

Iontophoretic fraction collection for Coupling
Capillary Zone Electrophoresis with Matrix-
assisted Laser Desorption/Ionization Mass
Spectrometry

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Impact of spotting on separation resolution and efficiency. To better assess the impact of the spotting process on the resolution, Table S-1 has been constructed. It relates the efficiency of several peaks as a function of the number of collected fractions. 5 peaks, covering the full separation window, have been chosen to study the impact of the spotting process on the separation efficiency. The chosen peaks are marked on the trace D of the Figure 2 in the main part of this manuscript.

	Fraction collection interval (s)			
	0	60	30	15
◇	566493	532340	519851	550647
□	61605	88577	77586	91000
X	165572	160540	144940	165237
○	108716	122963	107381	113958

Table S-1. Impact of the spotting process on the separation efficiency (number of theoretical plates/meter).

Same experimental conditions as in Figure 2 of the main part of the manuscript

Direct MALDI-TOF-MS of the test peptide mixture.

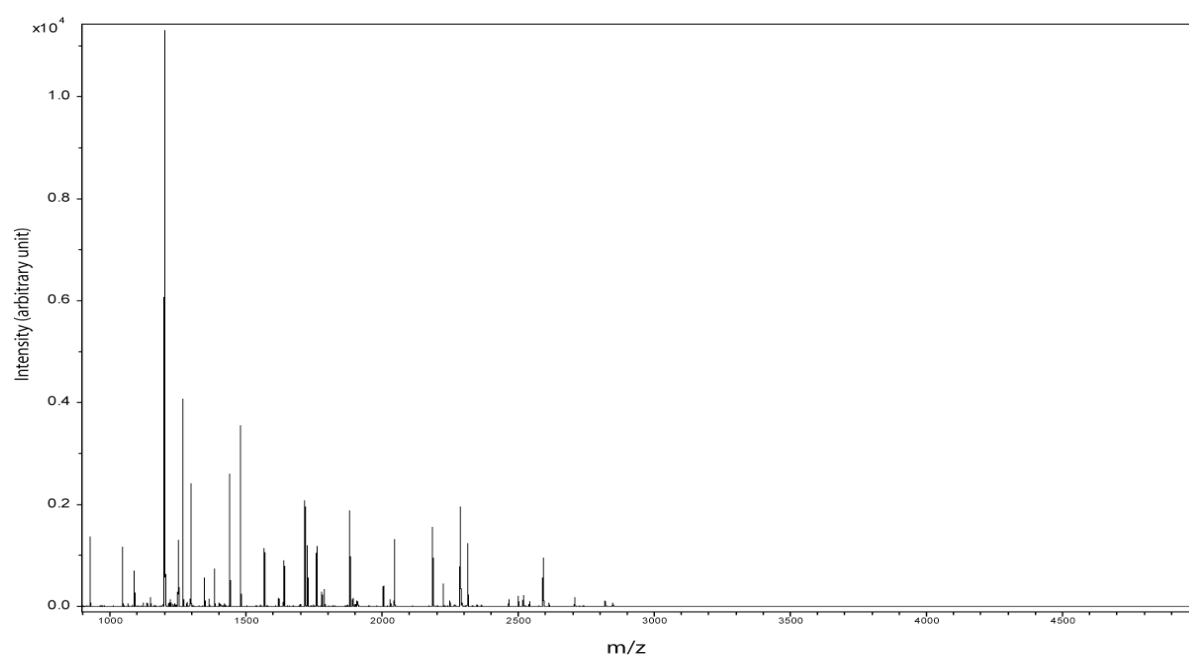


Figure S-1. MALDI-TOF mass spectrum of the total peptide mixture.

Sample: 0.5 μL of the Tryptic digest of total caseins (3.5 μM), α - $\square\square\square\square\square\square\square\square\square\square$ (1.75 μM), β - $\square\square\square\square\square\square\square\square\square\square$ (2.3 μM), bovine serum albumin (0.75 μM), ribonuclease A (3.75 μM). Sample and matrix application method as described in the materials and methods section.

