

HEAT TRANSFER OF ICE SLURRIES IN PIPES

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Some basic ideas of the heat transfer of ice slurries are discussed. The equations describing a flow of a Bingham fluid through a tube with a heat flux over the boundary are presented. The theory shows that familiar relations known in the field of heat exchangers, e.g. the logarithmic-mean temperature difference, do not always apply. - With a device, designed to measure the heat transfer coefficients in tubes, first experimental data on the specific heat capacity have been obtained. They are compared with model results. The velocity profile of a laminar ice slurry flow has been measured with an ultra-sound velocity profile sensor. Furthermore, ice particles of a water/ethanol ice slurry have been observed under a microscope.

1. INTRODUCTION

In classical refrigeration technology the working fluids usually show Newtonian behaviour. Therefore, the molecular friction can be described by a single quantity, the dynamic viscosity. The occurrence of fine-crystalline ice slurries - which are clearly Non-Newtonian fluids - request knowledge of basic rheology. In rheology scientists deal with complex mixtures (see for example Ref. [1]). A large number of these fluids can be described as pseudohomogenous multiphase mixtures. Ice slurries are multi-component and biphasic fluids.

The *time-independent viscous fluids* are classified as follows:

- 1) Newtonian fluids
- 2) Pseudoplastics
- 3) Dilatant fluids
- 4) Bingham fluids
- 5) Yield-pseudoplastics.

Furthermore there are *time-dependent fluids*, the

- 1) Thixotropic fluids (the shear stress decreases with time)
- 2) Rheopectic fluids (the shear stress increases with time).

A third group is named the *viscoelastic fluids*. They belong into the overlapping region of elasticity theory and fluid dynamics.

The listed time-independent fluids are limiting and idealized cases. Usually real fluids show a behaviour only approximately equal to one of these ideal fluids. The rheograms of the time-independent ideal fluids are shown in figure 1.

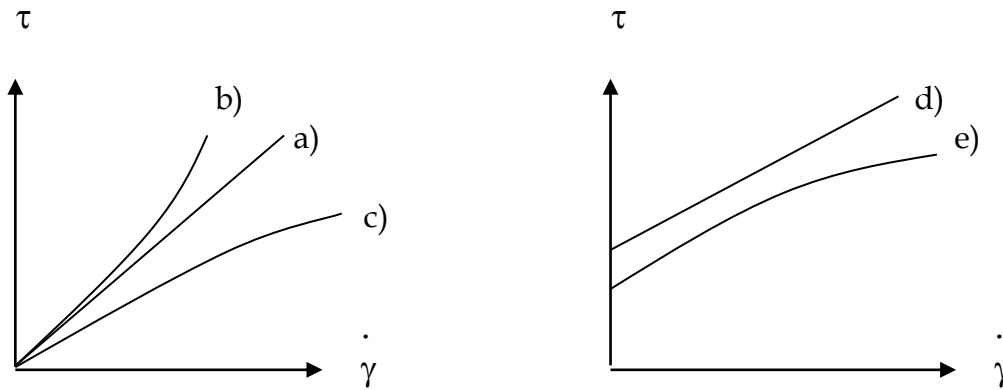


Figure 1: Rheograms of Newtonian fluid (a), dilatant fluid (b), pseudoplastic (c), Bingham fluid (d) and yield-pseudoplastic (e).

In some cases it is difficult to decide, which model description gives the best fit to the real occurring behaviour. In the case of an ice slurry based on a water/ethanol mixture some research groups favour the model describing pseudoplastics, where other groups give their preference to the Bingham theory. Discussions and an answer to this question are expected to be achieved in the new working party "Ice Slurries" of the International Institute of Refrigeration. In the Bingham model the shear stress is related to the shear velocity by the following formula (see figure 1)

$$\tau = \mu \dot{\gamma} + \tau_0, \tag{1/1}$$

where τ_0 describes the critical shear stress, characteristic for Bingham fluids. All theoretical considerations in this article are based on the Bingham model.

2. THE BASIC EQUATIONS

The continuity, Navier Stokes and the energy equation for time-dependent flows are

$$\frac{\partial \rho}{\partial t} + u \frac{d\rho}{dT} \frac{\partial T}{\partial x} + \rho \frac{\partial u}{\partial x} = 0 \tag{2/1}$$

$$\rho \frac{\partial u}{\partial t} + \rho u \frac{\partial u}{\partial x} + \frac{\partial p}{\partial x} - \frac{\mu}{r} \frac{\partial u}{\partial r} - \frac{1}{r} \tau_0 - \frac{d\mu}{dT} \left(\frac{\partial u}{\partial r} \frac{\partial T}{\partial r} \right) - \mu \frac{\partial^2 u}{\partial r^2} - \frac{d\tau_0}{dT} \frac{\partial T}{\partial r} = 0 \quad (2/2)$$

$$\rho c_p \frac{\partial T}{\partial t} + \rho u c_p \frac{\partial T}{\partial x} - u \frac{\partial p}{\partial x} - \frac{k}{r} \frac{\partial T}{\partial r} - \frac{dk}{dT} \left(\frac{\partial T}{\partial r} \frac{\partial T}{\partial r} \right) - k \frac{\partial^2 T}{\partial r^2} = 0, \quad (2/3)$$

where all the physical properties are functions of the temperature. In Ref. [2] they are derived from basic physical principles taking several approximative assumptions into consideration.

