

Identifying a Threshold Impurity Level for Organic Solar Cells: Enhanced First-Order Recombination Via Well-Defined PC₈₄BM Traps in Organic Bulk Heterojunction Solar Cells

Sarah R. Cowan, Wei Lin Leong, Natalie Banerji, Gilles Dennler, and Alan J. Heeger*

Small amounts of impurity, even one part in one thousand, in polymer bulk heterojunction solar cells can alter the electronic properties of the device, including reducing the open circuit voltage, the short circuit current and the fill factor. Steady state studies show a dramatic increase in the trap-assisted recombination rate when [6,6]-phenyl C₈₄ butyric acid methyl ester (PC₈₄BM) is introduced as a trap site in polymer bulk heterojunction solar cells made of a blend of the copolymer poly[N-9'-hepta-decanyl-2,7-carbazole-alt-5,5-(4',7'-di-2-thienyl-2',1',3'-benzothiadiazole) (PCDTBT) and the fullerene derivative [6,6]-phenyl C₆₁ butyric acid methyl ester (PC₆₀BM). The trap density dependent recombination studied here can be described as a combination of bimolecular and Shockley–Read–Hall recombination; the latter is dramatically enhanced by the addition of the PC₈₄BM traps. This study reveals the importance of impurities in limiting the efficiency of organic solar cell devices and gives insight into the mechanism of the trap-induced recombination loss.

1. Introduction

Recent reports in the literature show polymer bulk heterojunction (BHJ) solar cells demonstrate continuing promise: solar cells with near 100% internal quantum efficiency^[1] and with 6%–8% power conversion efficiencies^[1–3] have been demonstrated. There are many challenges to overcome on the way to competitively efficient devices, including the need for a detailed knowledge of charge transport and loss mechanisms in the complex phase-separated BHJ systems. Nevertheless, many of the perceived drawbacks of polymer materials such as atmospheric sensitivity^[4,5] and high band gap^[1–3,6–9] are being overcome, both in the lab and in field testing.

The BHJ material is commonly implemented by casting a polymer:fullerene blend from solution, resulting in distributed

junctions between the polymer donor and fullerene acceptor interfaces. The fullerene derivatives [6,6]-phenyl-C₆₁-butyric acid methyl ester (PC₆₀BM) and [6,6]-phenyl-C₇₁-butyric acid methyl ester (PC₇₁BM) have been widely used as the electron-accepting material. Small amounts of these fullerene materials have been found to quench luminescence in the highly absorptive and fluorescent semiconducting polymer materials.^[10] When mixed in large fraction, charge transfer from the polymer to the fullerene derivative is efficient, ultrafast, and largely irreversible.^[10–15] Ultrafast and highly efficient charge transfer is essential to high efficiency BHJ solar cell operation.

In 1985, H. W. Kroto, R. F. Curl, R. E. Smalley, et al. detected C₆₀ in carbon vapor from laser ablation of graphite.^[16]

Since the synthesis of macroscopic quantities of C₆₀, C₇₀, and the higher fullerenes in the early 1990s,^[17,18] fullerenes and soluble fullerene adducts have become highly studied research materials. The synthesis of PC₆₀BM in 1995^[19] led to the first polymer:fullerene bulk heterojunction solar cell^[20] within the year. Use of polymers or other small molecules as the electron-accepting material have not yet been as successful.^[21,22] [6,6]-phenyl-C₈₄-butyric acid methyl ester (PC₈₄BM) became briefly interesting as a solar cell acceptor^[23] due to its strong, broad absorption, and relatively high mobility.^[24] However, it was abandoned as an acceptor due to observed loss of photocurrent and increased recombination.

The electronic effect of traps and disorder – material properties inherent to polymer blend systems – has been studied in relation to the performance of organic light-emitting diodes,^[25] organic thin-film transistors,^[26] and organic photovoltaics.^[22,27–29] However, the role of impurities in the recombination process and the attainable open circuit voltage has only received moderate attention. In 2007, Blom and coworkers observed the modification of the open circuit voltage and its light intensity dependence due to trap-limited electron transport and trap-assisted recombination.^[29] In this work, we explore the effect of impurities via the controlled introduction of the PC₈₄BM molecule and propose a simple model which takes into account the density of trap states and the intensity-dependence

Dr. S. R. Cowan, Dr. W. L. Leong, Dr. N. Banerji, Prof. A. J. Heeger
Center for Polymers and Organic Solids
University of California
Santa Barbara, California, 93106, USA
E-mail: ajhe@physics.ucsb.edu

Dr. G. Dennler
IMRA Europe 220, rue Albert Caquot - BP 213 06904 SOPHIA-
ANTIPOLIS Cedex, France

DOI: 10.1002/adfm.201100514

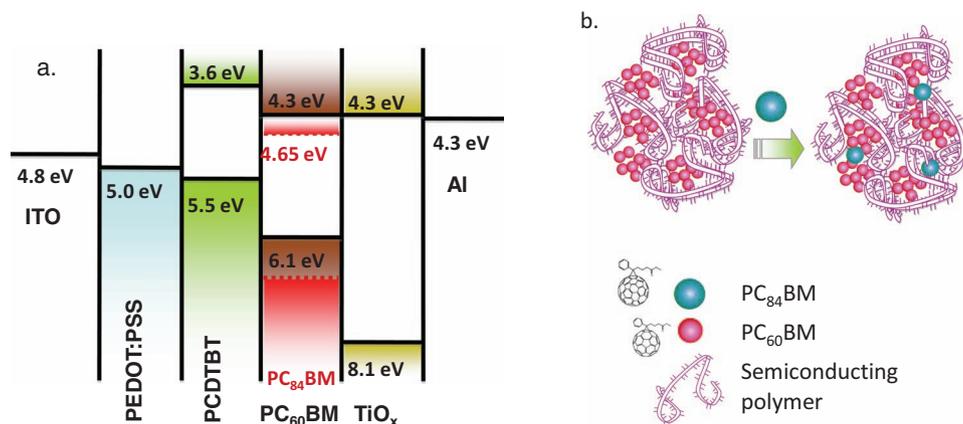


Figure 1. a) Energy level diagram of the device components with the LUMO level of PC₈₄BM (dotted line) within the band gap of the BHJ; b) schematic illustration of the addition of PC₈₄BM in small mass fraction to the polymer bulk-heterojunction solar cell; legend shows the chemical structure of PC₈₄BM and PC₆₀BM.

of two competing recombination processes: bimolecular and trap-assisted recombination. The present work is specifically designed to provide a road map toward optimizing material synthesis/purification for organic photovoltaics. We identify a threshold impurity level at which impurities begin to affect the device characteristics via the controlled introduction of an impurity energy state in the interfacial band gap.

The LUMO level of PC₈₄BM is 0.35 eV lower than that of PC₆₀BM (4.3 eV),^[23] as shown in **Figure 1a**. In the dilute concentrations used in this study, 0.01%, 0.1%, and 1% (and possibly even 10%) by weight, the PC₈₄BM molecule is expected to function as a localized electron trap within the fullerene component of the BHJ material, pictured in **Figure 1b**. The structure of PC₆₀BM and the likely configuration of the d2 isomer of PC₈₄BM^[23] are included in the **Figure 1b** legend.

The presence of PC₈₄BM in the BHJ matrix composed of the copolymer poly[N-9''-hepta-decanyl-2,7-carbazole-alt-5,5-(4',7'-di-2-thienyl-2',1',3'-benzothiadiazole) (PCDTBT) and PC₆₀BM is expected to increase charge trapping and recombination. Several physical processes may be facilitated by a trap state in the interfacial band gap:

1. Photons may be absorbed directly into the trap state energy. An excitation in the trap state may relax to the ground state or if charge separation occurs, the excitation may be converted to charge carriers.
2. Excitations in the polymer may be separated at a polymer: trap site interface, creating charge in the trap state.
3. Mobile charge carriers in the BHJ may be trapped by the impurity depending on the capture cross section of the trap.

Once trapped, the charge has a finite probability of de-trapping via a photon- or phonon-assisted transition back to the LUMO level or a finite probability of recombination to the ground state. The trapping/de-trapping process has been observed in ongoing time-resolved transient photoconductivity studies and impedance spectroscopy measurements,^[30] and will be clarified on the ultrafast time scale in this paper.

