

Structuration of zero-shrinkage LTCC using mineral sacrificial materials

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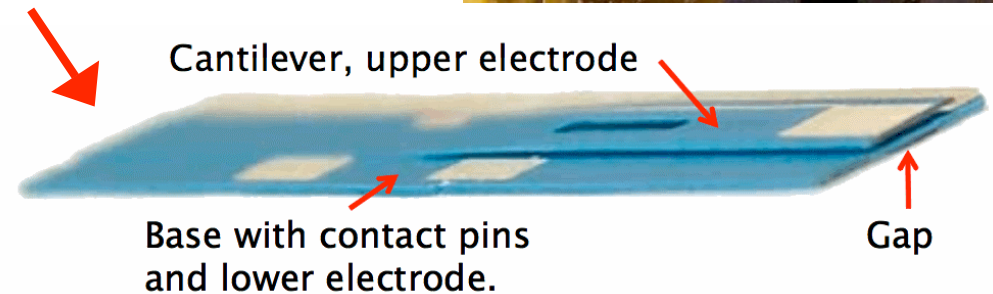
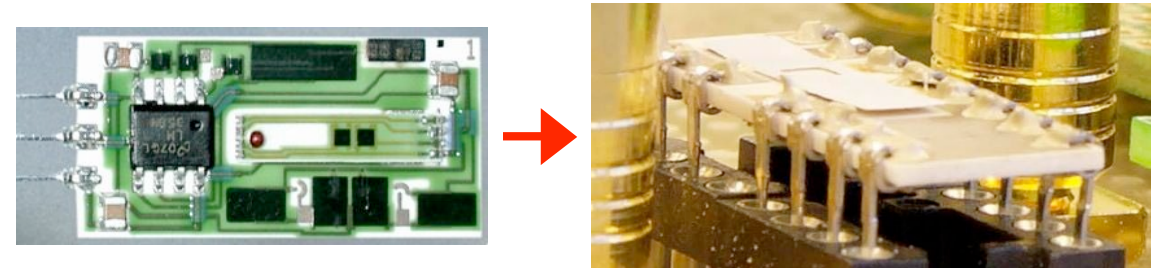
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1. Introduction - **Thick-film / LTCC structuration techniques**
2. Mineral sacrificial pastes - **Requirements & formulation**
3. New sacrificial materials - **Thick-film structures**
4. LTCC structures - **Processing and results**
5. Conclusions & outlook

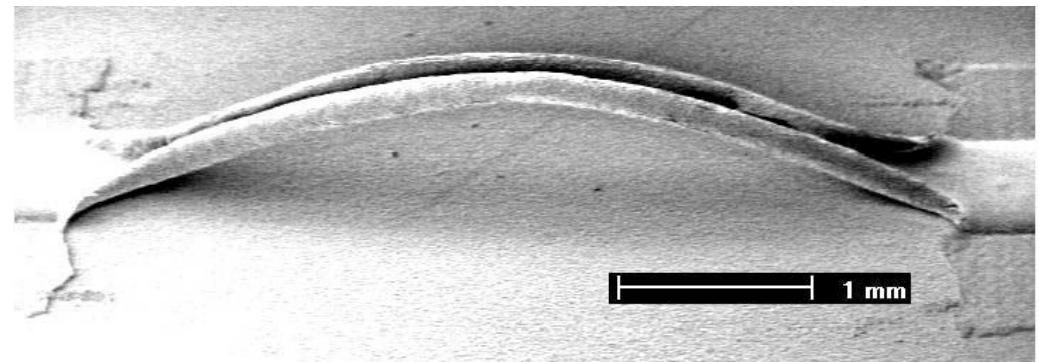
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1. Applications : cantilever & bridge

- Micro-force sensors
 - Thick-film or LTCC
 - Mineral sacrificial paste etched by acid

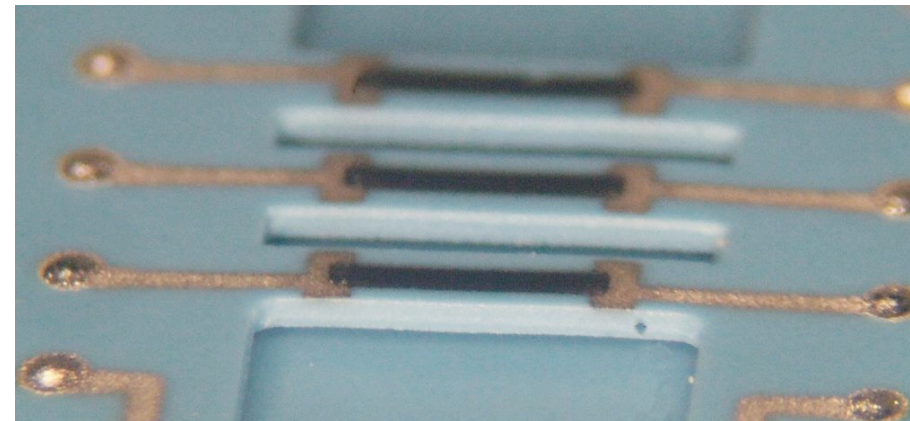
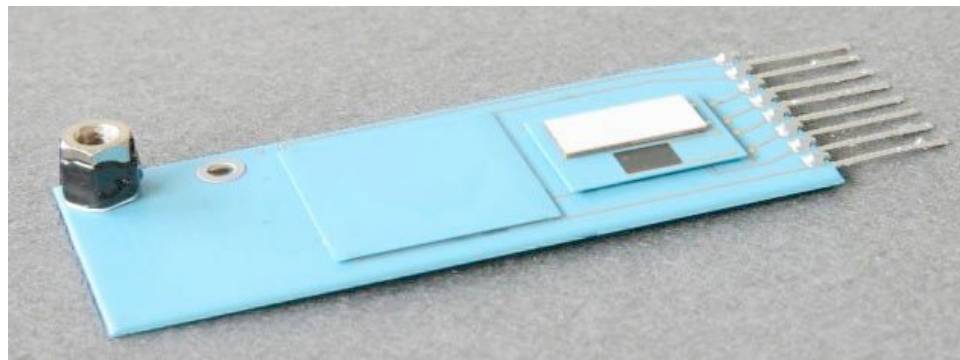
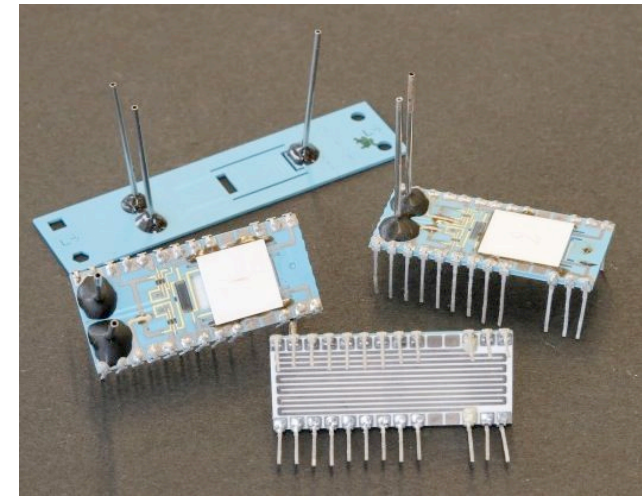


- Suspended thermistor
 - Flow sensing
 - Thermal actuator
 - Igniter
 - μ -Thruster
 - SOFC