3. THE CONTINUOUS-PROPERTIES MODEL

The kinematic boundary condition is assumed to be the no-slip boundary condition and the thermal condition may, for example, describe a constant heat flux density \dot{q} in x-direction

$$u = 0, \quad \frac{\partial T}{\partial r} = \frac{1}{k} \dot{q}, \quad r = R. \quad (3/1a-c)$$

Special case: no flow. Then

$$u \equiv 0. \quad (3/2)$$

Equation (2/1) vanishes and from (2/3) it follows

$$\frac{\partial T}{\partial t} = \alpha(T) \left(\frac{\partial^2 T}{\partial r^2} + \frac{1}{r} \frac{\partial T}{\partial r} \right) + \beta(T) \left(\frac{\partial T}{\partial r} \frac{\partial T}{\partial r} \right), \quad (3/3)$$

which is the continuous-properties model in cylindrical co-ordinates. The nonlinear coefficients α and β are defined in Ref. [3]. From (2/2) it follows that a further condition remains

$$\frac{\partial p}{\partial x} - \frac{1}{r} \tau_0 - \frac{d\tau_0}{dT} \frac{\partial T}{\partial r} = 0, \quad (3/4)$$

which is a relation between the pressure drop per unit length in downstream direction and the radial directional derivative of the critical shear stress.

4. HEAT TRANSFER MODEL

4.1 General theory

For a steady state from equation (2.3) and the definition of enthalpy $dh=c_p dT$ it follows that

$$\rho u dh \gg u dp \quad \Leftrightarrow \quad dh \gg \frac{1}{\rho} dp \quad (4.1/1a,b)$$

is fulfilled, the following formula applies to an infinitesimal small control volume spanned over the entire cross section of the pipe

$$\langle \rho u h \rangle_r (x + \Delta x) A_P - \langle \rho u h \rangle_r (x) A_P = \dot{q} \Delta A_S. \quad (4.1/2)$$

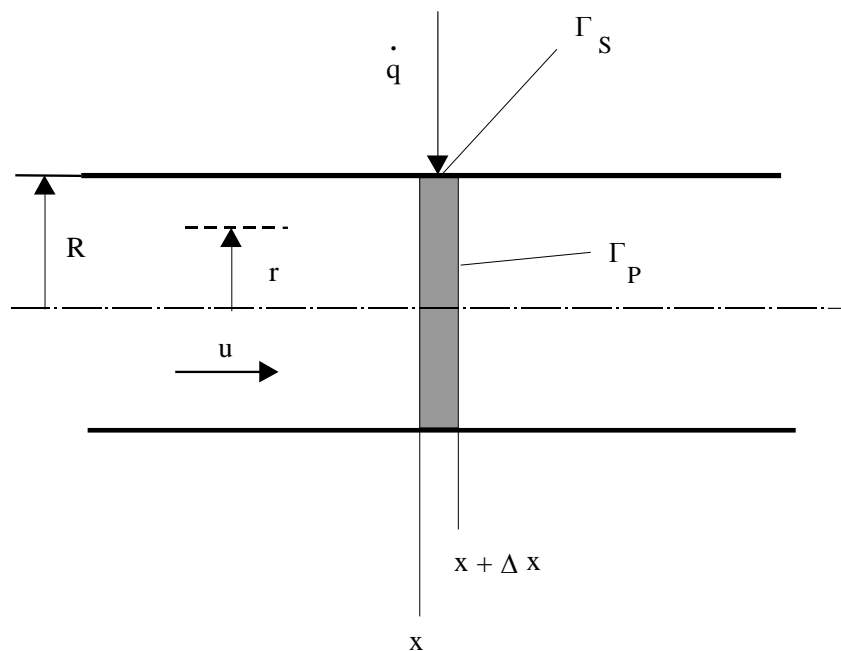


Figure 2: Control volume for the derivation of equations describing the heat transfer to a flowing ice slurry.