The presence of band tails in disordered organic materials means that thermalization of mobile charge likely precedes

a recombination event because these two processes occur at largely different time scales – femto/picoseconds^[31,32] vs. nano/microseconds, respectively.^[32–34] Because a recombination event necessarily involves an electron and hole, recombination rates are related to the spatial correlation between the two charge species. Unequal charge density of electrons and holes or a lack of spatial correlation between the charge species due to material disorder can lead to a recombination process rate-limited by one charge species. Classical studies of recombination from inorganic materials which describe recombination through a single defect state in the energy gap (Shockley-Read-Hall)^[35–37] and from multiple energy levels like those found in disordered materials (Sah-Shockley)^[38] will be discussed further in this paper. Recombination, trap-assisted or otherwise, in which electrons and holes have equal rates of recombination are characterized as second order, while trap-assisted or rate-limited recombination in which electrons and holes have unequal recombination rates can be characterized by a lower order.^[39]

In order to study the recombination mechanism in detail, by tailoring both charge trapping and recombination dynamics, we probe this system via steady state current-density voltage (*J*-*V*) measurements, intensity dependent *J*-*V* measurements, internal photon-to-electron conversion efficiency, and steady-state and time-resolved absorption measurements at short circuit and near open circuit where recombination plays a dominant role.

2. Increase in Recombination Due to Addition of PC₈₄BM Trap States

Figure 2a shows the *J*-*V* curves under 100 mW cm⁻² AM1.5G illumination from a filtered xenon lamp for representative devices constructed from PCDTBT:PC₆₀BM containing the PC₈₄BM impurity. **Figure 2b** shows the *J*-*V* curves in the dark for the same devices. The device structure is depicted in the inset in **Figure 2b**. Details of device fabrication and testing procedures are included in the Experimental Section. The extracted parameters from the *J*-*V* curves for these devices are listed in

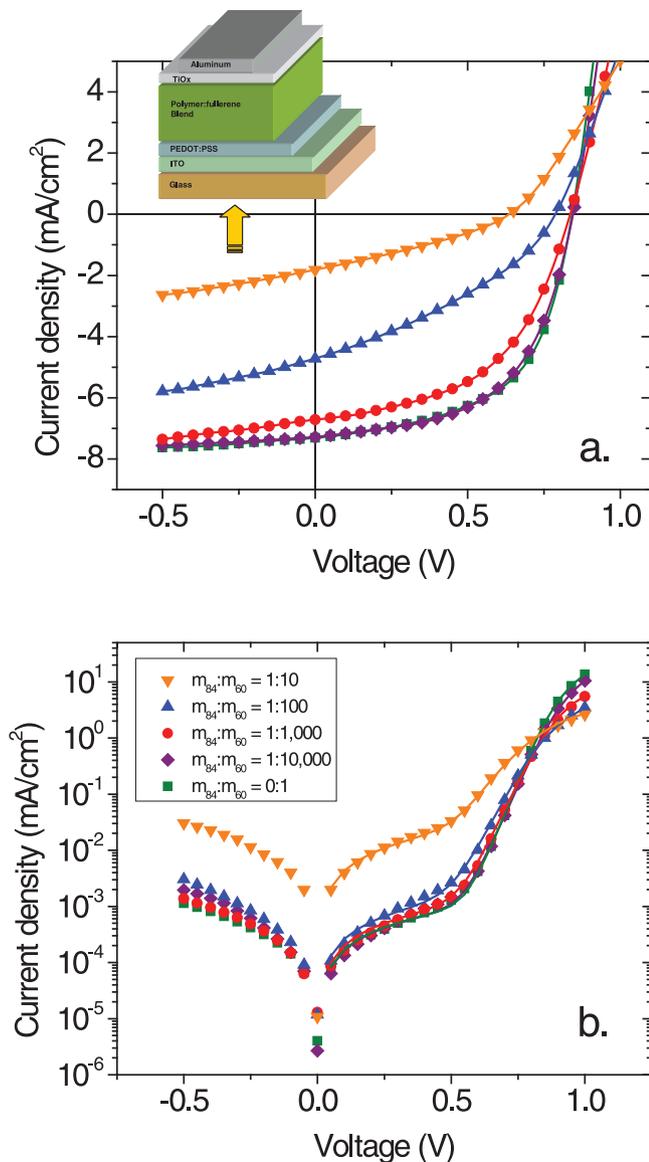


Figure 2. a) Steady state current density vs. voltage (J - V) curves for BHJ solar cell devices with small amounts of PC_{84}BM trap states under illumination by a 100 mW cm^{-2} solar simulator (lines are guides to the eye). b) Steady state dark current density vs. voltage in the dark (lines are fits to the Shockley diode equation) for BHJ solar cell devices with small amounts of PC_{84}BM trap states. Inset: Device structure of the BHJ PCDTBT: PC_{60}BM solar cells used in this study.

Table 1. The PC_{84}BM is added in mass fraction to the dissolved PCDTBT: PC_{60}BM solution, in the mass ratios indicated in Figure 2: pristine, 1 part in 10 000, 1 part in 1 000, 1 part in 100, and 1 part in 10. PC_{84}BM in lower weight concentrations, e.g., $m_{84}:m_{60} = 1:10,000$, are observed to have virtually no electronic effect on device performance. Devices made with $m_{84}:m_{60} = 1:1,000$ show a small but observable reduction in short circuit current, J_{sc} , and fill factor, FF . When the PC_{84}BM concentration increased to the ratio of $m_{84}:m_{60} = 1:100$ and $m_{84}:m_{60} = 1:10$, the short circuit current, fill factor, and the open circuit voltage are significantly reduced indicating enhanced recombination

Table 1. Light J - V characteristics for the devices used in this study.

Material	J_{sc} [mA cm^{-2}]	FF	V_{oc} [V]	PCE [%]	m
PCDTBT: PC_{60}BM pristine ($m_{84}:m_{60} = 0:1$)	7.31	0.61	0.83	3.67	1.15
PCDTBT: PC_{60}BM with PC_{84}BM ($m_{84}:m_{60} = 1:10,000$)	7.29	0.57	0.82	3.42	1.27
PCDTBT: PC_{60}BM with PC_{84}BM ($m_{84}:m_{60} = 1:1,000$)	6.71	0.59	0.82	3.22	1.34
PCDTBT: PC_{60}BM with PC_{84}BM ($m_{84}:m_{60} = 1:100$)	4.71	0.40	0.77	1.47	1.47
PCDTBT: PC_{60}BM with PC_{84}BM ($m_{84}:m_{60} = 1:10$)	1.82	0.34	0.64	0.40	1.62

loss. Although the J - V curve for devices made with $m_{84}:m_{60} = 1:10$ are included in Figure 2a–b, we note that aggregation and associated changes in morphology would be expected at this high concentration of PC_{84}BM .

The Shockley diode model modified for organic solar cells^[40] is fit to the measured dark J - V current of these devices in Figure 2(b). The fit parameters to the diode dark current are listed in Table 2. Details of the dark current fitting are included in the Supplementary Information. Analysis of the dark current from these devices confirms the qualitative analysis of the illuminated J - V measurement. The shunt resistance decreased by a factor of 25 while the series resistance increased by a factor of 12 by adding 10% by weight of the PC_{84}BM molecule.

Under forward bias, the diode dark current is mainly shaped by recombination. An efficient diode will have a high turn-on voltage above the band gap and efficiently conduct charge injected from the contacts. Practical diodes see a more limited charge injection efficiency due to charge trapping and recombination effects. The slope of the dark J - V characteristic above the turn-on voltage depends on two parameters: the diode ideality factor, n , and the reverse saturation current, J_0 . Both parameters are observed to increase significantly with increasing concentrations of PC_{84}BM , indicating that the addition of the trap state reduces the “quality” of the BHJ diode. The diode ideality factor, which phenomenologically describes the exponentiality of the diode dark current in forward bias, increases from $n = 1.4 \pm 0.2$ in the pristine device to $n = 2.0 \pm 0.2$ in the 10%

Table 2. Dark J - V fit parameters for the devices used in this study.