List of the identified peptides. For the different experiments related in the main part of this article, the following tables mention the experimental and theoretical masses of the detected peptides, the mass accuracy, the corresponding sequences and each peptide position within the considered protein. The number of missed cleavage as well as existing modifications is specified. Finally, for each table corresponding to a CE-MALDI-MS experiment, the fraction in which each peptide has been detected is indicated.

Table S-2: CE-MALDI-TOF-MS of the peptide mixture with direct spotting on a MALDI target

Experimental conditions for CE: HPC coated capillary, total/effective length 80/21.5 cm x 50 µm I.D. Voltage : 30 kV, current : 12 µA, UV absorbance at 200 nm. BGE: 10% acetic acid. Sample: Tryptic digest of total caseins (14 µM), α- κ -casein (11 µM), β-casein (9 µM), bovine serum albumin (3 µM), ribonuclease A (15 µM). Sample injection: 30 mbar 200 s. Leading electrolyte (ammonium acetate pH 4, 250 mM ionic strength) injection prior to sample injection : 30 mbar 40 s. See materials and methods section for MS experimental conditions

Direct MALDI-TOF analysis								
Protein	CE fraction	Experimental mass (Da)	Theoretical mass (Da)	Mass accuracy (ppm)	Position	Sequence	Number of missed cleavage	Modifications
αS1-casein		1267.709	1267.7045	3.5	106-115	YLGYLEQLLR	0	
		1384.718	1384.7299	8.6	38-49	FFVAPFPEVFGK	0	
		1759.918	1759.9449	15.3	23-37	HQGLPQEVLENLLR	0	
β-casein		2186.142	2186.1678	11.8	166-217	DMPIQAFLLYQEPVLPVPR	0	
		1251.706	1251.7096	2.9	46-55	YIPIQYVLSR	0	
κ-casein		1091.506	1091.519	10.1	141-144	LQDWLCEK	0	
		1200.64	1200.6524	10.3	118-127	VGINWLAHK	0	Cys_CAM: 139
α-lactalbumin		1779.792	1779.8404	27.2	128-141	ALCSEKLDQWLCEK	1	Cys_CAM: 130, 139
		1892.864	1892.9244	31.9	128-142	ALCSEKLDQWLCEKL	2	Cys_CAM: 130, 139
		2591.163	2591.1071	21.6	78-98	IWCKDDQNPSSNICNISCD K	1	Cys_CAM: 80, 92, 96
		2847.396	2847.432	12.6	118-141	VGINYVLAHKALCSEKLDQWLCEK	2	Cys_CAM: 130, 139
		1121.479	1121.468	9.8	77-85	WENGECAGK	0	Cys_CAM: 82
β-lactoglobulin		1635.745	1635.7746	18.2	141-154	TPEVDELEKFKDK	1	
		1715.769	1715.8057	21.4	165-178	LSFNPTLQEEQCHI	0	Cys_CAM: 176
		2313.233	2313.2587	11.1	57-76	VYVEELKPTPEGDLLELLOK	0	
		2707.449	2707.3759	27.0	31-56	VAGTWYSLAMAASDILLDAQSAPLR	0	
		2818.348	2818.2667	28.8	118-140	YLLFCMENSAPPEQSLAQCC LVR	0	Cys_CAM: 122, 135, 137
RNA		1150.62	1150.6215	1.3	27-36	KETAAAKFER	2	
		2224.051	2224.0855	15.5	131-150	HIIVAGEGNPYVPVHFDASV	0	Cys_CAM: 136
		2285.933	2285.922	4.8	93-111	NGQTNQYQSYSTMSITDCR	0	Cys_CAM: 98, 110
BSA		2517.226	2517.2224	1.4	65-87	CKPWNFVIESLADVQAVCS QK	0	Cys_CAM: 66, 84
		927.499	927.4934	6.0	161-167	YLVEIAR	0	
		1249.607	1249.6211	11.3	35-44	FKDLGEEHFK	1	
		1283.701	1283.7106	7.5	361-371	HPEYAVSVLLR	0	
		1399.662	1399.6926	21.9	569-580	TVMENFAVVDK	0	
		1419.674	1419.6936	13.8	89-100	SLHTLFGDELCK	0	Cys_CAM: 99
		1438.902	1438.8117	6.7	360-371	RHPEYAVSVLLR	1	
		1479.793	1479.7954	1.6	421-433	LGEYGFQNALIVR	0	
		1567.739	1567.7427	2.4	347-359	DAFLGSFLVEYSR	0	
		1639.908	1639.9377	18.1	437-451	KVQVSTPTLVEYSR	1	
		1724.802	1724.8346	18.9	469-482	MPCTEDYLSLILNR	0	Cys_CAM: 471
		1880.896	1880.9211	13.3	508-523	RPCFSALTPDETYPVK	0	Cys_CAM: 510
		1907.871	1907.9207	26.0	529-544	LFTFHADICTLPDTEK	0	Cys_CAM: 537
		2003.767	2003.7779	5.4	106-122	ETYGDMDQCEKQEPER	1	Cys_CAM: 114, 115
		2045	2045.0279	13.6	169-183	RHPYFVPELLYANK	1	
	2247.911	2247.9427	14.1	267-285	ECCHGDLLCADDRADLAK	1	Cys_CAM: 268, 269, 276	
	2541.201	2541.1674	13.2	118-138	QEPERNECFSLHKDDSPDLP K	2	Cys_CAM: 125	