The following average over the domain Γ_P - with area A_P - has been introduced

$$\langle \chi \rangle_r (t, x) = \frac{1}{A_P} \int_{\Gamma_P} \chi(t, x, r) dA_P, \quad \chi \in \{u, \rho, h\}. \quad (4.1/3)$$

In cylindrical co-ordinates the differential area and total area of the cross section are

$$dA_P = \int_0^R r d\varphi dr = 2\pi \int_0^R r dr, \quad A_P = \pi R^2, \quad (4.1/4a-c)$$

and the space-averaged quantity in the radial direction is

$$\langle \chi \rangle_r (t, x) = \frac{2}{R^2} \int_0^R \chi(t, x, r) r dr . \quad (4.1/5)$$

Furthermore we have

$$\Delta A_S = 2\pi R \Delta x . \quad (4.1/6)$$

From equations (4.1/2), (4.1/4c) and (4.1/6) a new equation is derived

$$\langle \rho u h \rangle_r (x + \Delta x) - \langle \rho u h \rangle_r (x) = \dot{q} \frac{\Delta A_S}{A_P} = 2\dot{q} \frac{\Delta x}{R} . \quad (4.1/7a,b)$$

Now, we introduce the definition of a mean weighted enthalpy density

$$\bar{h} := \frac{\langle \rho u h \rangle_r}{\langle \rho u \rangle_r} \quad \Leftrightarrow \quad \langle \rho u h \rangle_r = \bar{h} \langle \rho u \rangle_r . \quad (4.1/8a,b)$$

The inequality of Schwarz states that for positive functions f and g

$$\langle fg \rangle_r \leq \langle f \rangle_r \langle g \rangle_r . \quad (4.1/9)$$

Now a correlation coefficient is defined

$$R_{f,g} = \frac{\langle fg \rangle_r}{\langle f \rangle_r \langle g \rangle_r} \quad \Leftrightarrow \quad 0 \leq R_{f,g} \leq 1 , \quad (4.1/10a,b)$$

which is a direct result of equation (4.1/9).

Special case (CPI):

$$\langle \rho u h \rangle_r = \langle \rho u \rangle_r \langle h \rangle_r \quad \Leftrightarrow \quad R_{\rho u, h} = 1 . \quad (4.1/11a,b)$$

Assuming the enthalpy density to be positive (not as in Ref. [3]) and from the definition (4.1/8a) we can conclude that

$$\bar{h} = \langle h \rangle_r . \quad (4.1/12)$$

Furthermore it is clear that

$$\dot{m} = \langle \rho u \rangle_r A_P \quad \Leftrightarrow \quad \langle \rho u \rangle_r = \frac{\dot{m}}{A_P} \quad (4.1/13a,b)$$

describes the mass flow, which is not a function of the downstream co-ordinate x . This argumentation is correct because of mass conservation.

Inserting (4.1/8b) into (4.1/7a) we derive

$$\begin{aligned} \bar{h}(x + \Delta x) \langle \rho u \rangle_r (x + \Delta x) - \bar{h}(x) \langle \rho u \rangle_r (x) &= \bar{h}(x + \Delta x) \frac{\dot{m}}{A_P} - \\ \bar{h}(x) \frac{\dot{m}}{A_P} &= \dot{q} \frac{\Delta A_S}{A_P}, \end{aligned} \quad (4.1/14a,b)$$

a formula, which is obtained by also applying formula (4.1/13b).

Therefore one concludes that

$$\bar{h}(x + \Delta x) - \bar{h}(x) = \frac{\dot{q}}{\dot{m}} \Delta A_S = 2\pi R \frac{\dot{q}}{\dot{m}} \Delta x, \quad (4.1/15a,b)$$

respectively

$$\frac{\bar{h}(x + \Delta x) - \bar{h}(x)}{\Delta x} = 2\pi R \frac{\dot{q}}{\dot{m}}. \quad (4.1/16)$$

In the limit Δx equal to zero we have

$$\lim_{\Delta x \rightarrow 0} \frac{\bar{h}(x + \Delta x) - \bar{h}(x)}{\Delta x} = \frac{d\bar{h}}{dx} = 2\pi R \frac{\dot{q}}{\dot{m}}. \quad (4.1/17a,b)$$

4.2 Constant heat flux density

Now we are prepared to study the case, when a constant heat flux density is applied. Then from equation (4.1/17a,b), by an integration one obtains

$$\bar{h}(x) = 2\pi R \frac{\dot{q}}{\dot{m}} x + \bar{h}(0), \quad 0 \leq x \leq L. \quad (4.2/1)$$

This linear relationship is fulfilled in any case. Introducing a correlation function in analogy to the correlation coefficient (4.1/10a) leads to

$$\bar{h}(x) = \frac{R_{\rho u, h}(x) \langle \rho u \rangle_r \langle h \rangle_r(x)}{\langle \rho u \rangle_r} = R_{\rho u, h}(x) \langle h \rangle_r(x). \quad (4.2/2a,b)$$

Therefore one concludes that

$$\langle h \rangle_r(x) = \frac{1}{R_{\rho u, h}(x)} \bar{h}(x) = \frac{1}{R_{\rho u, h}(x)} \left[2\pi R \frac{\dot{q}}{\dot{m}} x + \bar{h}(0) \right]. \quad (4.2/3a,b)$$

The space-averaged enthalpy density across the pipe section is not necessarily linear.

