact for Silicon Solar Cells. *Nano Lett.* **2014**, *14*, 967–971.
- (2) Avasthi, S.; McClain, W. E.; Man, G.; Kahn, A.; Schwartz, J.; Sturm, J. C. Hole-Blocking Titanium-Oxide/Silicon Heterojunction and Its Application to Photovoltaics. *Appl. Phys. Lett.* 2013, 102, 203901.
- (3) Wan, Y.; Samundsett, C.; Bullock, J.; Hettick, M.; Allen, T.; Yan, D.; Peng, J.; Wu, Y.; Cui, J.; Javey, A.; *et al.* Conductive and Stable Magnesium Oxide Electron-Selective Contacts for Efficient Silicon Solar Cells. *Adv. Energy Mater.* **2016**, *7*, 1601863.
- Bivour, M.; Temmler, J.; Steinkemper, H.; Hermle, M. Molybdenum and Tungsten Oxide: High Work Function Wide Band Gap Contact Materials for Hole Selective Contacts of Silicon Solar Cells. Sol. Energy Mater. Sol. Cells 2015, 142, 34–41.
- (5) Gerling, L. G.; Mahato, S.; Morales-Vilches, A.; Masmitja, G.; Ortega, P.; Voz, C.; Alcubilla, R.; Puigdollers, J. Transition Metal Oxides as Hole-Selective Contacts in Silicon Heterojunctions Solar Cells. Sol. Energy Mater. Sol. Cells 2016, 145, 109–115.
- (6) Macco, B.; Vos, M. F. J.; Thissen, N. F. W.; Bol, A. A.; Kessels, W. M. M. Low-Temperature Atomic Layer Deposition of MoO_x for Silicon Heterojunction Solar Cells. *Phys. Status Solidi RRL – Rapid Res. Lett.* **2015**, *9*, 393–396.
- Bullock, J.; Hettick, M.; Geissbühler, J.; Ong, A. J.; Allen, T.; Sutter-Fella, C. M.; Chen, T.; Ota, H.;
 Schaler, E. W.; De Wolf, S.; *et al.* Efficient Silicon Solar Cells with Dopant-Free Asymmetric
 Heterocontacts. *Nat. Energy* 2016, *1*, 15031.
- (8) Wan, Y.; Samundsett, C.; Bullock, J.; Allen, T.; Hettick, M.; Yan, D.; Zheng, P.; Zhang, X.; Cui, J.; McKeon, J.; *et al.* Magnesium Fluoride Electron-Selective Contacts for Crystalline Silicon Solar Cells. *ACS Appl. Mater. Interfaces* **2016**, *8*, 14671–14677.
- (9) Zhang, Y.; Liu, R.; Lee, S.-T.; Sun, B. The Role of a LiF Layer on the Performance of Poly(3,4-Ethylenedioxythiophene):Poly(Styrenesulfonate)/Si Organic-Inorganic Hybrid Solar Cells. *Appl. Phys. Lett.* **2014**, *104*, 083514.
- (10) Xu, X.; Bullock, J.; Schelhas, L. T.; Stutz, E. Z.; Fonseca, J. J.; Hettick, M.; Pool, V. L.; Tai, K. F.; Toney, M. F.; Fang, X.; *et al.* Chemical Bath Deposition of P-Type Transparent, Highly Conducting (CuS)_x:(ZnS)_{1-x} Nanocomposite Thin Films and Fabrication of Si Heterojunction Solar Cells. *Nano Lett.* **2016**, *16*, 1925–1932.
- (11) Zielke, D.; Pazidis, A.; Werner, F.; Schmidt, J. Organic-Silicon Heterojunction Solar Cells on n-Type Silicon Wafers: The BackPEDOT Concept. *Sol. Energy Mater. Sol. Cells* **2014**, *131*, 110–116.