1. Applications : fluidics

- Chemical reactor
- Flowmeter
- Gas viscosity sensor



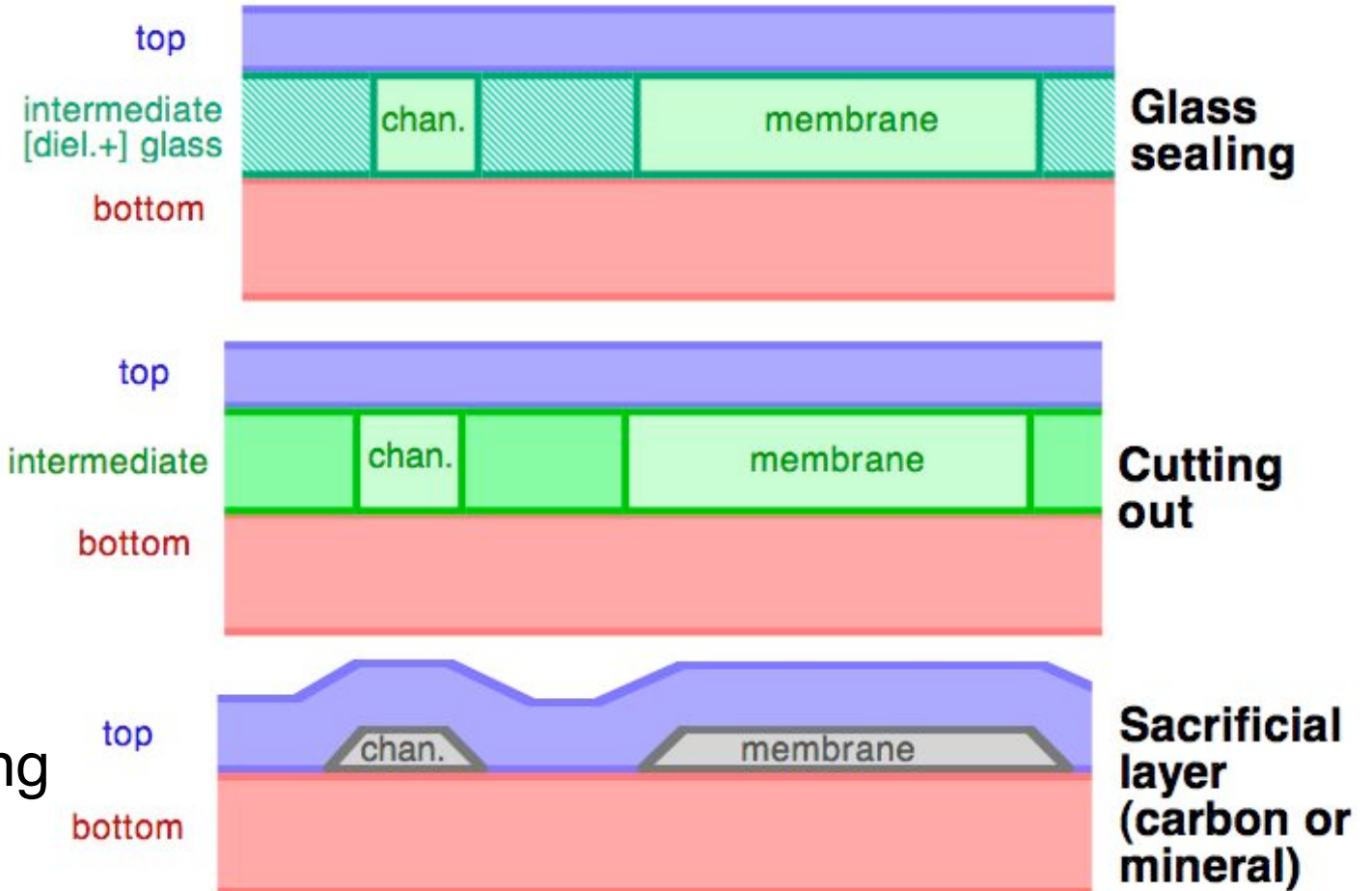
1. Structuration methods

Features

- Vias
- Channels
- Membranes
- Beams

Three types

- 1) Glass sealing
- 2) Cutting & stacking
- 3) Sacrificial layer



1. LTCC vs. alumina for sensors

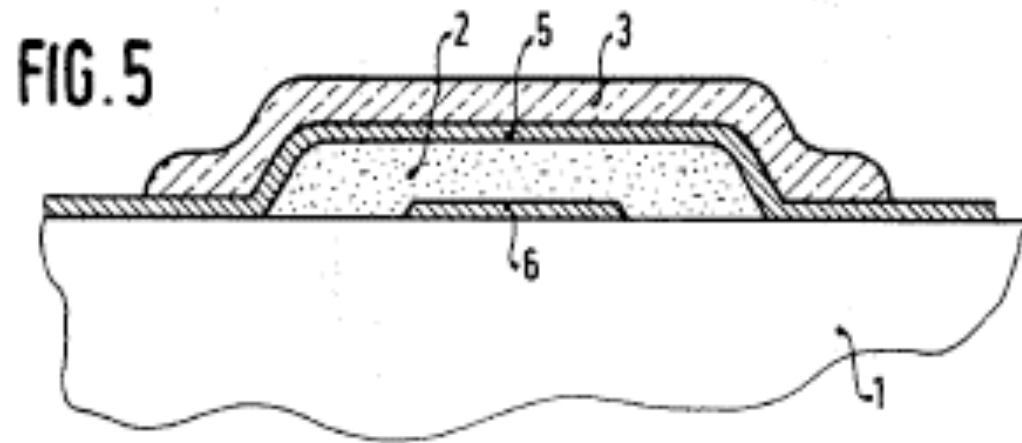
Material	LTCC (DP 951)	Al ₂ O ₃ (96%)	Ratio
Minimal thickness [mm]	0.04	0.17	0.24
Short-term strength [MPa]	320	600	0.53
10 year strength [MPa]	110	270	0.41
Young's modulus [GPa]	110	320	0.34
Thermal conductivity [W/m]	3	25	0.12
Design strain [ppm]	1'000	800	1.25
Flexural sensitivity [kN ⁻¹]	5.68	0.11	53
Thermal resistance [K/W]	8'333	235	35

- LTCC for thermal & low-range mechanical sensors
- **Thick-film dielectric \approx LTCC**