Special case (CP1):

Then

$$R_{\rho u, h}(x) = 1, \quad \forall x \in [0, L] \quad \Leftrightarrow \quad \langle h \rangle_r(x) = \left[2\pi R \frac{\dot{q}}{\dot{m}} x + \langle h \rangle_r(0) \right]. \quad (4.2/4a,b)$$

The linearity in x is also fulfilled for the non-weighted enthalpy density $\langle h \rangle_r$.

It will be a great challenge to obtain good experimental and numerical data to check if $R_{\rho u, h} \cong 1$. If this is not the case the problem becomes very difficult to solve, and (probably) no analytical solutions will be available. Then only numerical simulations can produce quantitative useful results.

Now Newton's law on heat transfer is applied. At this stage the surface temperature T_S is not yet assumed to be constant

$$\dot{q} = \alpha(x) [T_S(x) - \langle T \rangle_r(x)]. \quad (4.2/5)$$

Substitution into equation (4.1/17b) leads to

$$\frac{d\bar{h}}{dx} = 2\pi \frac{R}{\dot{m}} \alpha(x) [T_S(x) - \langle T \rangle_r(x)]. \quad (4.2/6)$$

To obtain a differential equation that can be solved here \bar{h} must be expressed by the temperature. As it already has been stressed, this leads to a very complicated relation. Therefore, only the special case (CP1) is investigated.

Special case (CP1):

Considering equation (4.2/6) and with the help of equation (4.1/12) one obtains

$$\frac{d}{dx} \langle h \rangle_r = 2\pi \frac{R}{\dot{m}} \alpha(x) [T_S(x) - \langle T \rangle_r(x)]. \quad (4.2/7)$$

Because of

$$\frac{d}{dx} \langle \chi \rangle_r = \frac{d}{dx} \chi \rangle_r \quad \chi \in \{\rho, u, h\} \quad (4.2/8)$$

from equation (4.2/7) it follows that

$$\langle c_p \frac{dT}{dx} \rangle_r = 2\pi \frac{R}{\dot{m}} \alpha(x) [T_S(x) - \langle T \rangle_r(x)]. \quad (4.2/9)$$

By applying (4.1/10a) we obtain

$$\langle c_p \frac{dT}{dx} \rangle_r = R \frac{\partial T}{c_p, \partial x} \langle c_p \rangle_r \langle \frac{dT}{dx} \rangle_r. \quad (4.2/10)$$

Special case (CP2):

$$R \frac{\partial T}{c_p, \partial x} = 1, \quad \forall x \in [0, L]. \quad (4.2/11)$$

Then from (4.2/9)

$$\langle \frac{dT}{dx} \rangle_r = 2\pi \frac{R}{\dot{m}} \frac{\alpha(x)}{\langle c_p \rangle_r(x)} [T_S(x) - \langle T \rangle_r(x)]. \quad (4.2/12)$$

4.3 Constant surface temperature

If the surface temperature is constant

$$T_S = const \quad \Rightarrow \quad \frac{dT_S}{dx} = 0. \quad (4.3/1a,b)$$

Therefore the derivative (4.3/1b) can be subtracted from the left side of equation (4.2/12). By applying (4.2/8) it follows that

$$\frac{d}{dx} \langle \Delta T \rangle_r(x) = -2\pi \frac{R}{\dot{m}} \frac{\alpha(x)}{\langle c_p \rangle_r(x)} \langle \Delta T \rangle_r(x), \quad (4.3/2)$$

where the following definition has been introduced

$$\langle \Delta T \rangle_r (x) = T_S - \langle T \rangle_r (x). \quad (4.3/3)$$

Equation (4.3/2) is identical to

$$\frac{d}{dx} \log[\Delta \langle T \rangle_r (x)] = -2\pi \frac{R}{\dot{m}} \frac{\alpha(x)}{\langle c_p \rangle_r (x)}. \quad (4.3/4)$$

By an integration one obtains

$$\log[\Delta \langle T \rangle_r (x)] = -2\pi \frac{R}{\dot{m}} \int \frac{\alpha(x)}{\langle c_p \rangle_r (x)} dx + c_1. \quad (4.3/5)$$

The boundary condition at the inlet of the heat exchanging pipe ($x=0$) is

$$c_1 = \log[\Delta \langle T \rangle_r (0)]. \quad (4.3/6)$$

This is inserted into (4.3/5)

$$\log \frac{\Delta \langle T \rangle_r (x)}{\Delta \langle T \rangle_r (0)} = -2\pi \frac{R}{\dot{m}} \int_0^x \frac{\alpha(y)}{\langle c_p \rangle_r (y)} dy. \quad (4.3/7)$$

At the outlet of the heat exchanger ($x=L$) one obtains

$$\log \frac{\Delta \langle T \rangle_r (L)}{\Delta \langle T \rangle_r (0)} = -2\pi \frac{R}{\dot{m}} \int_0^L \frac{\alpha(y)}{\langle c_p \rangle_r (y)} dy, \quad (4.3/8)$$

respectively

$$\frac{\Delta \langle T \rangle_r (L)}{\Delta \langle T \rangle_r (0)} = \exp \left[-2\pi \frac{R}{\dot{m}} \int_0^L \frac{\alpha(y)}{\langle c_p \rangle_r (y)} dy \right]. \quad (4.3/9)$$