Material	J_0 [mA cm^{-2}]	n	R_s [Ω]	R_{SH} [Ω]	RMS error [mA cm^{-2}]
PCDTBT: PC_{60}BM pristine ($m_{84}:m_{60} = 0:1$)	2.0×10^{-10}	1.41	6.6	5.90×10^5	0.082
PCDTBT: PC_{60}BM with PC_{84}BM ($m_{84}:m_{60} = 1:10,000$)	1.0×10^{-9}	1.54	7.4	6.05×10^5	0.127
PCDTBT: PC_{60}BM with PC_{84}BM ($m_{84}:m_{60} = 1:1,000$)	4.0×10^{-9}	1.65	18.3	5.00×10^5	0.090
PCDTBT: PC_{60}BM with PC_{84}BM ($m_{84}:m_{60} = 1:100$)	2.6×10^{-8}	1.80	37.0	3.80×10^5	0.098
PCDTBT: PC_{60}BM with PC_{84}BM ($m_{84}:m_{60} = 1:10$)	7.2×10^{-7}	2.00	82.0	2.30×10^4	0.051

by weight trap device. Increasing values of the ideality factor have been considered signature of increased recombination loss or increased disorder in the electronic states.^[41] The relation between the ideality factor and recombination processes in p-n junctions have been well-studied, both recently in organic photovoltaics^[42–44] and for over 50 years in inorganic p-n junctions.^[45–48] The diode ideality factor was first introduced by W. Shockley and colleagues in order to quantitatively describe the effect of Shockley–Read–Hall recombination in the bulk (as opposed to the interfaces) of a p-n junction.^[46] The effect of recombination on the diode current through surface or interface states has also been observed to lead to deviations of the ‘ideal’ ideality factor ($n = 1$) to the non-ideal case ($n = 2$).^[45,47] Here, we attribute the increase in the diode ideality factor to an increase in interfacial trap-assisted trapping and recombination due to the introduced PC₈₄BM recombination centers.

It is a well-studied phenomenon that the diode ideality factor is different in the light (m , Table 1) and the dark (n , Table 2).^[49,50] In this paper, we find this difference may be due to charge density dependent recombination mechanisms. An observation of orders of magnitude different charge density at short circuit and open circuit conditions has been put forward by Shuttle et al.^[51] and supported by Cowan et al.^[52] Increased charge densities will lead to increased correlation between the two charge species and an increased probability of recombination, especially bimolecular recombination due to the recombination rate’s square dependence on charge density for bimolecular recombination vs. a lower order for trap-assisted recombination. However, the marked increase in the diode ideality factor in the dark qualitatively reflects enhanced recombination due to the PC₈₄BM trap states or increased electronic disorder in the BHJ.

Internal photon-to-electron conversion efficiency (IPCE) measurements taken at short circuit ($V = 0$ V), and near open circuit ($V = 0.8$ V for the pristine, and $m_{84}:m_{60} = 1:10,000$ and $1:1,000$ devices, $V = 0.75$ V for the $m_{84}:m_{60} = 1:100$ device, and $V = 0.6$ V for the $m_{84}:m_{60} = 1:10$ device) are shown in Figure 3a,b. IPCE measures the percentage of incoming photons that are collected as electrons or holes at the electrodes. IPCE loss is due to i) photons not absorbed in the BHJ (reflected from the device or absorbed in a non-photoconductive layer), ii) bound electron-hole pairs which recombine before reaching, or at, the charge separating interface, and/or iii) mobile electrons/holes which recombine with holes/electrons within the BHJ material. The IPCE data at short circuit and near open circuit (see Figure 3a,b) show that the addition of the PC₈₄BM traps decreases the efficiency of photon to electron conversion. These data contrast with the observed absorption spectra in the Supplementary Information which show very similar absorption in the BHJ layer with and without the addition of 0.01–10% by weight of PC₈₄BM traps. The absorption of neat films of PCDTBT, PC₈₄BM, and PC₆₀BM are also plotted for reference. The two fullerene materials have similar absorption spectra both in magnitude and in wavelength dependence, accounting for the relatively small change in absorption spectra between the devices with and without traps. The observed decrease in IPCE efficiency at short circuit is attributed to the PC₈₄BM traps, which hamper charge transport due to enhanced disorder and reduce charge carrier mobility in the bulk heterojunction

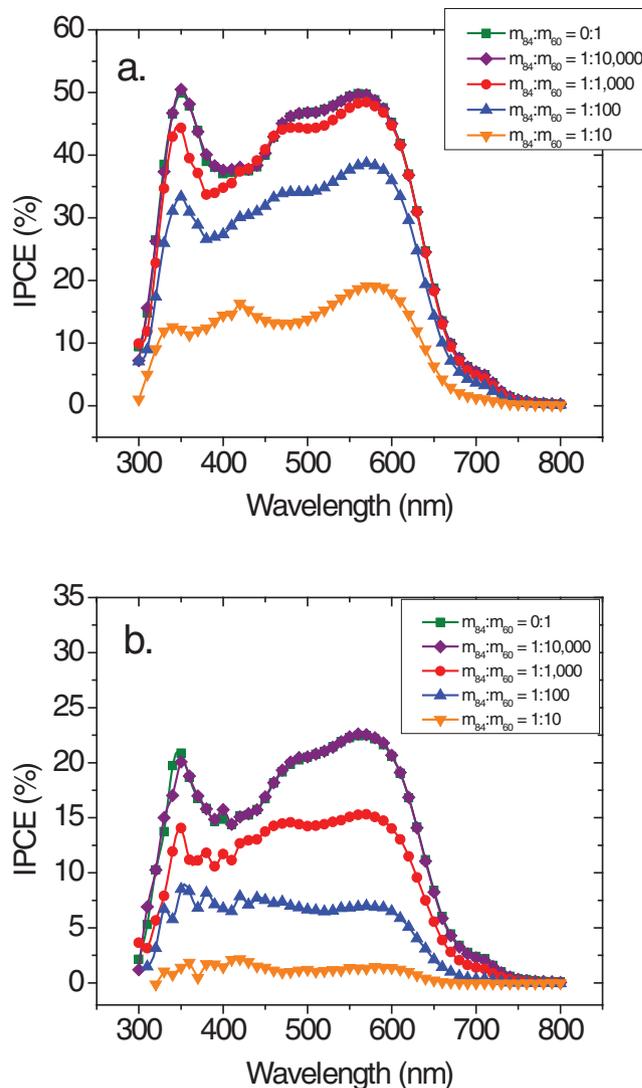


Figure 3. Internal photon conversion efficiency versus wavelength at a) short circuit conditions ($V = 0.0$ V) and b) approaching open circuit ($V = 0.8$ V for $m_{84}:m_{60} = 0:1$, $1:10,000$, and $1:1,000$, $V = 0.75$ V for $m_{84}:m_{60} = 1:100$, and $V = 0.6$ V for $m_{84}:m_{60} = 1:10$). The PC₈₄BM traps increase recombination in the BHJ solar cell devices, as does the decreasing internal field near open circuit.

material such that only a fraction of the carriers generated are swept out prior to recombination. Near open circuit, the conversion efficiency is even lower – with and without PC₈₄BM – due to a decrease in the built-in electric field.^[32]

IPCE measurements in Figure 3b were collected under an applied voltage 20–30 mV below the open circuit condition. In this measurement, the applied voltage nearly cancels the built-in field of the device. In a reduced electric field, the time needed for charge transport and collection is increased along with the probability of recombination. These data are consistent with the poor fill factor of the trap-filled devices in Figure 2a. According to the data in Figure 3, near open circuit, where the internal field is nearly zero, more than half of all photogenerated charge recombine before reaching the electrodes.

The total current density, J , in the solar cell is defined as the sum of the dark current, J_{dark} , and the photogenerated current, J_{photo} ,

$$J(I, V) = J_{\text{photo}}(I, V) + J_{\text{dark}}(V) \quad (1)$$

where the photogenerated current can be expressed as the product of the photon flux per unit volume, G , the thickness of the BHJ layer, d , the elementary charge, e , and the charge collection probability, P_C : $J_{\text{photo}}(I, V) = edG(I)P_C(I, V)$. IPCE measurements employ a chopped light source and lock-in detection and therefore measure only the photogenerated current. Hence at constant illumination, IPCE directly measures the probability of charge collection as a function of applied voltage. The IPCE measurements in Figure 3b show that charge collection probability decreases near the open circuit condition. Comparing the data in Figure 3a and 3b, IPCE is reduced by more than a factor of 2 near the open circuit condition with addition of 1% PC₈₄BM, clear evidence that the PC₈₄BM traps enhance recombination.

3. Ultrafast Transient Absorption Measurements

Our conclusions are confirmed via time-resolved transient absorption (TA) measurements on neat films (no metal contacts or device structure) of the BHJ material with the inclusion of small amounts of PC₈₄BM, shown in Figure 4. The BHJ samples were excited by a femtosecond pulsed laser at 400 nm.