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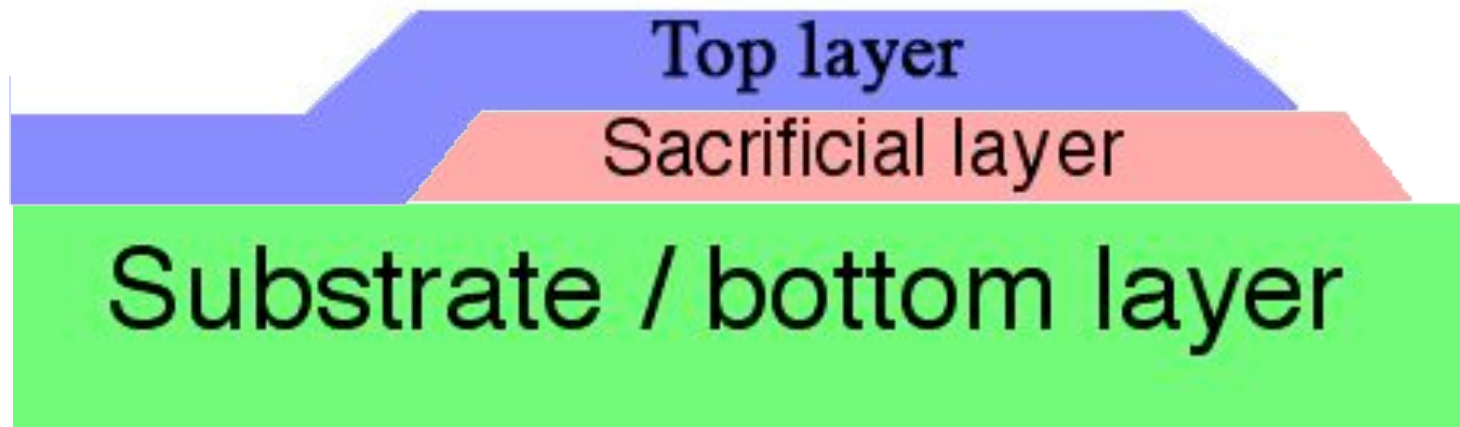
2. Mineral sacrificial paste vs. carbon

- Pioneering work in in the 1980's: **std. thick-film tech.**
 - Stecher et al., Bosch, Germany
 - Thick-film dielectric on carbon sacrificial paste
 - Alumina substrate
 - Complex steps to avoid sagging / swelling (N_2 , then air)
 - Already thought about using mineral sacrificial paste (MSP)!
- Too complex
 - Did not catch on

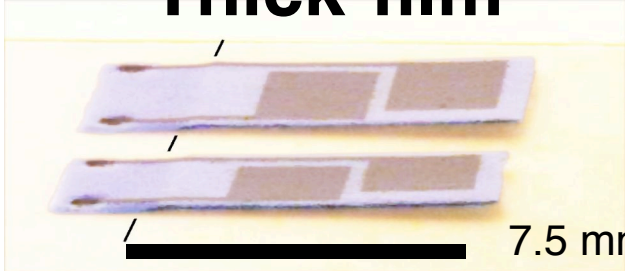



2. Mineral sacrificial paste - process

- Print [fire] mineral sacrificial paste (MSP) onto LTCC / ceramic substrate
 - Laminate (LTCC) / print (ceramic) top layer
 - Fire structure (co- or post-fire)
 - Chemically etch sacrificial layer - typ. **acetic / phosphoric acid**
- Requires relatively “open” structure



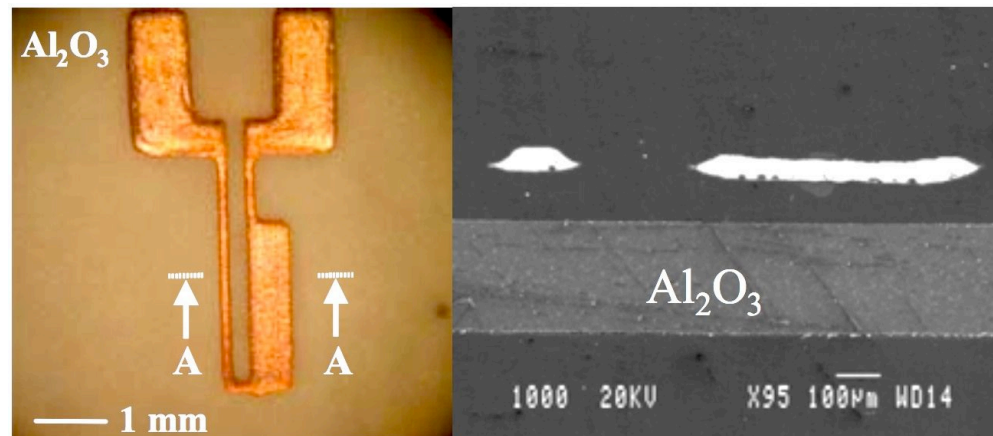
2. MSP requirements - thick-film & LTCC

Property	Thick-film 	LTCC 
Cohesion in fired state	Required (post) Desirable (co)	Desirable (avoid peeling)
Paste formulation	Allow overprinting	Not critical (except thin LTCC)
Shrinkage matching	Not critical	Required

2. Recent work : SrCO_3

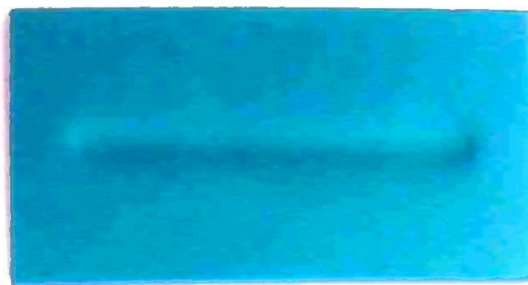
- **Thick-film** : Lucat, Ginet & al. (2006-2007)
 - SrCO_3 doesn't dissociate at $<1000^\circ\text{C}$, quite inert
 - CO_2 evolved in acid etching
 - No shrinkage: not compatible with std. LTCC
 - No cohesion / sintering: only co-fire
 - Situation quite similar to LTCC

2007 Lucat
(Thick-film ;
co-fired)

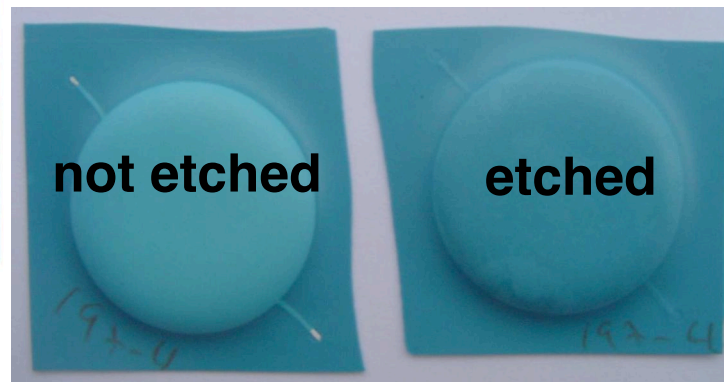


2. Recent work : $\text{CaCO}_3/\text{CaO} + \text{B}_2\text{O}_3$

- **LTCC+thick-film:** Birol, Fournier et al. (2005-2007)
 - Firing: dissociation of CaCO_3 to $\text{CaO} + \text{CO}_2$
 - CaO hydrates to $\text{Ca}(\text{OH})_2$ - **must etch fast**
 - Large amount not allowed - **reaction**
 - Good dissolution in acids ✓
 - Some limited cohesion (✓)
 - Little sintering (<10%) : **incompatible with std. LTCC**



Sintered LTCC + MSP
(reverse side)



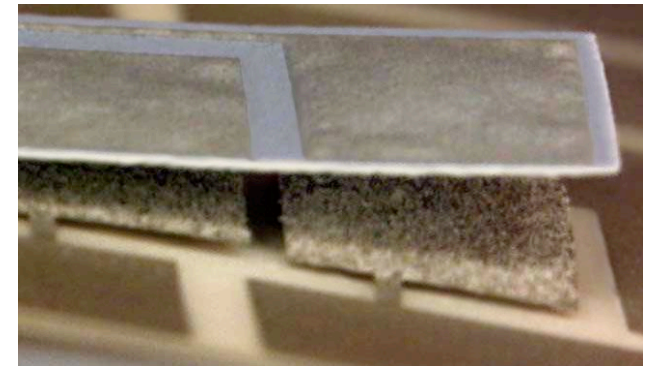
2006
Birol
(LTCC)