Special case (CP3):

$$\langle c_p \rangle_r (x) = c_p = \text{const}, \quad \forall x \in [0, L]. \quad (4.3/10)$$

For an ice slurry this applies when

$$\frac{\dot{q}A_S}{\dot{m}c_{p\min}} \ll 1, \quad c_{p\min} = \min_x \{ \langle c_p \rangle_r (x) \}, \quad x \in [0, L]. \quad (4.3/11)$$

When a large heat capacity per unit time is transported by convection through the tube and the heat impact over the wall is small, then the temperature of the ice slurry alters only slightly, and then the heat capacity is approximately constant.

Now equation (4.3/9) can be simplified

$$\frac{\Delta \langle T \rangle_r (L)}{\Delta \langle T \rangle_r (0)} = \exp \left[-2\pi \frac{RL \langle \alpha \rangle_x}{\dot{m} c_p} \right]. \quad (4.3/12)$$

In this equation the global heat transfer coefficient is introduced

$$\langle \alpha \rangle_x := \frac{1}{L} \int_0^L \alpha(x) dx. \quad (4.3/13)$$

Assuming special case (CP1) the totally transferred heat is

$$\dot{Q} = \langle \rho u h \rangle_r (L) - \langle \rho u h \rangle_r (0) = \dot{m} [\langle h \rangle_r (L) - \langle h \rangle_r (0)]. \quad (4.3/14a,b)$$

Because of the constant specific heat c_p one obtains

$$\dot{Q} = \dot{m} c_p [\langle T \rangle_r (L) - \langle T \rangle_r (0)]. \quad (4.3/15)$$

By adding and subtracting the surface temperature T_S this equation evolves to

$$\dot{Q} = \dot{m} c_p \{ [T_S - \langle T \rangle_r (0)] - [T_S - \langle T \rangle_r (L)] \}. \quad (4.3/16)$$

Substituting (4.3/3)

$$\dot{Q} = \dot{m} c_p [\Delta \langle T \rangle_r (0) - \Delta \langle T \rangle_r (L)]. \quad (4.3/17)$$

From (4.3/8), (4.3/10) and (4.3/13) one concludes that

$$\log \frac{\Delta \langle T \rangle_r (L)}{\Delta \langle T \rangle_r (0)} = -2\pi \frac{RL \langle \alpha \rangle_x}{\dot{m} c_p}. \quad (4.3/18)$$

Solving for the heat capacity per time unit one obtains

$$\dot{m} c_p = -2\pi \frac{RL \langle \alpha \rangle_x}{\log \frac{\Delta \langle T \rangle_r (L)}{\Delta \langle T \rangle_r (0)}}. \quad (4.3/19)$$

This is inserted into equation (4.3/17). Then we obtain

$$\dot{Q} = 2\pi RL \langle \alpha \rangle_x \langle \Delta T \rangle_{r,\log}, \quad \langle \Delta T \rangle_{r,\log} := \frac{\Delta \langle T \rangle_r (L) - \Delta \langle T \rangle_r (0)}{\log \frac{\Delta \langle T \rangle_r (L)}{\Delta \langle T \rangle_r (0)}}, \quad (4.3/20a,b)$$

which is the logarithmic-mean temperature difference (e.g. see Ref. [4] and [5]).

5. EXPERIMENTAL SET-UP

The objective of the project group FIFE at the EIVD in Yverdon mainly consists of the determination of the heat transfer coefficients of fine-crystalline ice slurries in axisymmetric heat exchangers for various operating conditions. The device is shown in figure 3.



Figure 3: The experimental device at the thermodynamics laboratory of the EIVD. From the left to the right there is a refrigerating machine (with an ice generator), a storage tank and an apparatus with a horizontal insulated heat exchanger pipe mounted on the top. This pipe can be inclined in steps of 15°.

Video and photographic pictures of flows are performed through a microscope. On each picture an ensemble of circa twenty ice particles can be observed.

The test facility “Coulis de glace”, with a refrigerating power of 9 kW, was designed and built at the thermal laboratory of engineering (LGT). A cylindrical aluminium heat exchanger with electric heating elements was designed, calculated and built together (see figure 4). This is our main part for the investigations of heat transfer phenomena in pipes.