The photoinduced absorption band between 820–1200 nm, which appears within the 150 fs time resolution, has previously been assigned in PCDTBT:fullerene blends to the signature of charged polarons.^[15] The contribution from absorption of directly excited PC₈₄BM molecules was found to be small in this spectral region by comparison with TA spectra of a pristine PC₈₄BM film. The decay dynamics of the polarons probed at 900 nm with different PC₈₄BM concentrations in the BHJ material are shown in the top panel of Figure 5. Without additional traps, the polarons decay slowly in the ns regime (with a 1.9 ns and a longer time constant). This slow decay appears as a plateau in the probed time window (τ_3 and τ_4). It should be noted that there is a large error on the value and relative amplitude of $\tau_3 = 1.9$ ns, since TA dynamics were only recorded up to 1.6 ns and there might have been some delay stage misalignment at long time-delays. Therefore, the amplitudes of τ_3 and τ_4 are summed in the table inset of Figure 4. With increasing PC₈₄BM concentration, fast polaron decay components of 2.3 ps and 60 ps appear and gain amplitude. They lead to 50% of polaron loss in the $m_{84}:m_{60} = 1:10$ sample and are ascribed to polaron trapping and/or recombination to the ground state via the added PC₈₄BM. There is also indication that the 1.9 ps component increases in amplitude relative to the long τ_4 plateau,

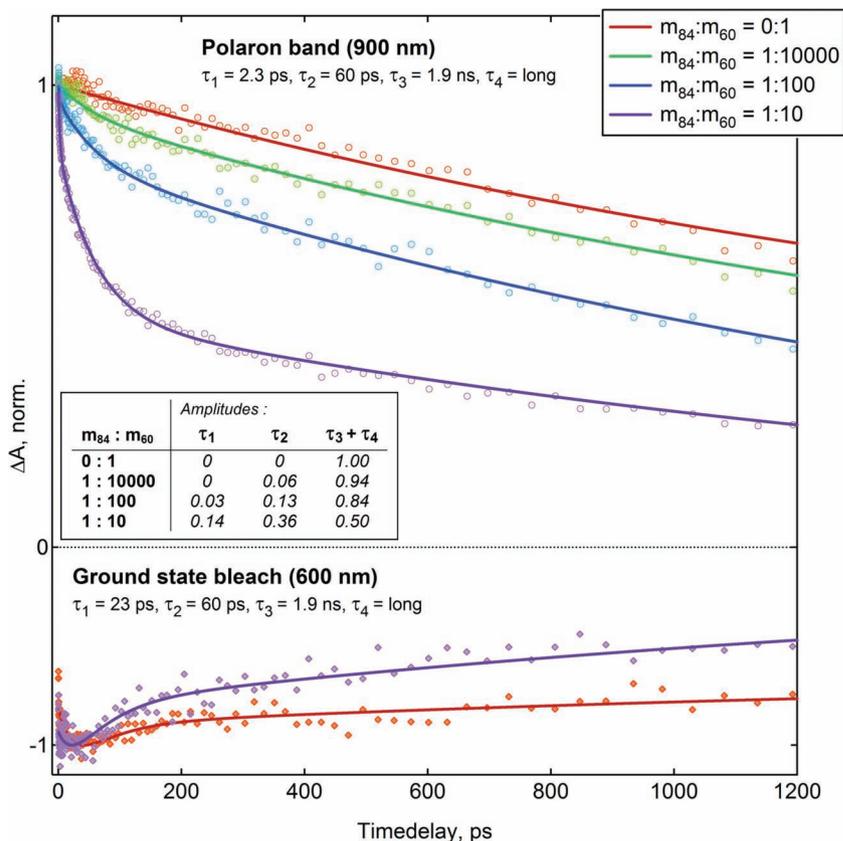


Figure 4. Femtosecond transient absorption dynamics of neat films of PCDTBT:PC₆₀BM with and without the PC₈₄BM molecule after excitation at 400 nm. The top panel shows the evolution of the polaron signature at 900 nm, while the bottom panel shows the polymer ground state recovery. Solid lines are best global multiexponential fits (time constants and 900 nm amplitudes are also shown). Increased polaron trapping and trap-induced recombination occurs in the presence of PC₈₄BM.

pointing, in addition, to slower trapping/recombination processes in presence of added PC₈₄BM.

Probing the PCDTBT ground state bleach at 600 nm reveals that there is some ground state recovery (evidence of recombination) with the 60 ps component, even in the sample without additional PC₈₄BM traps (bottom panel of Figure 4). This has previously been attributed in the PCDTBT:PC₇₁BM blend to recombination from interfacial trap states, which are populated in <1 ps during the initial charge separation.^[15] Figure 4 clearly shows that recombination to the ground state is increased with additional PC₈₄BM traps (1:10), as both the 60 ps and 1.9 ns decay components gain amplitude. This confirms that loss of polarons probed at 900 nm is not only due to trapping, but also partly due to recombination to the ground state from those traps. The different recombination time scales (most recombination occurs slower than can be measured here) are due to traps that are directly occupied during charge separation (following excitation of the polymer or of PC₈₄BM) and traps that are populated more slowly by mobile polaron trapping. Note that the 23 ps rise of the ground state bleach is caused by a spectral shift due to early excited-state relaxation. In summary, the TA experiments provide direct evidence that the PC₈₄BM molecules in

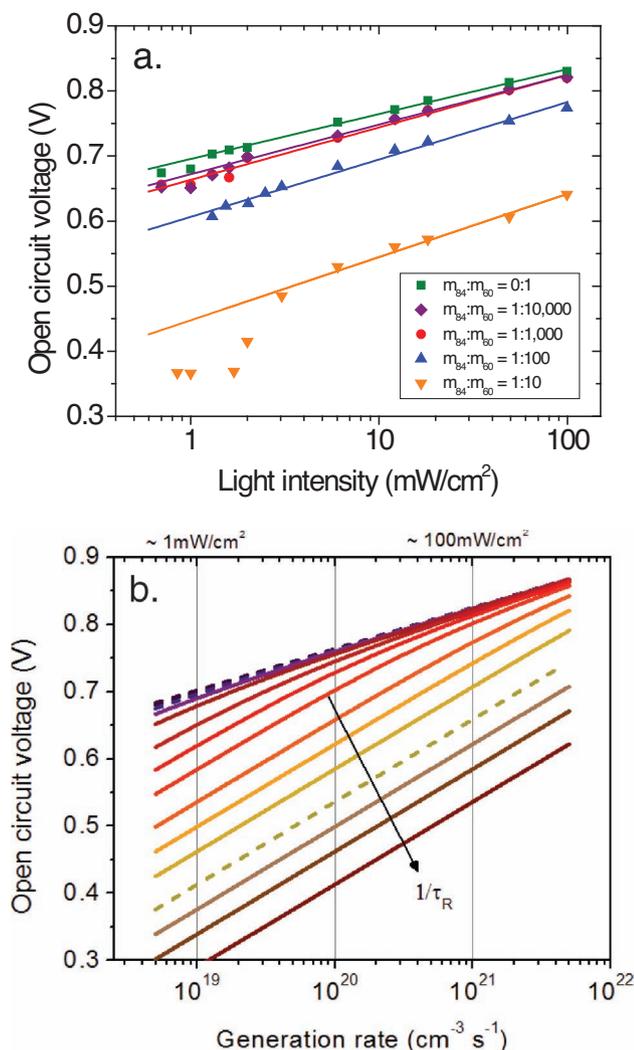


Figure 5. a) Light intensity dependence of the open circuit voltage (V_{oc}) measured in the pristine device and with the addition of small concentrations by weight of PC₈₄BM. Solid lines are the best fit by linear least squares to the data. b) Generation rate dependence of the open circuit voltage by the bimolecular/SRH recombination model. Color scale from purple to brown indicates logarithmically increasing trap densities, with “slope” tending from kT/e (dotted purple line) to $2kT/e$ (dotted tan line).

the PCDTBT:PC₆₀BM matrix act as recombination centers for mobile carriers.

4. Trap-Assisted Recombination at the Open Circuit Condition

In the following section, we present and discuss data obtained from intensity dependent steady-state measurements in order to determine the recombination order in the pristine and the trap-containing devices.