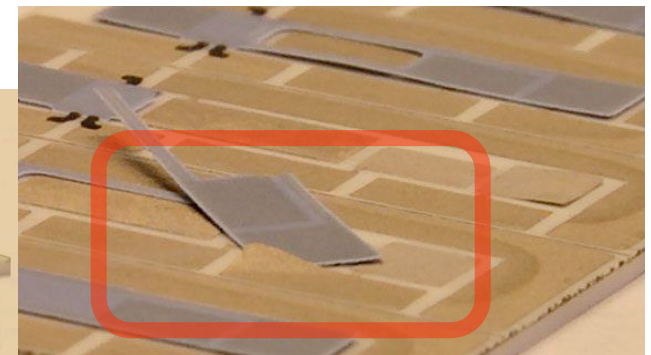
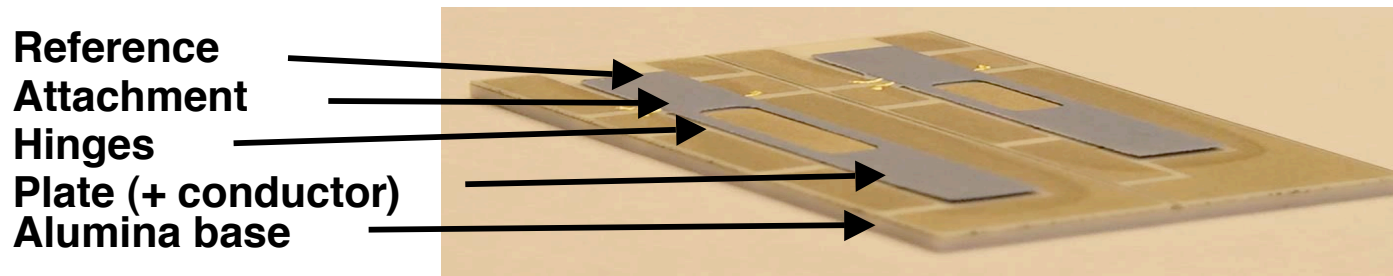
2. Recent work : CaO + borax

■ Thick-film: Fournier et al. (2006-2007)

- B_2O_3 replaced by borax: less volatile, melts at $\approx 740^\circ C$ ✓
- More dense and cohesive, can achieve good sintering ✓
- **High contents** → **problems:**
 - Slower dissolution in acids
 - Excessive reactions

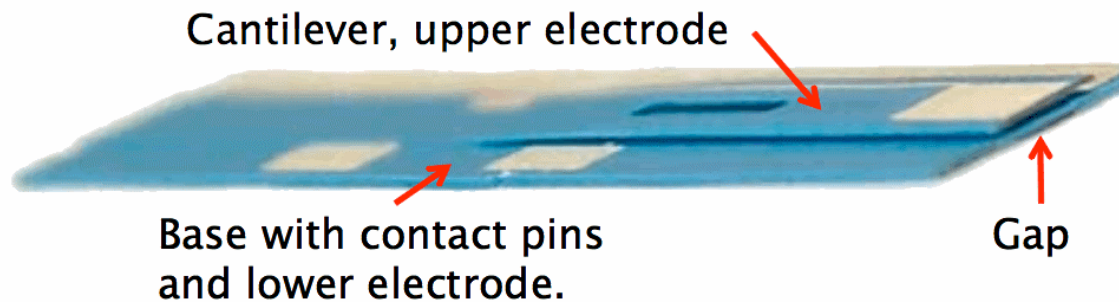


Thick-film dielectric (ESL 4913) μ -force sensor

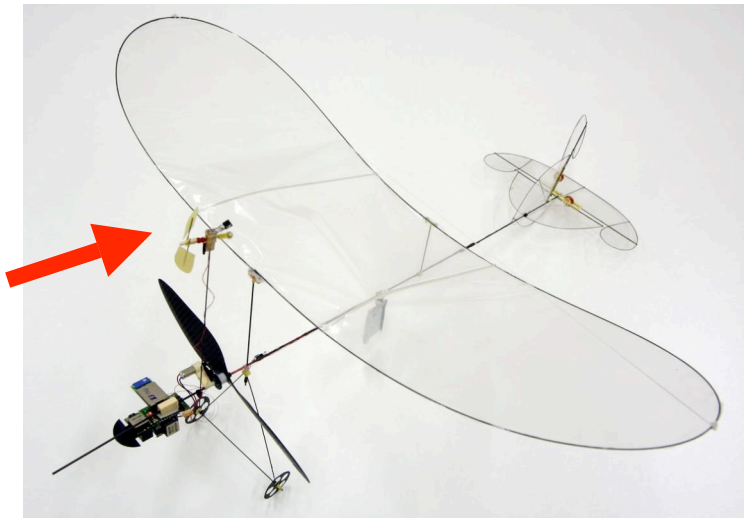


2. Recent work : CaCO_3 + carbon

- **LTCC: Fournier et al. (2008)**
 - Decomposition of CaCO_3 gives some shrinkage
 - Additional shrinkage introduced with carbon
 - Can \pm match LTCC shrinkage but **must add 2/3 carbon**
 - **No cohesion**



LTCC (DuPont 951) μ -force sensor



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3. New materials: MgO + CaB₂O₄

- **MgO - magnesium oxide**

- Alkaline earth oxide - similar to CaO / SrO
- Oxide reasonably stable vs. humidity ("dead burned")
- Well-known in MEMS literature (surface micromachining)
- No sintering at common firing temperatures

- **CaB₂O₄**

- Available as CaB₂O₄ · 2H₂O
- Significant sintering & cohesion
- Issues with high B₂O₃ + H₂O contents

- **Mixes: intermediate properties**



**Zero-
shrinkage
LTCC**

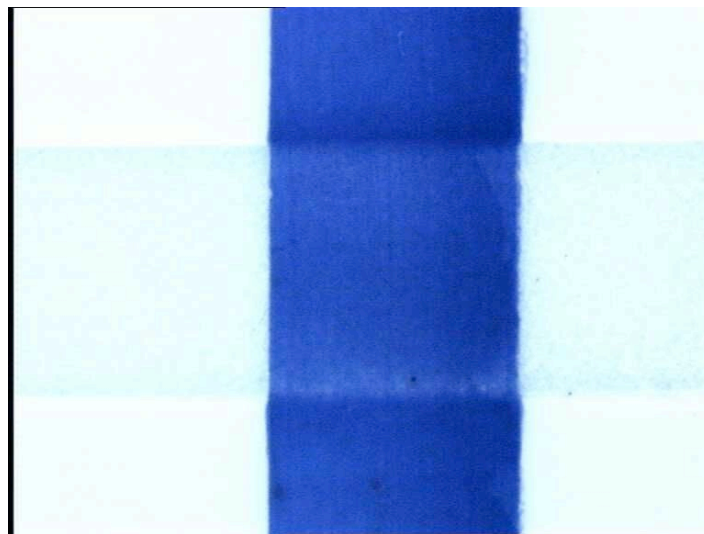


**Standard
LTCC**

3. MgO + thick-film process

■ Sequential firing of MSP + dielectric

- B_2O_3 or some borax ($Na_4B_2O_7$): cohesion
- Dielectric paste adapted to porous underlayer (LTCC: ✓)
- No lateral shrinkage \approx zero-shrinkage LTCC