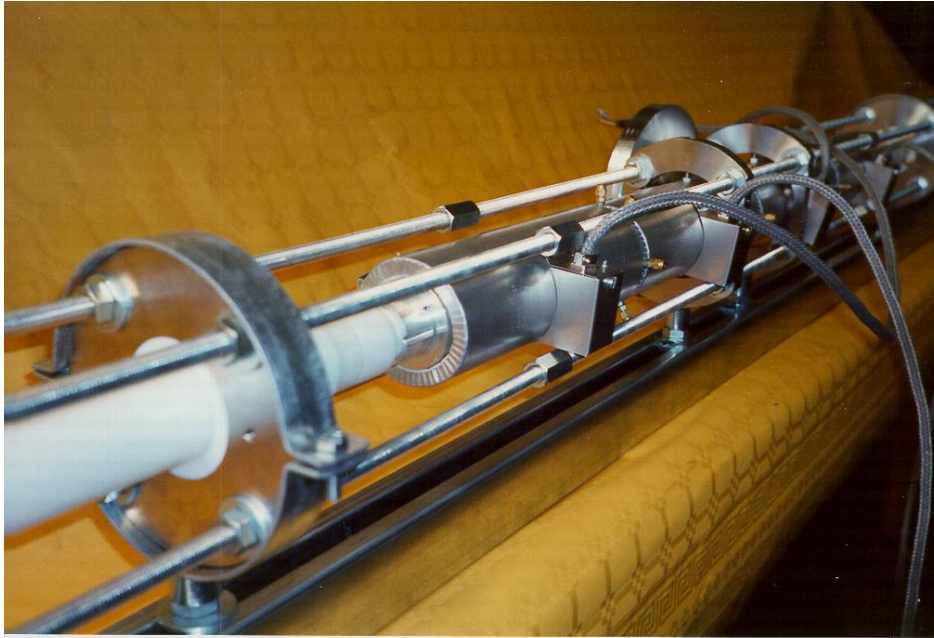


Figure 4: The test facility to measure heat transfer coefficients and temperature profiles in tubes.

The heat exchanger is composed of three distinct elements and has a total length of three meters. The aluminium tube, where the heat transfer occurs, has a total length of one meter. The interior diameter is 23 mm and the outside diameter 60 mm. The cylinder is surrounded by electric heating elements with a maximum total heating power of 9 kW. An input and output insulating piece, each composed of three elements of teflon, have an interior diameter of 23 mm and a length of 333 mm. They are added to obtain steady flow conditions at the inlet and outlet of the heat exchanger.

The electric heating of the pipe leads to a constant heat flux density in the downstream direction. Another research group – see Christensen and Kauffeld (Ref. [7]) - has realised a constant surface temperature boundary condition in their experimental apparatus. This is achieved by an evaporation of a refrigerant on the outer surface, e.g. of a pipe.

The fluid temperatures and the temperatures at the interior surface in each of five sections of the heat exchanger are measured and determined.

In the apparatus all the temperature measurements are carried out with Pt 100 sensors. For preliminary tests, as described in Ref. [6], water was used as a working fluid.

The specific heat is determined by measuring the power of the electric heating, the mass flow at the inlet and the fluid temperatures in the inlet and outlet of the heat exchanger. Four temperature sensors are placed at equidistant positions over one section to evaluate a space-averaged value. The measurements are confirmed by sensors with higher inertia and with a sensor length equivalent to the section diameter. The relative fluctuations of the temperature signals are approximately 0.17 % (taking the average temperature as a basis for the calculation).

6. MEASUREMENTS

6.1 Specific heat

The specific heat is determined and compared with results of the physical properties model (see Ref. [3]). The agreement of the compared experimental and numerical data is good (see figure 5).

The numerically calculated specific heat reaches a maximum value of 53.200 kJ/kg and the measured value is 46.700 kJ/kg K. The temperature at which the specific heat increases corresponds to the expected freezing starting point of water/ethanol-ice solidification. Above the freezing region the calculated specific heat is constant, with a value of 4.353 kJ/kg K, and in comparison a mean value of 4,156 kJ/kg K is measured. On the left of the maximum the measured specific heat also decreases in agreement with numerical results.

During a cooling process with approximately constant cooling power in the single-phase liquid mixture the temperature decreases rapidly, because only sensible heat must be released. When the freezing domain is reached also latent heat must be removed, therefore, the temperature decreases more slowly. This leads to a kink, when the ice creation sets in. This kink is taken to determine the freezing point. With 11 mass- % ethanol content the ice particles are formed at -4.48°C (see figure 6).

Numerical calculation results show an onset of freezing at a temperature of -4.45°C . The discontinuity in the calculated specific heat occurs at this temperature (see figure 5).