Table S-3: Direct-MALDI-TOF-MS of the peptide mixture

Sample: 0.5 µL of the Tryptic digest of total caseins (3.5 µM), α- κ -casein (1.75 µM), β-casein (2.3 µM), bovine serum albumin (0.75 µM), ribonuclease A (3.75 µM). Spotting method as described in the materials and methods section.

Numerical model for diffusion-migration simulations. The diffusion-migration numerical simulations are decoupled in 2 steps, following a supporting electrolyte excess assumption. First, the electric potential distribution is calculated using the classical Laplace equation (1). Then, the calculated electric field is used in the transient Nernst-Planck equation (2) in order to determine the time evolution of the concentration distribution of the sample plug species.

$$\nabla \cdot (j) = \nabla \cdot (-\sigma \nabla \phi) = 0 \quad (1)$$

$$\frac{\partial c_i}{\partial t} + \nabla \cdot \left(-D_{app} \nabla c_i - \frac{z_i F}{RT} D_i c_i \nabla \phi \right) = 0 \quad (2)$$

where j is the electrical current density, ϕ the electrical potential, σ the solution conductivity, c_i the species concentration, D_i their diffusion coefficient, z_i their charge, F the Faraday constant, R the gas constant and T the temperature. In the droplet, an apparent diffusion coefficient D_{app} is introduced to take into account the microscopic chaotic motion, or “spontaneous” convection, following a concept introduced by Levich.¹ The effect of this microscopic fluid motion is non-negligible in macroscopically still solutions and acts as an apparent diffusion coefficient depending on the distance y from a solid wall:

$$D_{app} = D_i \left(1 + 1.507 \left(\frac{y}{\delta} \right)^4 \right) \quad (3)$$

where δ is the thickness of the classical Nernst layer. Eq (3) illustrates that the diffusion coefficient is equal to its theoretical value in the stagnant layer ($y < \delta$) and vary in y^4 when y is larger than the diffusion layer thickness δ .

The solution is assumed to be isothermal without any macroscopic convection. To decouple the electric field calculation from the transport equation, a uniform conductivity is assumed due to the presence of the BGE.