Recombination has been studied extensively over the past 10 years in polymer solar cells and in particular, in polymer:fullerene mixtures. First order (geminate,^[53,54] trap-based,^[29,55–58] and rate-limited bimolecular) and second order (bimolecular)^[50–52,59,60] recombination are expected to play a

role in the complex dynamics of charge separation, transfer, and transport. Geminate recombination has been shown to play a negligible role in optimized organic devices.^[54,61] Bimolecular recombination has been found to be a dominant recombination mechanism for a number of polymer:fullerene devices. When the energy gap is large compared to the thermal energy, the bimolecular recombination rate, R_b , is proportional to the product of the density of photogenerated electrons, n_e , and holes, n_h ,

$$R_b = \gamma n_e n_h \quad (2)$$

where the BJJ bimolecular recombination rate constant, $\gamma = \zeta \frac{\epsilon(\mu_e + \mu_h)}{\epsilon}$, has been found to be reduced by orders of magnitude from the theoretical Langevin recombination coefficient,^[62] where ζ is the reduction factor, $\mu_{e(h)}$ is the charge mobility, and ϵ is the effective dielectric constant of the material.

Herein, we wish to probe the threshold impurity level required for trap-assisted recombination to play a dominant role in these devices via intentionally introduced impurity states. Intensity dependent measurements have been shown to provide insight into the order of the recombination mechanism in recent studies.^[39,50,52] Shockley, Read, and Hall (SRH) described recombination through a single-electron trap with a discrete energy level. There are two widely used extensions of SRH trap theory: i) Sah-Shockley theory describes recombination through trap states with multiple charge levels;^[38] ii) the Taylor and Simmons approximation describes recombination via a continuum of energy states as found in disordered materials.^[63] Theoretically, these mathematical treatments are interesting to consider in terms of the disordered bulk heterojunction. However, our modeling results show that the simplified version of SRH theory is sufficient to describe the effects observed here. All the PC₈₄BM molecules function as similar energy trap states, hence a single trapping level is more likely as a first approximation, rather than a distribution of trap states. The SRH recombination rate, R_t , can be written as

$$R_t = \frac{n_e n_h - n_i^2}{\tau_e (n_e + n_{e,traps}) + \tau_h (n_h + n_{h,traps})} \quad (3)$$

where $n_{e(h)}$ is the density of electrons(holes) in the LUMO(HOMO) of the fullerene(polymer), $n_i^2 = n_e^0 n_h^0$ where n_e^0 is the density of electrons in the LUMO of the fullerene at equilibrium, n_h^0 is the density of holes in the polymer HOMO at equilibrium, and n_i is the intrinsic carrier density at equilibrium, $n_{e(h)}$ is the density of trapped electrons(holes), and $\tau_{e(h)} = \sigma_{e(h)} v_{e(h)}$ is the recombination lifetime of mobile electrons(holes) under the assumption that an electron moving at its thermal velocity, $v_{e(h)}$, and inside of the capture cross-section, $\sigma_{e(h)}$, of the trap will be trapped. A more complete derivation of this expression has been previously published.^[52] Here, we consider a special case: i) both bimolecular and trap-assisted recombination contribute, ii) the thermal charge density and thermal population of traps is assumed to be small such that $n_e n_h \gg n_i^2$ and $n_{e(h)} \gg n_{e(h),traps}$, (3) the density of electrons and holes is balanced such that $n_e \sim n_h$, (4) at the open circuit condition, all charge recombines such that generation flux, G , is equal to the recombination flux, R , at open circuit: $G = R(V_{oc})$. We implicitly assume that the photon flux is efficiently converted into electron-hole pairs because geminate recombination

has been shown to play a small role in two polymer:fullerene devices.^[54] We note that this assumption may be less relevant as the concentration of PC₈₄BM increases. Under these assumptions, the effective recombination rate may be written as

$$R(V_{oc}) = G = \frac{n_{oc}}{\tau_R} + \gamma n_{oc}^2 \quad (4)$$

where n_{oc} is the charge density at open circuit and $\tau_R = \tau_e + \tau_h$. Equation (4) can be solved for the charge density at open circuit in terms of the recombination rate coefficients and the generation rate, G , of electron-hole pairs:

$$(n_{oc})^2 = \frac{1}{4\gamma^2\tau_R^2} \left(1 + (1 + 4\gamma G\tau_R^2) - 2(1 + 4\gamma G\tau_R^2)^{1/2} \right) \quad (5)$$

Under illumination and at the open circuit condition, the open circuit voltage, V_{oc} , equals the difference between the quasi-Fermi levels within the polymer and fullerene phase-separated domains:

$$V_{oc} = \frac{1}{e} \left(E_{LUMO}^{fullerene} - E_{HOMO}^{polymer} - \frac{kT}{e} \ln \left(\frac{n_e n_h}{N_c N_v} \right) \right) \quad (6)$$

where $N_{c(v)}$ is the density of conduction states at the band edge of the polymer and fullerene, k is the Boltzmann coefficient, T is the absolute temperature. The form of the equation for V_{oc} , assuming purely bimolecular recombination, has been extensively shown to be a universal equation for optimized polymer:fullerene solar cells.^[52,60] For a bulk recombination process incorporating both bimolecular recombination and trap-assisted recombination, V_{oc} can be expressed as a combination of Equation (5) and (6). A simplified case of SRH recombination predicts a slope of $\delta V_{oc} = 2kT/e$ in the plot of V_{oc} versus the natural logarithm of the light intensity, rather than a slope of $\delta V_{oc} = kT/e$ as shown previously for bimolecular recombination.^[52,60] Here, we probe the intensity-dependent electronic characteristics of the two processes acting simultaneously.

In Figure 5a, the open circuit voltage as a function of $\ln(I)$ is plotted where I is the incident light intensity. Data are shown for the pristine PCDTBT:PC₆₀BM solar cell and for the similar devices with PC₈₄BM traps. The pristine, 1:10,000, 1:1,000, 1:100, and 1:10 devices show a slope (as listed in Table 1) of $\delta V_{oc} = (1.15 \pm 0.02)kT/e$, $(1.27 \pm 0.02)kT/e$, $(1.34 \pm 0.02)kT/e$, $(1.47 \pm 0.02)kT/e$, and $(1.62 \pm 0.02)kT/e$, respectively, indicating that increasing concentrations of traps enhance the relative importance of first-order recombination. We attribute the change in the slope to an increase in trap-assisted recombination. In Figure 5b, the predicted electron-hole pair generation-rate dependence of the open circuit voltage is plotted via the bimolecular/SRH recombination model. The color scale from purple to brown indicates logarithmically increasing trap-assisted charge recombination rates (inverse lifetimes), proportional to the density of trap states.

The model plotted in Figure 5b uses the parameters given here: $E_{LUMO}^{fullerene} - E_{HOMO}^{polymer} = 1.2$ eV, $N_c = N_v = 5 \times 10^{19} \text{ cm}^{-3}$,^[50] $T = 310$ K, $\gamma = 5 \times 10^{-13} \text{ cm}^3 \text{ s}^{-1}$,^[51] $G = 10^{19} - 10^{21} \text{ cm}^{-3} \text{ s}^{-1}$, with τ_R in the range $\tau_R = 2 \mu\text{s} - 2$ ms. The range of generation rates here is estimated from the magnitude of the photocurrent at short circuit at 1 sun, 0.1 sun, and 0.01 sun. The generation flux may be estimated from the previously presented relation: $G(I) = J_{photo}(I)/edP_C$. For example at 1 sun, $J_{photo} = 7.3 \text{ mA cm}^{-2}$, $d \sim 100$ nm, and $P_C \sim 1$. Therefore we can estimate $G = 5 \times 10^{21}$. Recombination times here do not correspond to the rates observed in the transient absorption measurements, as the TA times show mobile carrier trapping and recombination of directly (<1 ps) populated traps only on the picosecond time scale. Recombination times do, however, correspond well to transient photocurrent results published by Leong et al. elsewhere.^[30] Recombination times fit to the data collected here are $\tau_R = 2$ ms (devices with low concentrations of PC₈₄BM) to $\tau_R = 2 \mu\text{s}$ (devices with high concentrations of PC₈₄BM). When the trap-assisted recombination lifetime is of order 2 ms, trap-assisted recombination is not likely to effect the device performance, as most charge is collected before this long time scale.^[32] When the trap-assisted recombination lifetime approaches the microsecond regime, trap-assisted recombination is more likely to occur on a scale which significantly decreases the amount of photogenerated current collected. The logarithmic 'slope' of the model tends from kT/e at high generation rates and/or low trap densities, to $2kT/e$ at low generation rates and/or high trap densities as indicated in Figure 5b.