- (12) Nagamatsu, K. A.; Avasthi, S.; Sahasrabudhe, G.; Man, G.; Jhaveri, J.; Berg, A. H.; Schwartz, J.; Kahn, A.; Wagner, S.; Sturm, J. C. Titanium Dioxide/Silicon Hole-Blocking Selective Contact to Enable Double-Heterojunction Crystalline Silicon-Based Solar Cell. *Appl. Phys. Lett.* **2015**, *106*, 123906.
- (13) Holman, Z. C.; Descoeudres, A.; Barraud, L.; Fernandez, F. Z.; Seif, J. P.; De Wolf, S.; Ballif, C. Current Losses at the Front of Silicon Heterojunction Solar Cells. *Photovolt. IEEE J. Of* **2012**, *2*, 7– 15.
- (14) Richter, A.; Glunz, S. W.; Werner, F.; Schmidt, J.; Cuevas, A. Improved Quantitative Description of Auger Recombination in Crystalline Silicon. *Phys. Rev. B* **2012**, *86*, 165202.
- (15) Battaglia, C.; de Nicolas, S. M.; De Wolf, S.; Yin, X.; Zheng, M.; Ballif, C.; Javey, A. Silicon Heterojunction Solar Cell with Passivated Hole Selective MoO_x Contact. *Appl. Phys. Lett.* **2014**, 104, 113902.

- Allen, T. G.; Bullock, J.; Jeangros, Q.; Samundsett, C.; Wan, Y.; Cui, J.; Hessler-Wyser, A.; De Wolf, S.; Javey, A.; Cuevas, A. A Low Resistance Calcium/Reduced Titania Passivated Contact for High Efficiency Crystalline Silicon Solar Cells. *Adv. Energy Mater.* 2017, *7*, 1602606.
- (17) Bullock, J.; Zheng, P.; Jeangros, Q.; Tosun, M.; Hettick, M.; Sutter-Fella, C. M.; Wan, Y.; Allen, T.; Yan, D.; Macdonald, D.; *et al.* Lithium Fluoride Based Electron Contacts for High Efficiency N-Type Crystalline Silicon Solar Cells. *Adv. Energy Mater.* **2016**, *6*, 1600241.
- (18) Yang, X.; Weber, K.; Hameiri, Z.; De Wolf, S. Industrially Feasible, Dopant-Free, Carrier-Selective Contacts for High-Efficiency Silicon Solar Cells. *Prog. Photovolt. Res. Appl.* **2017**, *25*, 896–904.
- (19) Masmitjà, G.; Gerling, L. G.; Ortega, P.; Puigdollers, J.; Martín, I.; Voz, C.; Alcubilla, R. V₂O_x-Based Hole-Selective Contacts for c-Si Interdigitated Back-Contacted Solar Cells. *J. Mater. Chem. A* **2017**, *5*, 9182–9189.
- (20) Geissbühler, J.; Werner, J.; Martin de Nicolas, S.; Barraud, L.; Hessler-Wyser, A.; Despeisse, M.; Nicolay, S.; Tomasi, A.; Niesen, B.; De Wolf, S.; *et al.* 22.5% Efficient Silicon Heterojunction Solar Cell with Molybdenum Oxide Hole Collector. *Appl. Phys. Lett.* **2015**, *107*, 081601.
- (21) Essig, S.; Dréon, J.; Rucavado, E.; Mews, M.; Koida, T.; Boccard, M.; Werner, J.; Geissbuhler, J.; Löper, P.; Morales-Masis, M.; et al. Towards Annealing-Stable Molybdenum-Oxide-Based Hole-Selective Contacts for Si Photovoltaics. Accepted for publication in Sol. Rapid Res. Lett. 2018, DOI: 10.1002/solr.201700227.
- (22) Descoeudres, A.; Holman, Z. C.; Barraud, L.; Morel, S.; De Wolf, S.; Ballif, C. >21%; Efficient Silicon Heterojunction Solar Cells on n- and p-Type Wafers Compared. *Photovolt. IEEE J. Of* **2013**, *3*, 83– 89.
- (23) Liu, M.; Johnston, M. B.; Snaith, H. J. Efficient Planar Heterojunction Perovskite Solar Cells by Vapour Deposition. *Nature* **2013**, *501*, 395.