This 2-D axi-symmetrical model is implemented on the finite element software Flux-Expert® (Astek, Grenoble, France) and operated on a Linux PC. The geometry is a cross-section of the end of the capillary immersed in the drop (50 and 375 μm internal and external diameters, 1mm length with 500 μm immersed in the drop with 500 μm water below the end of the capillary). As standard values, a microscopic diffusion coefficient of $10^{-10} \text{ m}^2 \cdot \text{s}^{-1}$ is taken for the peptides with a net charge of 3. The diffusion layer thickness, previously determined to be 230 μm for $\text{Fe}(\text{CN})_6^{4-}$, has been approximated to 150 μm for the present value of the peptide

diffusion coefficient ($\delta \propto D^{0.25}$).¹ The electrical field intensity in the capillary is 350 V/cm (370 V imposed between the top of the capillary and the electrode, assumed to be an equipotential). A mesh size from 5 μm in the capillary to 0.2 μm near the immersed edge and a time step of 0.5 ms are used for all the calculations.

Simulation of the iontophoretic spotting process. To improve the understanding of the reported iontophoretic spotting process, diffusion-migration simulations have been carried out. The spotting process has been simulated from the moment where a 1 mm sample plug arrives at the outlet of the capillary (Images A and F of the Figure S-2) and further performed to simulate a total collection time of 40 seconds. A peptide presenting average values of electrophoretic mobility and diffusion coefficient has been chosen for the calculation. (See the supplementary information for further details)

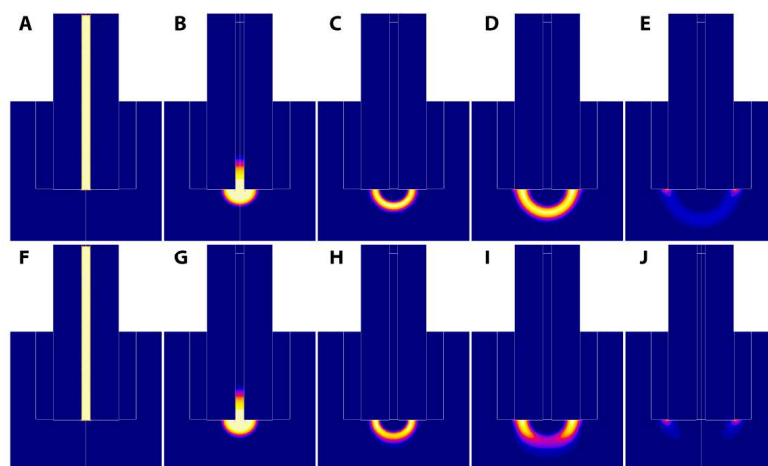


Figure S-2. Peptide concentration isovalues during the iontophoretic spotting process.

Figure S-2 represents the concentration isovalues of the chosen model peptide as a function of the spotting time in two different cases. The first case (Images A to E) deals with a situation where the spontaneous convection is considered negligible. In a second case (Images F to J), this effect is taken into account following Eq (3). Even if this increases the model accuracy, it has to be noticed that the macroscopic convection is undoubtedly playing a more important role in the present case. Indeed, in addition to the evaporation process, each

capillary manipulation (sequential dip and removal) certainly affects the droplet steadiness. Nevertheless, the iontophoretic spotting process can be well understood. As suggested by Figure S-2, the process can be divided into three steps. At first, the migrating peptide enters the collecting droplet by migrating in a crown-like zone. Then, depending on the magnitude of the convection, the crown-shape is more or less distorted in the vertical direction. Indeed, as can be seen on the images C,D and H,I, the further away the species are from the capillary, the higher the convection and the higher the dispersion. Finally, it can be observed that some of the collected species migrate toward the electrode (Images D,I, and E, J) where those species will be irreversibly attracted, the experimental drawback being certainly an increased carryover of the collected species from fraction to fraction. If the images E and J of Figure S-2 are compared, it can be suggested that the amount of peptides attracted by the electrodes is lower when the spontaneous convection is taken into account.