The recombination mechanism is light-intensity dependent. For a given recombination lifetime, or trap density, the model predicts that the recombination mechanism changes continuously from trap-assisted recombination at low generation rates to bimolecular recombination at high generation rates. At higher generation rates, bimolecular recombination dominates because of the increase in the electron-hole product and because traps are more likely to be filled by an electron via the trapping mechanism or by photoexcitation. At low generation rates, partially occupied traps acting as recombination centers are the dominant recombination mechanism. In the conditions of low generation rates of electron-hole pairs, absorption directly into the trap state is less likely and the number of empty traps will increase, increasing the probability of charge trapping and decreasing the recombination time, τ_R .

Figure 6 illustrates the intensity dependence of the open circuit voltage in a new way: the intensity-dependent V_{oc} data are plotted as a function of PC₈₄BM trap concentration, and the bimolecular/SRH recombination model is overlaid. Via the two-recombination mechanism model, the trap-assisted recombination rate ($1/\tau_R$) is plotted vs. open circuit voltage for discrete values of the generation rate of electron-holes pairs, visualized in color on the graph from low generation rates (black and blue) to high generation rates (brown and purple). The generation rate contour lines which fit the intensity-dependent data at 1 sun (dashed brown), 0.1 sun (dashed yellow), and 0.01 sun (dashed blue) are $G = 10^{21} \text{ cm}^{-3} \text{ s}^{-1}$, $G = 10^{20} \text{ cm}^{-3} \text{ s}^{-1}$, and $G = 10^{19} \text{ cm}^{-3} \text{ s}^{-1}$, respectively. Cross-sections through this plot at constant trap density/recombination rate reproduce Figure 5a–b. The special-case bimolecular/SRH recombination model derived above is a good fit to the intensity-dependent V_{oc} data

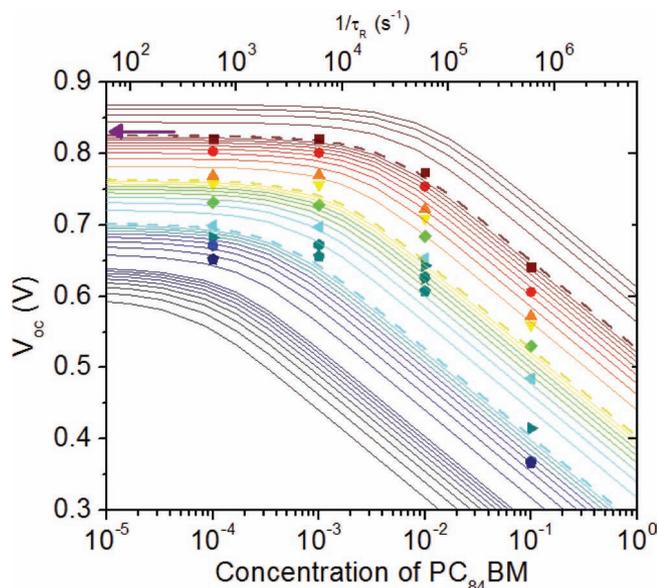


Figure 6. Intensity dependent V_{oc} plotted as a function of PC₈₄BM trap concentration (data points, lower x-axis). Bimolecular/SRH recombination model fits to the data (lines, upper x-axis): the trap-assisted recombination rate plotted vs. the generation rate of excitons with color from blue to brown indicating an increase in generation rate. Arrow indicates the V_{oc} at 1 sun for the trap-free PCDTBT:PC₆₀BM solar cell devices. Dotted lines correspond to logarithmic increases in generation rate (G (brown) = 10^{21} cm⁻³ s⁻¹, G (yellow) = 10^{20} cm⁻³ s⁻¹, G (blue) = 10^{19} cm⁻³ s⁻¹).

within experimental error. The model predicts that recombination mechanisms strongly depend on density of charge carriers in the BHJ system. In this visualization of the data, it becomes clear that light intensity/charge density and the density of traps simultaneously control the dominant recombination mechanism. Two limiting cases exist in Figure 6: 1) At high light intensity and/or low trap density, recombination is bimolecular. 2) At low light intensity and/or high trap density, recombination becomes first order (dominated by trap-assisted loss). Factors which control the cross-over from one recombination mechanism to the other are the strength of bimolecular recombination rate coefficient, the carrier density, the occupation and capture cross-section of the traps, and the thermal velocity of charge carriers. When only a low number of traps exist in the system, the probability of a carrier being trapped will be lower than the case where the number of traps is increased by orders of magnitude. Statistically, carriers will be less likely to be captured by a trap under increased charge carrier density conditions where charge is more spatially correlated and bimolecular decay is dominant. As the carrier density is reduced, the fraction of traps relative to the amount of mobile charge will increase, also increasing the probability of trap-assisted recombination. This mechanism could explain the markedly different electronic characteristics seen in the same photodiode in the dark and under illumination, especially the difference in ideality factors in the light, m , and the dark, n .

The open circuit voltage is expected to be reduced by two effects. The first effect that we expect to reduce the open circuit voltage is that which we observe: bimolecular and trap-assisted

recombination lower the average occupation density of the LUMO level (i.e. lower the Fermi level). The second effect that we expect to reduce the open circuit voltage (in polymer:fullerene devices with a high concentration of trap states) is Fermi level reduction due to the lower energy state of the introduced PC₈₄BM molecules. However, as the V_{oc} reduction is easily and comprehensively explained by the recombination model alone, we hypothesize that the second effect does not play a large role.

Hence, we conclude that the trap-assisted recombination rate (proportional to the trap density) and the generation rate (proportional to incident light intensity) control the cross-over regime from bimolecular to trap-assisted recombination, as has been preliminarily discussed.^[52] Above the cross-over point at low trap densities and high light intensities, the open circuit voltage is relatively unaffected by the density of traps within the BHJ due to disordered band tails or intentionally introduced traps. Below the cross-over point at high trap densities, such as those potentially found in BHJ materials with impurities left over from the synthesis and device processing, recombination through traps becomes the dominant loss mechanism and results in reduction in the open circuit voltage under operating illumination levels.

5. Conclusions

We have quantified the role of impurities in reducing the fill factor, short circuit current, and open circuit voltage of polymer-based solar cell devices. PC₈₄BM, when used as a well-defined trap state impurity in the PCDTBT:PC₆₀BM BHJ solar cell, negatively effects all measurable properties of the BHJ solar cell. Additionally, these purposefully introduced PC₈₄BM traps provide a definitive measure of trap-assisted recombination in polymer:fullerene solar cells confirmed via steady state intensity dependent current-voltage measurements. Charge transport studies investigating the effect of traps on time-resolved transport measurements and A.C. impedance measurements of this materials system are ongoing.^[30]

In making devices from newly synthesized materials, the initial tests showing absorption, fluorescence, energy levels, and conductivity are indicative of promise alone. The suitability of a material for efficient devices is often only determined by trial and error in the device processing laboratory. Why are initially promising polymeric materials often found to be completely unsuitable for device applications? The answer is complex. Issues of molecular weight, solubility, surface energy interactions, and intercalation of fullerene into polymeric sidechains have been found to play a role. We determine that dilute impurities act as an important loss mechanism that may be eliminated via further purification processes.

6. Experimental Section

Polymer-fullerene solar cells were fabricated using blends of the copolymer poly[N-9'-hepta-decanyl-2,7-carbazole-alt-5,5-(4',7'-di-2-thienyl-2',1',3'-benzothiadiazole) (PCDTBT) and the fullerene derivatives, [6,6]-phenyl C₆₁ butyric acid methyl ester (PC₆₀BM) and [6,6]-phenyl C₈₄ butyric acid methyl ester (PC₈₄BM). Devices of reproducible quality were fabricated on indium tin oxide (ITO) coated glass substrate with the

following structure: ITO-coated glass substrate/poly(3,4 ethylenedioxythiophene):poly(styrenesulfonate) (PEDOT:PSS)/polymer:fullerene blend/titanium sub-oxide (TiO_x)/aluminum (Al).