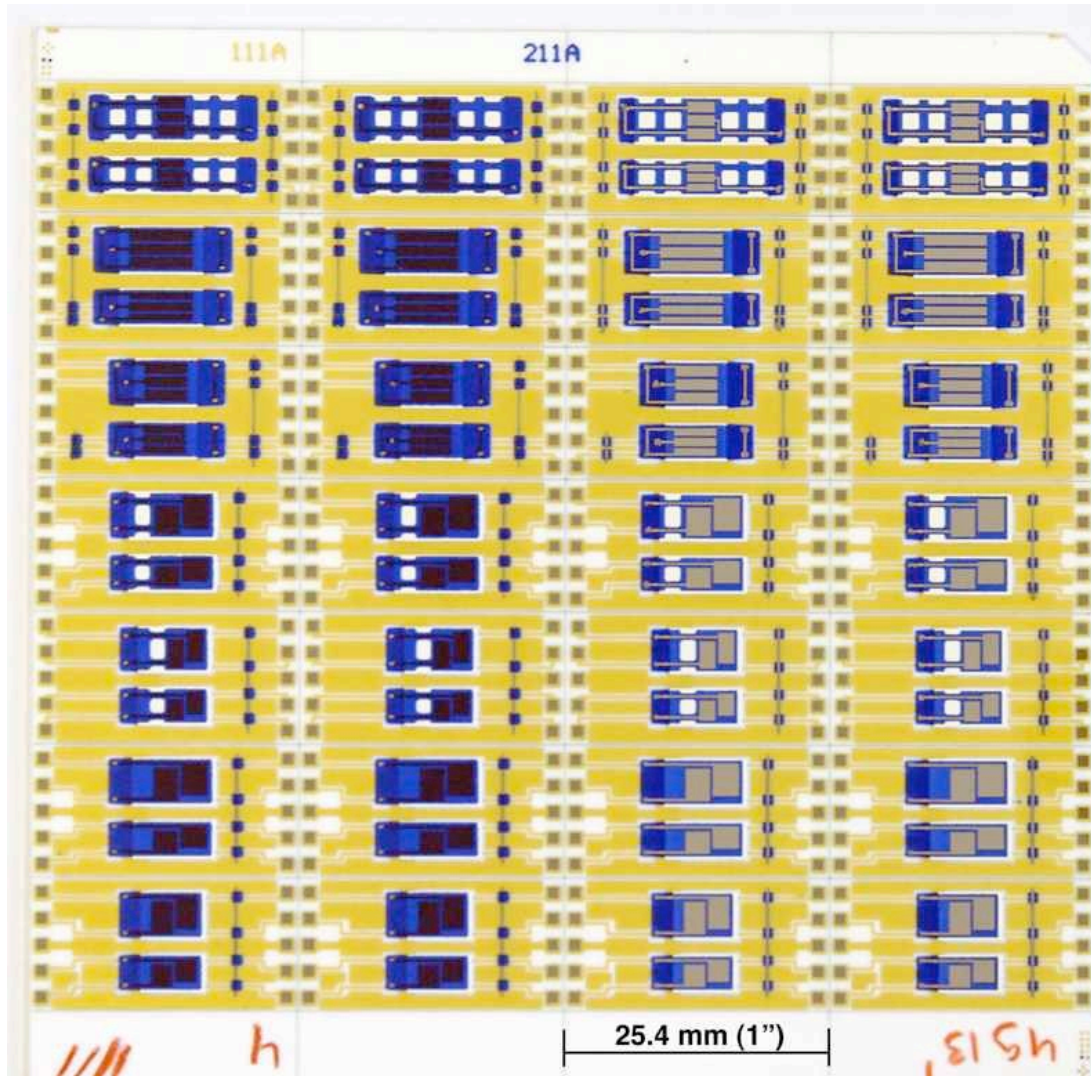


ESL 4913 over MgO + borax

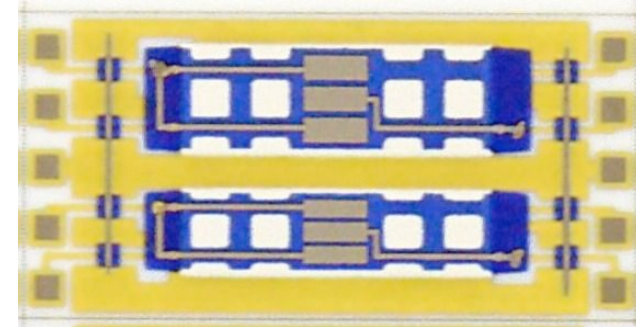
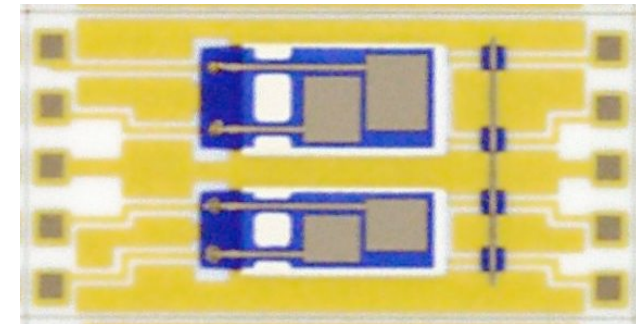


ESL 4913 over MgO + B_2O_3

3. MgO + thick-film test structures



- Structures for capacitive force sensors / actuators
- Cantilevers or bridges
- Different sizes / shapes



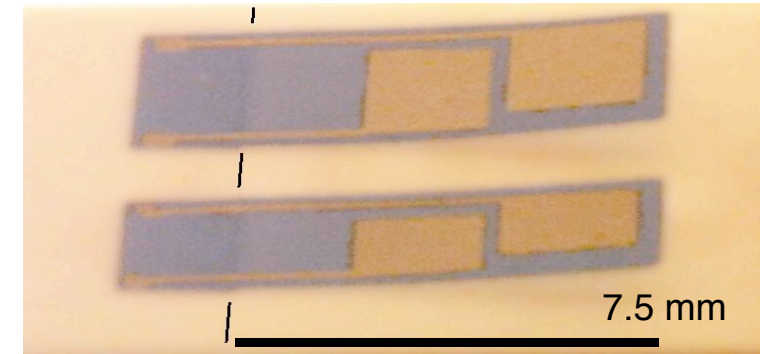
3. Sacrificial layer removal

- Dip into acid (\approx 1 day) @ RT
 - MgO not very reactive
 - "Dead-burned" or additives on surface
 - 10% phosphoric or acetic acid not optimised
- Rinse with tap water
- Neutralise acid with TRIS buffer
- Rinse with deionised water
- Rinse with isopropanol
- Dry @ 60°C in oven