- Hsu, W.; Sutter-Fella, C. M.; Hettick, M.; Cheng, L.; Chan, S.; Chen, Y.; Zeng, Y.; Zheng, M.; Wang, H.-P.; Chiang, C.-C.; *et al.* Electron-Selective TiO₂ Contact for Cu(In,Ga)Se₂ Solar Cells. *Sci. Rep.* 2015, *5*, 16028.
- Yin, X.; Battaglia, C.; Lin, Y.; Chen, K.; Hettick, M.; Zheng, M.; Chen, C.-Y.; Kiriya, D.; Javey, A.
 19.2% Efficient InP Heterojunction Solar Cell with Electron-Selective TiO₂ Contact. ACS Photonics 2014, 1, 1245–1250.
- (26) Boccard, M.; Yang, X.; Weber, K.; Holman, Z. C. Passivation and Carrier Selectivity of TiO₂ Contacts Combined with Different Passivation Layers and Electrodes for Silicon Solar Cells. In 2016 IEEE 43rd Photovoltaic Specialists Conference (PVSC); 2016; pp. 2403–2407.
- (27) Cho, J.; Debucquoy, M.; Recaman Payo, M.; Malik, S.; Filipič, M.; Radhakrishnan, H. S.; Bearda, T.; Gordon, I.; Szlufcik, J.; Poortmans, J. Contact Resistivity Reduction on Lowly-Doped n-Type Si Using a Low Workfunction Metal and a Thin TiO_x Interfacial Layer for Doping-Free Si Solar Cells. *Energy Procedia* **2017**, *124*, 842–850.
- Bullock, J.; Ota, H.; Wang, H.; Xu, Z.; Essig, S.; Samundsett, C.; Yan, D.; Hettick, M.; Morales-Masis, M.; Wan, Y.; *et al.* Microchannel Contacting of Crystalline Silicon Solar Cells. *Sci. Rep.* 2017, *7*, 9085.
- (29) Bae, D.; Seger, B.; Vesborg, P. C. K.; Hansen, O.; Chorkendorff, I. Strategies for Stable Water Splitting via Protected Photoelectrodes. *Chem. Soc. Rev.* **2017**, *46*, 1933–1954.
- (30) Richards, B. S.; Richards, S. R.; Boreland, M. B.; Jamieson, D. N. High Temperature Processing of TiO₂ Thin Films for Application in Silicon Solar Cells. J. Vac. Sci. Technol. Vac. Surf. Films 2004, 22, 339–348.
- (31) Alén, P.; Vehkamäki, M.; Ritala, M.; Leskelä, M. Diffusion Barrier Properties of Atomic Layer Deposited Ultrathin Ta₂O₅ and TiO₂ Films. *J. Electrochem. Soc.* **2006**, *153*, G304–G308.
- (32) Plagwitz, H.; Nerding, M.; Ott, N.; Strunk, H. P.; Brendel, R. Low-Temperature Formation of Local Al Contacts to a-Si:H-Passivated Si Wafers. *Prog. Photovolt. Res. Appl.* **2004**, *12*, 47–54.

- (33) Bullock, J.; Yan, D.; Wan, Y.; Cuevas, A.; Demaurex, B.; Hessler-Wyser, A.; De Wolf, S. Amorphous Silicon Passivated Contacts for Diffused Junction Silicon Solar Cells. J. Appl. Phys. 2014, 115, 163703.
- (34) Bullock, J.; Cuevas, A.; Yan, D.; Demaurex, B.; Hessler-Wyser, A.; De Wolf, S. Amorphous Silicon Enhanced Metal-Insulator-Semiconductor Contacts for Silicon Solar Cells. *J. Appl. Phys.* **2014**, *116*, 163706.
- (35) Schroder, D. K. *Semiconductor Material and Device Characterization*; 3rd edition.; John Wiley & Sons, pp. 389-465: Hoboken, New Jersey, USA, 2006.