PCDTBT:PC₆₀BM blend films were cast from a solution of PCDTBT:PC₆₀BM (1:4) in an anhydrous 1,2-dichlorobenzene:chlorobenzene solvent mixture (3:1) with a polymer concentration of 7 mg/mL. Small amounts of PC₈₄BM were dissolved in a low concentration solution with chlorobenzene, and pipetted into the BHJ solution in substitution for a small portion of the chlorobenzene solvent. Glass/ITO substrates were cleaned by sonication for 10 min each in soapy deionized water, deionized water, acetone, and isopropanol. Substrates were then heated above 100 °C overnight to evaporate the water and residual solvent. Substrates were UV-ozone plasma treated for 60 min prior to PEDOT:PSS deposition. PEDOT:PSS, acquired from H. C. Stark as Clevios PH, was filtered with a 0.45 micrometer PTFE filter and spun onto the substrate in air at 5000 RPM for 40 sec. The PEDOT:PSS film was then annealed in air at 140 °C for 10 min to evaporate the solvent (water). BHJ films were cast and annealed in a nitrogen environment. The polymer:fullerene solutions were spin cast with no filtration at 3500 RPM and annealed after casting at 70 °C for 10 min. An amorphous solution-processable TiO_x layer was spin cast in air onto all the devices as a buffer layer and an optical spacer.^[64] The TiO_x film was annealed in air at 80 °C for 10 min to oxidize the TiO_x precursor film and to evaporate solvent. A 100 nm aluminum electrode was vacuum-deposited at pressure 3×10^{-6} torr, and devices were not annealed after deposition of the aluminum electrode. Long term illumination at 1 sun intensity in our device test rig heats devices to 35 °C. Devices were encapsulated for testing in air with a UV-curable epoxy and covered with a glass slide. Encapsulated devices degraded 15% over the six month test period, with the majority of degradation occurring in the first 2 weeks.

Current density–voltage (J – V) characteristics of the devices were measured using a Keithley 236 Source Measure Unit. Solar cell performance used a Newport Air Mass 1.5 Global (AM 1.5G) full spectrum solar simulator with an irradiation intensity of 100 mW cm^{-2} . The 100 mW cm^{-2} spectrum of incident light was spectrum and intensity matched with an Ocean Optics USB4000 spectrometer calibrated for absolute intensity via a deuterium tungsten halogen calibration standard lamp with NIST-traceable calibration from 350–1000 nm. The intensity of the lamp was modulated with a series of 2 neutral density filter (NDF) wheels of 6 filters apiece, allowing for 35 steps in intensity from 0.4 mW cm^{-2} – 100 mW cm^{-2} . Intensity of light transmitted through each filter combination was independently measured via the Ocean Optics USB4000 spectrometer calibrated for absolute intensity. Error is introduced while modulating the full solar spectrum with “grey” filters, which non-linearly reduce the solar spectrum, especially at high filter optical densities. Scatter in the data specific to the density filters result in error in the fit of V_{oc} vs $\ln(I)$ of ± 0.001 –equivalent to an uncertainty of ± 12 degrees in temperature or ± 0.04 in the “slope.” Therefore, external temperature measurement is necessary to reduce experimental error in the fitting – as previously implemented.^[52]

IPCE measurements were performed with a PV Measurements, Inc. QE77 Solar Cell QE Measurement System. In the IPCE measurement, illumination intensity mimics the 100 mW cm^{-2} AM1.5G spectrum, as calibrated with a silicon reference cell from $\lambda = 300$ – 800 nm . Absorption measurements were performed with a Beckman Coulter UV-visible spectrophotometer on films on slide glass. A piece of slide glass was used as a reference sample and subtracted from the absorption of the organic film on slide glass.

Transient absorption (TA) spectra were recorded for the spin-cast solid samples using femtosecond pulsed laser pump-probe spectroscopy. The 400 nm pump beam was generated by frequency doubling the 800 nm output of a Ti:sapphire laser system with a regenerative amplifier providing 100 fs pulses at a repetition rate of 1 kHz. The pump intensity per pulse was kept low, around $30 \mu\text{J cm}^{-2}$, in order to minimize annihilation and other intensity-dependent processes due to high laser power. The probe beam in the visible and near-infrared range (400–1100 nm) consisted of a white light continuum, generated by passing a portion of the 800 nm

amplified Ti:sapphire output through a 1 mm thick sapphire plate. The probe intensity was always less than the pump intensity and the spot size was much smaller (about 0.4 mm diameter). The probe pulses were time delayed with respect to the pump pulses using a computerized translation stage. The probe beam was split before the sample into a signal beam (transmitted through the sample and crossed with the pump beam) and a reference beam. The signal and reference were detected with a pair of 163 mm spectrographs (Andor Technology, SR163) equipped with a 512×58 pixels back-thinned CCD (Hamamatsu S07030-0906) and assembled by Entwicklungsbüro Stresing, Berlin. To improve sensitivity, the pump light was chopped at half the amplifier frequency, and the transmitted signal intensity was recorded shot by shot. It was corrected for intensity fluctuations using the reference beam. The transient spectra were averaged until the desired signal-to-noise ratio was achieved. The polarization of the probe pulses was at magic angle relative to that of the pump pulses. All spectra were corrected for the chirp of the white-light probe pulses. The FWHM of the response function was about 150 fs. Samples were kept under dynamic vacuum ($<10^{-4}$ mbar) during the TA measurements.

Supporting Information

Supporting Information is available from the Wiley Online Library or from the author.

Acknowledgements

This research was supported by the US Army General Technical Services (LLC/GTS-S-09-1-196) and the Department of Energy (DOE ER46535). S. Cowan was supported by the Center for Energy Efficient Materials, an Energy Frontier Research Center funded by the U.S. Department of Energy, Office of Science, Office of Basic Energy Sciences under award Number DE-SC0001009. The authors thank Dr. Robert Street and Dr. Anshuman Roy for many important comments and discussions, and S. Cowan thanks Dr. Jay Hyun Lee for discussions regarding dark current fitting. N. Banerji thanks the Swiss National Science Foundation for Fellowship support (fellowship for prospective researchers PBGE2-125859). The materials used for the fabrication of the solar cells were provided by Dr. D. Waller of Konarka Technologies.

Received: March 8, 2011

Published online: June 21, 2011

- [1] S. H. Park, A. Roy, S. Beaupre, S. Cho, N. Coates, J. S. Moon, D. Moses, M. Leclerc, K. Lee, A. J. Heeger, *Nat. Photonics* **2009**, *3*, 297.
- [2] H.-Y. Chen, J. Hou, S. Zhang, Y. Liang, G. Yang, Y. Yang, L. Yu, Y. Wu, G. Li, *Nat. Photonics* **2009**, *3*, 649.
- [3] J. Hou, H.-Y. Chen, S. Zhang, R. I. Chen, Y. Yang, Y. Wu, G. Li, *J. Am. Chem. Soc.* **2009**, *131*, 15586.
- [4] J. A. Hauch, P. Schilinsky, S. A. Choulis, R. Childers, M. Biele, C. J. Brabec, *Sol. Energy Mater. Sol. Cells* **2008**, *92*, 727.
- [5] E. Voroshazi, B. Verreert, T. Aernouts, P. Heremans, *Sol. Energy Mater. Sol. Cells* **2011**, *95*, 1303.
- [6] C. P. Chen, S. H. Chan, T. C. Chao, C. Ting, B. T. Ko, *J. Am. Chem. Soc.* **2008**, *130*, 12828.
- [7] J. H. Hou, H. Y. Chen, S. Q. Zhang, G. Li, Y. Yang, *J. Am. Chem. Soc.* **2008**, *130*, 16144.
- [8] Z. Zhu, D. Waller, R. Gaudiana, M. Morana, D. Muhlbacher, M. Scharber, C. Brabec, *Macromolecules* **2007**, *40*, 1981.
- [9] Y. Lee, T. P. Russell, W. H. Jo, *Org. Electron.* **2010**, *11*, 846.
- [10] N. S. Sariciftci, L. Smilowitz, A. J. Heeger, F. Wudl, *Science* **1992**, *258*, 1474.