3. MgO + thick-film process

■ Achievements

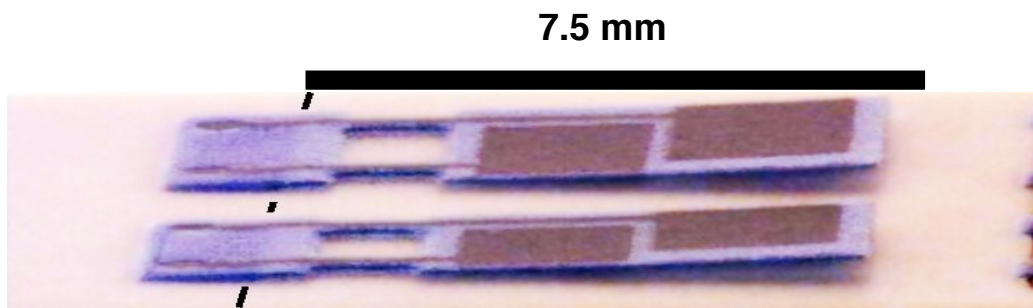
- Very slender structures
- Straightforward process
- Dielectrics : good acid resistance



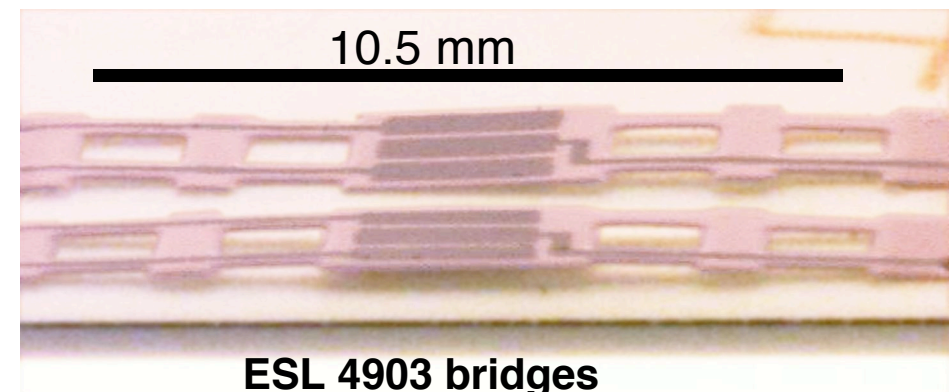
ESL 4913 cantilevers

■ Issues

- Differential TCE with alumina (bridges)
- Internal stresses: metallisation - dielectric



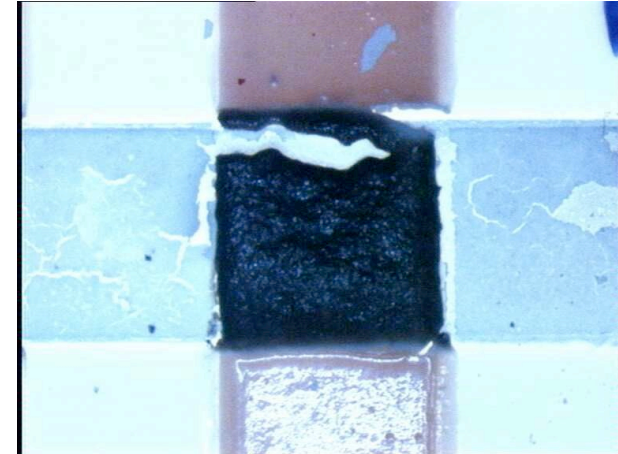
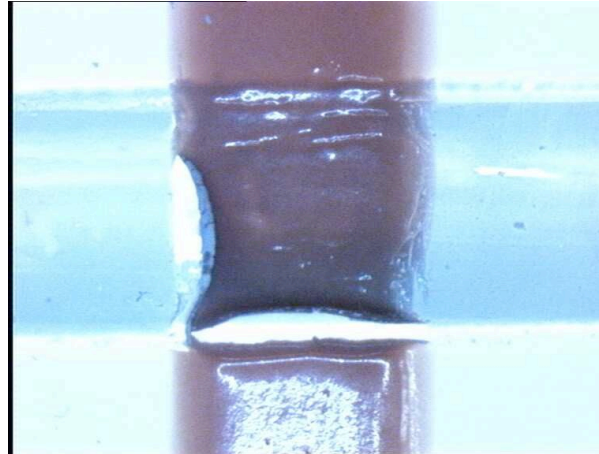
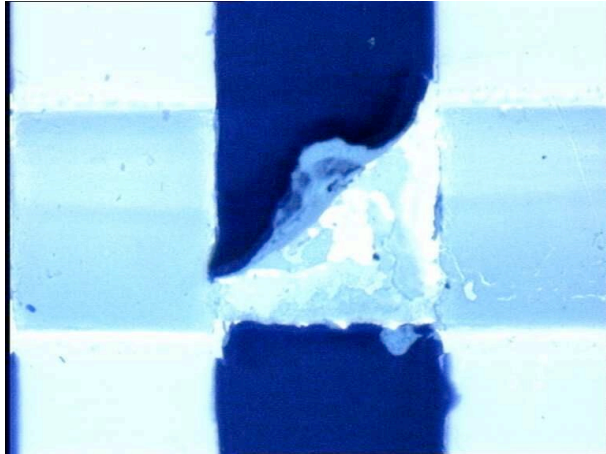
ESL 4913 cantilevers



ESL 4903 bridges

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4. MgO + H₃BO₃ - co-firing issues



ESL 4913 (left) / 4903 (right) over MgO+H₃BO₃
Insufficient cohesion of MgO in spite of B₂O₃

ESL 4903 over MgO+H₃BO₃
Too high H₃BO₃: reaction

- Balance: ensuring cohesion / avoiding reactions

4. Flatness of LTCC



Single LTCC sheets fired @ 875°C

- Good flatness for DP 951 and HL2000
- HL800 more problematic (needs thicker structures)

4. Flatness of LTCC with MSP

DP 951

HL2000

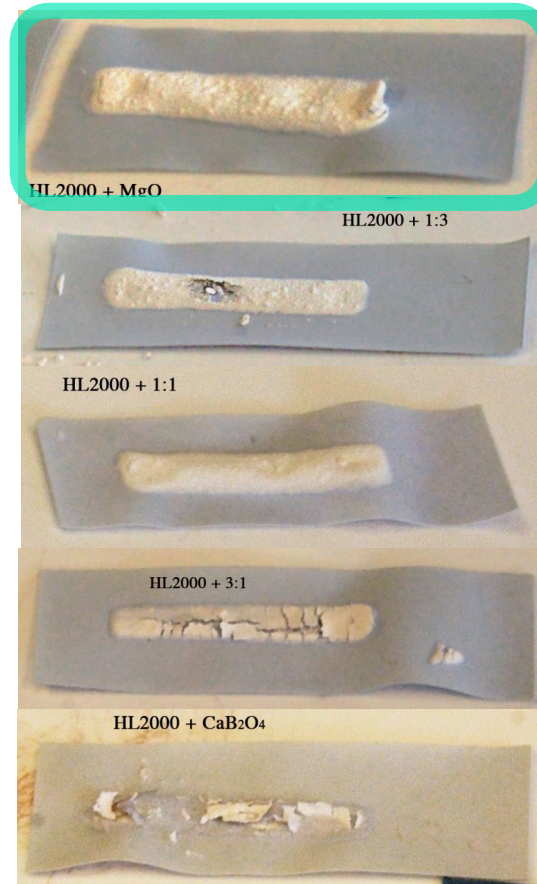
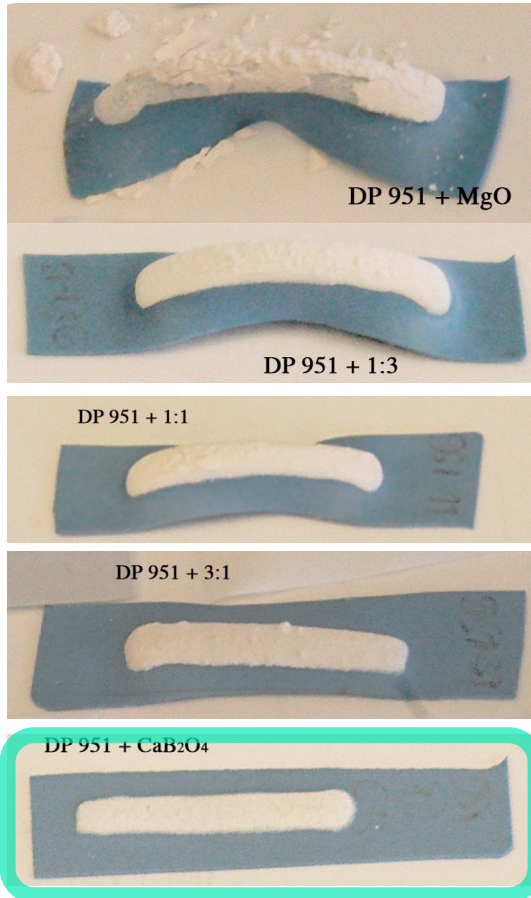
MgO