All measurements are obtained under steady state conditions. The heat flux and flow rate have no influence on the specific heat, as it is confirmed by the different measurement series in figure 5. For the serie No. 1 the heat flux is 78200 W/m^2 and the flow rate 0.70 kg/s ,

Specific heat
(kJ/kg K)

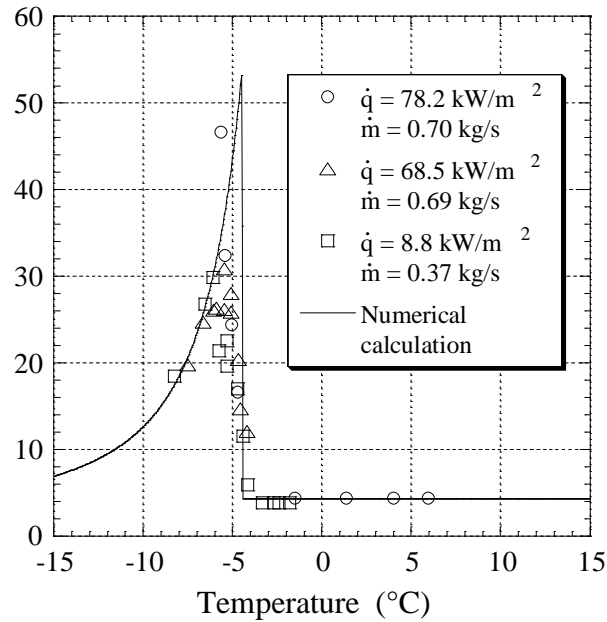


Figure 5: The specific heat of an ice slurry produced with a 11 mass-% water/ethanol mixture. The figure shows a comparison between experimental data and theoretical results.

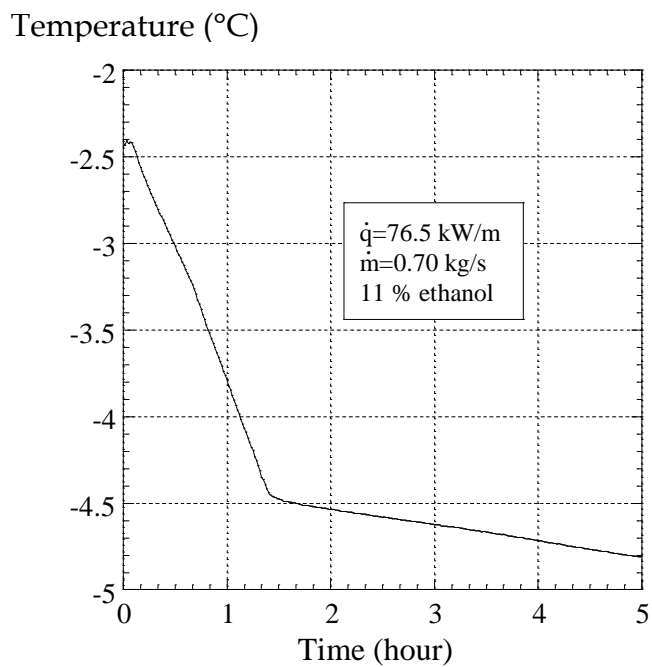


Figure 6: Temperature measurement during the cooling process. Water solidification is characterized by a change of the slope of the temperature in function of time. This phenomena permits to determine the temperature when freezing occurs very exactly.

for the serie No. 2 the heat flux is 68500 W/m^2 and the flow rate $0,69 \text{ kg/s}$. For the serie No. 3 the experimental quantities are 8800 W/m^2 and $0,37 \text{ kg/s}$. The specific heat is directly related to the slope of the enthalpy (see Ref. [3]). At the freezing point the discontinuous increase of the specific heat corresponds to a kink in the enthalpy curve.

6.2 Velocity profile

The Hagen-Poiseuille flow of Newtonian fluids is characterized by a parabolic velocity profile. Bingham fluids show a plug in the velocity profile, which is located symmetrically

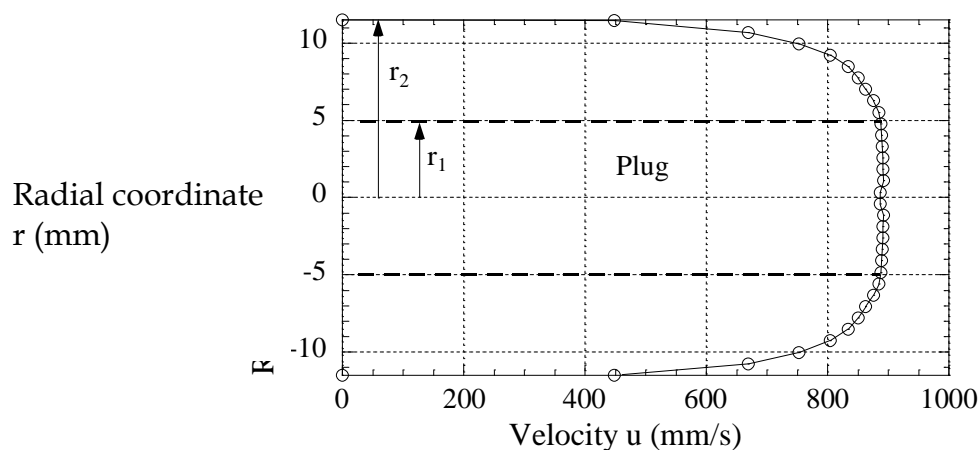


Figure 7: Velocity profile of ice slurry composed with 11 mass-% water/ethanol in a cylindrical tube.

to the axis (see figure 7). This velocity profile was obtained with an ultrasonic velocity profile meter (UVP) applied to a flow in a plexiglas tube. A laminar flow of 0.35 kg/s with a temperature of $-4.6 \text{ }^\circ\text{C}$ constitutes the experimental condition for this acquisition. The experimental result has been obtained by taking an average of 1028 successive profiles. In Ref. [8] the formulas of a laminar Bingham flow are presented. It can be seen that an experimental determination of r_1 together with the pressure derivative in the downstream direction allows the calculation of the dynamic viscosity and critical shear stress. In a diploma work, performed by Didier Vuarnoz, this method is used to determine viscous behaviour of an ice slurry as a function of temperature.