To further prove this last point, we have evaluated the amount of species, which can efficiently enter the collection droplet. To do so, we have calculated the flux of migrating and diffusing species across a virtual horizontal barrier. Experimentally, this barrier could be considered as the thickness of the film of liquid, which remains on the capillary when it is lifted from the droplet after a given collection time. As it was not experimentally possible to determine precisely this thickness, the calculations have been performed for different plausible values, ranging from 50 to 300 micrometers. An integration of the species flux as a function of the spotting time was carried out to estimate the percentage of species efficiently transferred to the collection droplet, the remaining corresponding to the proportion of species being potentially carried over till the next fraction. This simulation has been carried out for three collection periods, ranging from 10 to 40 seconds and the results are shown on Figure S-3.

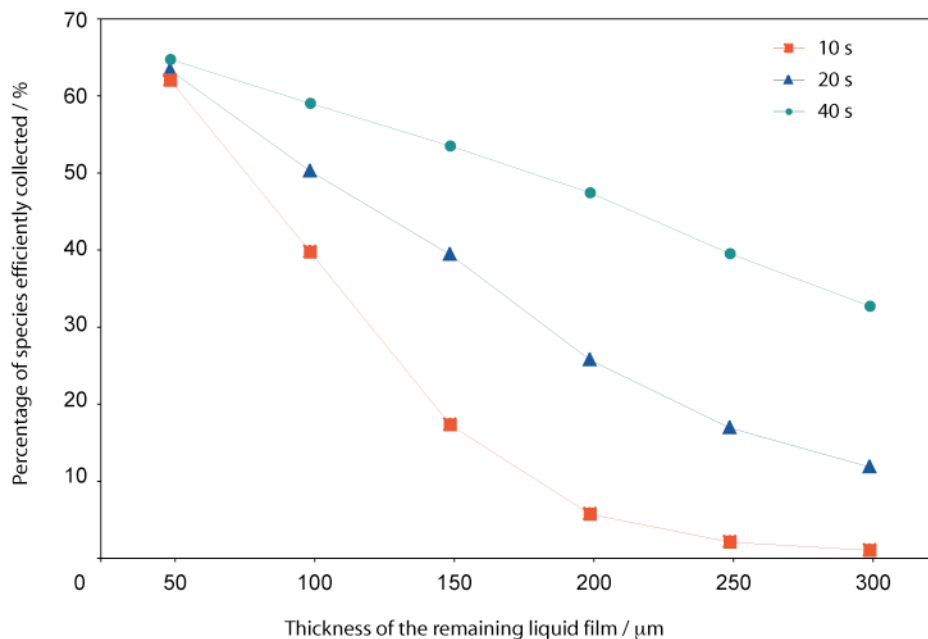


Figure S-3. Proportion of species efficiently collected as a function of the thickness of the remaining liquid film on the capillary outlet.

Figure S-3 demonstrates that the thickness of the film of liquid, which remains at the tip of the capillary, at the end of a collection period is certainly one of the most important parameter to be optimized when iontophoretic spotting is used. Indeed, for a short collection time of 10 s, the percentage of efficiently collected species can vary from about 60% to less than 10% when the film thickness is increased from 50 to 200 μm , as the species don't have time to cross the remaining film. This last case leads potentially to a significant carryover of the collected species from fraction to fraction. Then, if the collection time is gradually increased, it can be seen that the percentage of collected species increases. If this approach can constitute a solution for the analysis of a rather simple mixture, it is not the case for more complex ones, as the highest peak capacity will be required. Considering the results of this simulation part, it can be concluded that the coating of the outer outlet part of the separation capillary, has to be properly optimized as it can influence a lot the obtained results. Especially, in the case of electrophoretic separations carried out under high electric field or under conditions that induce large electrophoretic mobilities of the separated analytes, the outer coating of the

capillary outlet should insure the lowest film thickness while allowing current application when the capillary is moved from fraction to fraction.

1. Amatore, C.; Szunerits, S.; Thouin, L.; Warkocz, J. S. *J. of Electroanal. Chem.* **2001**, 500, 62-70