1:3

1:1

3:1

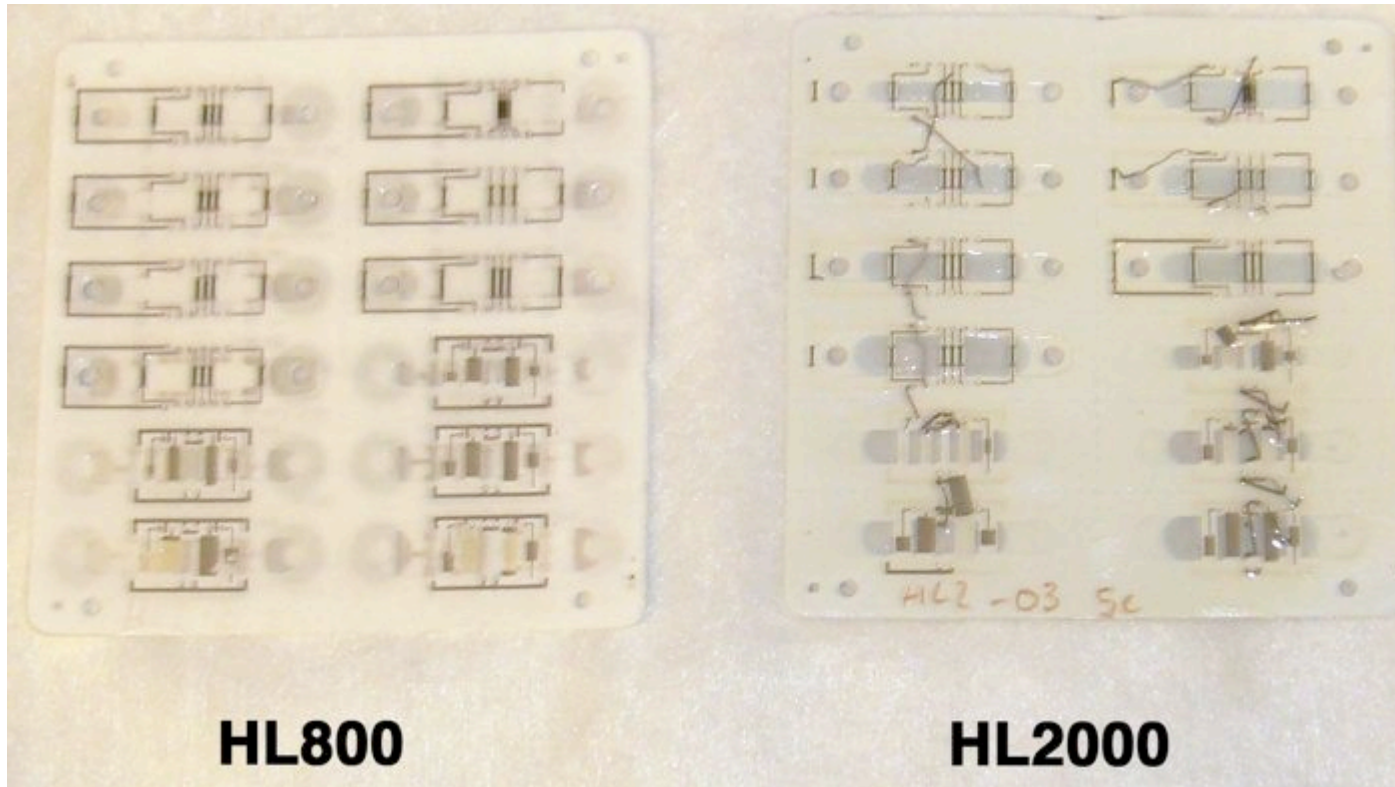
CaB₂O₄



- Expected behaviour!
- DP951 (std.) compatible with shrinkage of CaB₂O₄
- HL2000 better with MgO-rich MSPs

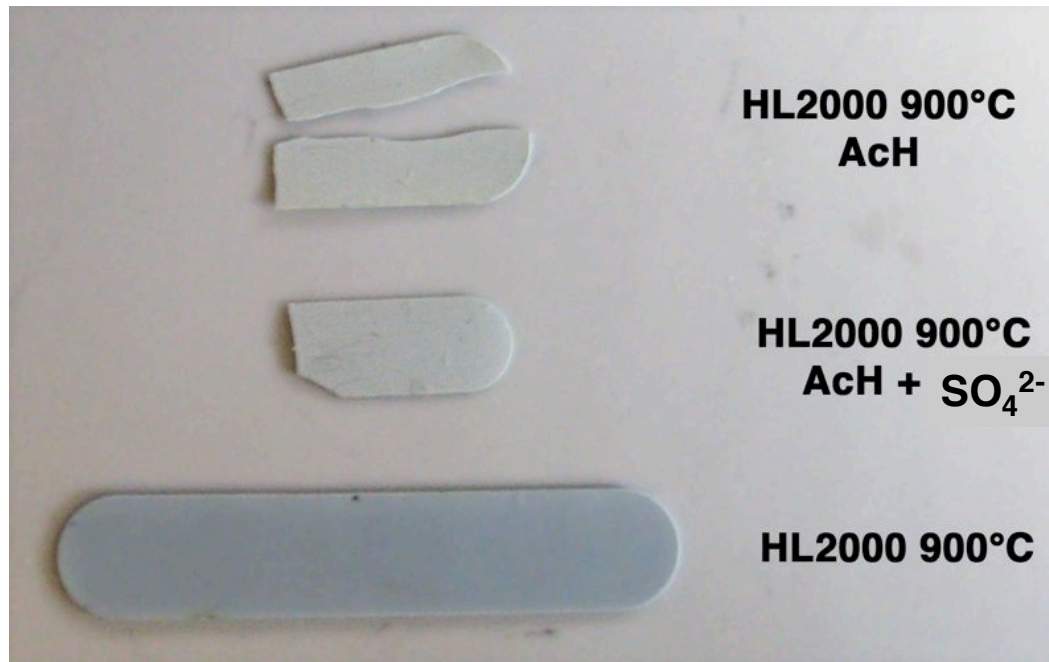
Deformation of LTCC + CaB₂O₄:MgO MSP mixes