6.3 Ice particles visualization

A microscopic visualization of the ice slurry permits to study the geometry of the ice particles (see figure 8). Physical properties as viscosity and critical shear stress and the heat transfer coefficients depend on the ice concentration and maybe also on the size of

the ice particles. The presented visualization of an ice slurry has been performed at -4.7 °C. The photograph shows particles with diameters of the order of $100\ \mu\text{m}$. Statistics on these ice particles will be presented at the next IIR workshop on ice slurries in Paris.

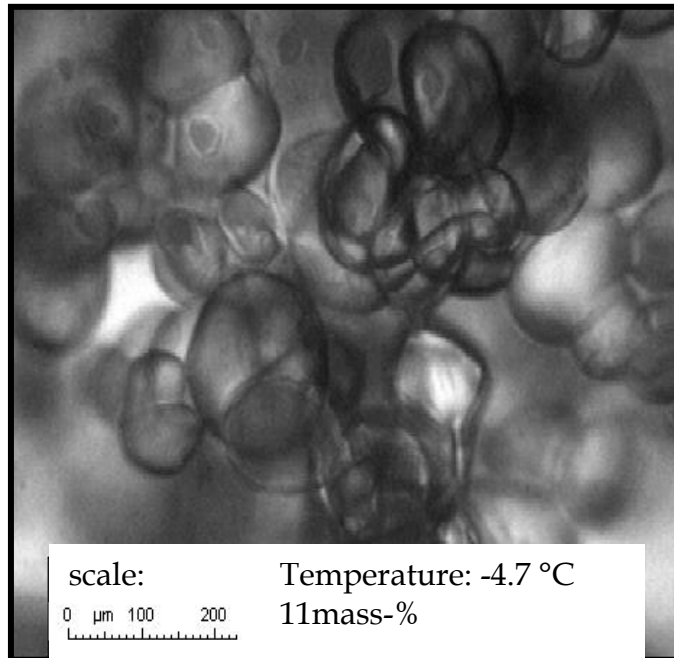


Figure 8: Ice particles in a ice slurry suspension with a temperature of -4.7 °C and 11 mass-% water/ethanol.

7. CONCLUSIONS AND OUTLOOK

A theoretical analysis of the heat transfer problem in a pipe is presented. It shows that one has to be very careful applying well-known relations of the theory of heat exchangers. When a constant heat flux density occurs across the tube wall, a weighted enthalpy density is a linear function of the downstream distance. In general the temperature averaged over the pipe section will not show such linear behaviour downstream. When the surface temperature is constant three conditions (critical points CP1, CP2 and CP3, which have been outlined in this article) must apply. Only then the logarithmic-mean temperature difference is valid. It is certain that in some practical applications this concept is meaningless.

The experimental set-up to measure heat transfer coefficients and temperature profiles in tubes is very briefly discussed. More informations are found in an internal research report of the FIFE group (see Ref. [6]).

First measurements of the specific heat capacity as a function of temperature, performed with the new device "COULIS DE GLACE", are compared with model results. In Ref. [3]

the theoretical model is described in detail. The agreement between experimental data and numerical results is good.

First measurements on velocity profiles with an ultra sound velocity profile meter are very encouraging. An average of an ensemble of 1028 velocity profiles clearly shows a rectangular central part symmetric to the axis and a parabolic decay toward the wall as predicted by the Bingham theory.

Microscopic photographs of ice particles in a water/ethanol ice slurry have been performed. They also show spheroidal shapes similar to those observed by Kauffeld et al. (see Ref. [9]). Their ice particles have been created in a 3 % Mg Cl₂ suspension.

NOMENCLATURE

Standard

A	area
c_p	specific heat
f	positive function
g	positive function
h	enthalpy density
k	thermal conductivity
\dot{m}	mass flow
p	pressure
\dot{q}	heat flux density
r	radial co-ordinate
R	correlation coefficient
R	radius of pipe
t	time
T	temperature
u	velocity
x	space co-ordinate downstream

Greek

α	heat transfer coefficient
α	nonlinear coefficient
β	nonlinear coefficient
χ	general variable
Δ	increment
φ	angle in azimuthal direction
$\dot{\gamma}$	shear velocity

μ	dynamic viscosity
ρ	density
τ	shear stress
τ_0	critical shear stress

Indices

min	minimal
P	cross section of pipe
S	surface of pipe

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