- [11] A. A. Bakulin, J. C. Hummelen, M. S. Pshenichnikov, P. H. M. van Loosdrecht, *Adv. Funct. Mater.* **2010**, *20*, 1653.
- [12] C. J. Brabec, G. Zerza, G. Cerullo, S. De Silvestri, S. Luzzati, J. C. Hummelen, S. Sariciftci, *Chem. Phys. Lett.* **2001**, *340*, 232.
- [13] B. Kraabel, J. C. Hummelen, D. Vacar, D. Moses, N. S. Sariciftci, A. J. Heeger, F. Wudl, *J. Chem. Phys.* **1996**, *104*, 4267.
- [14] S. Morita, A. A. Zakhidov, K. Yoshino, *Solid State Commun.* **1992**, *82*, 249.
- [15] M. Tong, N. E. Coates, D. Moses, A. J. Heeger, S. Beaupre, M. Leclerc, *Phys. Rev. B* **2010**, *81*, 125210.
- [16] H. W. Kroto, J. R. Heath, S. C. O'Brien, R. F. Curl, R. E. Smalley, *Nature* **1985**, *318*, 162.
- [17] F. Diederich, R. Ettl, Y. Rubin, R. L. Whetten, R. Beck, M. Alvarez, S. Anz, D. Sensharma, F. Wudl, K. C. Khemani, A. Koch, *Science* **1991**, *252*, 548.
- [18] W. Kratschmer, L. D. Lamb, K. Fostiropoulos, D. R. Huffman, *Nature* **1990**, *347*, 354.
- [19] J. C. Hummelen, B. W. Knight, F. Lepeq, F. Wudl, J. Yao, C. L. Wilkins, *J. Org. Chem.* **1995**, *60*, 532.
- [20] G. Yu, J. Gao, J. C. Hummelen, F. Wudl, A. J. Heeger, *Science* **1995**, *270*, 1789.
- [21] M. M. Koetse, J. Sweelssen, K. T. Hoekerd, H. F. M. Schoo, S. C. Veenstra, J. M. Kroon, X. N. Yang, J. Loos, *Appl. Phys. Lett.* **2006**, *88*, 083504.
- [22] M. M. Mandoc, W. Veurman, L. J. A. Koster, M. M. Koetse, J. Sweelssen, B. de Boer, P. W. M. Blom, *J. Appl. Phys.* **2007**, *101*, 104512.
- [23] F. B. Kooistra, V. D. Mihailetschi, L. M. Popescu, D. Kronholm, P. W. M. Blom, J. C. Hummelen, *Chem. Mat.* **2006**, *18*, 3068.
- [24] K. Shibata, Y. Kubozono, T. Kanbara, T. Hosokawa, A. Fujiwara, Y. Ito, H. Shinohara, *Appl. Phys. Lett.* **2004**, *84*, 2572.
- [25] H. Meyer, D. Haarer, H. Naarmann, H. H. Horhold, *Phys. Rev. B* **1995**, *52*, 2587.
- [26] Z. Chiguvare, V. Dyakonov, *Phys. Rev. B* **2004**, *70*.
- [27] J. M. Frost, F. Cheynis, S. M. Tuladhar, J. Nelson, *Nano Lett.* **2006**, *6*, 1674.
- [28] C. R. McNeill, N. C. Greenham, *Appl. Phys. Lett.* **2008**, *93*.
- [29] M. M. Mandoc, F. B. Kooistra, J. C. Hummelen, B. de Boer, P. W. M. Blom, *Appl. Phys. Lett.* **2007**, *91*, 263505.
- [30] W. L. Leong, S. R. Cowan, A. J. Heeger, *Adv. Energy Mat.* **2011**, doi: 10.1002/aenm.201100196.
- [31] N. Banerji, S. Cowan, M. Leclerc, E. Vauthey, A. J. Heeger, *J. Am. Chem. Soc.* **2010**, *132*, 17459.
- [32] S. R. Cowan, R. A. Street, S. Cho, A. J. Heeger, *Phys. Rev. B* **2011**, *83*, 035205.
- [33] N. S. Christ, S. W. Kettlitz, S. Valouch, S. Zuffe, C. Gartner, M. Punke, U. Lemmer, *J. Appl. Phys.* **2009**, *105*, 104513.
- [34] M. Punke, S. Valouch, S. W. Kettlitz, N. Christ, C. Gartner, M. Gerken, U. Lemmer, *Appl. Phys. Lett.* **2007**, *91*, 071118.
- [35] W. Shockley, W. T. Read Jr., *Phys. Rev. B* **1971**, *4*, 502.
- [36] R. N. Hall, *Physical Review* **1951**, *83*, 228.
- [37] R. N. Hall, *Physical Review* **1952**, *87*, 387.
- [38] C.-T. Sah, W. Shockley, *Physical Review* **1958**, *109*, 1103.
- [39] A. Maurano, R. Hamilton, C. G. Shuttle, A. M. Ballantyne, J. Nelson, B. O'Regan, W. Zhang, I. McCulloch, H. Azimi, M. Morana, C. J. Brabec, J. R. Durrant, *Advanced Materials* **2010**, *22*, 4987.
- [40] H. Hoppe, N. S. Sariciftci, *J. Mater. Res.* **2004**, *19*, 1924.
- [41] C. Vanberkel, M. J. Powell, A. R. Franklin, I. D. French, *J. Appl. Phys.* **1993**, *73*, 5264.
- [42] N. C. Giebink, G. P. Wiederrecht, M. R. Wasielewski, S. R. Forrest, *Phys. Rev. B* **2010**, *82*, 155305.
- [43] P. Schilinsky, C. Waldauf, J. Hauch, C. J. Brabec, *J. Appl. Phys.* **2004**, *95*, 2816.
- [44] C. Waldauf, M. C. Scharber, P. Schilinsky, J. A. Hauch, C. J. Brabec, *J. Appl. Phys.* **2006**, *99*, 104503.
- [45] C.-T. Sah, *IRE Trans. Electron Devices* **1962**, *ED-9*, 94.
- [46] C.-T. Sah, R. N. Noyce, W. Shockley, *Proc. IRE* **1957**, *45*, 1228.
- [47] M. A. Wolf, M. Wolf, *IEEE Trans. Electron Devices* **1984**, *31*, 684.
- [48] D. Zhu, J. R. Xu, A. N. Noemaun, J. K. Kim, E. F. Schubert, M. H. Crawford, D. D. Koleske, *Appl. Phys. Lett.* **2009**, *94*, 081113.
- [49] M. A. Hamdy, R. L. Call, *Sol. Cells* **1987**, *20*, 119.
- [50] L. J. A. Koster, V. D. Mihailetschi, R. Ramaker, P. W. M. Blom, *Appl. Phys. Lett.* **2005**, *86*, 123509.
- [51] C. G. Shuttle, B. O'Regan, A. M. Ballantyne, J. Nelson, D. D. C. Bradley, J. R. Durrant, *Phys. Rev. B* **2008**, *78*, 113201.
- [52] S. R. Cowan, A. Roy, A. J. Heeger, *Phys. Rev. B* **2010**, *82*, 245207.
- [53] M. Hallermann, S. Haneder, E. Da Como, *Appl. Phys. Lett.* **2008**, *93*, 053307.
- [54] R. A. Street, S. Cowan, A. J. Heeger, *Phys. Rev. B* **2010**, *82*, 121301.
- [55] A. Liu, S. Zhao, S. B. Rim, J. Wu, M. Konemann, P. Erk, P. Peumans, *Adv. Mater.* **2008**, *20*, 1065.
- [56] V. D. Mihailetschi, L. J. A. Koster, J. C. Hummelen, P. W. M. Blom, *Phys. Rev. Lett.* **2004**, *93*, 216601.
- [57] R. A. Street, M. Schoendorf, *Phys. Rev. B* **2010**, *81*, 205307.
- [58] D. Veldman, O. Ipek, S. C. J. Meskers, J. Sweelssen, M. M. Koetse, S. C. Veenstra, J. M. Kroon, S. S. van Bavel, J. Loos, R. A. J. Janssen, *J. Am. Chem. Soc.* **2008**, *130*, 7721.
- [59] C. Groves, N. C. Greenham, *Phys. Rev. B* **2008**, *78*, 155205.
- [60] L. J. A. Koster, E. C. P. Smits, V. D. Mihailetschi, P. W. M. Blom, *Phys. Rev. B* **2005**, *72*, 085205.
- [61] K. Maturova, S. S. van Bavel, M. M. Wienk, R. A. J. Janssen, M. Kemerink, *Nano Lett.* **2009**, *9*, 3032.
- [62] M. P. Langevin, *Ann. Chim. Phys.* **1903**, *28*, 433.
- [63] J. G. Simmons, G. W. Taylor, *Phys. Rev. B* **1971**, *4*, 502.
- [64] A. Roy, S. H. Park, S. Cowan, M. H. Tong, S. N. Cho, K. Lee, A. J. Heeger, *Appl. Phys. Lett.* **2009**, *95*, 013302.