4. Acid resistance of LTCC



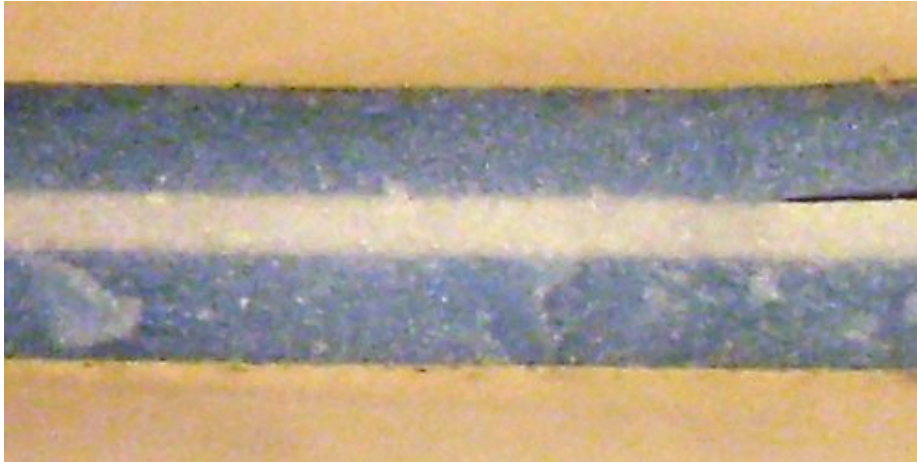
- DP951
 - Itself : OK
 - Attacked by CaB_2O_4
- HL800: some attack
- HL2000: destroyed

4. Acid resistance of LTCC

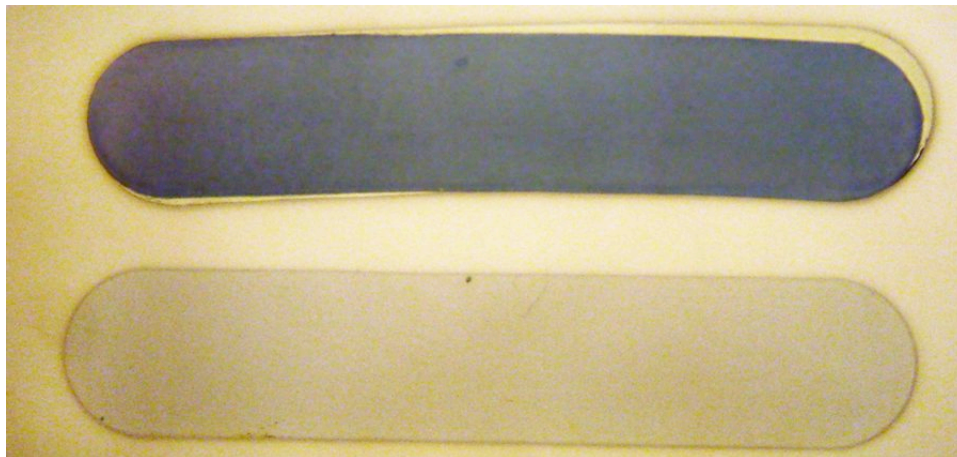


- Increase LTCC firing temperature to 900°C
- Buffering of acetic acid solution (NaAc:HAc)
- Add SO₄²⁻ ions
 - Mg soluble
 - Ca, Sr, Ba... insoluble
 - Better selectivity
 - With MgO MSP

4. Acid resistance of LTCC



- DP951-HL-DP951 sandwich structures
- Good acid resistance
- $\approx 1/5$ HL800 or HL2000 sufficient to block shrinkage $< 1\%$!

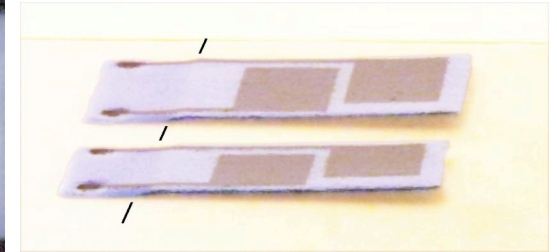
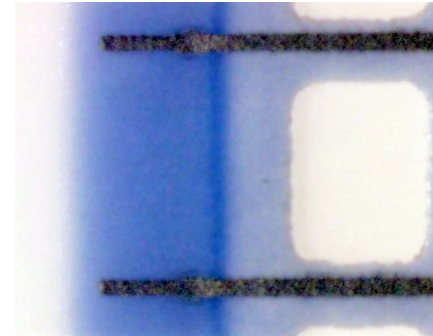


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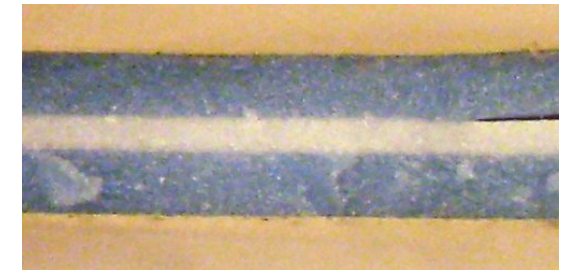
■ Thick-film : success

- MgO-based sacrificial layer
- Standard firing sequence



■ LTCC : promising results

- Good shape stability with HL materials
- Promising improvements in acid resistance
 - Sandwich structures with DP951
 - Optimised parameters

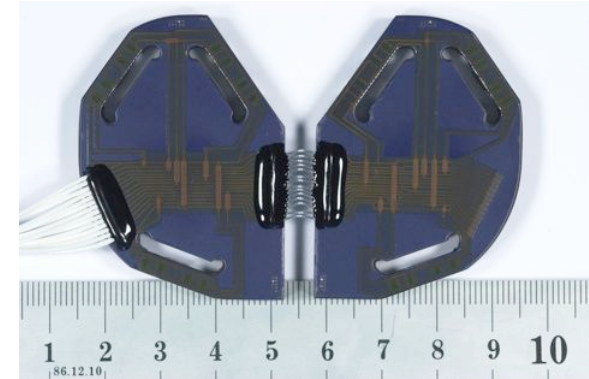
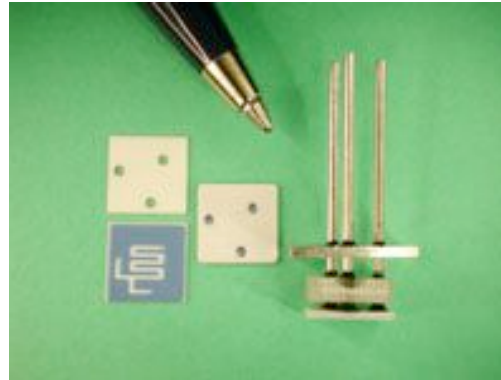


5. Outlook

- Improve acid resistance of zero-shrink LTCC
 - Firing studies
 - Etchants with better selectivity
 - Sandwich structures where applicable

- Optimise MSPs for LTCC
 - MgO-based (soluble in sulfate etchants)
 - Additives to impart limited cohesion

Merci



Thank you !

